



# What's Wrong With My Chromatography ?

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LC Columns & Consumables

Technical Support

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# Outline of Presentation

## I. Common Problems, Symptoms, and Solutions

- Baseline Shifts/Noise
- Peak Shapes
- Pressure
- Retention Shifts

## II. Preventative Maintenance/Good Practices

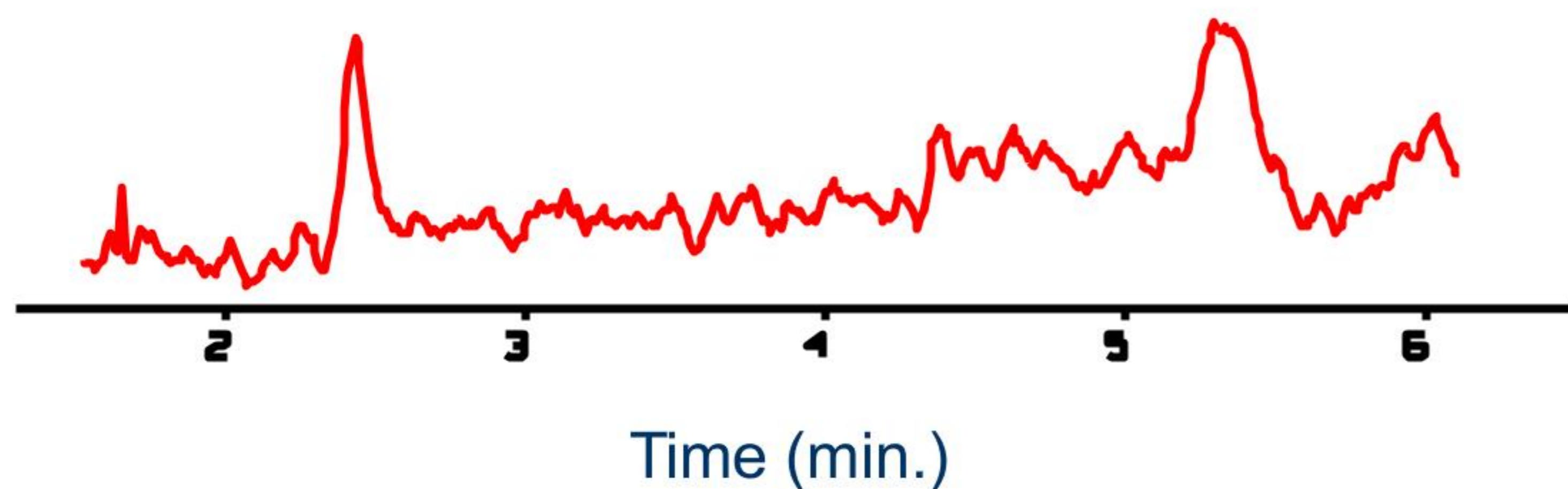
- In-line Devices
- Sample Preparation
- Operational Limitations

# Common Problems

## Noisy Baselines

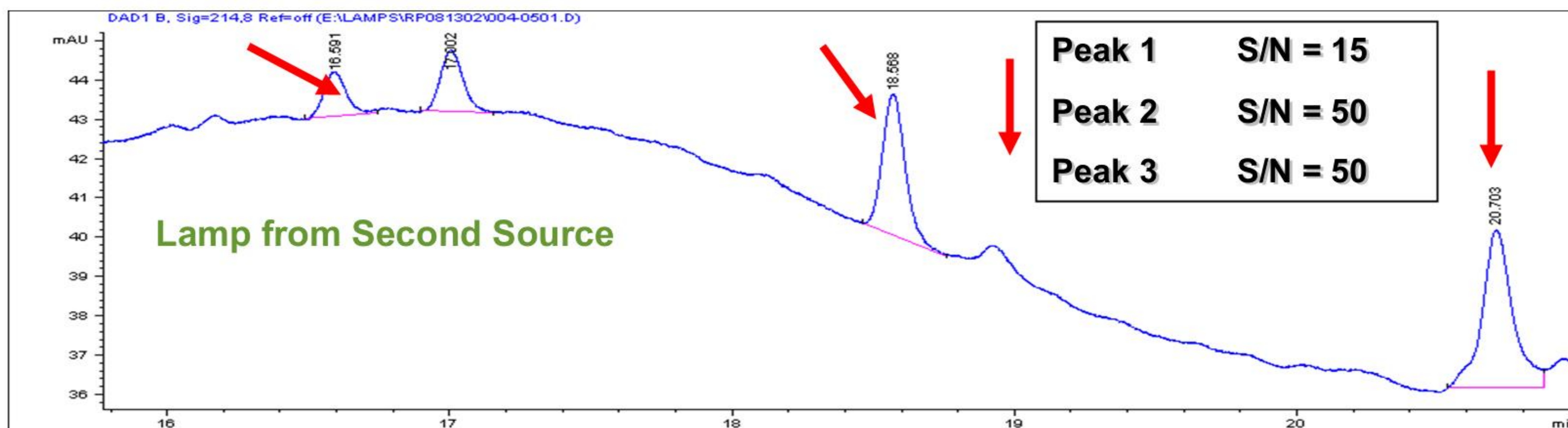
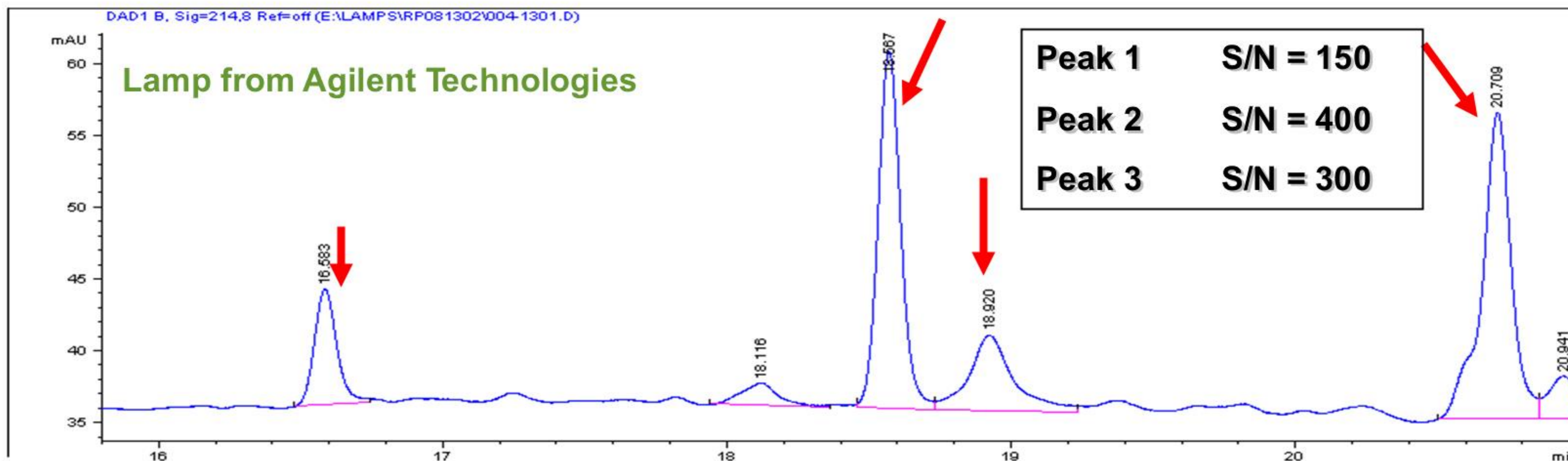
### Possible Causes

- Dirty flow cell
- Lamp failing
- Pulses from pump (if periodic)
- Temperature effects on detector
- Air bubbles passing through detector



# Non-spec Lamp

Chromatographic Results with Second Source Lamp at 214 nm



# Drifting Baseline



## Possible Causes

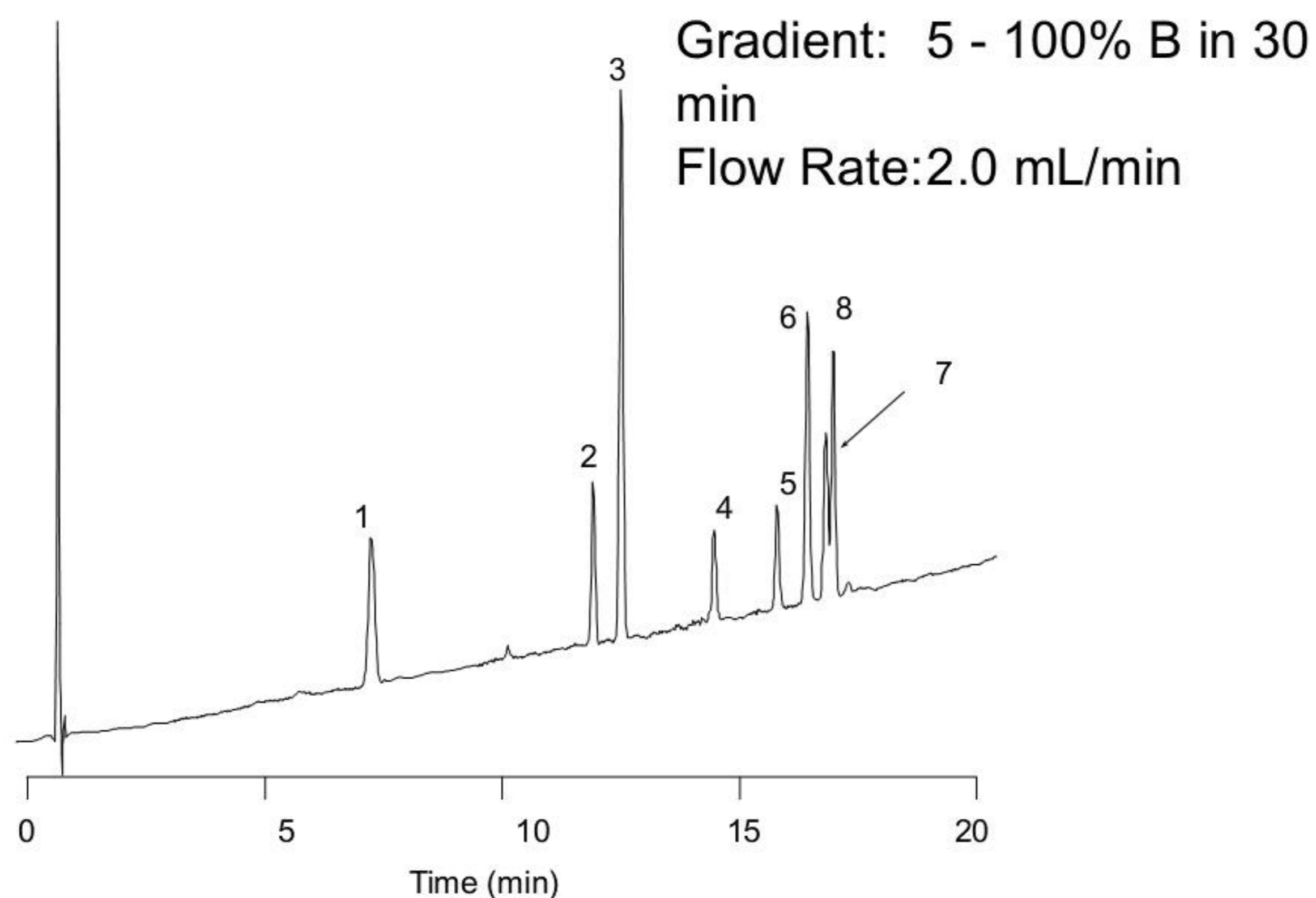
- Gradient elution
  - Mobile phase component
- Temperature unstable (RID)
- Contamination in mobile phase
- Mobile phase not in equilibrium with column
- Contaminant bleed in system
  - Hardware chemical compatibility

# Effect of TFA on Baseline

A: 0.1% TFA in H<sub>2</sub>O    B: 0.1% TFA in ACN    Temperature: 35°C

Sample: 1. Phenacetin    2. Tolmetin    3. Ketoprofen    4. Fenoprofen    5. Ibuprofen    6. Phenylbutazone    7. Mefenamic acid    8. Flufenamic acid

**Eclipse XDB-C8**  
4.6 x 150 mm, 5 µm



✓ Wavelength

• 215nm

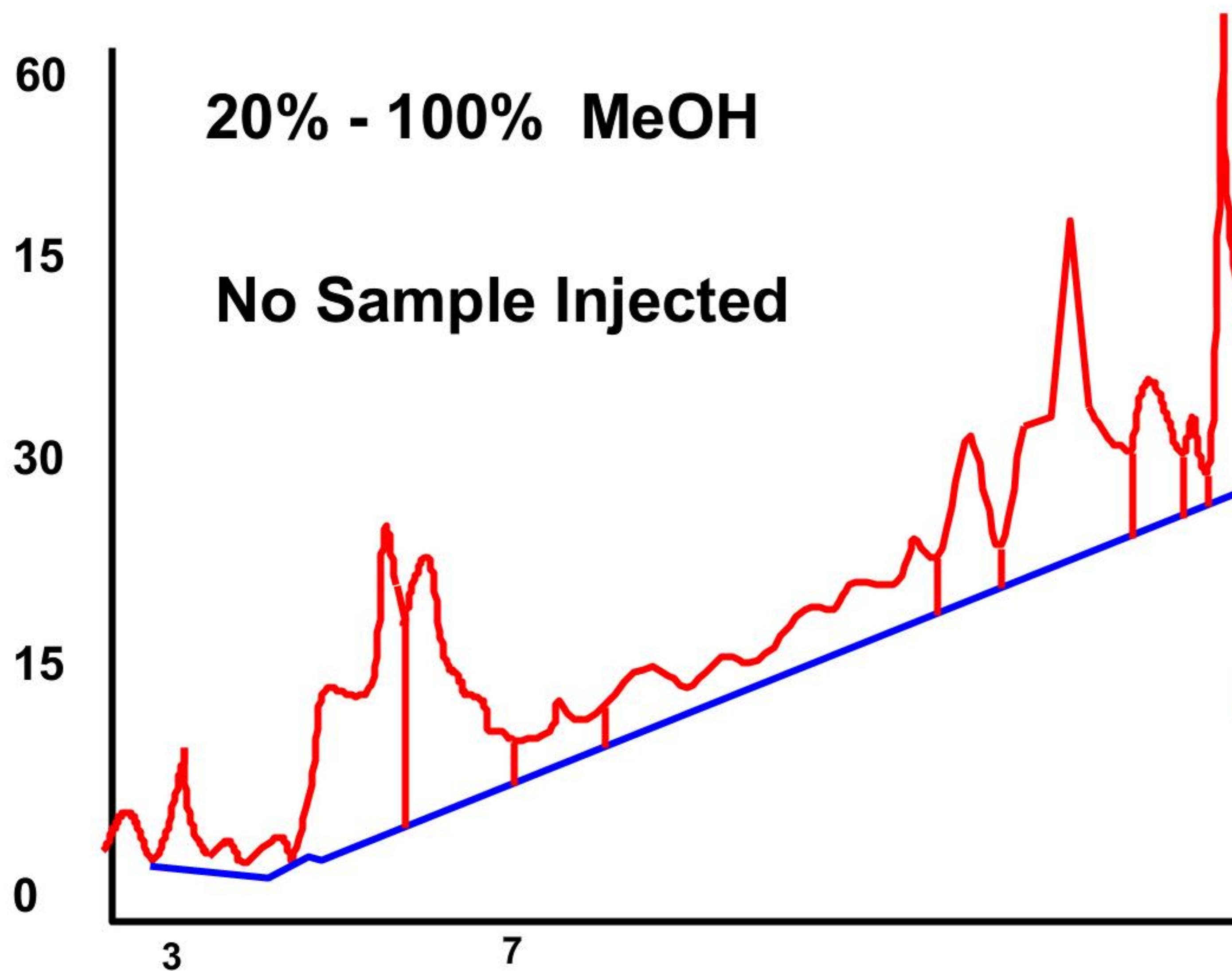
• 254nm less impact

✓ Adjust TFA concentration in Solvent B to level baseline

➤ Know the UV Cutoff of your mobile phase components

# Peak Shape

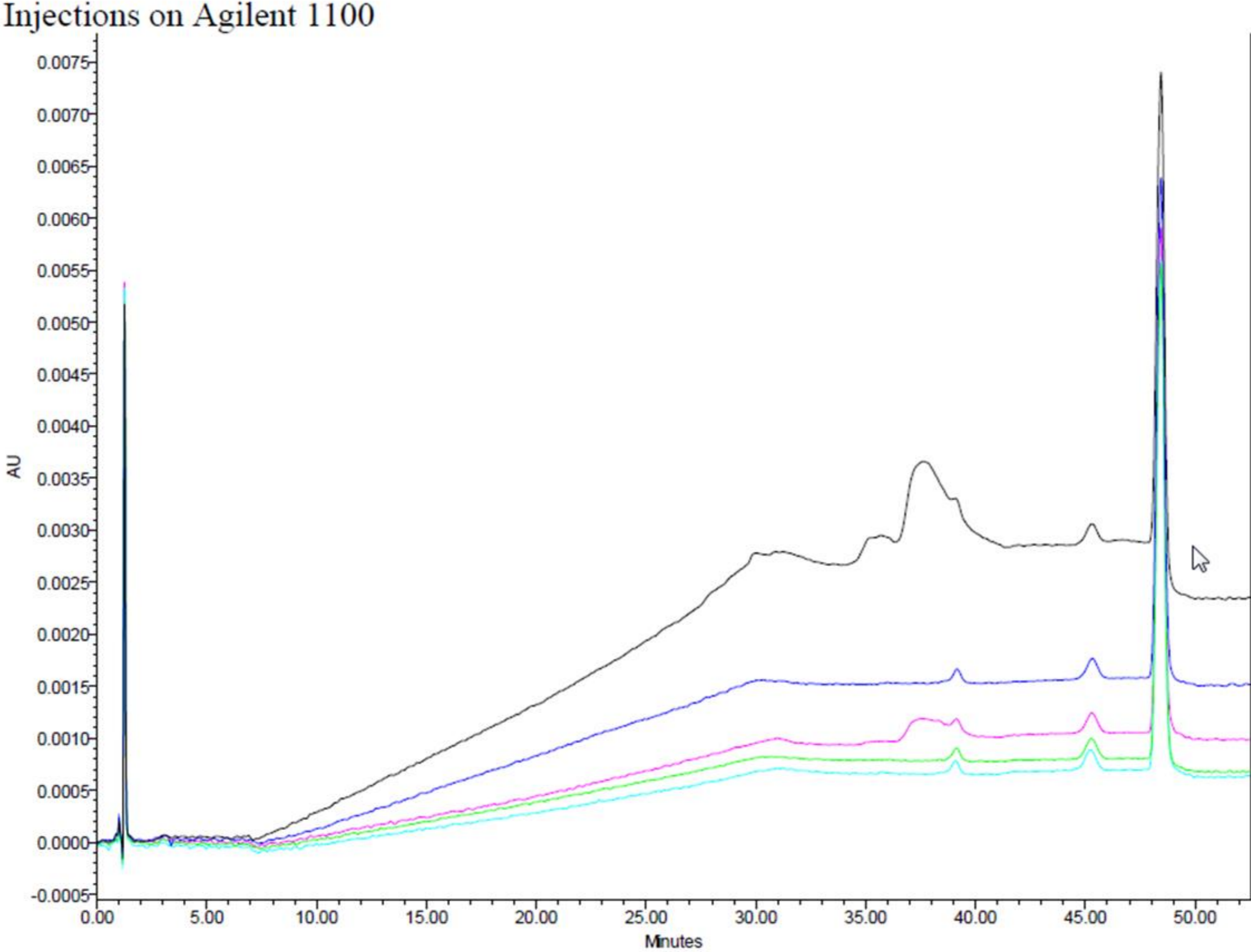
Ghost Peaks – Appear When No Sample is Injected



## Problem

- Dirty mobile phase
- Sample carryover
  - May imply poor recovery
- Peak from an early run (isocratic)

# Solvent Contamination

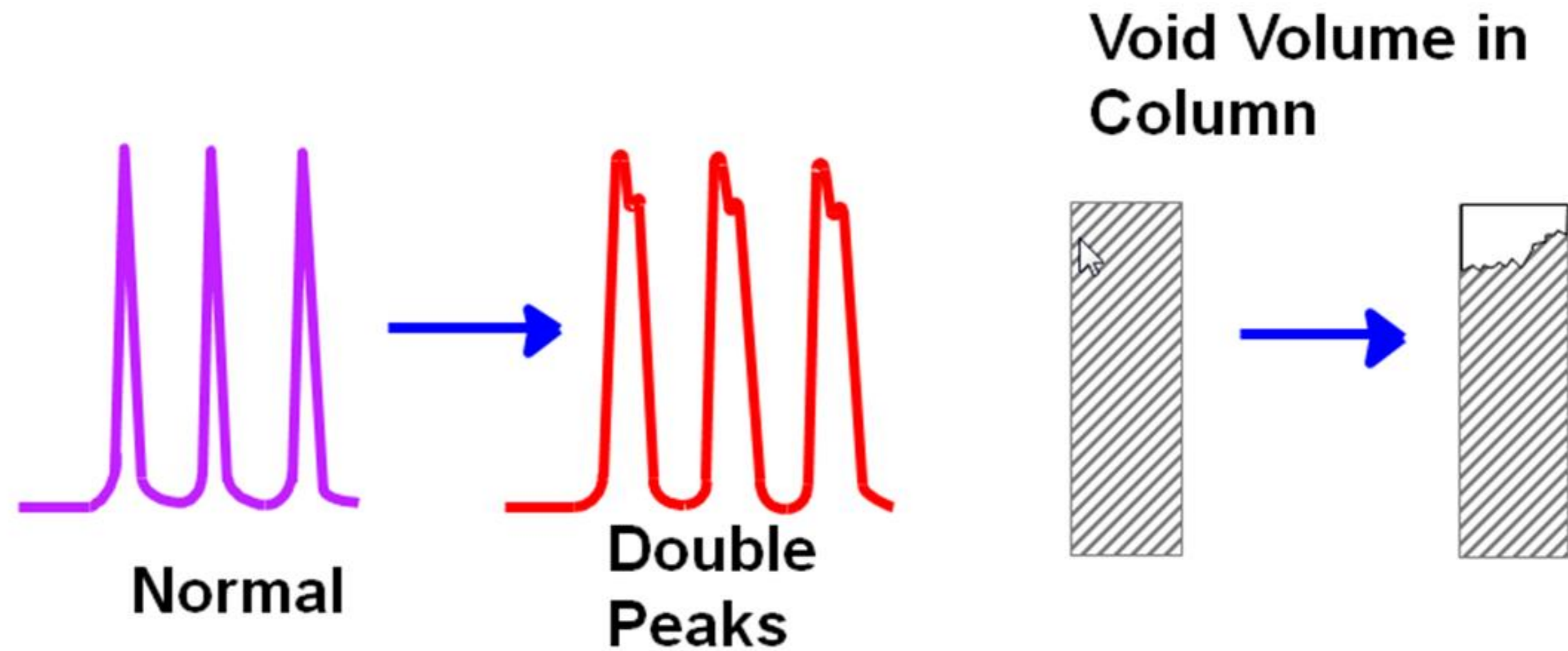


- Solvent Source 1
- Solvent Source 2
- Solvent Source 3
- Solvent Source 4, Lot 1
- Solvent Source 4, Lot 2



# Peak Shape

## Double Peaks



### Possible Causes

- Void in column
- Partially plugged frit
- Only one peak
- Sample solvent mismatch

# Strong Sample Solvent Can Compromise Peak Shape

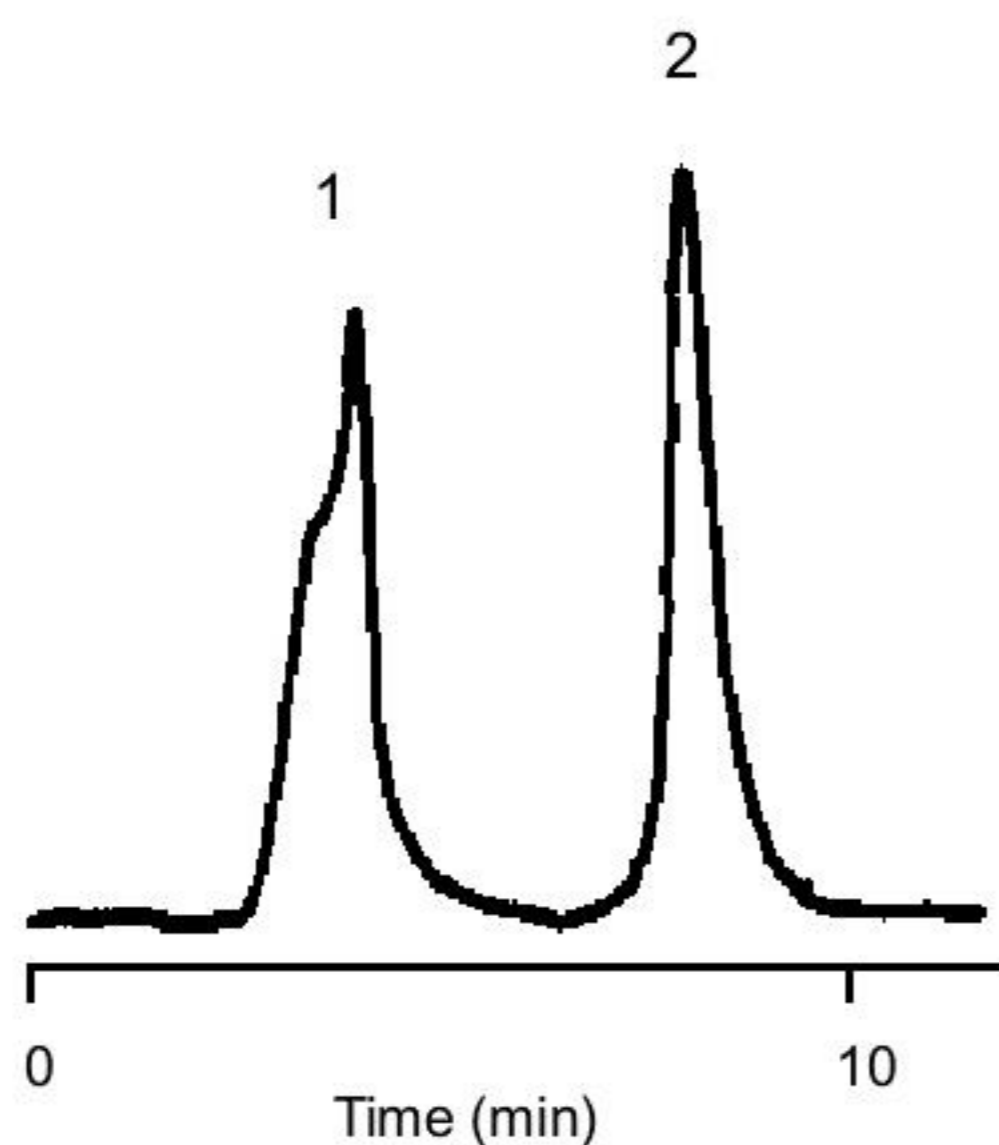
**Column:** ZORBAX SB-C8, 4.6 x 150 mm, 5  $\mu$ m

**Mobile Phase:** 82% H<sub>2</sub>O:18% ACN

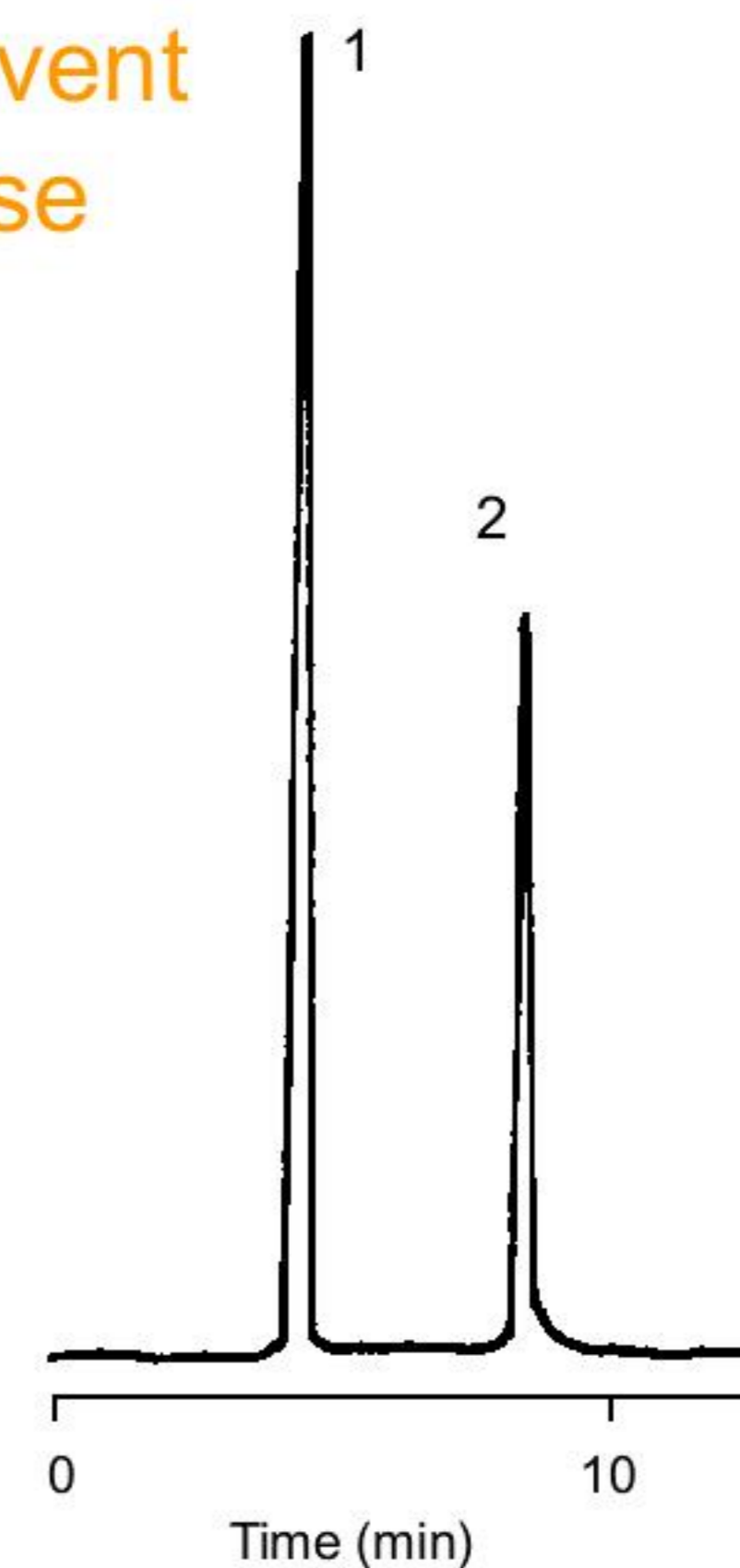
**Injection Volume:** 30  $\mu$ L

**Sample:** 1. Caffeine 2. Salicylamide

A. Sample Solvent  
100% Acetonitrile

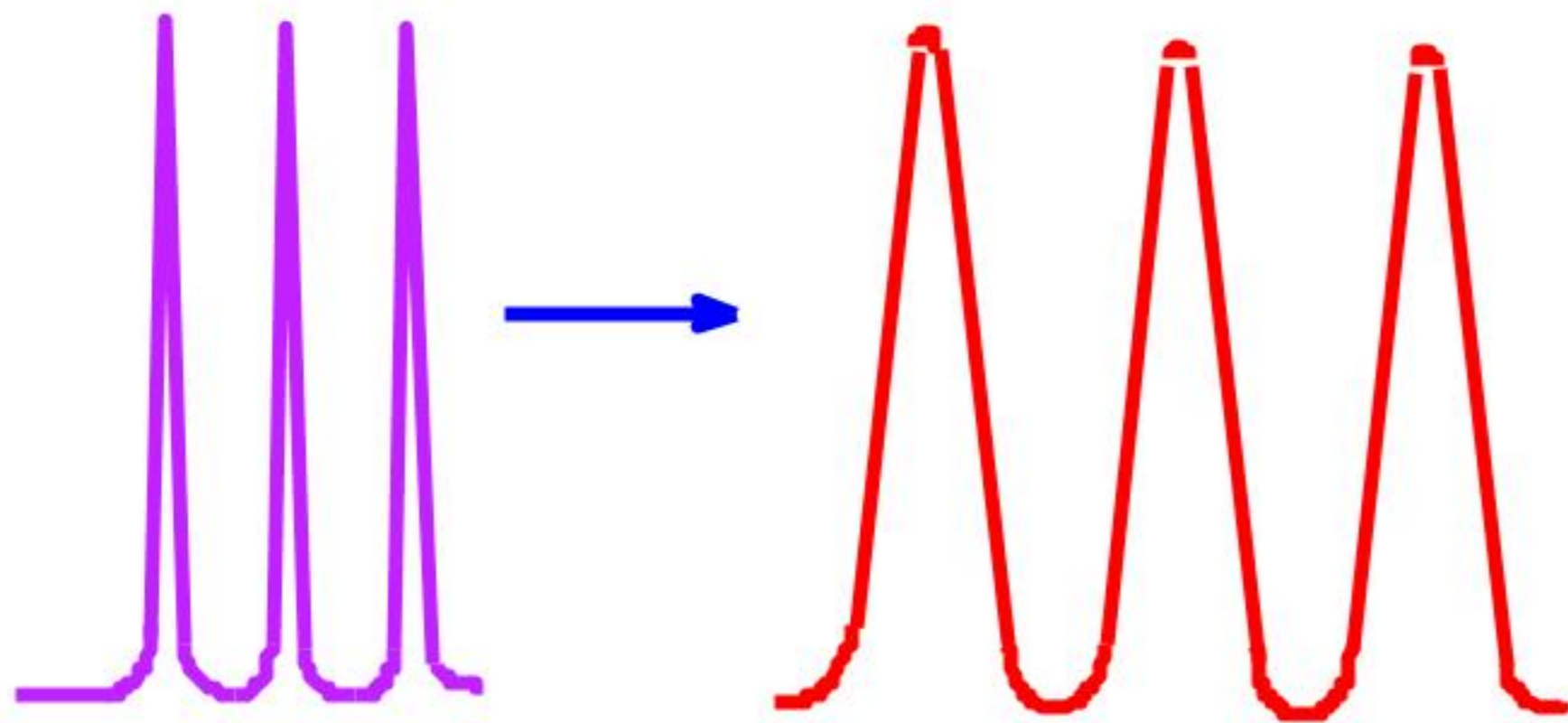


B. Sample Solvent  
Mobile Phase



# Peak Shape

## Broad Peaks

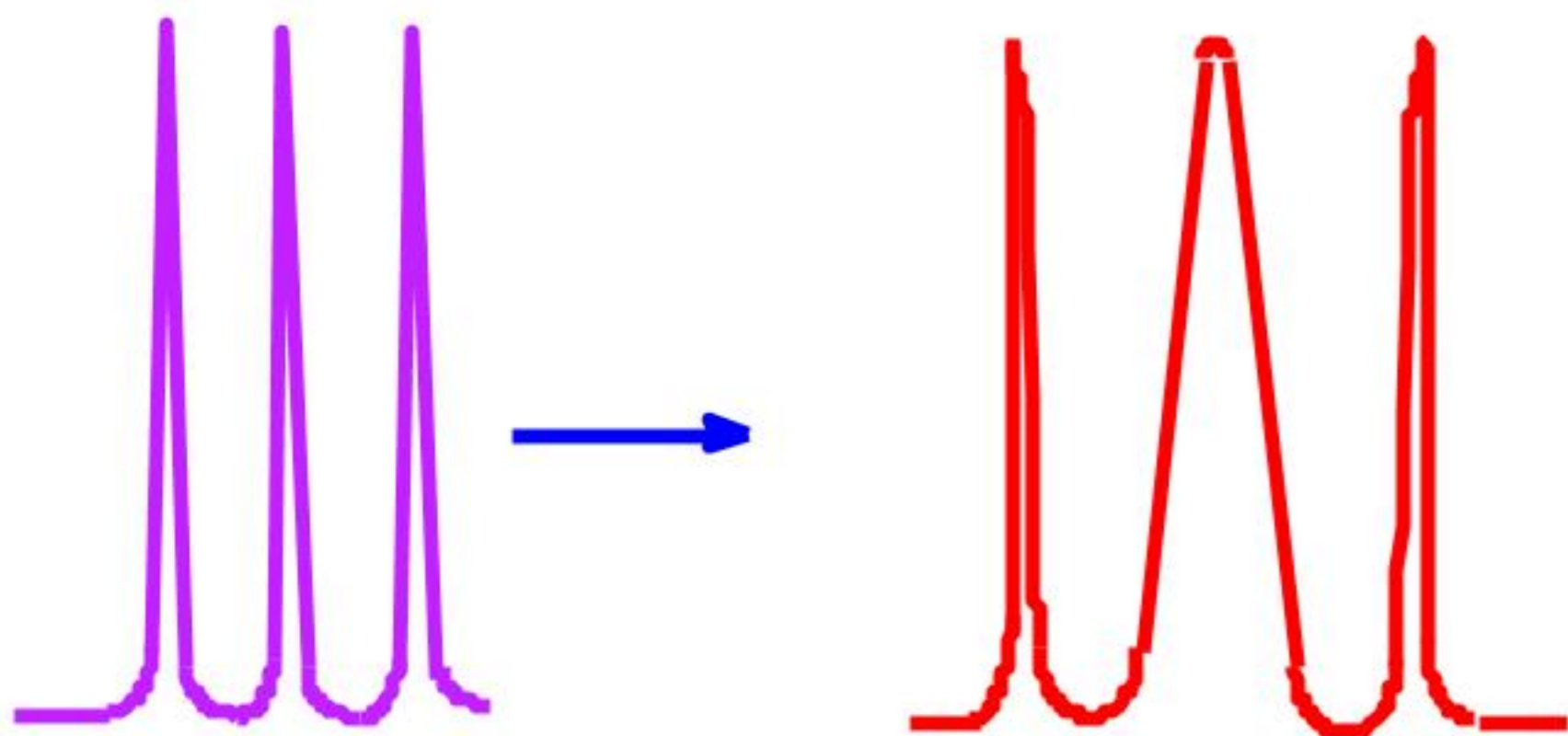


### All Peaks Broadened

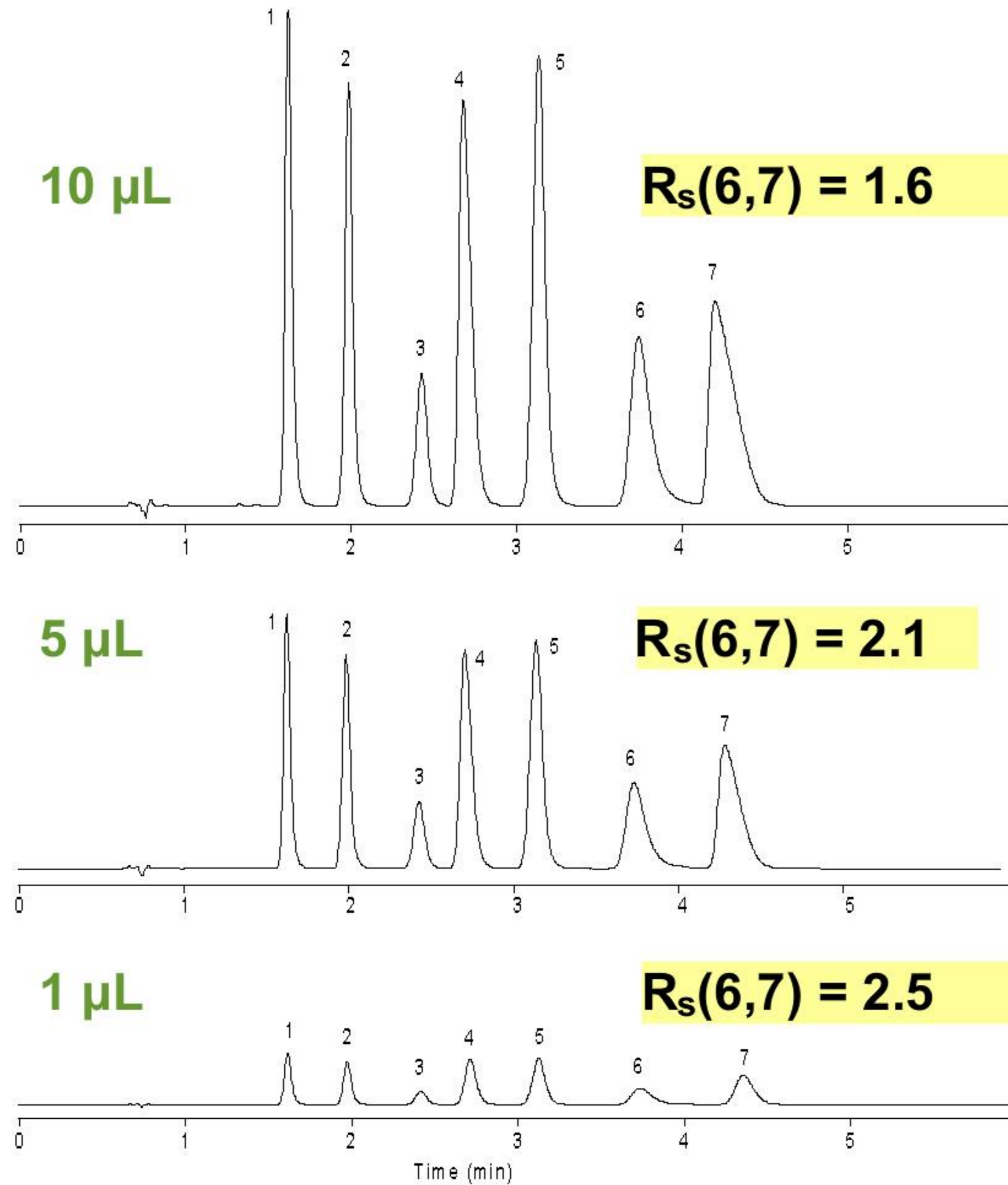
- Loss of column efficiency
- Large injection volume/mass
- High viscosity mobile phase
- Sample solvent mismatch

### Some Peaks Broadened

- Late elution from previous sample (isocratic)
- High MW sample – Protein or polymer



# Injection Volume Robustness



Column: ZORBAX Rapid Resolution Eclipse XDB-C8  
4.6 x 75 mm, 3.5 µm

Mobile Phase: 44% 25 mM phosphate, pH 7.00  
56% methanol

Flow Rate: 1.0 mL/min

Temperature: 25°C

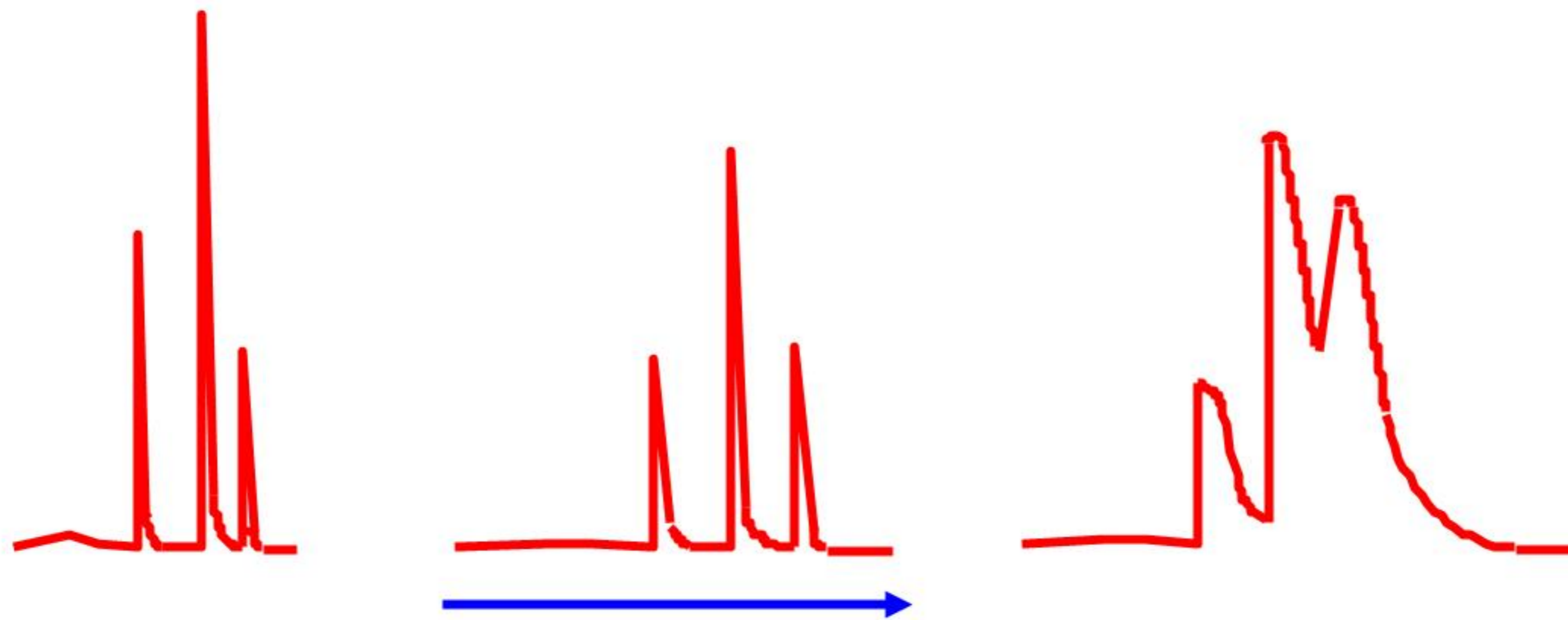
Detection: UV 250 nm

Sample:

1. ketoprofen
2. ethyl paraben
3. hydrocortisone
4. fenoprofen
5. propyl paraben
6. propranolol
7. ibuprofen

✓ Varying injection volume can sometimes reveal lack of robustness for resolution and peak shape.

# Extra Column Volume



**Increasing extra-column volume**

- **Use short, small internal diameter tubing between the injector and the column and between the column and the detector**
- **Make certain all tubing connections are made with matched fittings**
- **Use a low-volume detector cell**
- **Inject small sample volumes**

# Peak Broadening

Differences in Detector Flow Cell Volume Can Affect N and  $R_s$

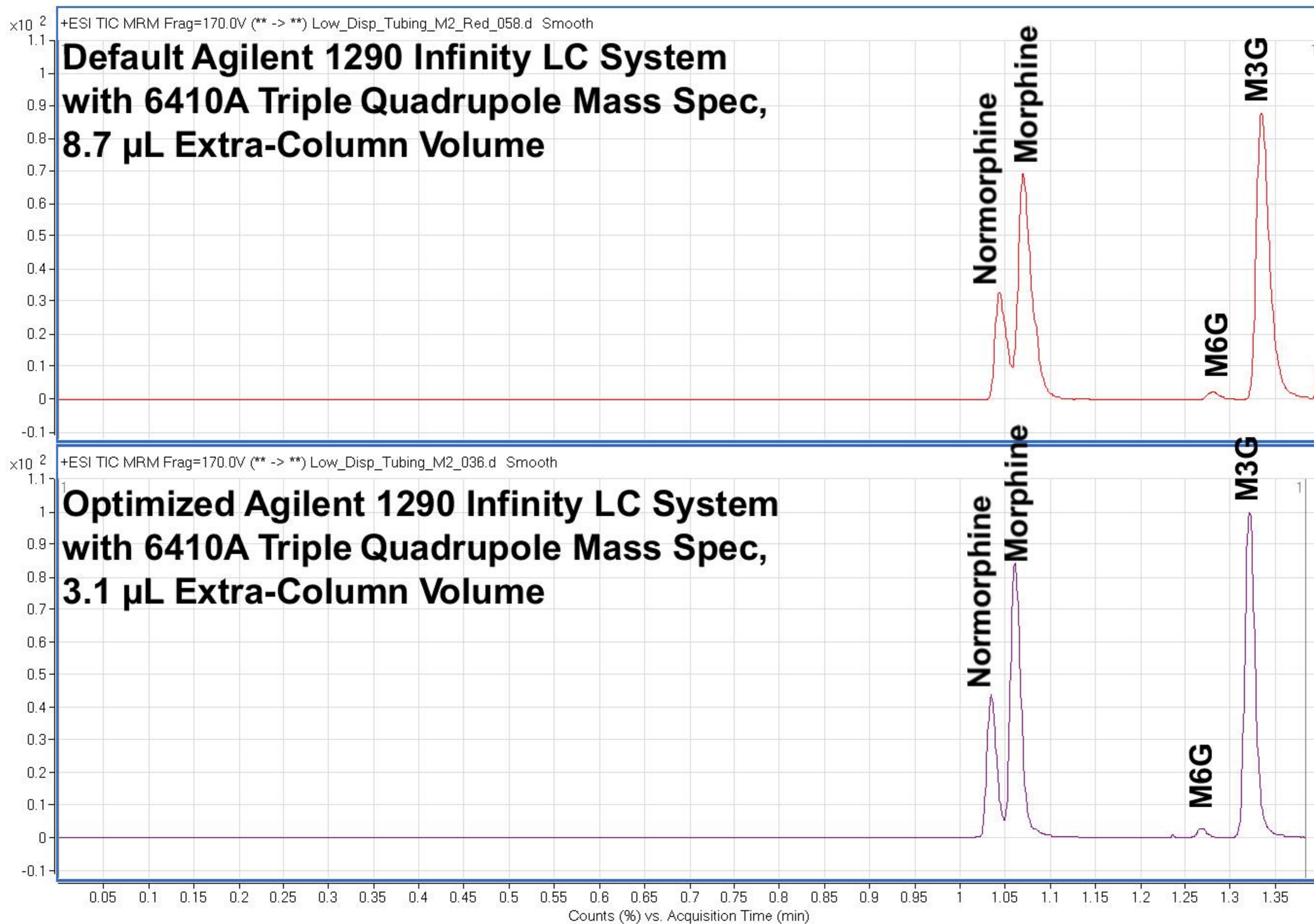
Scenario: ZORBAX Rapid Resolution Column: 75 mm, 3.5- $\mu\text{m}$ ; Flow Rate: 1mL/min;  $k = 3$

Flow Cell Volume	Band Broadening* (4.6 mm)	Band Broadening* (2.1 mm**)
1.7 $\mu\text{L}$	0.3%	6%
8 $\mu\text{L}$	6%	138%
14 $\mu\text{L}$	19%	423%

\*Versus 8571 theoretical plates (HPLC Calculations Assistant, Version 2.1, Savant Audiovisuals)

\*\*Flow Rate, 0.2 mL/min

# Analysis of Morphine and Metabolites by LC/MS/MS using an RRHD HILIC Plus Column



A: 10mM ammonium formate, pH 3.2  
B: acetonitrile/100mM ammonium formate, pH 3.2 (9:1)  
0.4 mL/min  
t(min) 0.0 0.25 1.00  
%B 100 100 55  
0.1 µL injection  
TCC: 25 °C  
MS: ESI+, dMRM, 250 C, 11 L/min, 30 psi, 4000 V, 200 V delta EMV, 40 ms cycle time  
Agilent ZORBAX RRHD HILIC Plus, 2.1 mm x 50 mm, 1.8 µm  
Sample:  
1. Normorphine  
2. Morphine  
3. Morphine-6-b-D-glucuronide (M6G)  
4. Morphine-3-b-D-glucuronide (M3G)

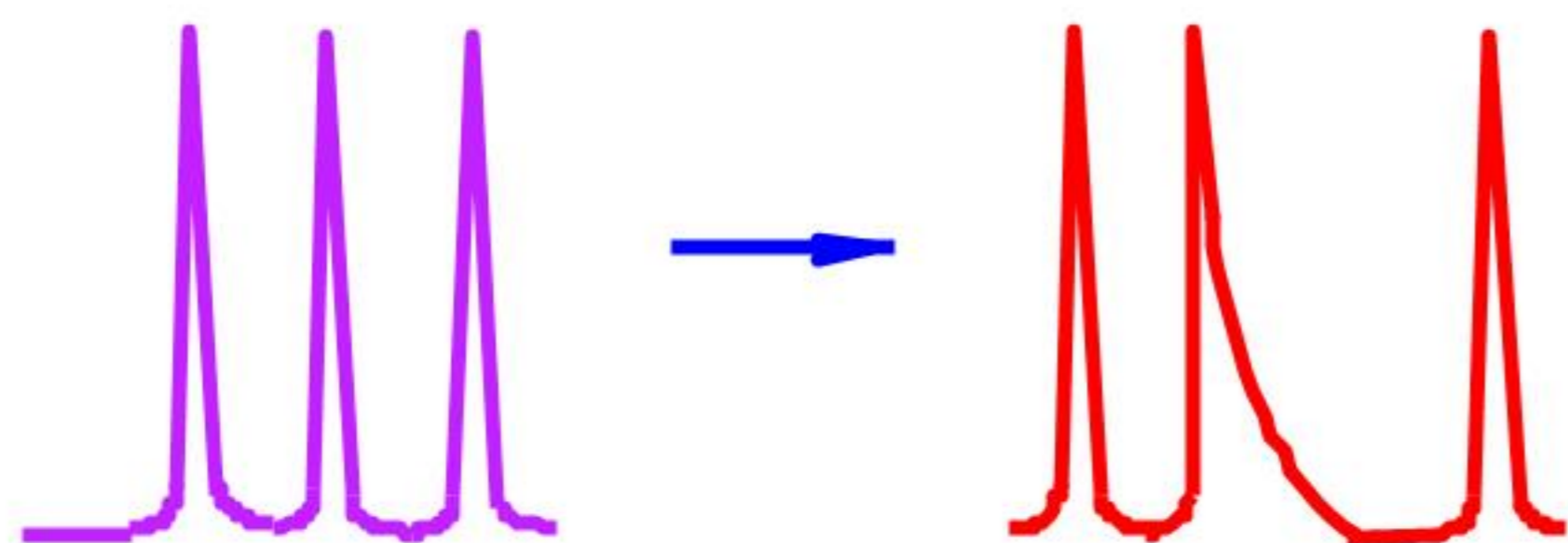
# Peak Shape

## Tailing Peaks, $>1.5$

### Causes

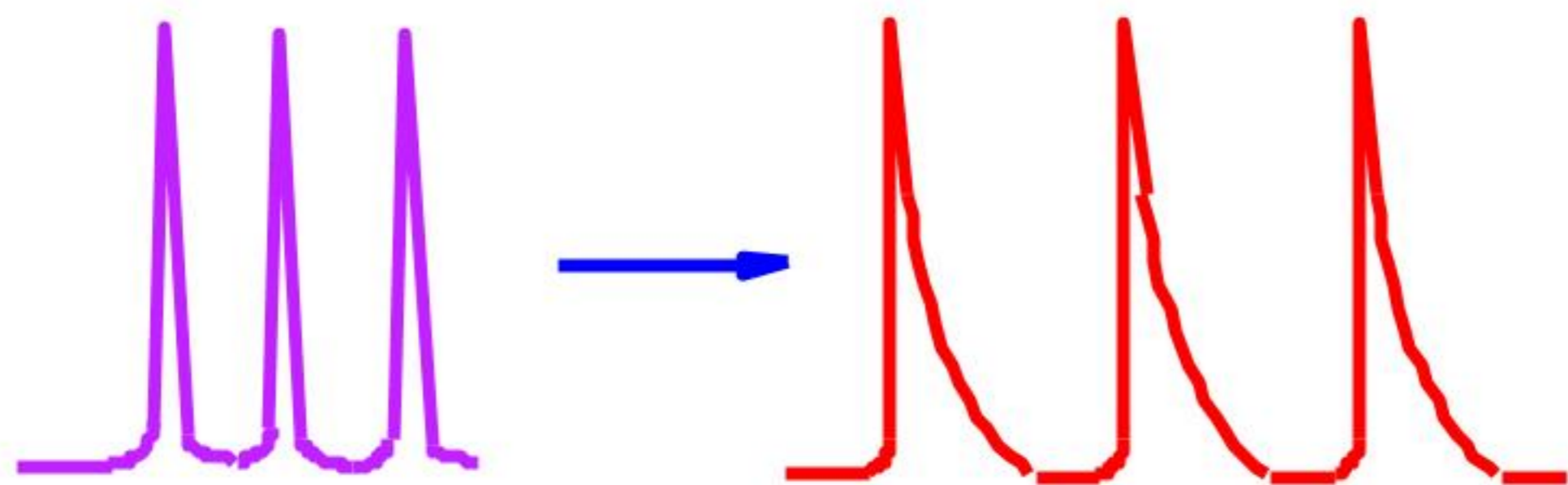
#### Some Peaks Tail

- Secondary effects
  - Residual silanol interactions
- Small peak eluting on tail of larger peak



#### All Peaks Tail

- Extra column effects
- Bad column
  - Column
  - Frit
- Contamination
- Metals
- Inappropriate sample size or solvent





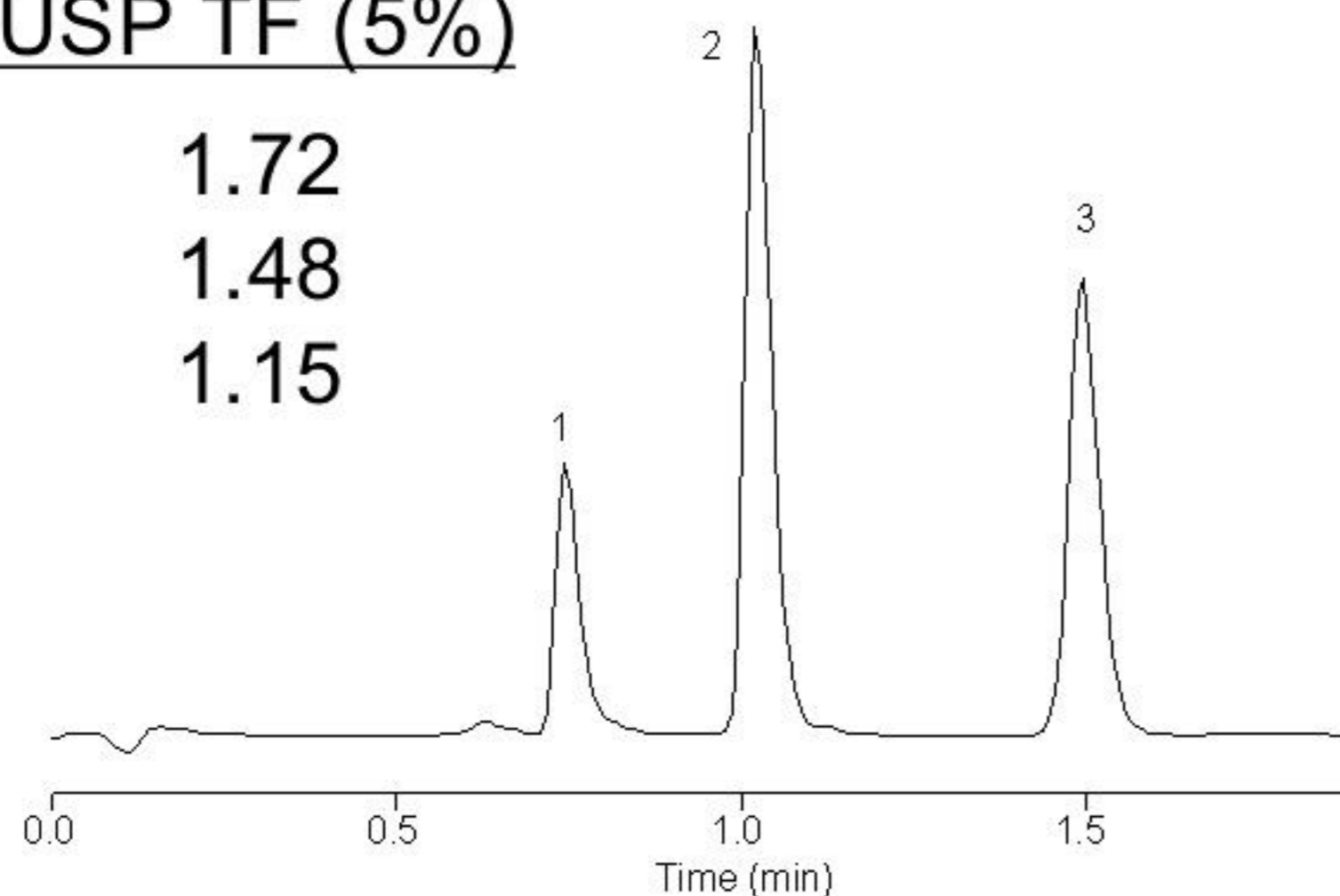
# Peak Tailing

## Injector Seal Failure

Column: Bonus-RP, 4.6 x 75 mm, 3.5 mm  
Temperature: R.T. Detection: UV 254 nm

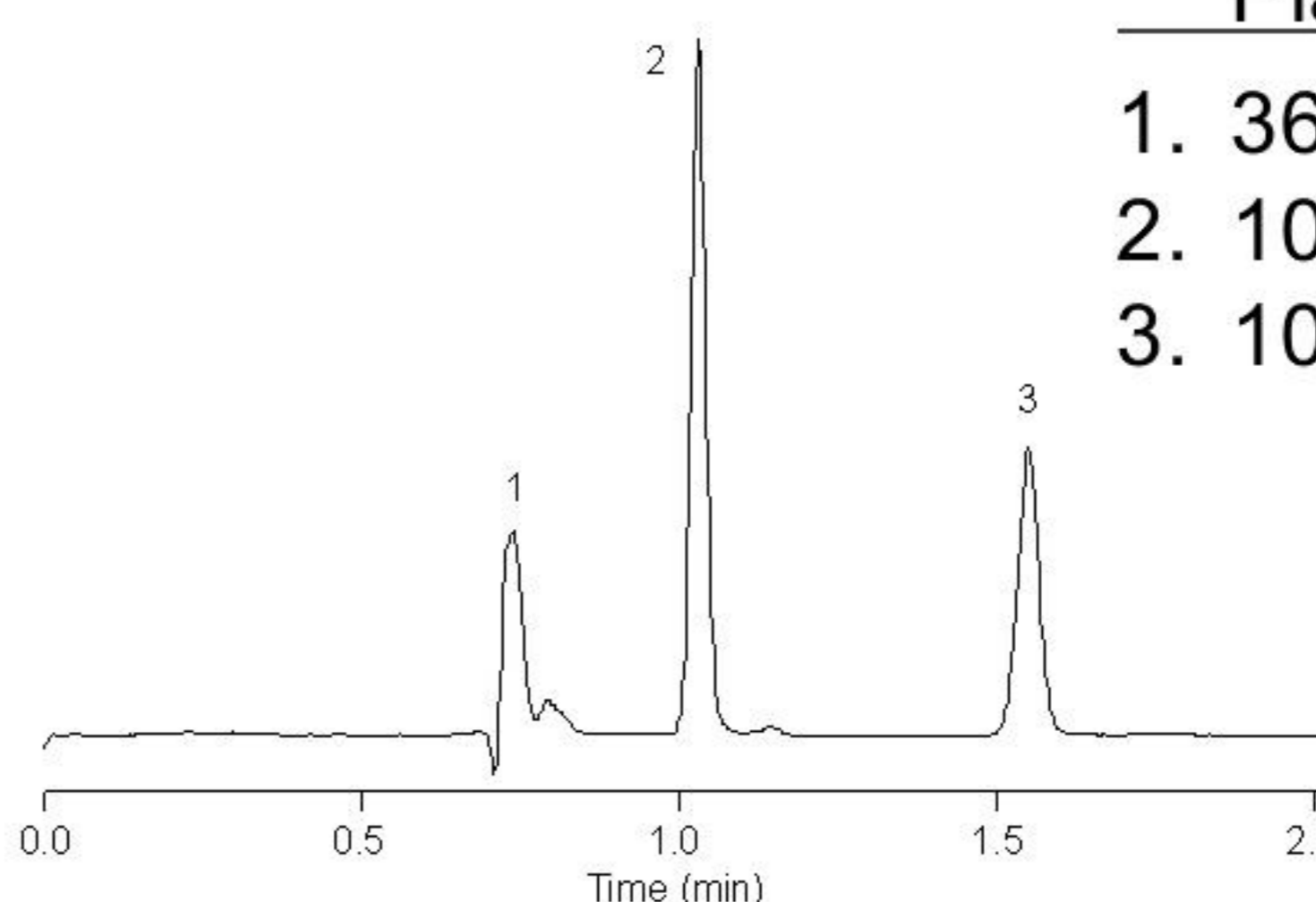
Mobile Phase: 30% H<sub>2</sub>O : 70% MeOH Flow Rate: 1.0 mL/min  
Sample: 1. Uracil 2. Phenol 3. N,N-Dimethylaniline

	Plates	USP TF (5%)
1.	2235	1.72
2.	3491	1.48
3.	5432	1.15



Before

	Plates	USP TF (5%)
1.	3670	1.45
2.	10457	1.09
3.	10085	1.00

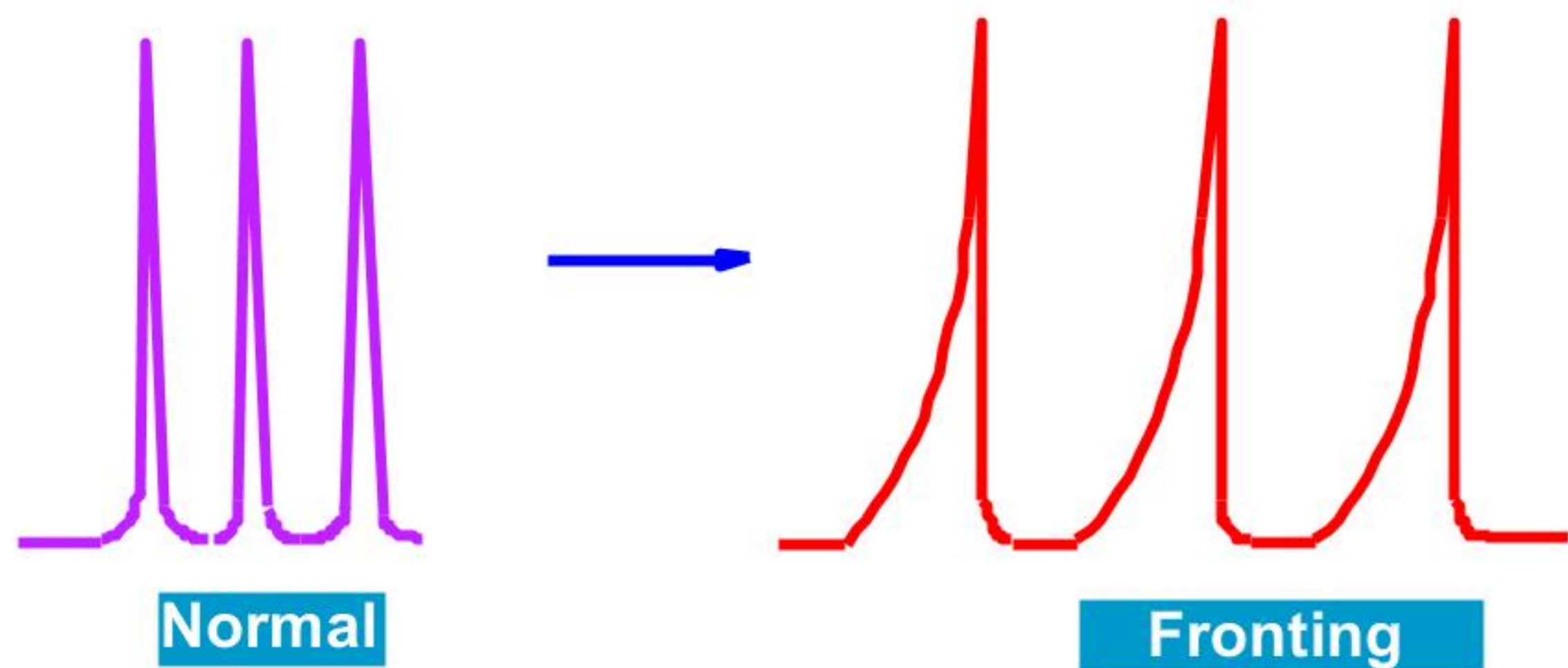


After replacing rotor seal  
and isolation seal

➤ Overdue instrument maintenance can cause peak shape problems.

# Peak Shape

## Fronting Peaks, Symmetry $< 0.9$



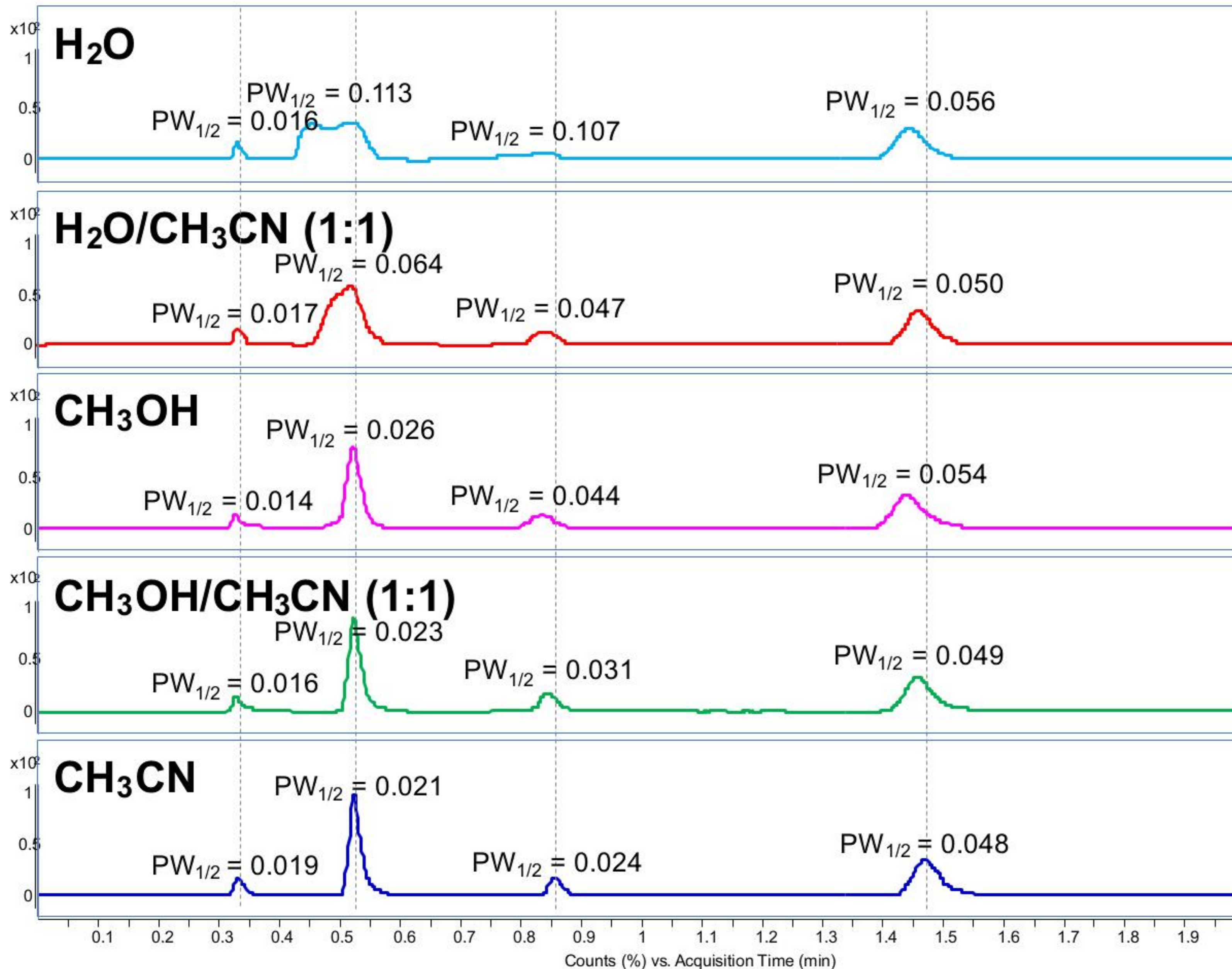
### Causes

- Small band eluting before large band
- Column void
- Inappropriate sample solvent

# Injection Solvent Effects - HILIC

## H<sub>2</sub>O, CH<sub>3</sub>OH, CH<sub>3</sub>CN

### LC/MS Separation of Vitamin B Related Compounds



Agilent 1290 Infinity LC System  
Agilent 6410A LC/MS  
Agilent ZORBAX RRHD HILIC Plus 2.1 x 50 mm, 1.8 μm

Acetonitrile / 100 mM ammonium formate pH 3.2 (9:1)  
0.4 mL/min, Pressure: 135 bar  
Isocratic elution  
Injection Volume: 1 μL of 5 μg/mL sample  
Column: 25 °C  
MS: ESI+, SIM, 200 °C, 10 L/min, 30 psi, 4000 V, 15 ms dwell time

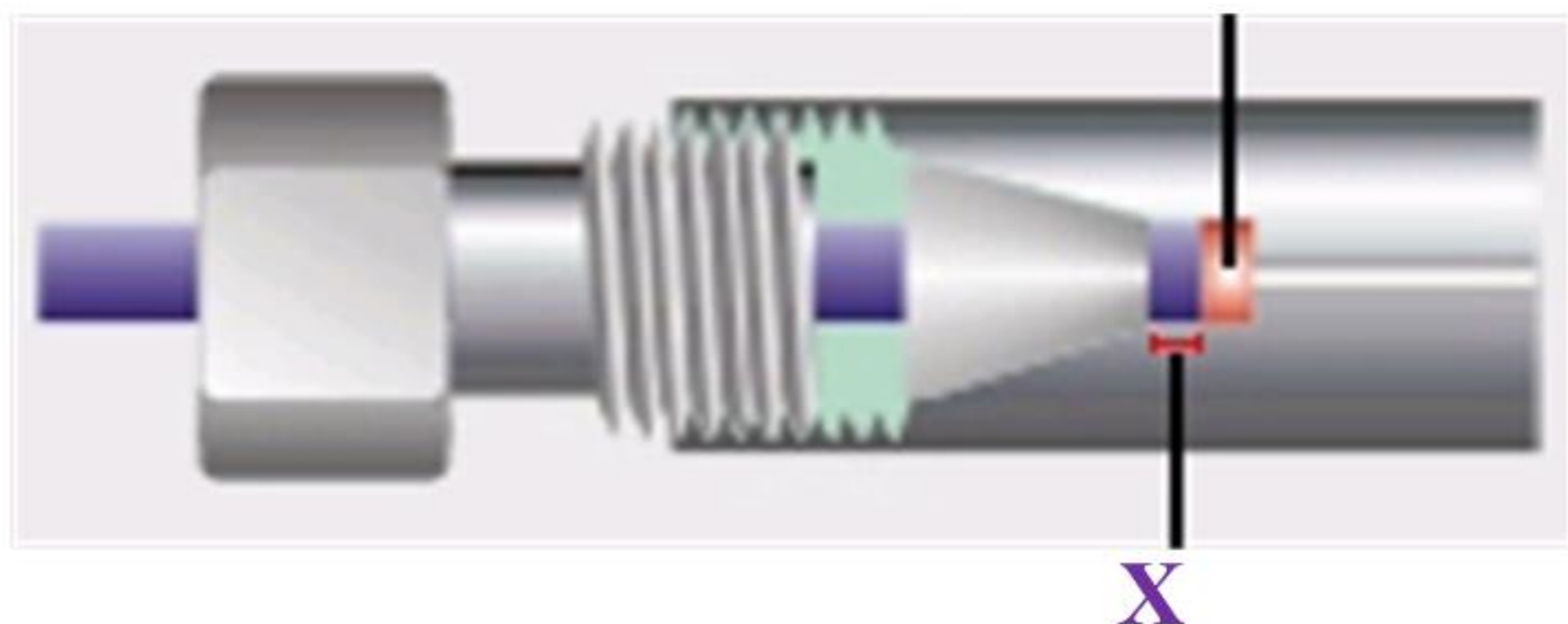
Sample:  
4-Aminobenzoic acid, m/z 138 (Frag 110 V)  
Nicotinamide, m/z 123 (Frag 130 V)  
Riboflavin, m/z 377 (Frag 160 V)  
Nicotinic acid, m/z 124 (Frag 130 V)

➤ **Strong injection solvents negatively affect peak shape and retention**

# Poorly Made Connections = Peak tailing/fronting

**Wrong ... too short**

Mixing Chamber

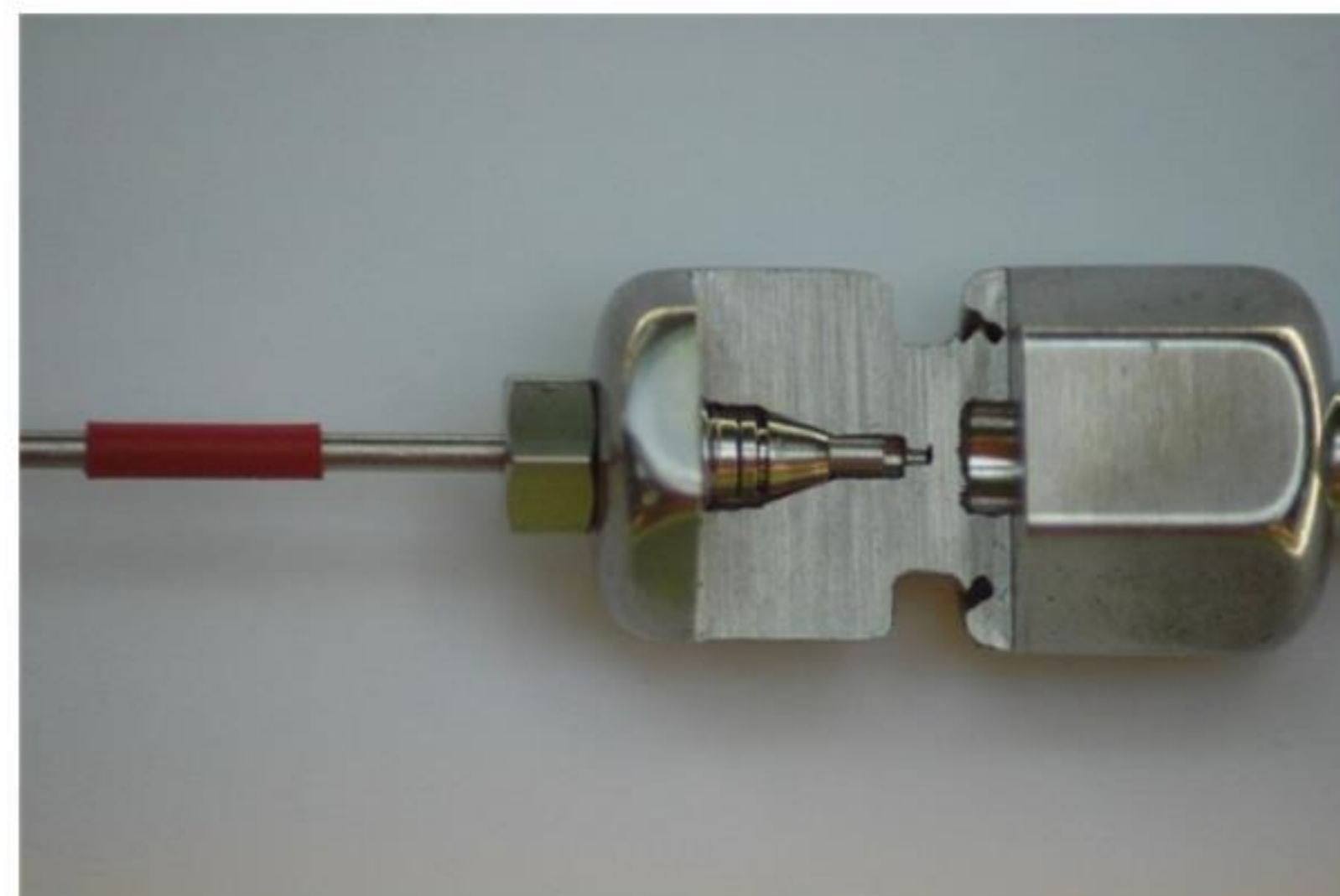
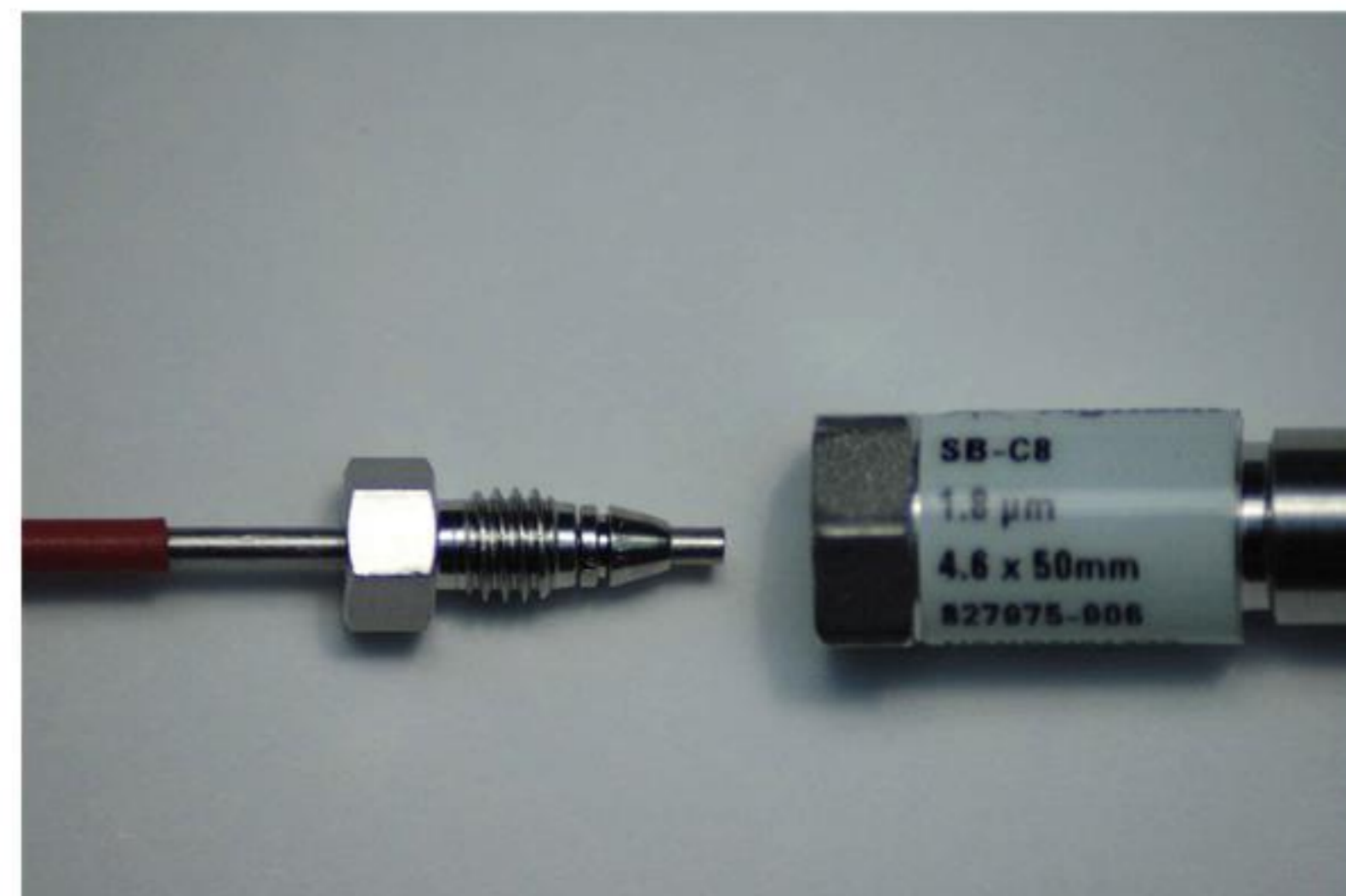


If Dimension **X** is too short, a dead-volume, or mixing chamber, will occur.

This can broaden or split peaks and/or cause tailing.

It will typically affect all peaks, but especially early eluting isocratic peaks

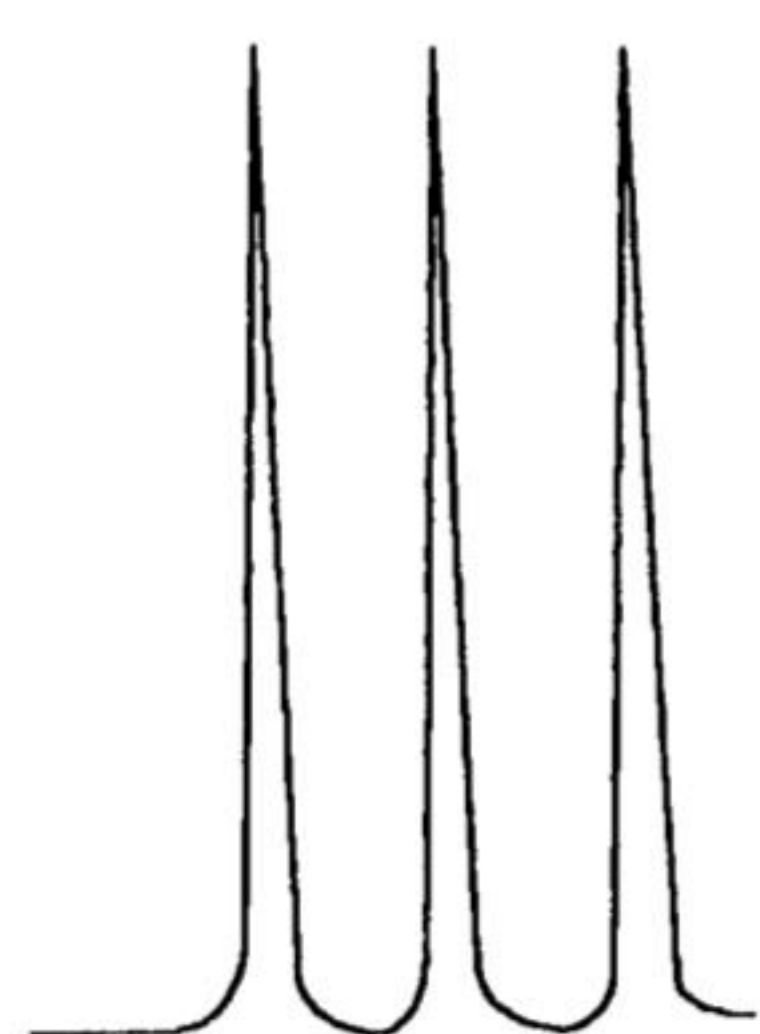
For information on making proper connections check out **The LC Handbook, Pub. No. 5990-7595EN**



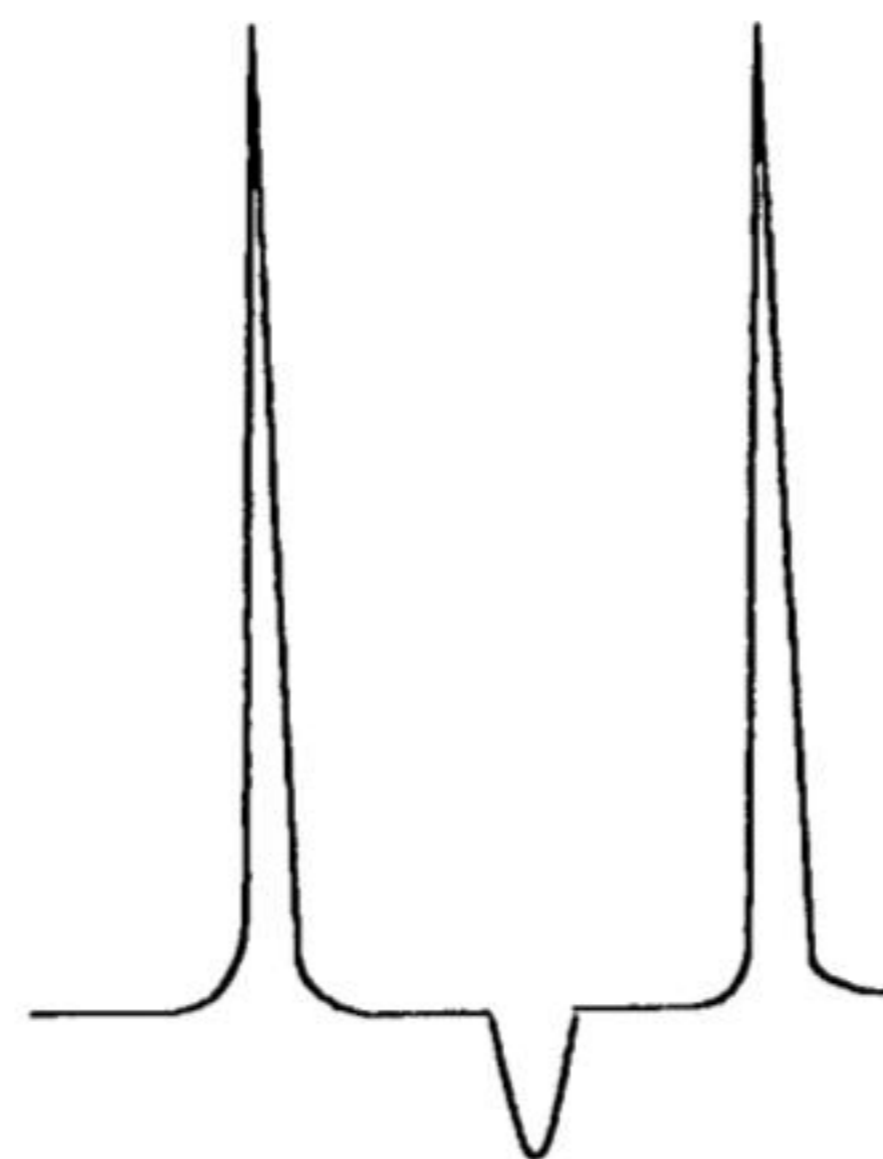
➤ ***New A-Line fittings and capillaries can prevent this***

# Peak Shape

## Negative Peaks



**Normal**



**Negative**

### Causes

- Absorbance of sample is less than the mobile phase
- Equilibrium disturbance when sample solvent passes through the column
- Normal with Refractive Index Detectors
- Indirect UV detection



# Peak Shape Issues

## “Usual Suspects”

- ✓ Partially blocked inlet frit
- ✓ Sample solvent strength
- ✓ Injection volume
- ✓ Sample load
- ✓ Secondary interactions
- ✓ Hardware failure (rotor, stator)
- ✓ Ghost peaks
- ✓ Metal interactions/chelation
- ✓ No or insufficient mobile phase pre-heating
- ✓ Extra column volume
- ✓ Mobile Phase



# Troubleshooting Pressure

## I. Continuously Increasing Pressure With No Injections

- Pump Seals
- Mobile Phase Particulates
- Mobile Phase Solubility
- Mobile Phase Unstable (polymerization)
- Column Void Formation (use condition dependent)

## II. Increasing Pressure With Sample Injections

- Sample Particulates
- Sample Not Soluble in Mobile Phase
- Sample Components Irreversibly Bound to Stationary Phase

# Determining the Cause and Correcting

- Check pressure with/without column - many pressure problems are due to blockages in the system or at the guard or in-line filter

## If Column pressure is high:

- Wash column
  - ✓ Eliminate column contamination and plugged packing
  - ✓ high molecular weight/adsorbed compounds
  - ✓ precipitate from sample or buffer
- Back flush column
  - ✓ Clear plugged frit
- Change frit – Not recommended



# Column Cleaning

Flush with stronger solvents than your mobile phase

## Reversed-Phase Solvent Choices in Order of Increasing Strength

- Mobile phase without buffer salts
- 100% Methanol
- 100% Acetonitrile
- 75% Acetonitrile:25% Isopropanol
- 100% Isopropanol
- 100% Methylene Chloride\*
- 100% Hexane\*

Use at least 10 column volumes of each solvent for analytical columns

\* When using either Hexane or Methylene Chloride the column must be flushed with Isopropanol before returning to your reversed-phase mobile phase.

# Protein/Peptide Samples

## **If precipitate might be present (i.e., aggregated protein, cellular material, polymer)**

- Attempt to clear blockage with appropriate solubilizing solvent
  - 0.1% TFA / 80% acetonitrile, 6M Guanidine HCl, THF, HFIP
  - See Appendix for other suggestions

# Microbial Growth

## ➤ Potential problems

- **Increased system pressure or pressure fluctuations**
- **Increased column pressure, premature column failure**
- Can mimic application problems
- Gradient inaccuracies
- Ghost peaks

## ➤ Prevent and/or Reduce Microbial Growth

- Use freshly prepared mobile phase
- Filter
- Do not leave mobile phase in instrument for days without flow
- Always discard “old” mobile phase
  - Do not add fresh mobile phase to old
- Use an amber solvent bottle for aqueous mobile phase
- If possible, can add
  - 5% organic added to water can be used to reduce bacterial growth
  - Few mg/l sodium azide

*Check your instrument manual for guidelines*

# Column Troubleshooting

## Retention Shifts

All Peaks Shift to Lower Retention (acids, bases, neutrals)

- Loss of bonded phase
- Mobile phase unstable (less likely)
- Solvent delivery system (flow rate or mixing)

All Peaks Shift to Greater Retention

- Loss of organic solvent in aqueous/organic mix
- Column change (less likely)
- Solvent delivery system (flow rate or mixing)

Ionic Peaks Shift Retention

- Loss of volatile MP component (ionic strength, pH shift)
- Column Change (bonded phase or contamination)

# Retention Shifts

## Mobile Phase Related Problems

- **Make fresh, compare to aged**
  - **pH**
  - **conductivity**
  - **chromatographic test**

## Column Related Problems

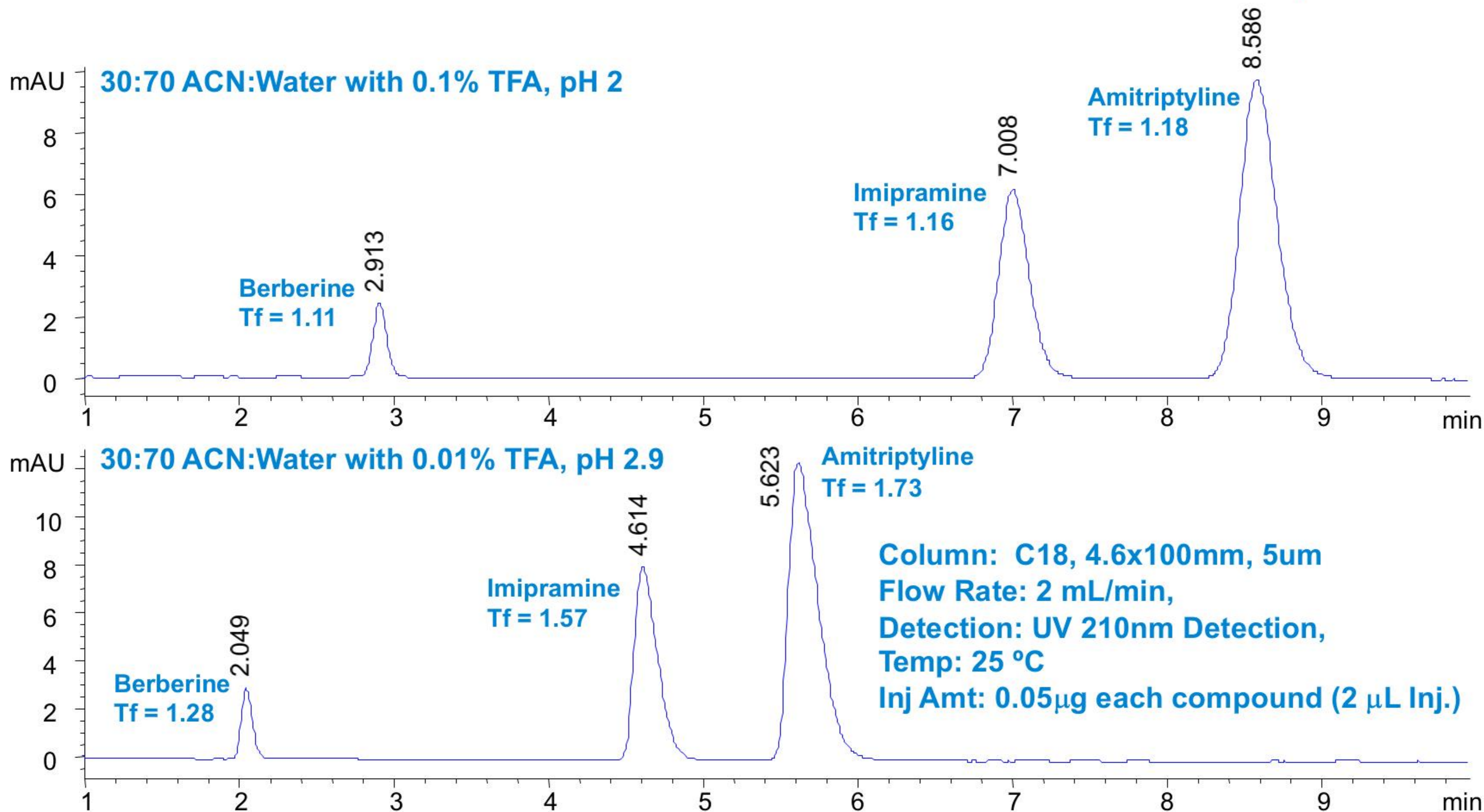
- **Test new column**
- **Test current column with test mixture or e.g., Toluene**
- **"Wash" column and retest**
- **Consider effect of sample matrix**

# Separation Conditions That Can Cause Changes in Retention\*

Condition	Change	Retention	Change
Flow Rate	+/- 1%	$t_R$	+/- 1%
Temperature	+/- 1 deg C	$t_R$	+/- 1 to 2%
% Organic	+/- 1%	$t_R$	+/- 5 to 10%
pH	+/- 0.01%	$t_R$	+/- 0 to 1%

\*excerpted from "Troubleshooting HPLC Systems", J. W. Dolan and L. R. Snyder, p 442.

# Change in Volatile Buffer Concentration Can Cause Shifts in Retention Time and Peak Shape



# Mobile Phase Preparation

➤ Small changes in mobile phase strength can have a large effect on retention

- ✓ HPLC grade or better
- ✓ Buffer prep procedure
  - Be consistent
    - Document process
  - See appendix

Volume % of solvents can depend on preparation

Specified volume ACN added to a 1 L volumetric and made to volume with H<sub>2</sub>O

≠

Specified volume H<sub>2</sub>O added to a 1 L volumetric and made to volume with ACN

≠

500 ml H<sub>2</sub>O added to 500 ml ACN

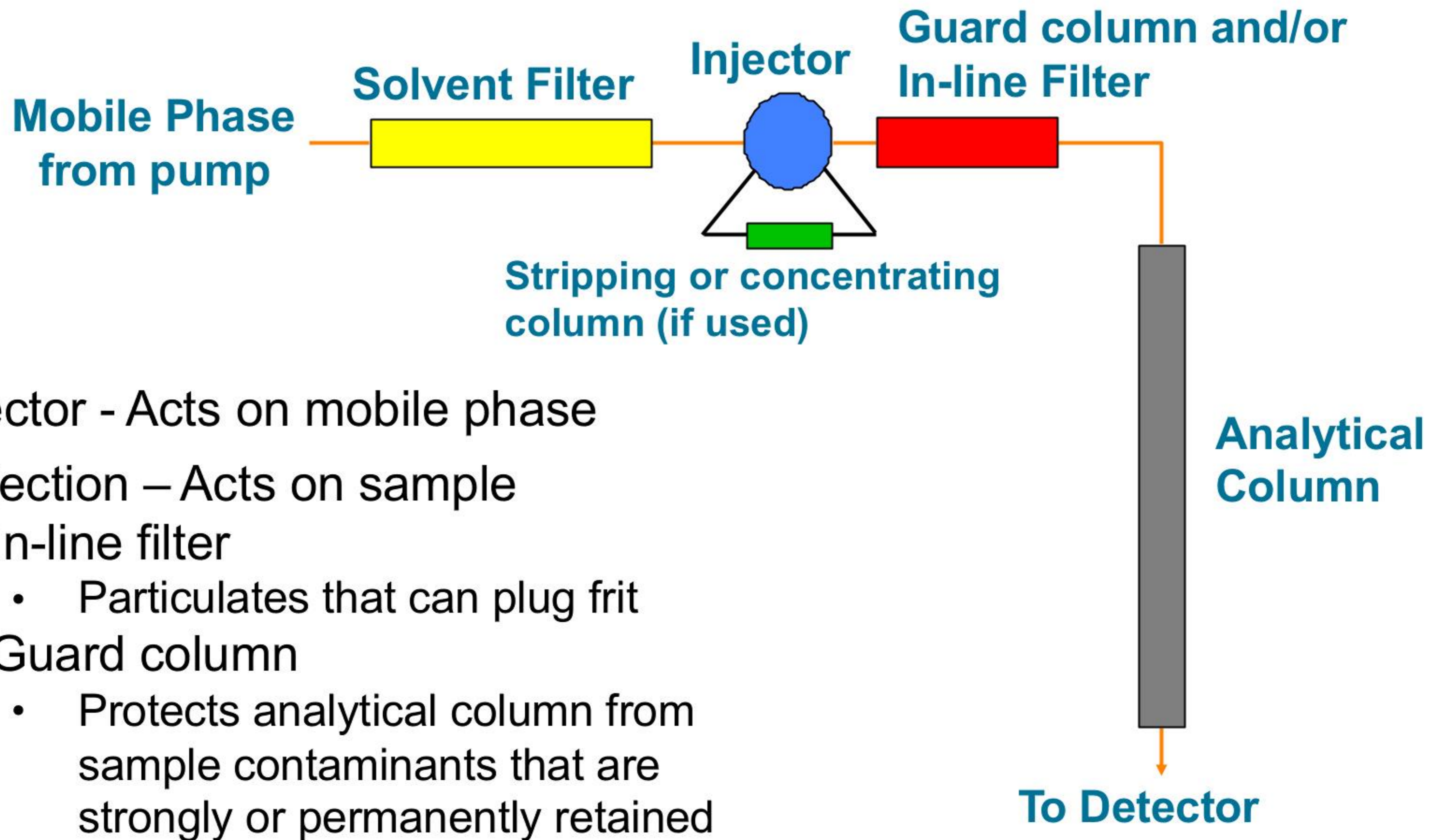
- ✓ Degree of contraction is affected by the relative quantities of each
- ✓ Temperature



# II. Preventing HPLC Column Problems

- ✓ **In-line devices**
  - **Pre-column**
  - **In-line Filter**
  - **Guard Column**
  
- ✓ **Sample Preparation**
  
- ✓ **Operational Conditions**

# In-line Filters & Guard Columns



Pre-injector - Acts on mobile phase

Post Injection – Acts on sample

- In-line filter
  - Particulates that can plug frit
- Guard column
  - Protects analytical column from sample contaminants that are strongly or permanently retained

# II. Preventing HPLC Column Problems

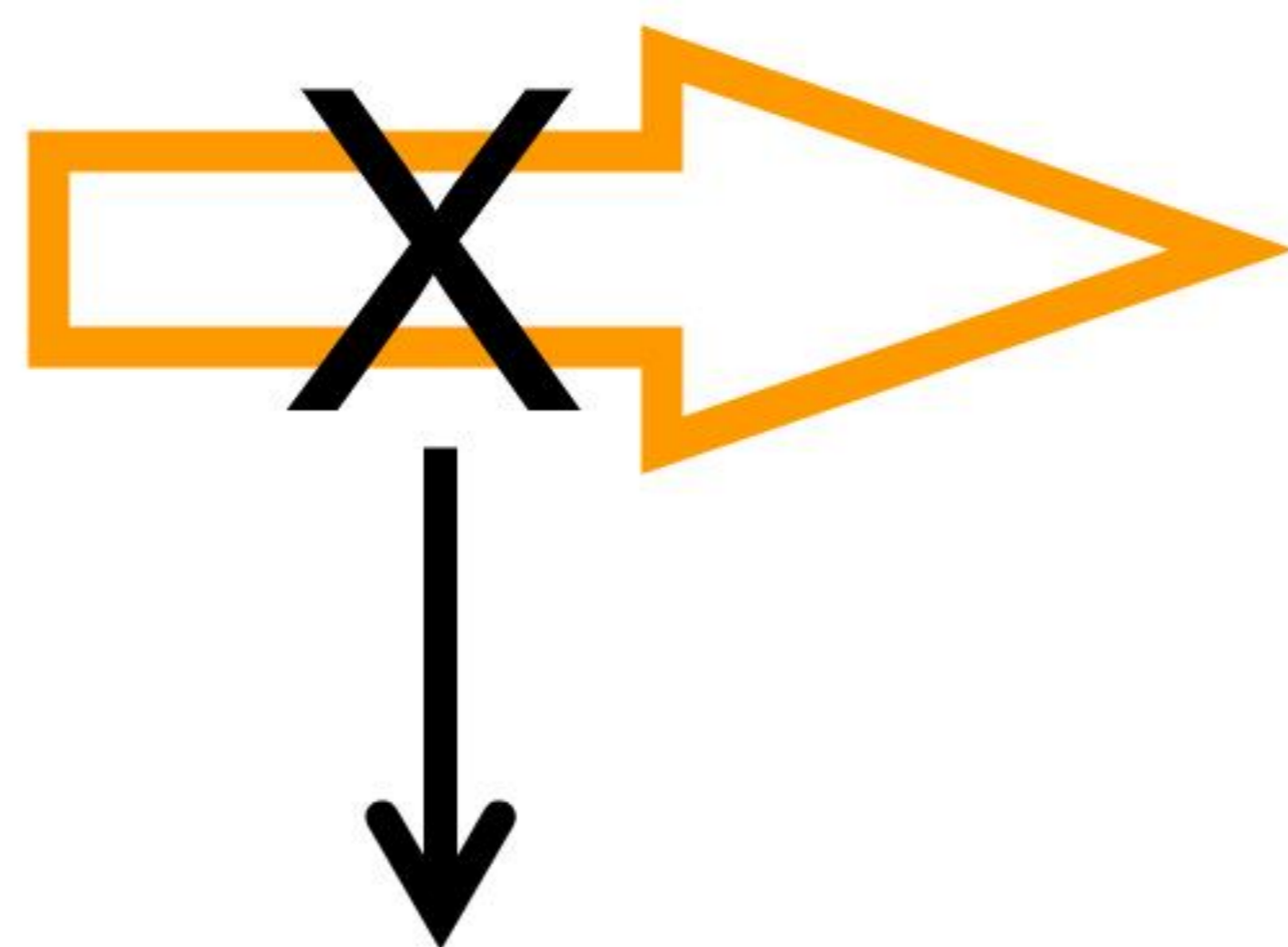
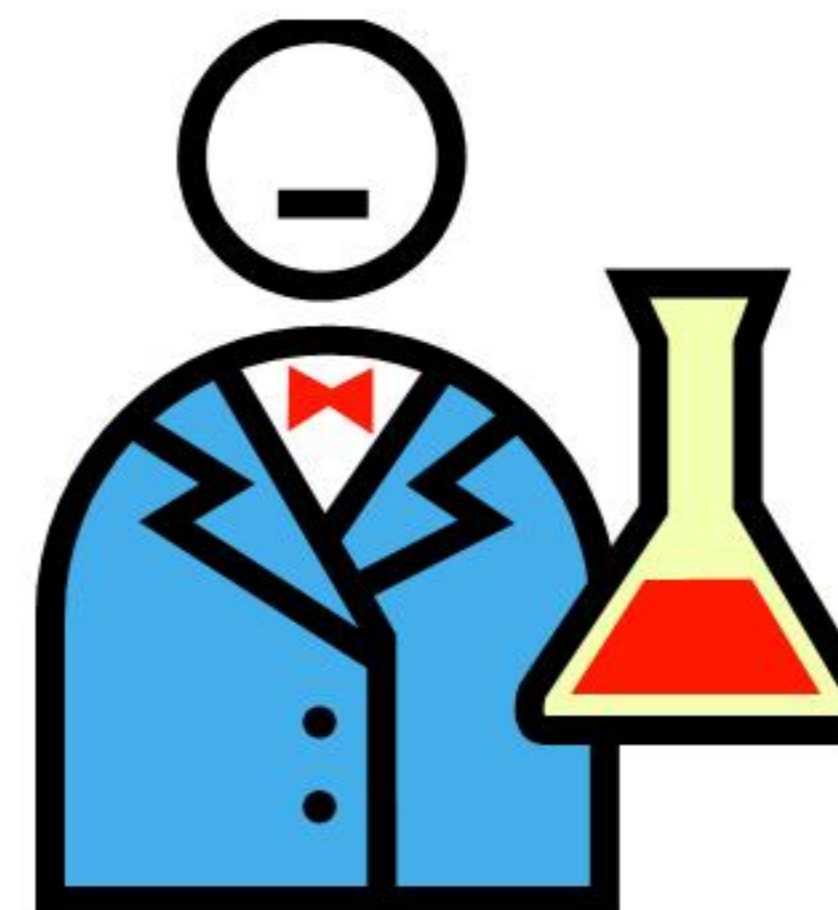
- ✓ In-line devices
  
- ✓ **Sample Preparation**
  - **SPE**
    - **Analyte Adsorption (Bind-Elute)**
    - **Matrix Adsorption (Interference Removal)**
  - **Filtration**
    - **Syringe**
    - **In-line**
  
- ✓ Operational Conditions

# Why is Sample Prep Required?

Collected Sample



Analyze



**Current Sample - Unsuited for further analysis**

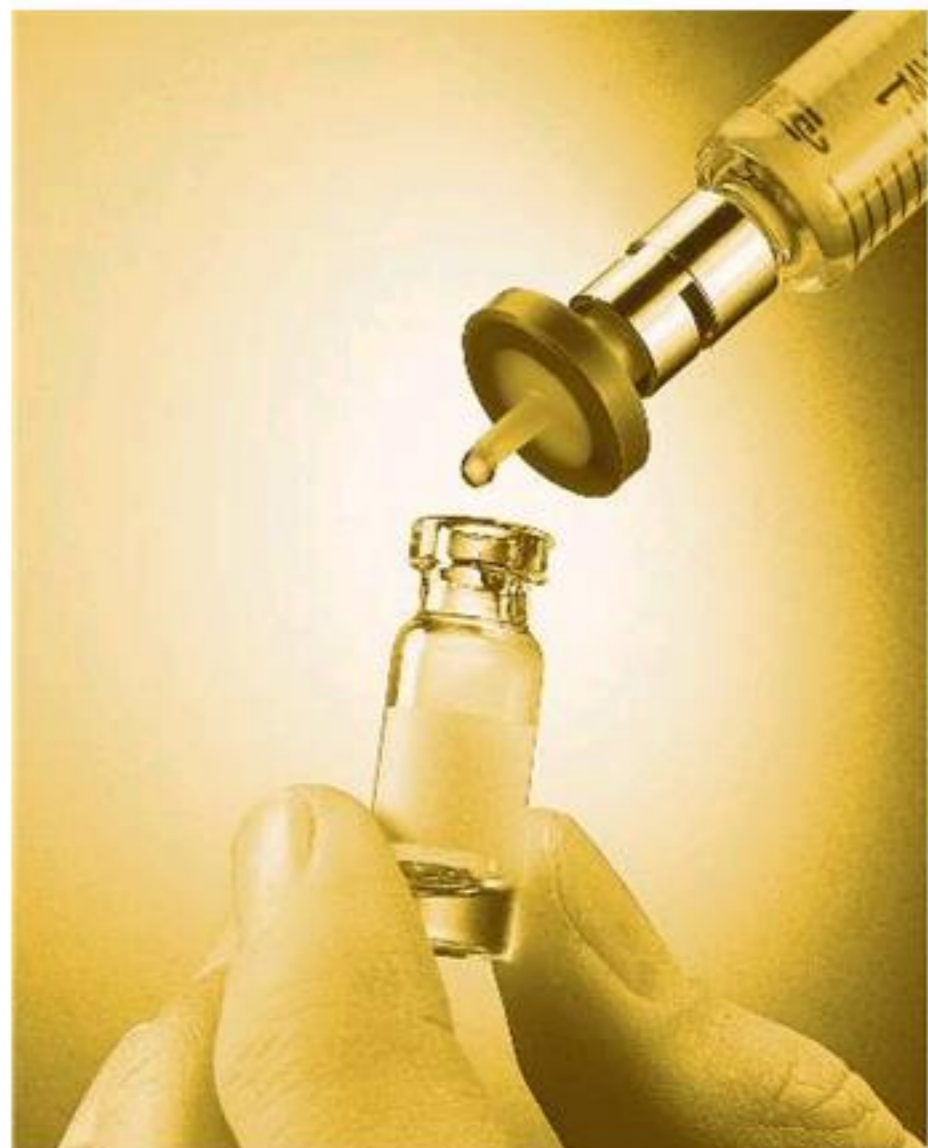
## Why?

- Too dirty
  - Contains other sample matrix components that interfere with the analysis
- Too dilute
  - Analyte(s) not concentrated enough for quantitative detection, present in low level
- Too dangerous
  - Contaminants can be 'column killers'



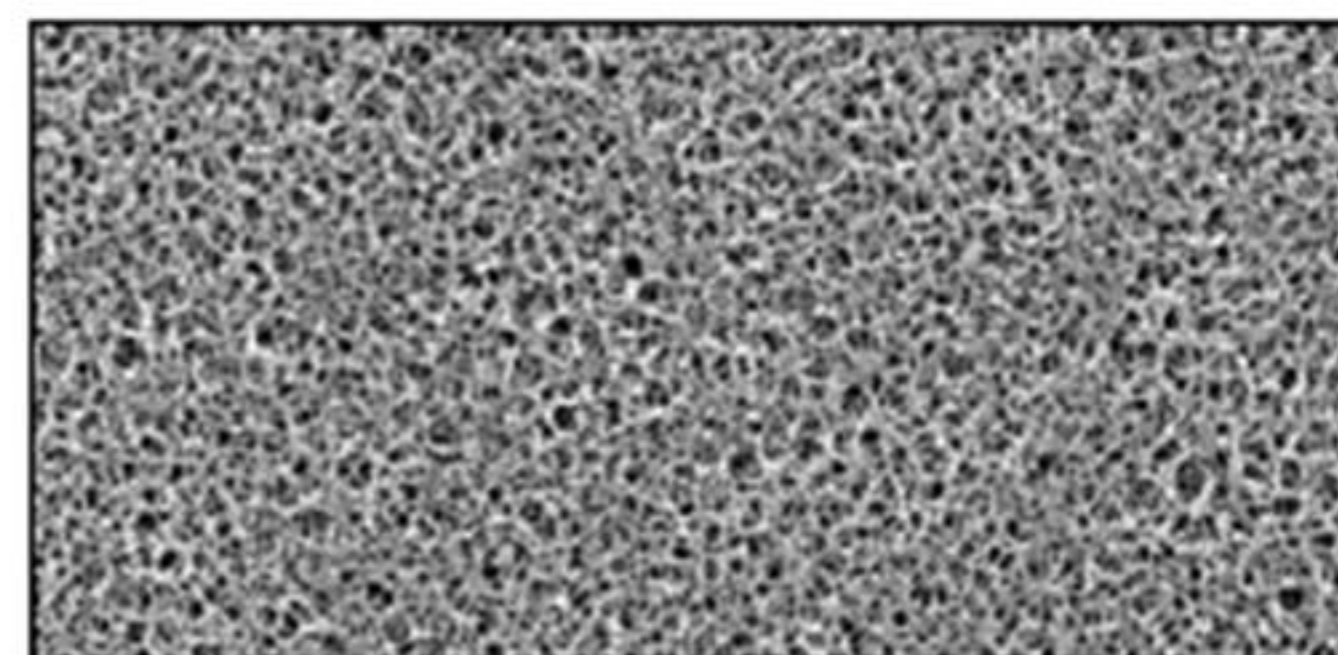
# Inexpensive Filters

## Prevent Column Frit Plugging



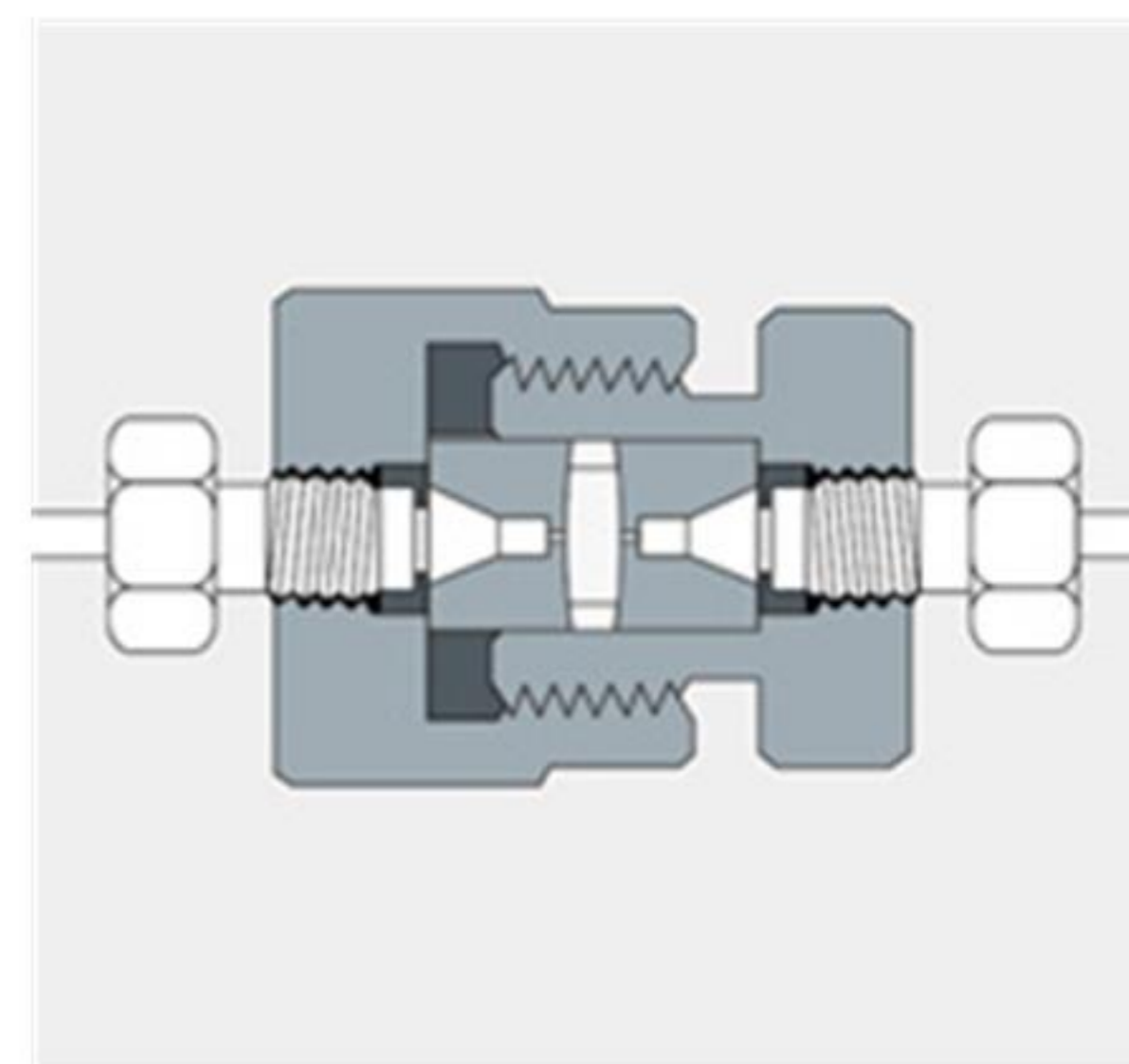
E.g., Regenerated Cellulose (RC)

- **Universal hydrophilic membrane, compatible with most solvents - aqueous and organic**
- **High purity, extremely low extractables and binding**
- **More uniform surface**
- **Choose appropriate pore size**



### In-line filters

- **Easy to Use and replace**
- **Frits available in 0.2, 0.5 and 2.0 $\mu$  porosity**
- **Much Less expensive than a Column**
- **Easier and Faster to Replace than a Column Frit**



# II. Preventing HPLC Column Problems

- ✓ In-line devices
- ✓ Sample Preparation
- ✓ **Operational Conditions**

# Mobile Phase pH and Buffers

## Why Are These So Important in HPLC?

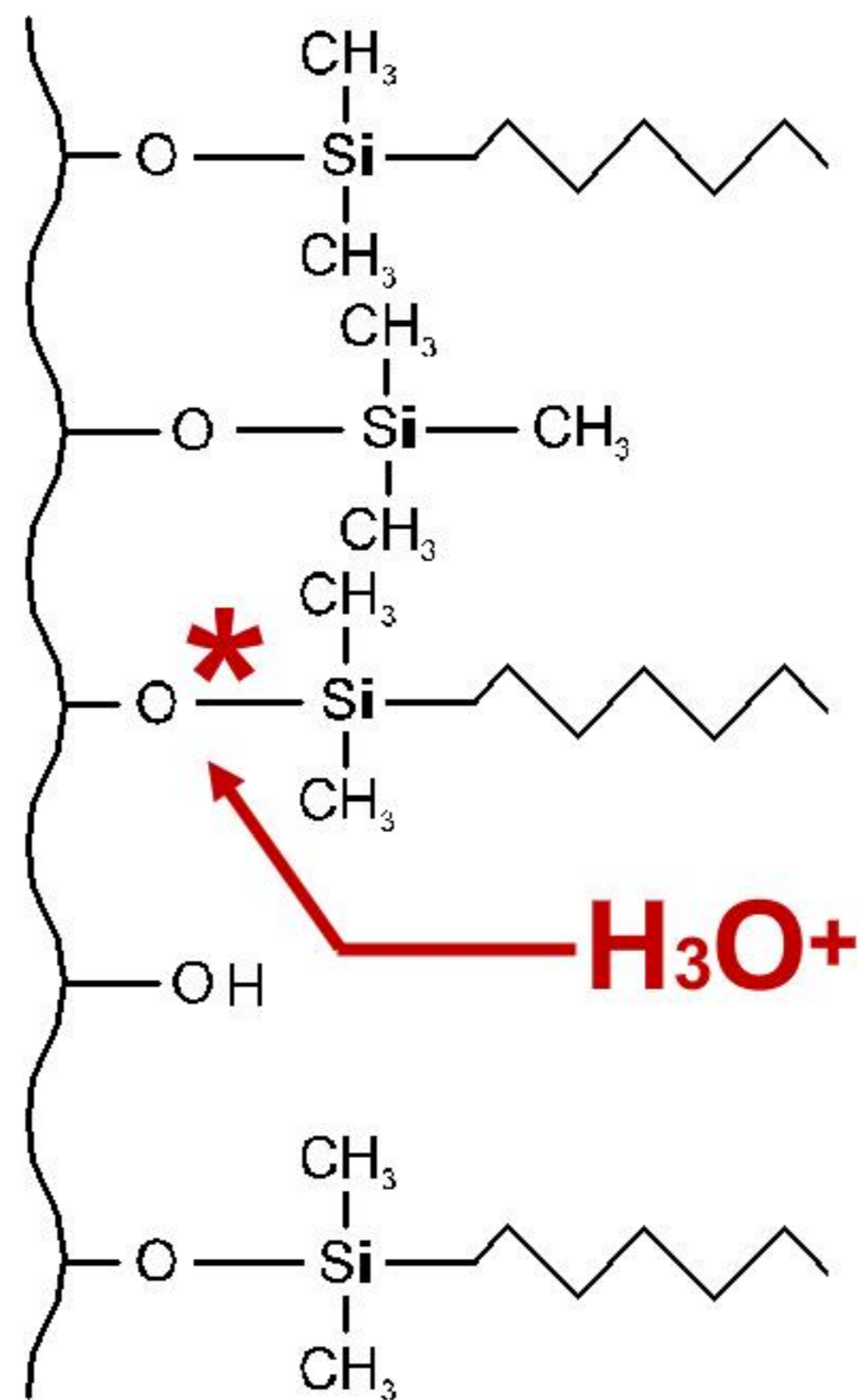
- pH - Effects Ionization
  - Silica surface of column
  - Sample components of interest
- Buffers
  - Resist changes in pH and maintain retention
  - Improve peak shape for ionizable compounds
- Effects Column Life
  - Low pH strips bonded phase
  - High pH dissolves silica

# Operational Conditions

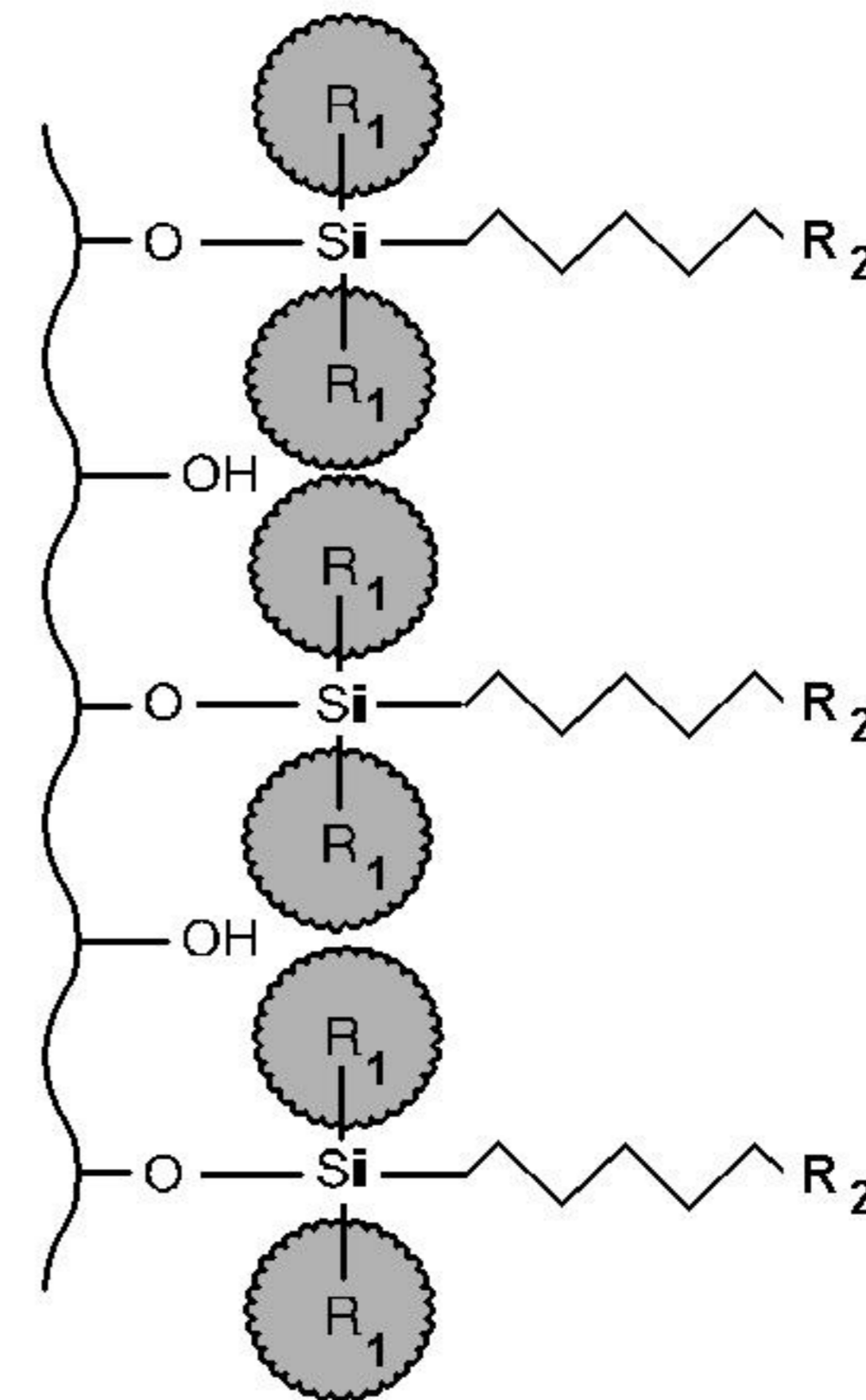
## Mobile Phase Effects on Column Life

Low pH (1-3) - Bonded phase loss by acid catalyzed hydrolysis

### Conventional



### StableBond



\* Hydrolytically sensitive siloxane bond

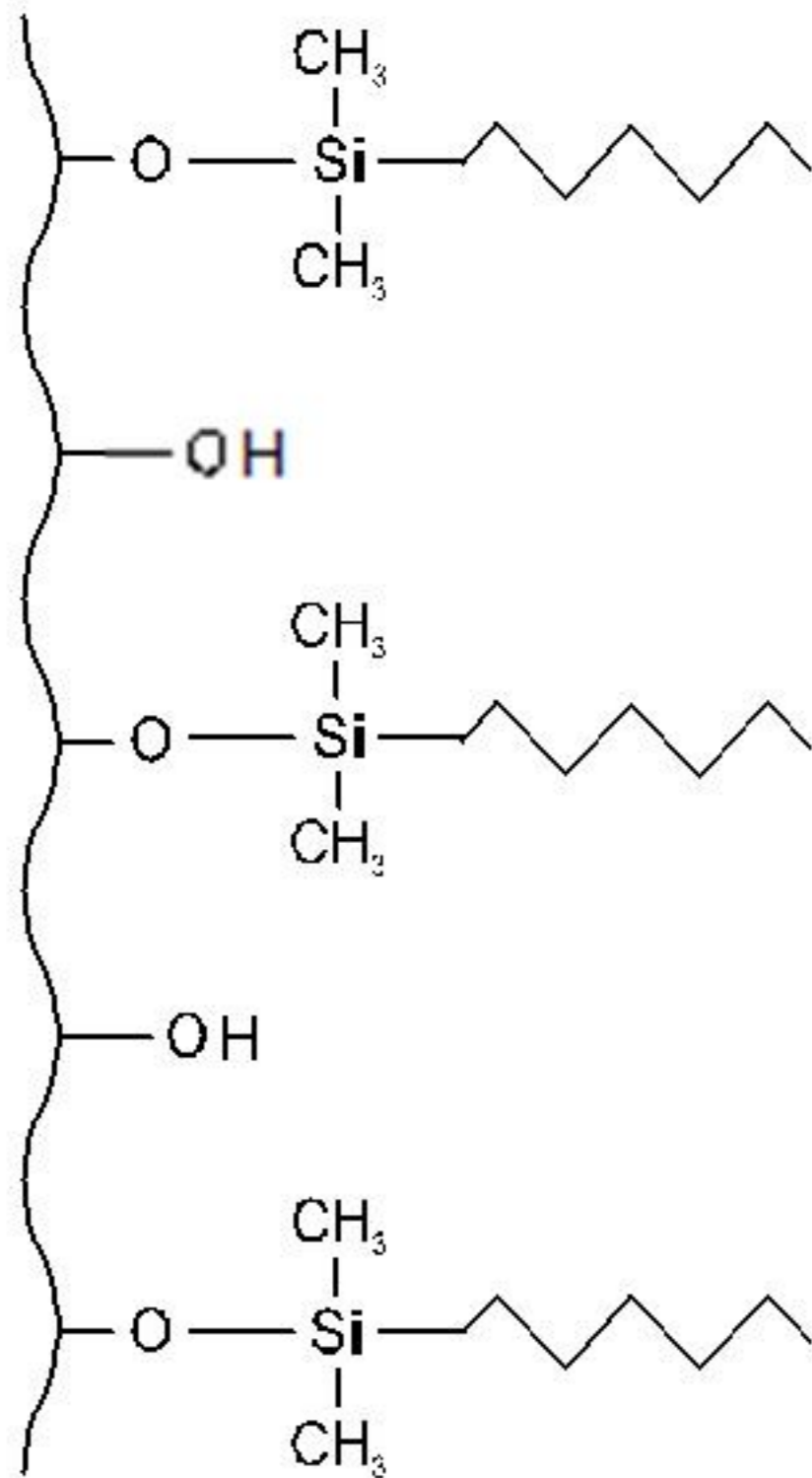


# Operational Conditions

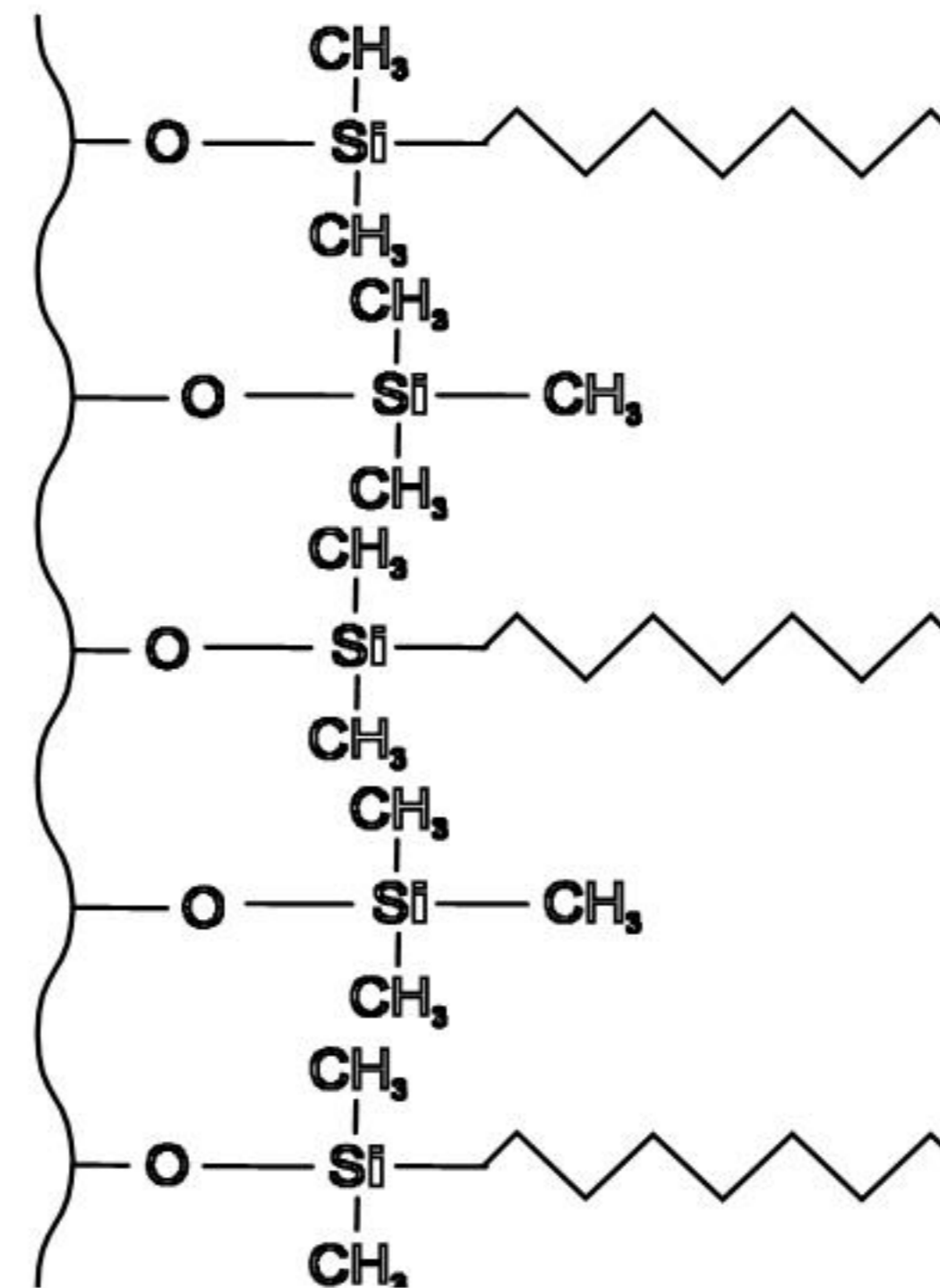
## Mobile Phase Effects on Column Life

### High pH >7- Silica Dissolution

#### Non-endcapped



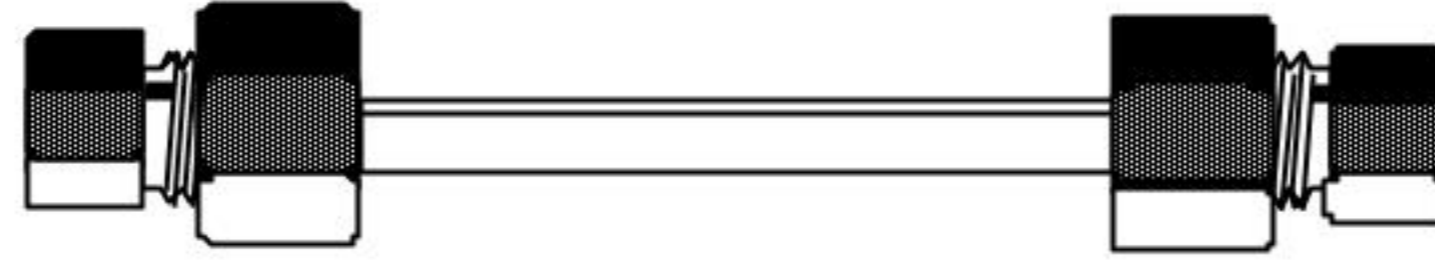
#### Endcapped and/or modified surface Eclipse Plus, Poroshell 120 EC, HpH



#### Dissolution Rate Increases With

- pH increase
- Temperature increase
- Phosphate > Borate > Organic Buffers

# Good Column Practices



- ✓ Filter buffers.
- ✓ Investigate effects of sample solvent on solubility and separation.
- ✓ Pretreat samples which contain strongly retained components of no interest.
- ✓ Awareness of column packing limits
  - pH
  - Temperature
  - Chemical compatibility
- ✓ Use fresh aqueous solutions, consider use of a bio-stat (sodium azide).
- ✓ Flush column periodically with strong solvent
- ✓ To store column, purge buffers and leave in appropriate solvent (ACN)
- ✓ Avoid physically mishandling columns: banging, dropping, or over tightening fittings

# Troubleshooting Summary

- Most HPLC column problems can be prevented provided proper precautions are taken
  - ✓ Mobile phase must be soluble and particulate free
  - ✓ Use guard columns and in-line filters; change as needed
  - ✓ Use appropriate sample preparation procedures
  - ✓ Use appropriate column cleaning procedure
  - ✓ Use appropriate columns for operating conditions (e.g., SB: pH 1-6, Poroshell 120 HpH: pH 3-11)
  - ✓ Keep record of column backpressure & important chromatographic parameters (e.g. R, N, k')
  - ✓ Store column in organic solvent (acetonitrile) or other recommended by manufacturer
- Problems are not always associated with the column and may be caused by instrument or experimental condition issues



# Agilent Technical Support

800-227-9770 (US & Canada)

Options 3, 3, 2

Email: [lc-column-support@agilent.com](mailto:lc-column-support@agilent.com)

[www.agilent.com/chem](http://www.agilent.com/chem)



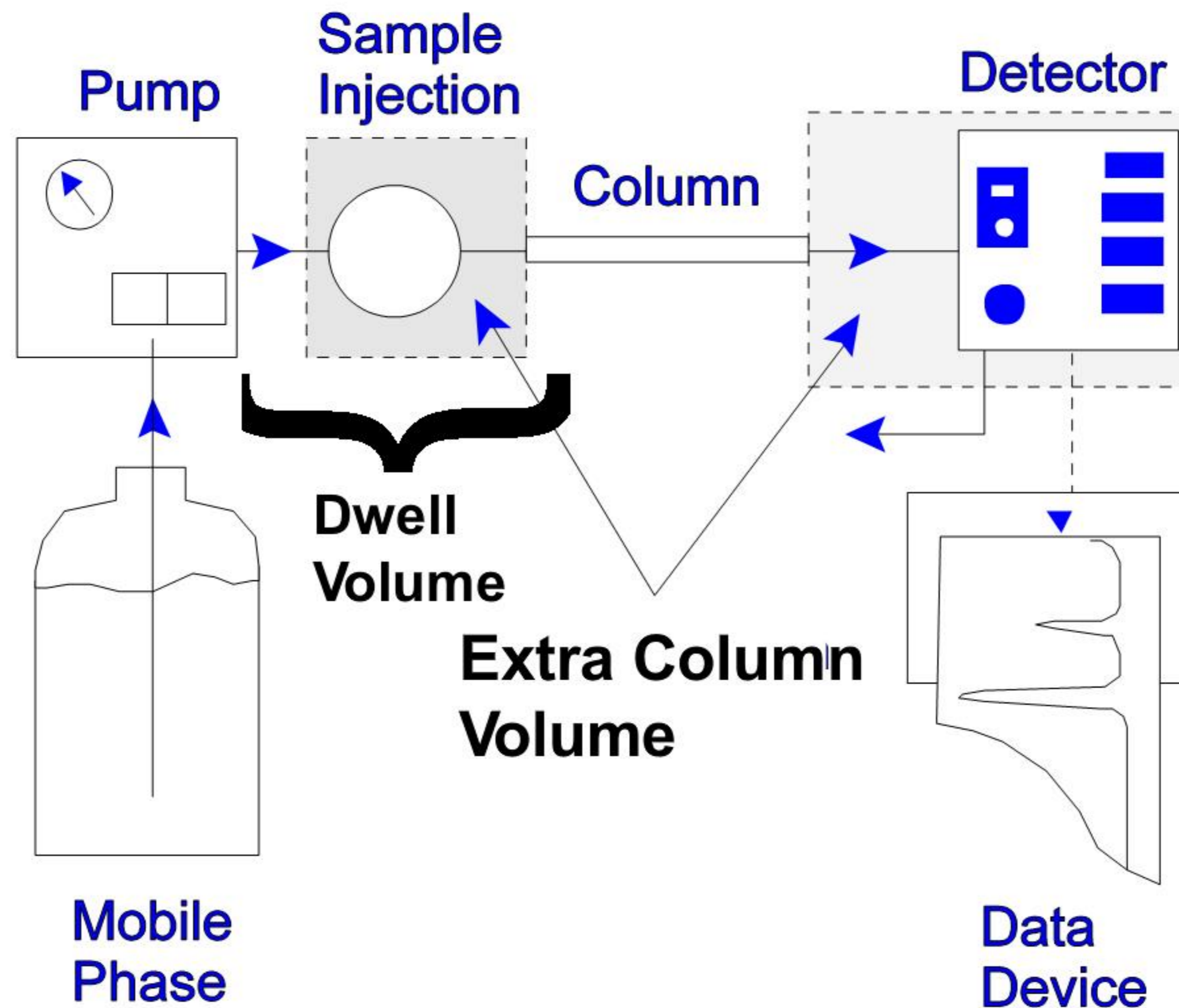
**THANK YOU FOR ATTENDING**

# Appendix

# Typical Column Problems

1. High back pressure
  - a) Plugged frits
2. Reproducibility
3. Peak shape
4. Sample recovery
5. Loss of resolution
6. Lifetime
7. Leaks

# Extra Column Volume and Dwell Volume



**ECV = injection volume + connecting tube volume + fitting volume + detector cell volume**

**Dwell Volume = volume from formation of gradient to top of column**

# Solubilization Solvents for Proteins/Peptides

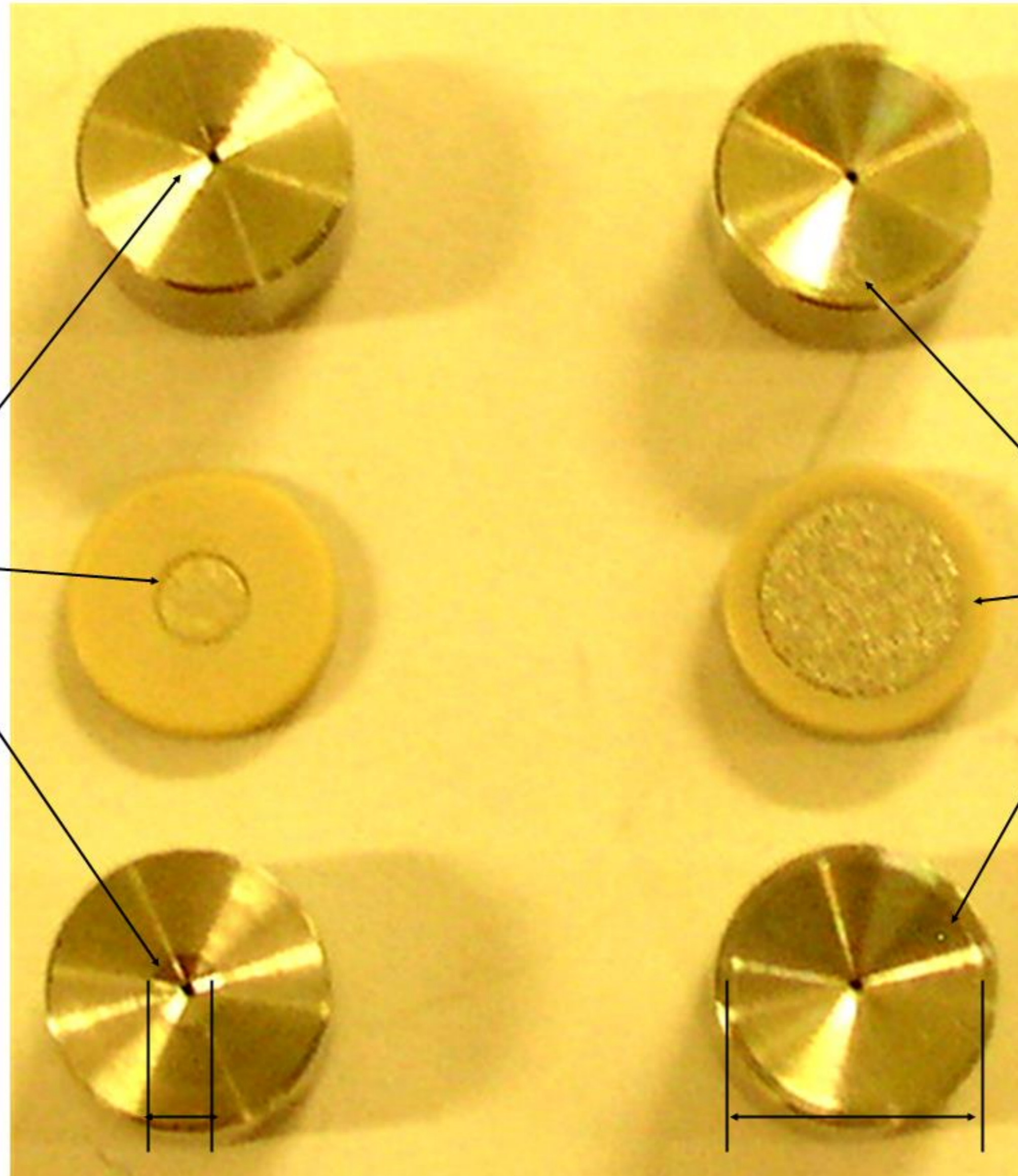
## In Order of Weakest to Strongest

- Water / phosphate buffer
- Dilute Acid (TFA, HOAc or HCl)
- Neutral pH 6-8 M Guanidine-HCl or isothiocyanate
- 5 % HOAc / 6 M Urea
- Dilute Acid + aqueous / organic solvents (ACN, MeOH, THF)
- Dilute Base (Ammonium Hydroxide)
- Neat Organic Solvents- ACN, MeOH, THF
- 99 % Formic Acid
- HFIP or HFIP / aqueous mixtures
- 100 % TFA
- DMSO or 0.1 - 1 % TFA in DMSO
- Formamide



# High Pressure In-line Filter Kit

For **2.1mm Frits** use inserts with **small cone**



For **4.6mm Frits** use inserts with **big cone**

## *Assembling the High Pressure Filter Kit*

Put the first insert into the frit housing.



Place the frit on top of this insert



Then place the second insert on top of the frit

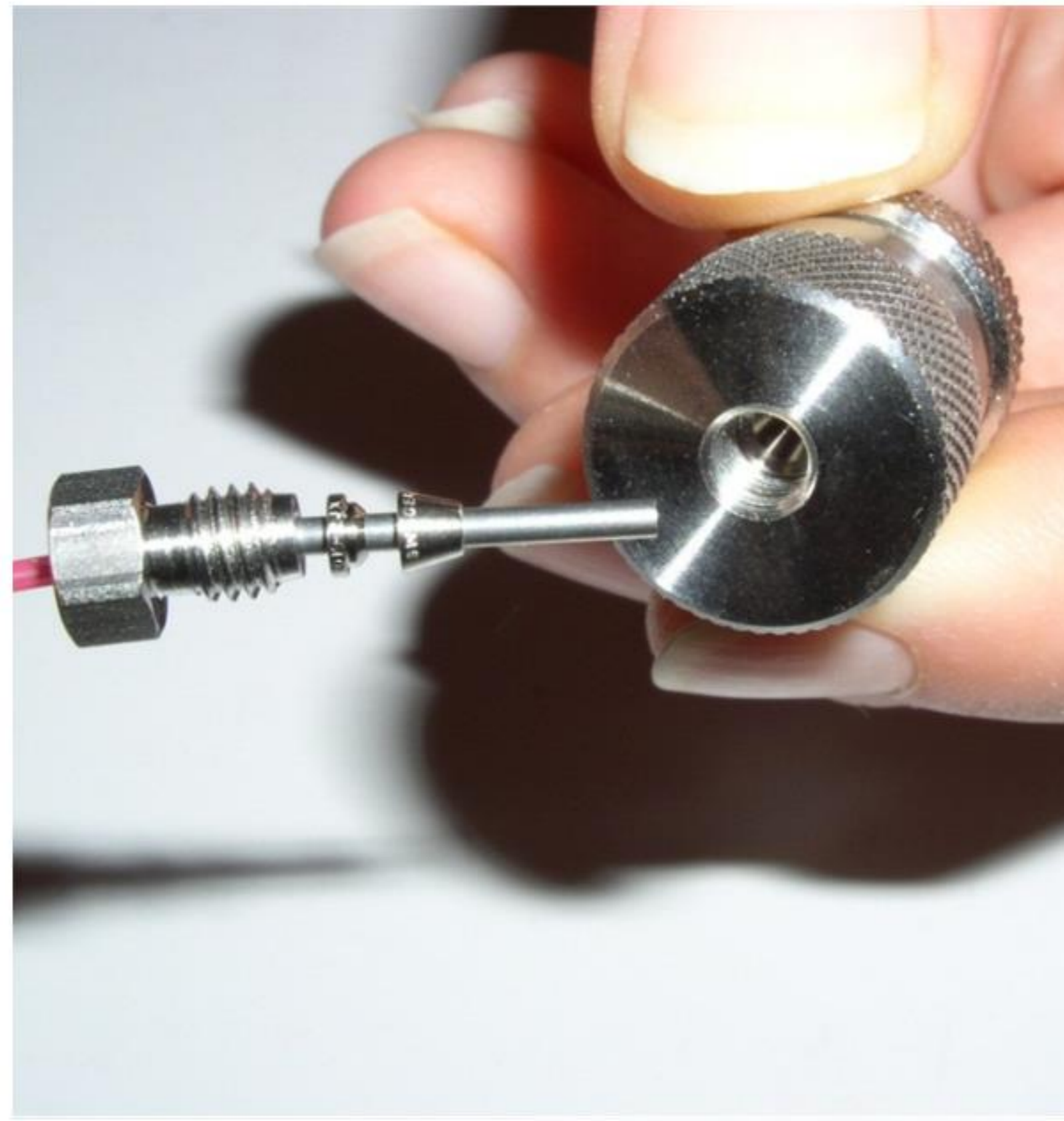


# Assembling the High Pressure Filter Kit

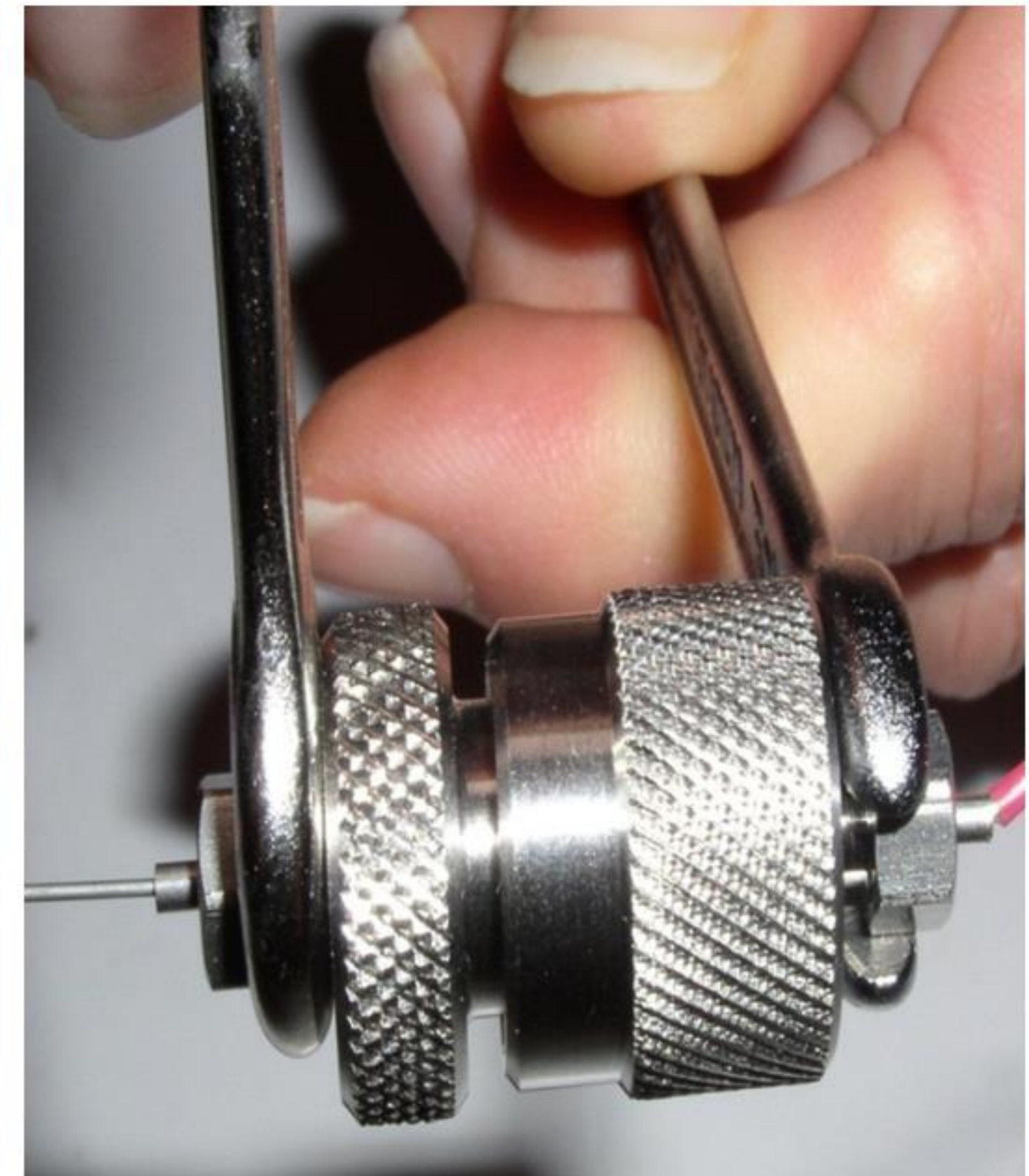
Close the frit housing,  
screw finger tight



Slide the fitting, back and front  
ferrule onto the capillary. Insert  
the capillary into the frit housing  
bore, then tighten the fitting with  
your fingers.  
Push the capillaries all the way in  
into the bore.  
Connect the second capillary.



Ensure that both capillaries are  
still pushed all the way in into the  
bore. Then tighten both fittings at  
the same time with two 1/4"  
spanners. This compresses the frit  
assembly and assures a leak tight  
connection.



# SPE Modes

## Analyte Adsorption (Bind-Elute )

Analyte(s) retained ( $K_D \gg 1$ )

Matrix unretained ( $K_D \sim 0$ )  
and/or strongly retained ( $K_D \gg 1$ )

Pre-concentration factor

Cleaner extracts

Load at 1-3 drops/sec (recovery  $\propto 1/\text{flow}$ )

Capacity issues may be more important

## Matrix Adsorption (Interference Removal)

Analyte(s) unretained ( $K_D \sim 0$ )

Matrix retained ( $K_D \gg 1$ )

No pre-concentration advantage

Eluates may not be as clean

Sample loading may gravity fed

Used less often than analyte adsorption

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# The Agilent LC Column and Sample Prep Navigator can help you find the optimal products for your separation. Find your LC solution!

The screenshot shows the Agilent LC Column and Sample Prep Navigator website. At the top, there are navigation links: 'BACK', 'GET YOUR COLUMN SELECTION GUIDE', and 'GET HELP'. The main content area is titled 'WHAT CAN WE HELP YOU WITH?' and includes the instruction 'Choose from these options to narrow your search'. There are four orange buttons with the following text: 'Option 1: I already have a column or sample prep method.', 'Option 2: Column recommendation based on method parameters.', 'Option 3: I would like to see compound-specific references.', and 'Option 4: USP method.'. To the right of these buttons is a vertical banner for the 'Agilent LC Column Selection Guide' with the text 'SOLUTIONS FOR YOUR SAMPLE SEPARATIONS' and 'Get all the details! Request your Agilent Column Selection Guide'. The footer of the page includes the Agilent logo and the slogan 'The Measure of Confidence', along with the text 'Agilent Technologies' and 'Contact us now | Privacy Statement | Terms of Use | United States only | © Agilent, 2010-2013'.

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This tool enables users to find an Agilent column and sample prep product recommendation in four easy ways:

1. By entering a competitive column or sample prep method (for a comparable Agilent cross-reference)
2. By entering some information about their analysis (i.e., instrument type, sample type, pKa of sample, hydrophobicity of sample, etc.)
3. By compound look-up
4. By USP code.