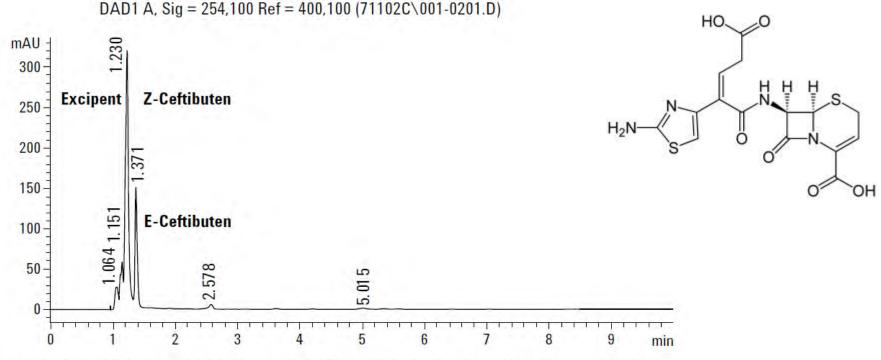
Polar Analytes: C18 Didn't Work, Now What?

Mark Powell Agilent Technologies Technical Support



C18 Doesn't Always Work...



Instrument: Agilent 1100 Series HPLC; Temp: ambient; Column: Alkyl-C18, $4.6 \times 150 \text{ mm}$, 5 µm; Mobile phase: 2% ACN, 98% 10 mM ammonium acetate, pH 5.4; Flow rate: 1 mL/min; Injection volume: 5 µL; Diode array detector: 254 nm; Reference: 400 nm; Bandwidth: 100 nm



Pore Dewetting or Phase Collapse

•Alkyl phases such as C8 or C18 can exhibit poor retention or reproducibility of retention in low organic mobile phases

•Phenomenon known as pore dewetting or phase collapse

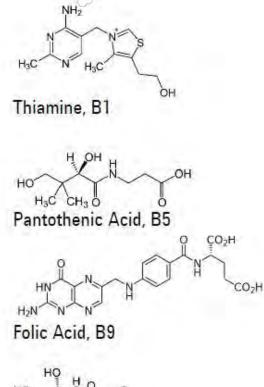
•Onset can be unpredictable

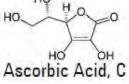
•A method robustness issue often mistaken as a column or lot issue

•See Przybyciel and Majors, *LCGC* **20**(6), 516-523 (2002).

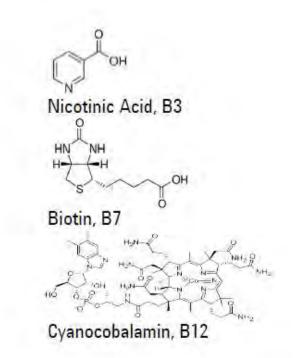


Water Soluble Vitamins

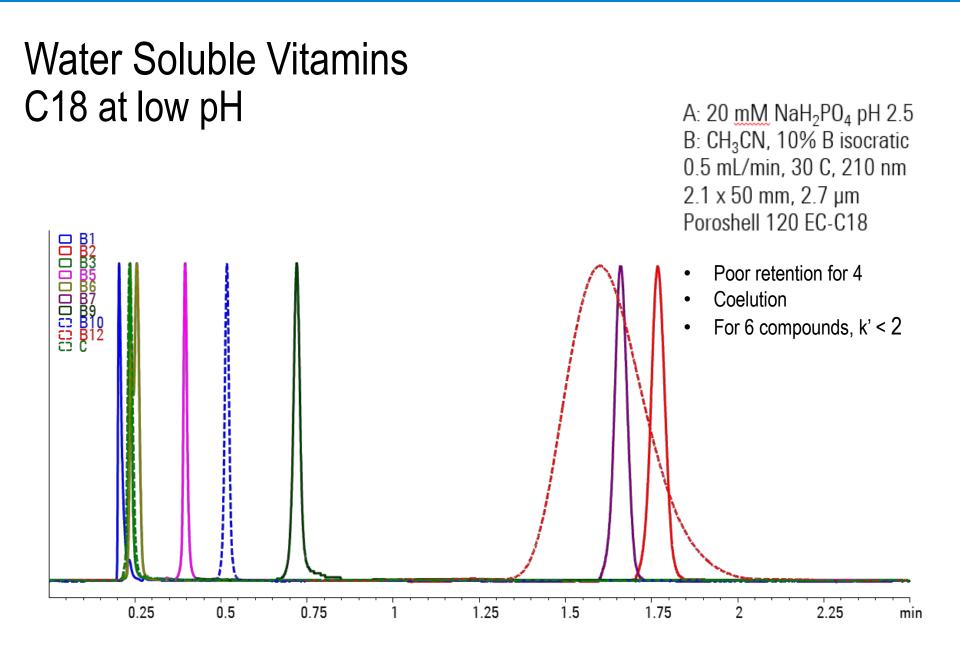














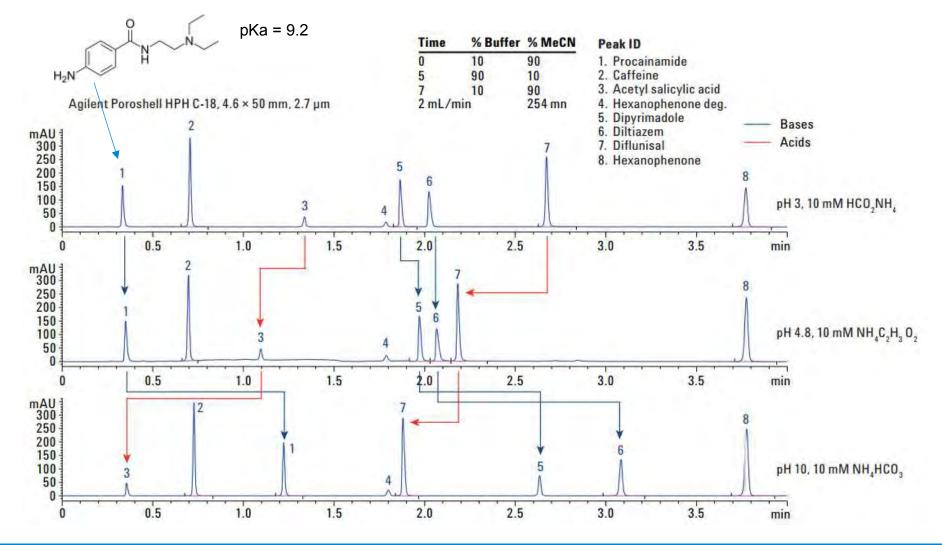
Now What?

Adjust mobile phase pH

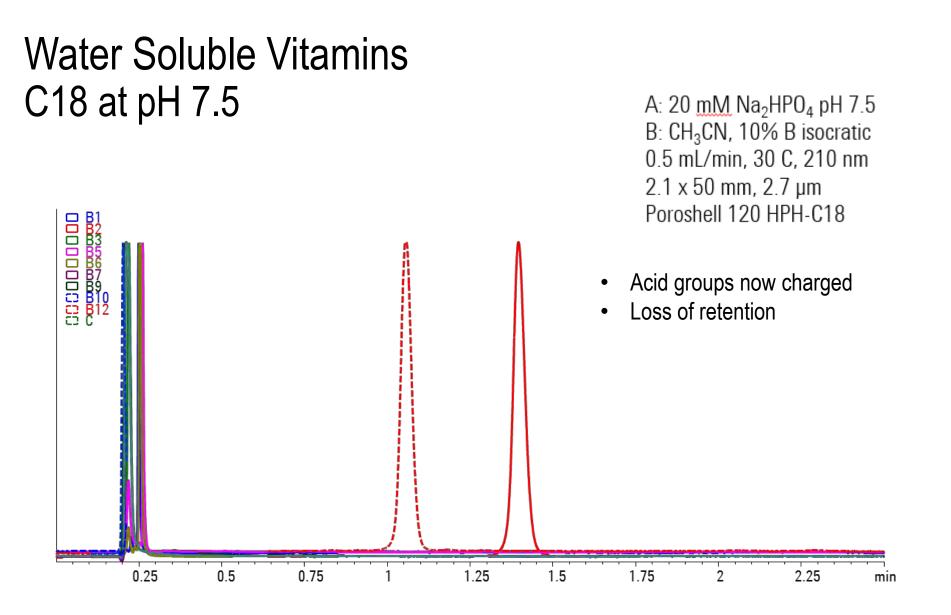
- Ion-pair chromatography
- •Alternate column choice
- •HILIC (hydrophilic interaction chromatography)



Why try higher pH? Poroshell HPH-C8 or C18









Now What?

•Adjust mobile phase pH

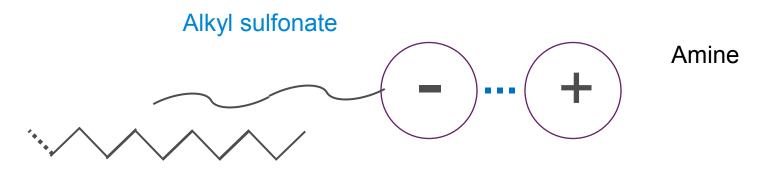
Ion-pair chromatography

- •Alternate column choice
- •HILIC



Ion-Pair Chromatography

Similar to reversed-phase, but an ion-pairing reagent is added to the mobile phase



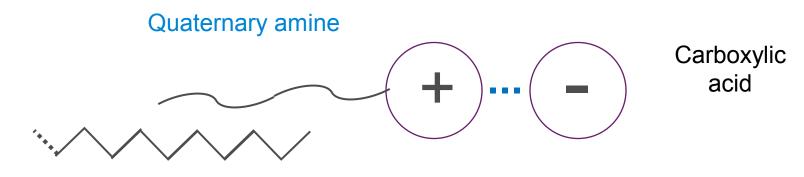
Non-polar stationary phase

Non-polar alkyl chain will adsorb into the non-polar stationary phase
Polar part of the ion-pairing reagent will "stick-out" into the mobile phase



Ion-Pair Chromatography

Similar to reversed-phase, but an ion-pairing reagent is added to the mobile phase



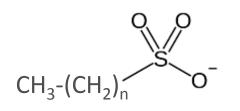
Non-polar stationary phase

Non-polar alkyl chain will adsorb into the non-polar stationary phase
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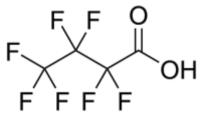


Some Common Ion-Pairing Reagents

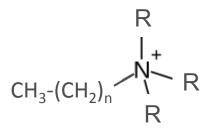
Pairs with Cations



Alkyl sulfonates



Heptafluorobutyric acid (HFBA) **Pairs with Anions**



Quaternary amines



Ion-Pair Chromatography Suggested Experimental Conditions

Column: C8 or C18

Mobile Phase:

- Organic often methanol
- Aqueous Buffered with appropriate IP reagent
- Temperature controlled between 35° and 60°C

Cations – bases

Buffer: 25 – 50 mM phosphate, pH 2- 3 IP reagent: 10-100 mM heptane sulfonate

Anions – acids

Buffer: 25 - 50 mM phosphate, pH 6 - 7

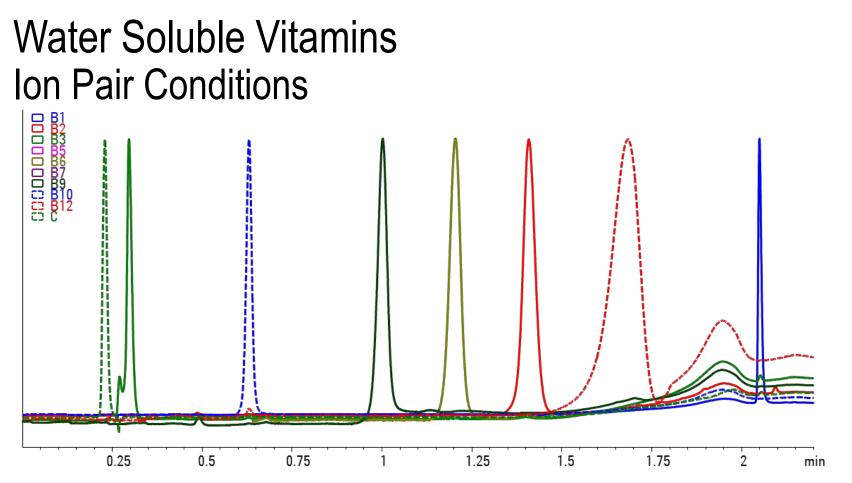
IP reagent: 10-40 mM tetrabutyl ammonium phosphate



Ion-Pair Chromatography Limitations

- •Higher level of complexity than RP, so generally chosen only if needed
- •Requires careful control of IP reagent, pH, temperature
- •Gradient methods are more difficult than RP
- •Equilibration is much slower than RP
- Column dedicated to IP
- •IP reagent in the injection solvent





EC-C18

A: 1.5 g sodium 1-heptanesulfonate + 0.2 mL triethylamine + 7.5 mL acetic acid + 992.5 mL water

B: CH3CN

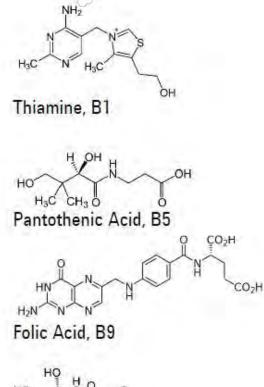
0.5 mL/min,10% B for 1 minute, then 10-40% B in 1 minute injection volume: varies according to signal strength TCC: 30 C

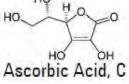
260, 8 nm Ref Off, 8 nm slit, 80 Hz

The ion pairing reagent increased retention for most compounds •6 compounds have k' > 2

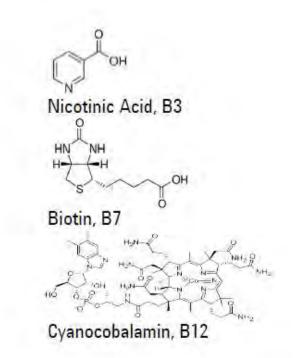
•B5 and B7 could not be detected due to low signal and high background noise at 210 nm (not detectable at 260 nm)

Water Soluble Vitamins



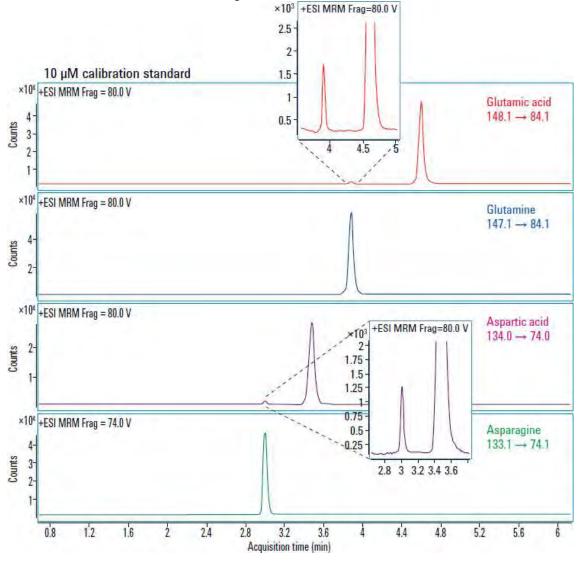


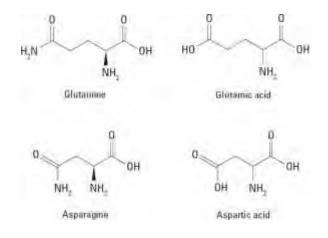






Amino Acids by Ion-Pair





ZORBAX SB-C18 RRHT, 1.8 μm, 3 x 50 mm, 25 °C, 1 μL inj 0.4 mL/min A: water/ 0.5 % FA + 0.3% HFBA B: ACN/0.5% FA + 0.3% HFBA 0 to 5% B over 5 minutes

5991-0904EN

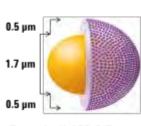


Now What?

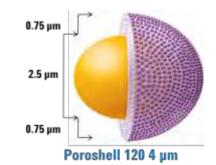
- Adjust method conditions
- Ion-pair chromatography
- Alternate column choiceHILIC



Phase Choices

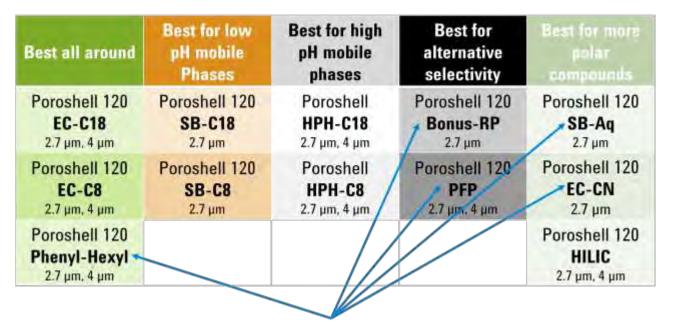


Poroshell 120 2.7 µm



Efficiency 90% of < 2 μm TPP Pressure 50% of < 2 μm TPP 2 μm inlet frit

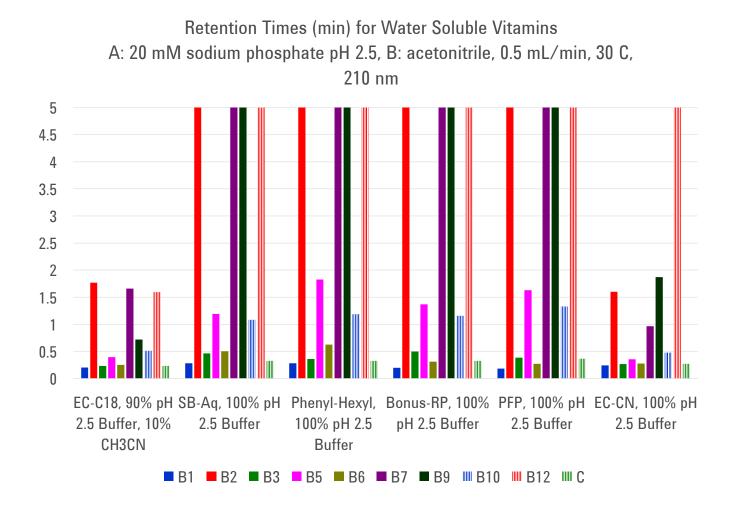
Efficiency 2x 5 µm TPP Pressure often below 200 bar 2 µm inlet frit



These phases can be used with 100% aqueous mobile phases to improve retention of highly polar analytes in RPLC mode

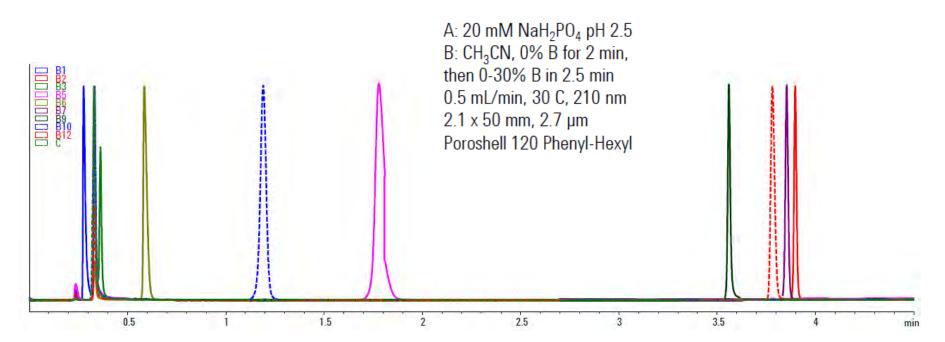


Water Soluble Vitamins Alternative Phases





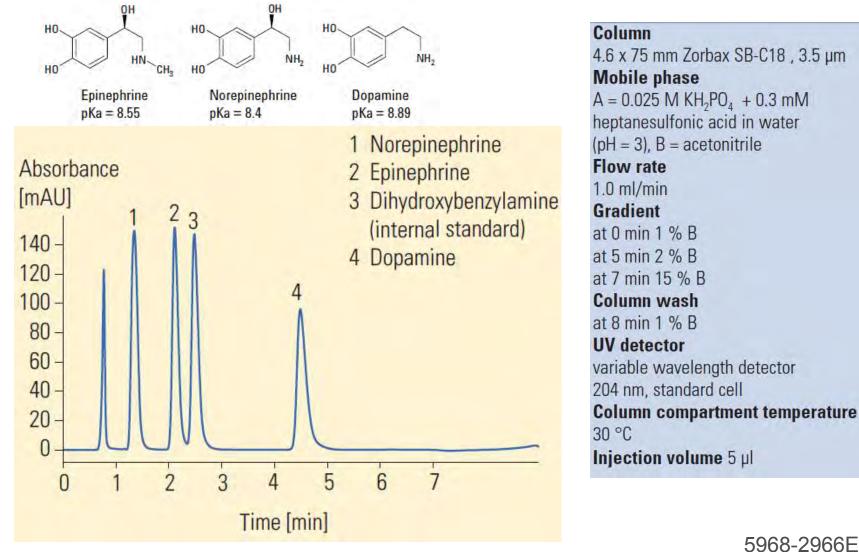
Water Soluble Vitamins Phenyl-Hexyl



- Phenyl-Hexyl has the best retention
- 7 compounds have k'> 2;
- C18 analysis had only 4 compounds with k' > 2

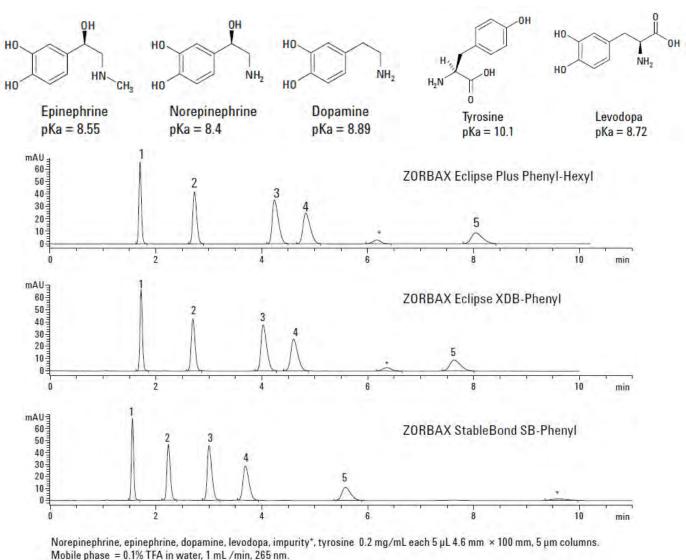


Catecholamines by Ion-Pair





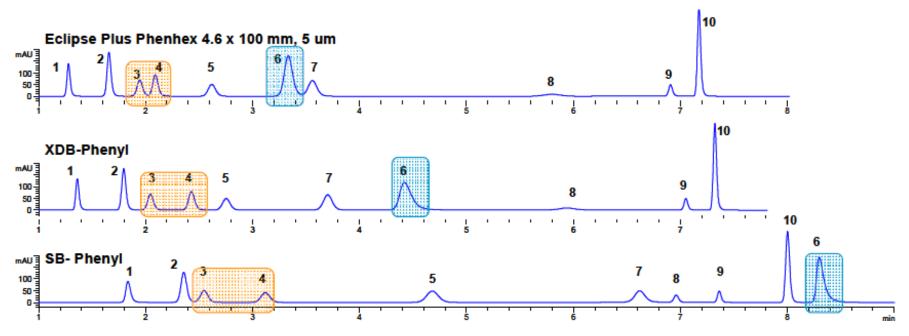
Catecholamines on Phenyl Phases



5990-3616EN



Nucleobases and Nucleosides



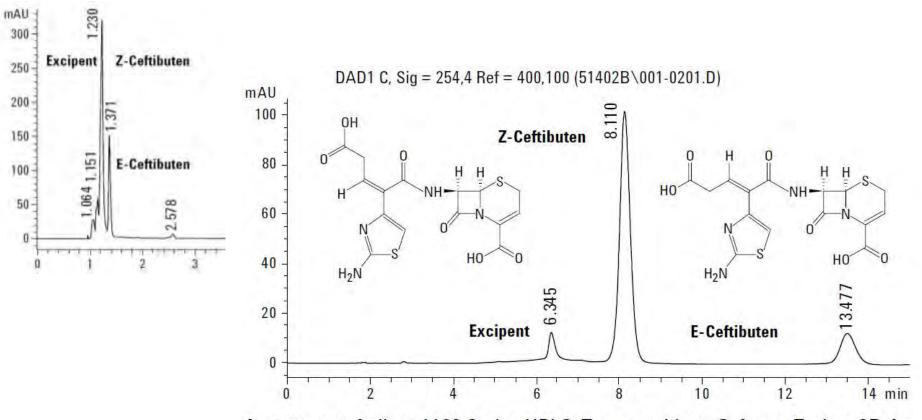
- 1. Cytosine 6. Adenine Thymine
- 2. 7. Uracil
- 3. Cytidine 8. Guanosine
- 4. Guanine 9. Thymidine
- 5. Uridine 10. Adenosine

A: 20 mM ammonium acetate, pH 4.5 B: methanol 1 mL/min, 254 nm

| Time (min) | % B |
|------------|-----|
| 0 | 1 |
| 4 | 1 |
| 6 | 50 |



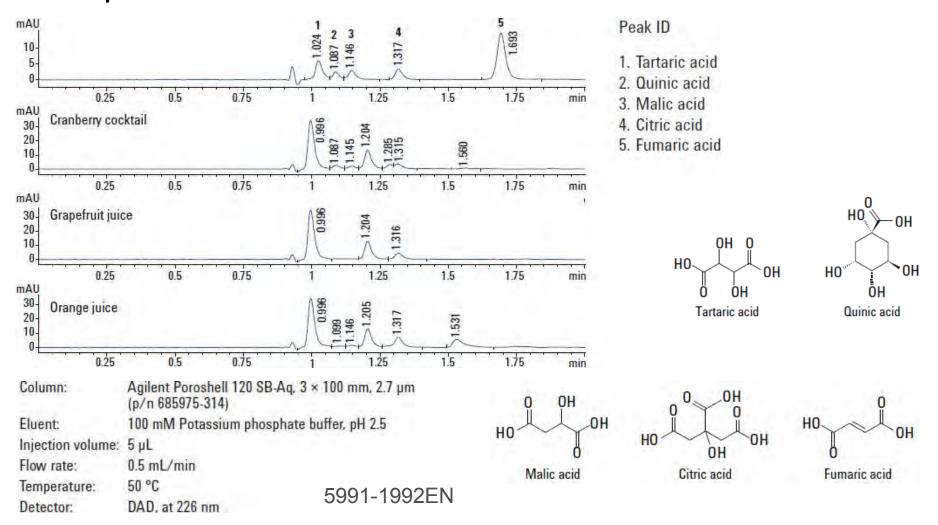
Our First Example, now with SB-Aq



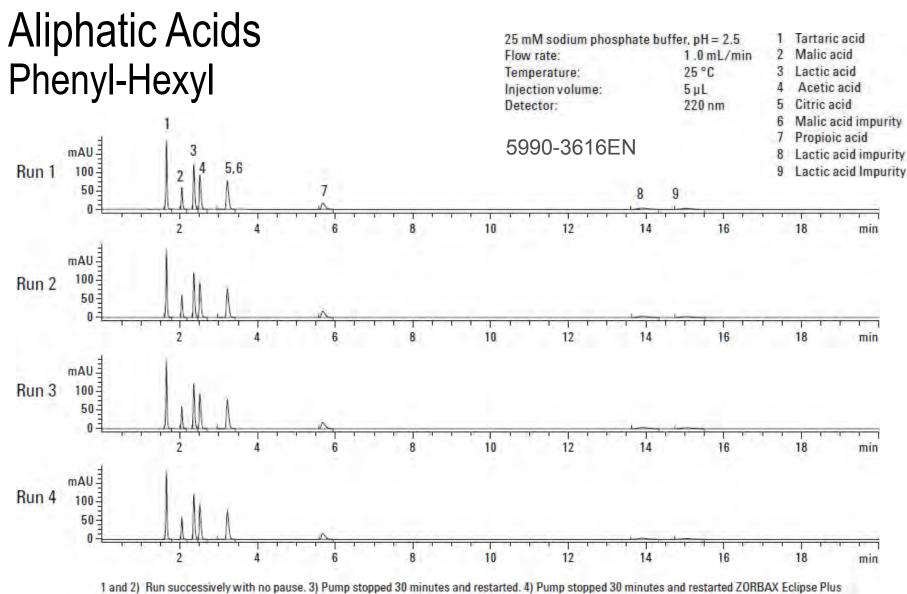
Instrument: Agilent 1100 Series HPLC; **Temp:** ambient; **Column:** Zorbax SB-Aq, $4.6 \times 150 \text{ mm}$, $5 \mu \text{m}$ (part number 883975-914); **Mobile phase:** 100% 10 mM ammonium acetate, pH 5.4; **Flow rate:** 1 mL/min; **Injection volume:** 5 μ L; **Diode array detector:** 254 nm; **Reference:** 400 nm; **Bandwidth:** 100 nm



Aliphatic Acids SB-Aq







Phenyl-Hexyl 4.6 mm × 150 mm 3.5 micron, p/n 959961-912.



What Now?

- Adjust method conditions
- Ion-pair chromatography
- •Alternate column choice

HILIC (Hydrophilic Interaction Chromatography)



HILIC Hydrophilic Interaction Chromatography

- Polar stationary phase:
 - Silica
 - Amine
 - Amide
 - Diol

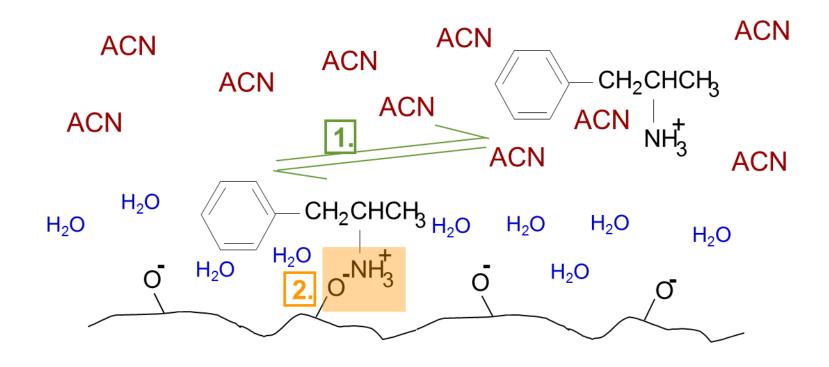


HILIC Hydrophilic Interaction Chromatography

- Polar stationary phase:
 - Silica
 - Amine
 - Amide
 - Diol
- Polar mobile phase:
 - Water is the strong solvent
 - THF<acetone<ACN<iPrOH<EtOH<MeOH<water
 - Typically ACN/water
 - •Buffer controls ionization of analyte and stationary phase
 - •Typically ammonium acetate or ammonium formate



How Does HILIC Work on Silica?



Partitioning in and out of adsorbed water layer
 Ion exchange with silanols



HILIC Advantages

 Good peak shape for basic compounds where RP may give tailing and/or low efficiency

- Low viscosity mobile phases with high organic content allow the use of higher flow rates and/or long columns
- Enhanced detection sensitivity with MS
- Can directly inject ACN extracts from C18 SPE cartridges



HILIC Challenges

- Slower equilibration than RPLC
 - Particularly true for bare silica columns
 - Longer to equilibrate initially
 - Longer to equilibrate when mobile phase changes for gradients or method development are required
- Peak distortion with mobile phase / sample solvent mismatch

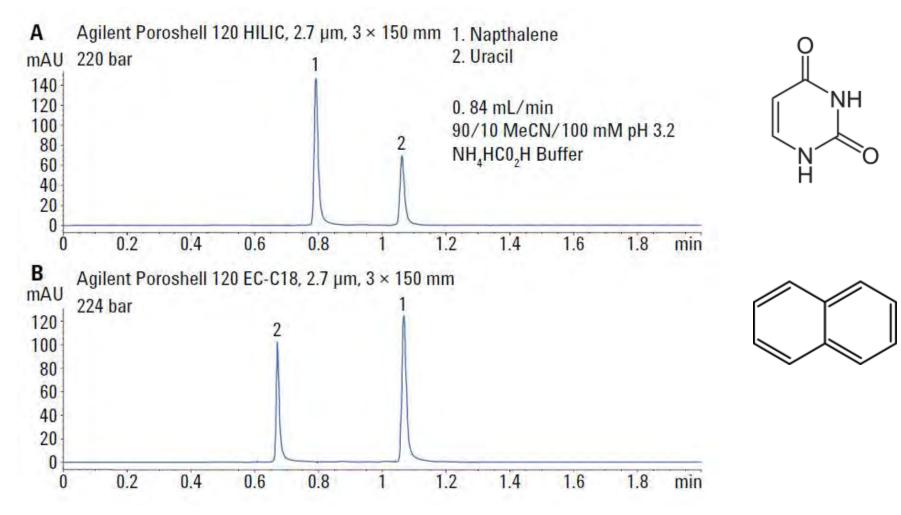


Typical Conditions

- Silica column (ZORBAX Rx-SIL, HILIC Plus, Poroshell 120 HILIC)
- Water (at least 2-3%, ~ 25%)/ACN
- Buffer (e.g., ammonium acetate)
- pH control

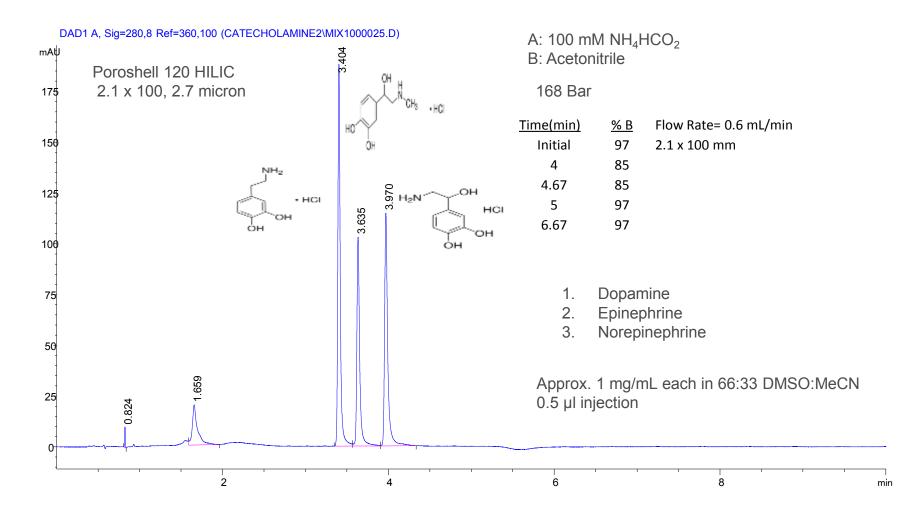


HILIC – comparison with C18



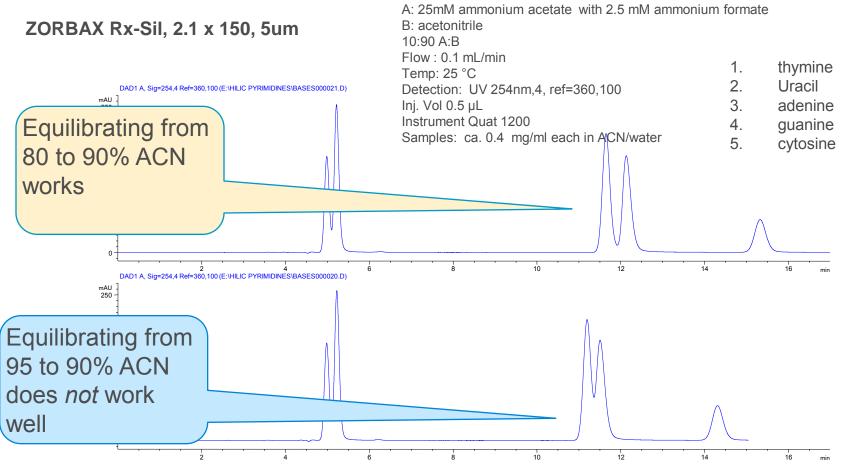


HILIC Separation of Catecholamines Poroshell 120 2.1 x 100, 2.7 micron





Equilibrate from high aqueous to low Critical factor when changing mobile phases





Water Soluble Vitamins HILIC

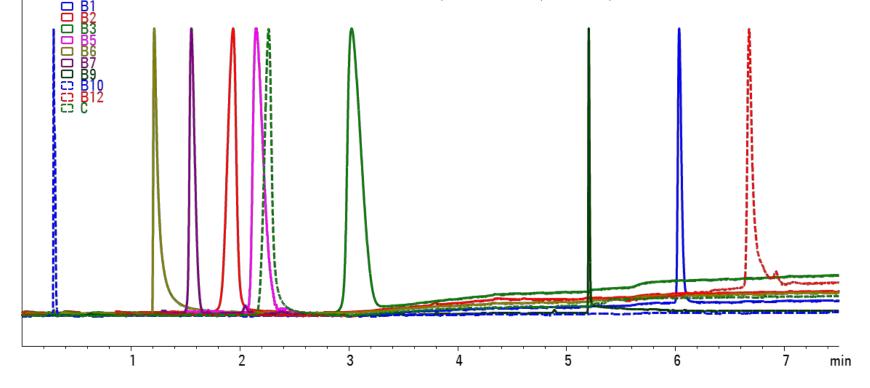
Poroshell 120 HILIC

A: 100 mM Ammonium Formate in H2O pH 3.0 with Formic Acid B: CH3CN

0.5 mL/min, 97% B for 2.5 minutes, then 97-60%D in 5 minutes injection volume: varies according to signal strength

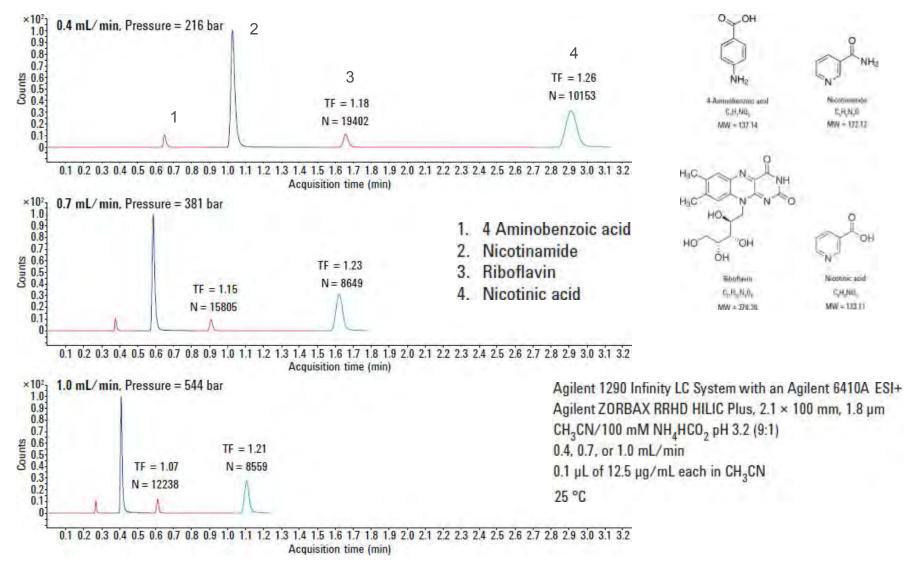
TCC: 30 C

260/210, 8 nm Ref Off, 8 nm slit, 80 Hz



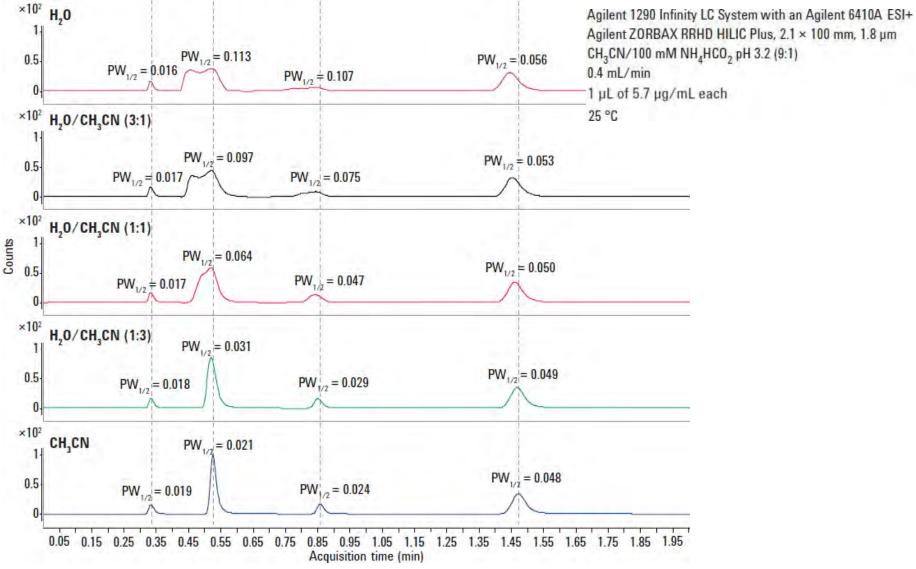


HILIC Separation





HILIC and Choice of Sample Solvent





Summary

- •What do you do when your analyte is too polar?
- •Stick with reversed-phase but
 - Adjust pH of mobile phase
 - Phenyl-Hexyl, Bonus-RP, SB-Aq, EC-CN
- •Consider HILIC
- Ion-pair chromatography

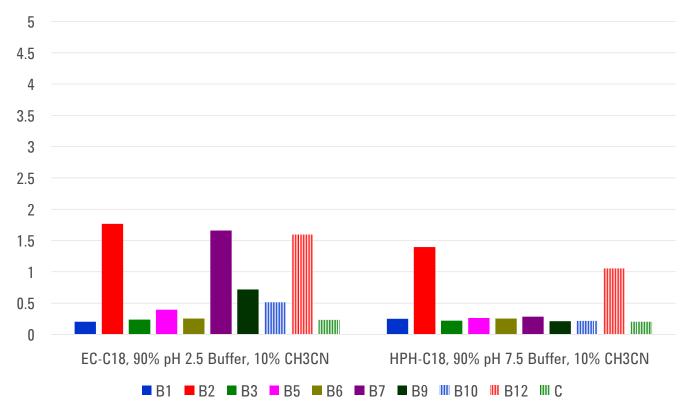


Additional slides

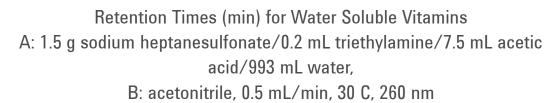


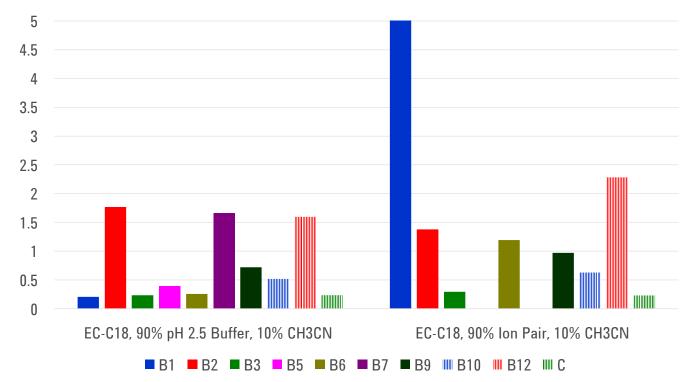
Water Soluble Vitamins C18 at pH 2.5 vs 7.5

Retention Times (min) for Water Soluble Vitamins A: 20 mM sodium phosphate pH 2.5 or 7.5, B: acetonitrile, 0.5 mL/min, 30 C, 210 nm











Ion-Pair Parameters

- •IP reagent
 - •Longer alkyl chain--more readily adsorbed by stationary phase
 - •Choose alkyl length which gives best separation (more retention of amines with octanesulfonate than hexanesulfonate)
 - •Select cationic ion-pairing reagent for anions (e.g., acids)
 - •Select anionic ion-pairing reagent for cations (e.g., amines)
 - •Not both together
- •IP Concentration
 - Increase retention with increasing IP concentration
 - •Increase concentration with %B non-linear adsorption
- •pH
- Buffer concentration
- •Choice of organic modifier
- •%B
- •Temperature

