



Pyrolysis of Rubber with Antioxidant 6-PPD

Application Note Rubber & Adhesive

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Rubber products are complex polymeric materials which generally include copolymers, fillers, antioxidants, plasticisers and other additives. Automobile tire rubbers are good examples, since they usually contain significant amounts of carbon black, hydrocarbon oil and antioxidants in a matrix that may include styrene, butadiene and isoprene. Nevertheless, a single pyrolysis run can provide significant information about such a complex material.

Figure 1 shows the pyrolysis at 700°C of a sample of tire rubber. The matrix is an isoprene/butadiene copolymer, and early in the pyrogram the monomers and dimers from the copolymer may be seen. Later in the chromatogram is a peak for the antioxidant 6-PPD [N-(1,3-dimethyl butyl)-N'-phenyl-p-phenylenediamine]. 6-PPD has a molecular weight of 268, and produces the mass spectrum seen in Figure 2.

Other antioxidants, especially hindered phenolics, are also frequently seen in the pyrolysis of rubber samples. Some of these, like BHT, are present as the parent molecule. Other antioxidant molecules are too large to permit GC analysis intact, and are seen as fragments, just as for the polymer itself.

At the end of the pyrogram is a region of unresolved peaks (marked 6 in Figure 1) which results from the oil added to the rubber in making the tire. Consequently, from just a single pyrolysis run, the monomer composition of the rubber, the antioxidant used and the oil added may be determined.

Instrument Conditions

Pyroprobe

Filament: 700°C, 10 seconds
Interface: 300°C
Valve Oven: 300°C

GC/MS

Column: 5% phenyl (30m x 0.25mm x 0.25µm)
Carrier: Helium, 50:1 split
Inlet: 300°C
Oven: 40°C for 2 minutes
10°C/min to 300°C

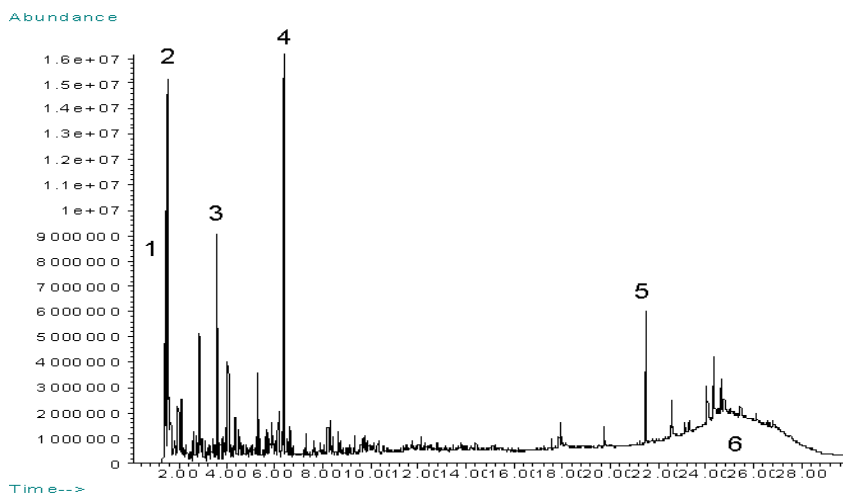


Figure 1. Pyrolysis of tire rubber at 700°C. Peak 1=butadiene, 2=isoprene, 3=butadiene dimer, 4=isoprene dimer, 5= 6-PPD, 6=Hydrocarbon oil.

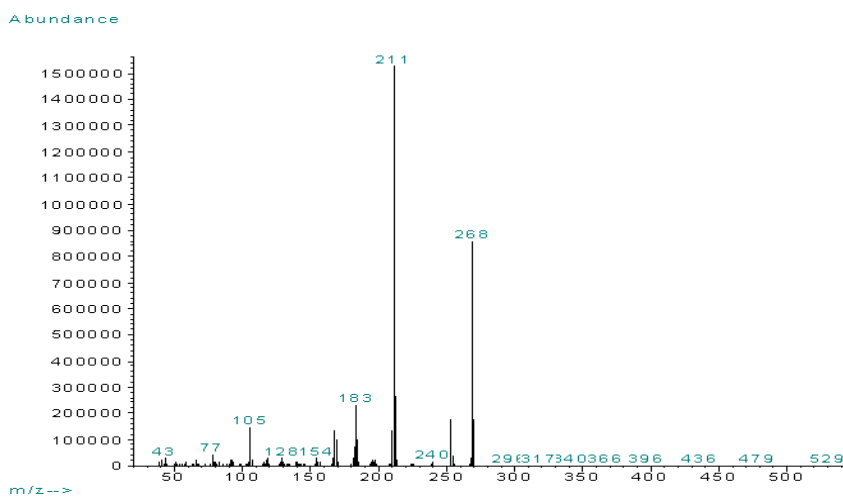


Figure 2. Mass spectrum of peak at 21.5 min.

FOR MORE INFORMATION CONCERNING THIS APPLICATION,
WE RECOMMEND THE FOLLOWING READING:

Temperature as a Sample Preparation Tool in the Analysis of Materials by GC-MS, T. P. Wampler, LC.GC, 17 (1999) S14.