

The GC Column

How to Choose the Correct Type and Dimension

Simon Jones
Application Engineer

Things to Consider...

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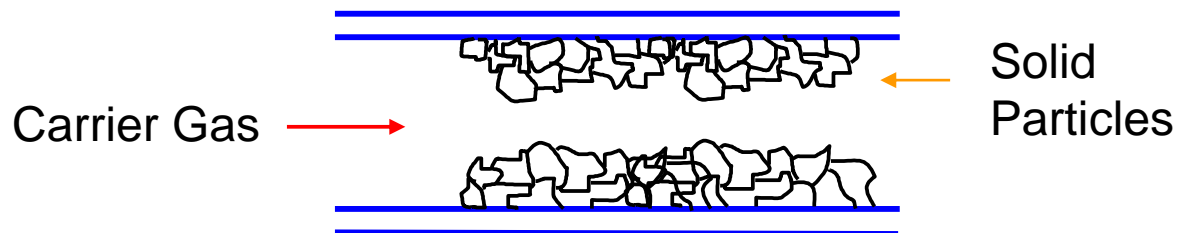
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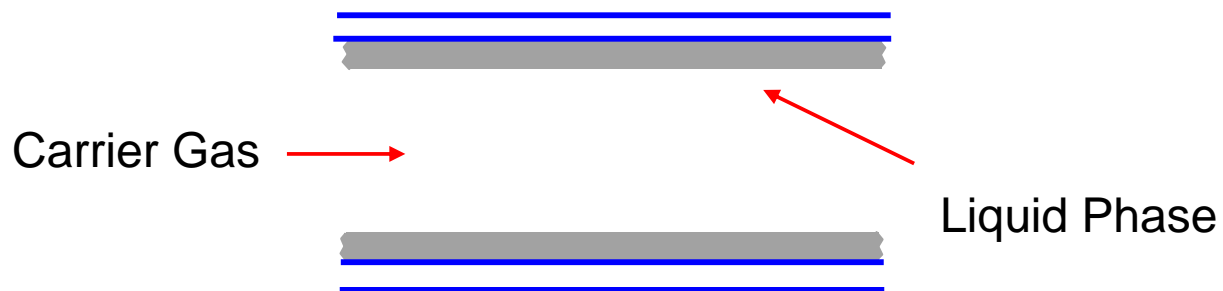
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CAPILLARY COLUMN TYPES

Porous Layer Open Tube (PLOT)

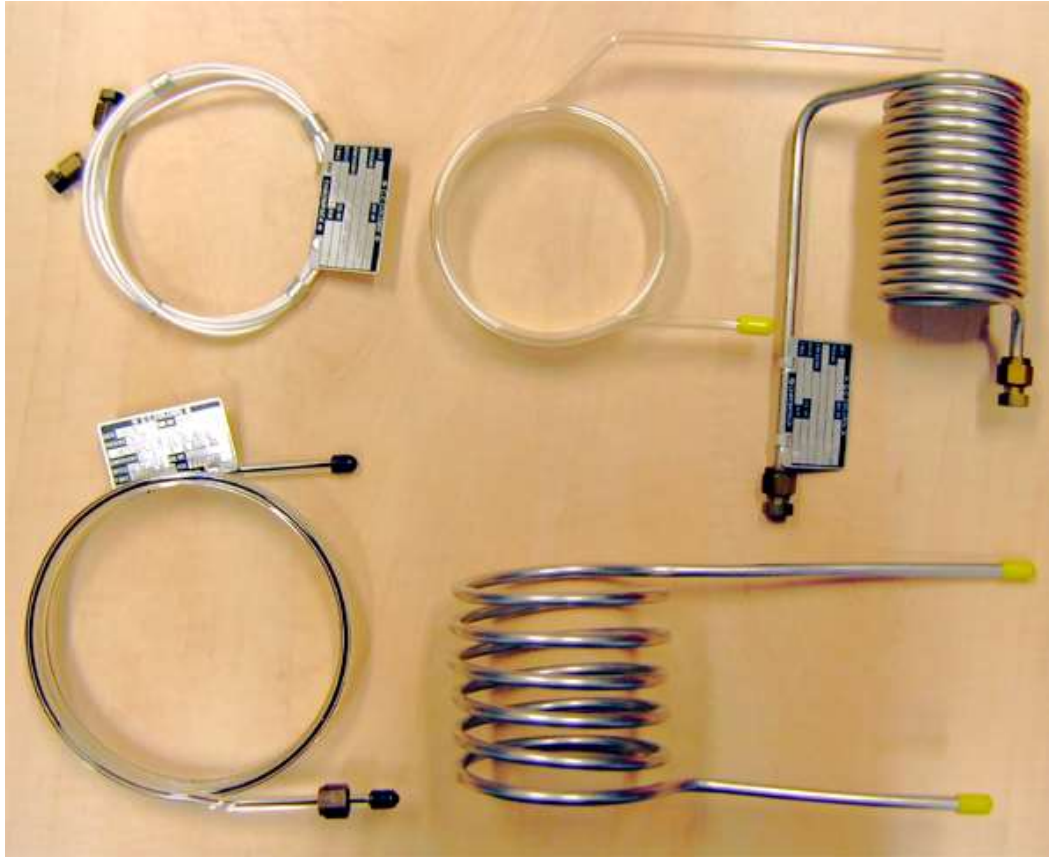


Wall Coated Open Tube (WCOT)



Packed Columns

1950 Introduction with the first gas chromatographs



Packed Column Designs and Materials

Packed Column Anatomy

Packed Columns

- 1 – 12 m length
- Internal Diameter 0.5 – 4mm
- Tubing
 - Stainless Steel, UltimetTM SS, Glass, Nickel, PTFE

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- 1 – 12 m length
- Internal Diameter 0.5 – 4mm
- Tubing
 - Stainless Steel, UltimetTM SS, Glass, Nickel, PTFE
- Packing
 - Coated packing
 - Inert, solid support (diatomaceous earth) coated with liquid stationary phase (e.g. OV-1, SE-30, Carbowax 20M, FFAP)
 - Porous packing
 - Porous polymers (PoraPak Q, N, HayeSep Q, R, S, etc.)
 - Porous carbons (Carboxens, Carbosieves, Carbotraps)

Packed Column Anatomy

Packed Columns

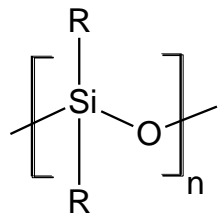
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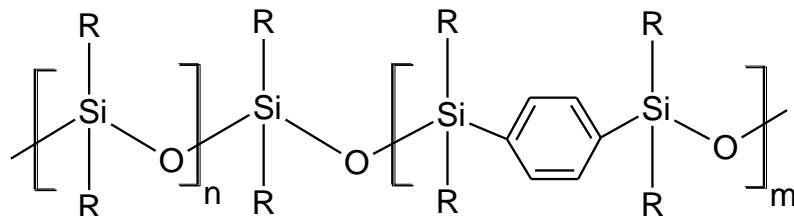
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STATIONARY PHASE POLYMERS

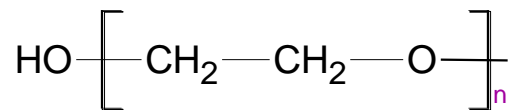


Siloxane

R=methyl, phenyl, cyanopropyl, trifluoropropyl



Siarylene backbone



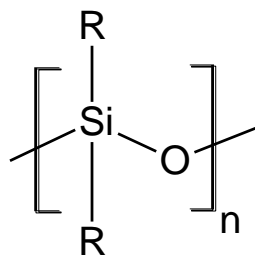
Polyethylene Glycol

Stationary Phase

% Substitution -- polysiloxanes

% = # of sites on silicon atoms occupied

Balance is methyl



Siloxane

R=methyl, phenyl, cyanopropyl, trifluoropropyl

Stationary Phase

Poly(ethylene) Glycol



100% PEG (DB-WAX)

Less stable than polysiloxanes

Unique separation characteristics

Poly(Ethylene) Glycol Modified

- Base deactivated (CAM)
- Acid Modified (DB-FFAP)
- Extended Temperature Range

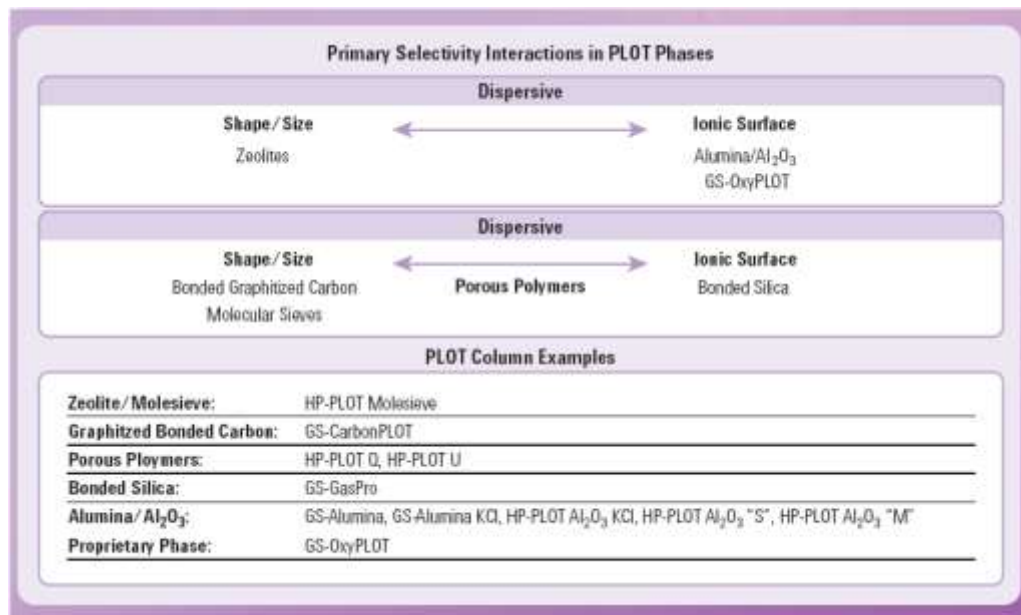
WCOT Column Types

Agilent J&W has over 50 different stationary phase offerings

Low Polarity			Mid Polarity			High Polarity		
CP-Sil 2	DB & HP-1ms UI	DB & HP-5ms UI	DB-XLB	DB-225ms	DB-ALC1	HP-88	DB-WAX	CP-TCEP
DB-MTBE	DB & HP-1ms	DB & HP-5ms	VF-Xms	DB-225	DB-Dioxin	CP-Sil 88	DB-WAXetr	
CP-Select CB MTBE	VF-1 ms	VF-5ms	DB-35ms UI	CP-Sil 43 CB	DB-200	DB-23	HP-INNOWax	
	DB & HP-1	DB & HP-5	DB & VF-35ms	VF-1701 ms	VF-200ms	VF-23 ms	VF-WAXms	
	CP-Sil 5 CB	CP-Sil 8 CB	DB & HP-35	DB-1701	DB-210		CP-Wax 57 CB	
	Ultra 1	Ultra 2	DB & VF-17ms	CP-Sil 19 CB	DX-4		DB & HP-FFAP	
	DB-1ht	VF-DA	DB-17	HP-Blood Alcohol			DB-WAX FF	
	DB-2887	DB-5.625	HP-50+	DB-ALC2			CP-FFAP CB	
	DB-Petro/ PONA	DB & VF-5ht	DB-17ht	DX-1			CP-WAX 58 FFAP CB	
	CP-Sil PONA CB	CP-Sil PAH CB	DB-608				CP-WAX 52 CB	
	DB-HT SimDis	Select Biodiesel	DB-TPH				CP-WAX 51	
	CP-SimDis	SE-54	DB-502.2				CP-Carbowax 400	
	CP-Volamine		HP-VOC				Carbowax 20M	
	Select Mineral Oil		DB-VRX				HP-20M	
	HP-101		DB-624				CAM	
	SE-30		VF-624ms					
			CP-Select 624 CB					
			DB-1301					
			VF-1301ms					
			CP-Sil 13 CB					

PLOT Column Types

PLOT columns are primarily, but not exclusively, used for the analysis of gases and low boiling point solutes (i.e., boiling point of solute is at or below room temperature).

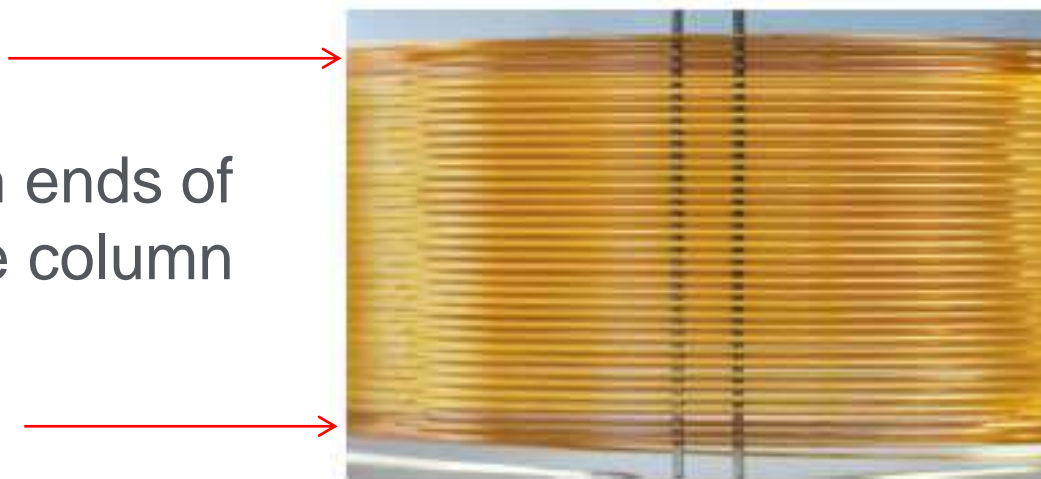


- Agilent J&W PLOT columns begin with the designation of
 - GS (Gas Solid) or
 - HP-PLOT followed by a specific name
 - CP (ChromPack) followed by name
 - **10 stationary phases**
 - GS-OxyPLOT / CP-Lowox
 - GS-Alumina
 - HP-PLOT Al₂O₃ "M"
 - HP-PLOT Al₂O₃ "S"
 - HP-PLOT Al₂O₃ "KCl" / CP-AL₂O₃/KCl
 - HP-PLOT MoleSieve / CP-Molsieve 5A
 - GS-CarbonPLOT / CP-CarboBOND
 - HP-PLOT Q / CP PoraBOND Q
 - HP-PLOT U / CP-PoraBOND U
 - GS-GasPro / CP-SilicaPLOT

- GS-OxyPLOT: oxygenates
- HP-PLOT Molsieve: O₂, N₂, CO, Methane
- HP-PLOT Alumina and GS-Alumina: complex hydrocarbon gas matrices, ethylene and propylene purity, 1,4-butadiene
- HP-PLOT Q: freons, sulfides
- HP-PLOT U: C₁ to C₇ hydrocarbons, CO₂, Polar Hydrocarbons
- GS-GasPro: freons, sulfurs, inorganic gases
- GS-CarbonPLOT: inorganic and organic gases

Integrated Particle Trap PLOT Columns

Particle trap is on both ends of the column



On the front end to help facilitate backflushing without blowing particles back into the inlet / valve

Specialty Phases

Columns developed for particular applications

Examples: DB-UI 8270D, DB-624UI
<467>, DB-VRX, DB-MTBE, DB-TPH,
DB-ALC1, DB-ALC2, DB-HTSimDis, DB-
Dioxin, Select Low Sulfur, CP-Volamine,
Select PAH, DB-EUPAH, DB-CLP1 & 2,
DB-Select 624 UI 467, CP-LowOx, Select
Permanent Gases.....

Ultra Inert Phases

DB-1msUI

HP-1msUI

DB-5msUI

HP-5msUI

DB-17msUI

DB-624UI

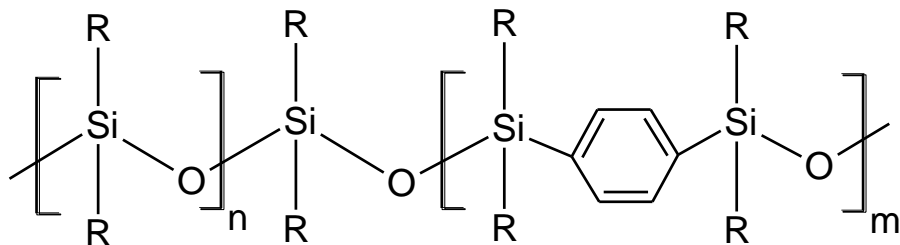
DB-Select 624UI 467

DB-WAXUI

Same Selectivity, more Inertness!

Three Types Of Low Bleed Phases

- Phases tailored to “mimic” currently existing polymers
Examples: DB-5ms, DB-35ms, DB-17ms, VF-1701ms

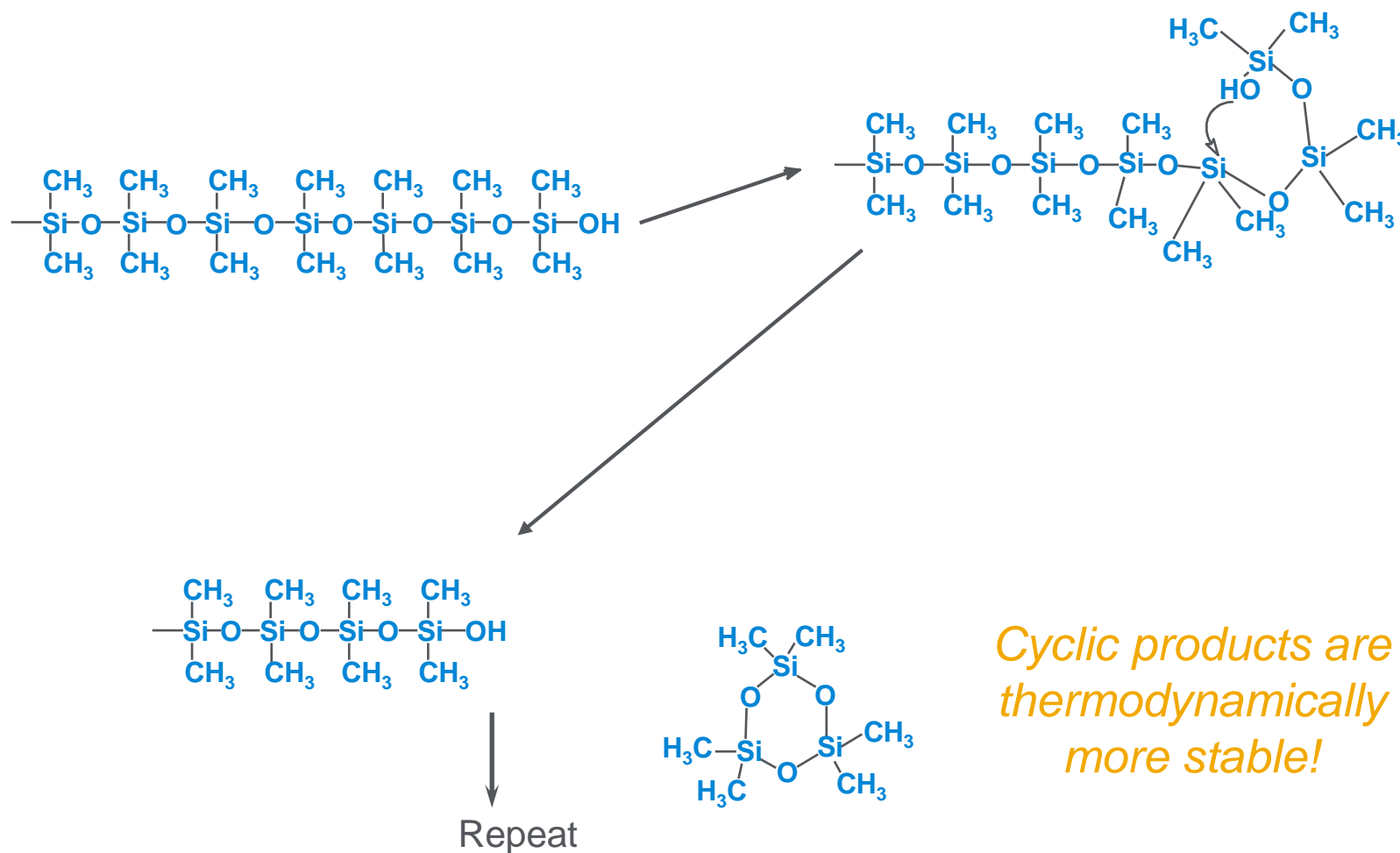


Siarylene backbone

- New phases unrelated to any previously existing polymers
Examples: DB-XLB
- Optimized manufacturing processes
DB-1ms, HP-1ms, HP-5ms, VF-5ms

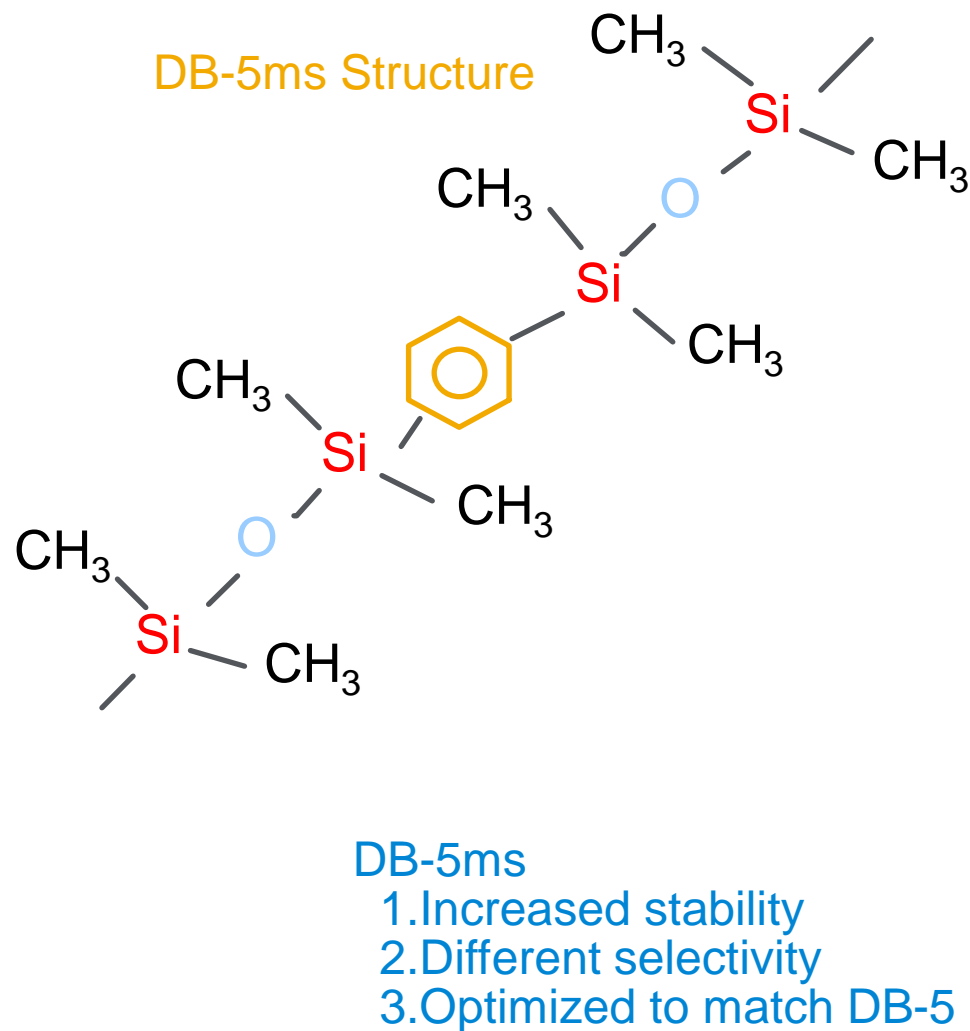
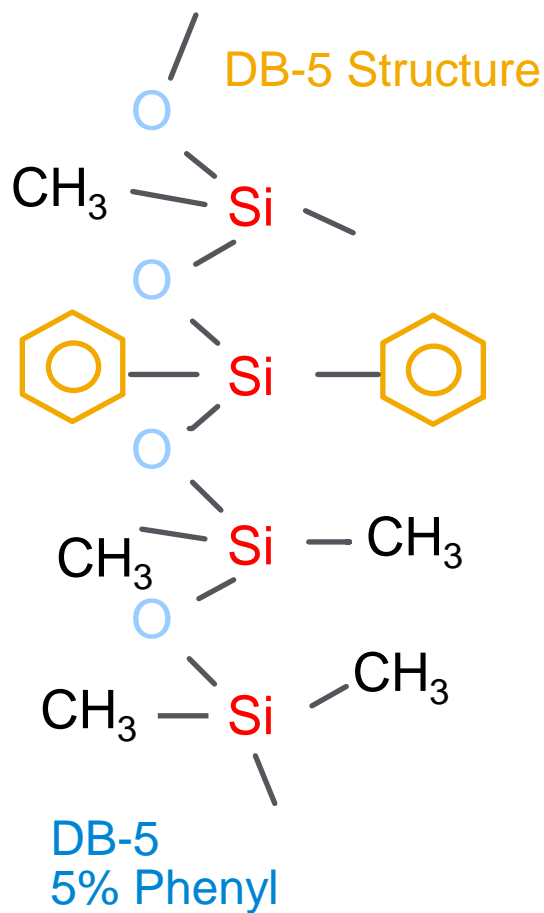
What is Column Bleed???

“Back Biting” Mechanism of Product Formation



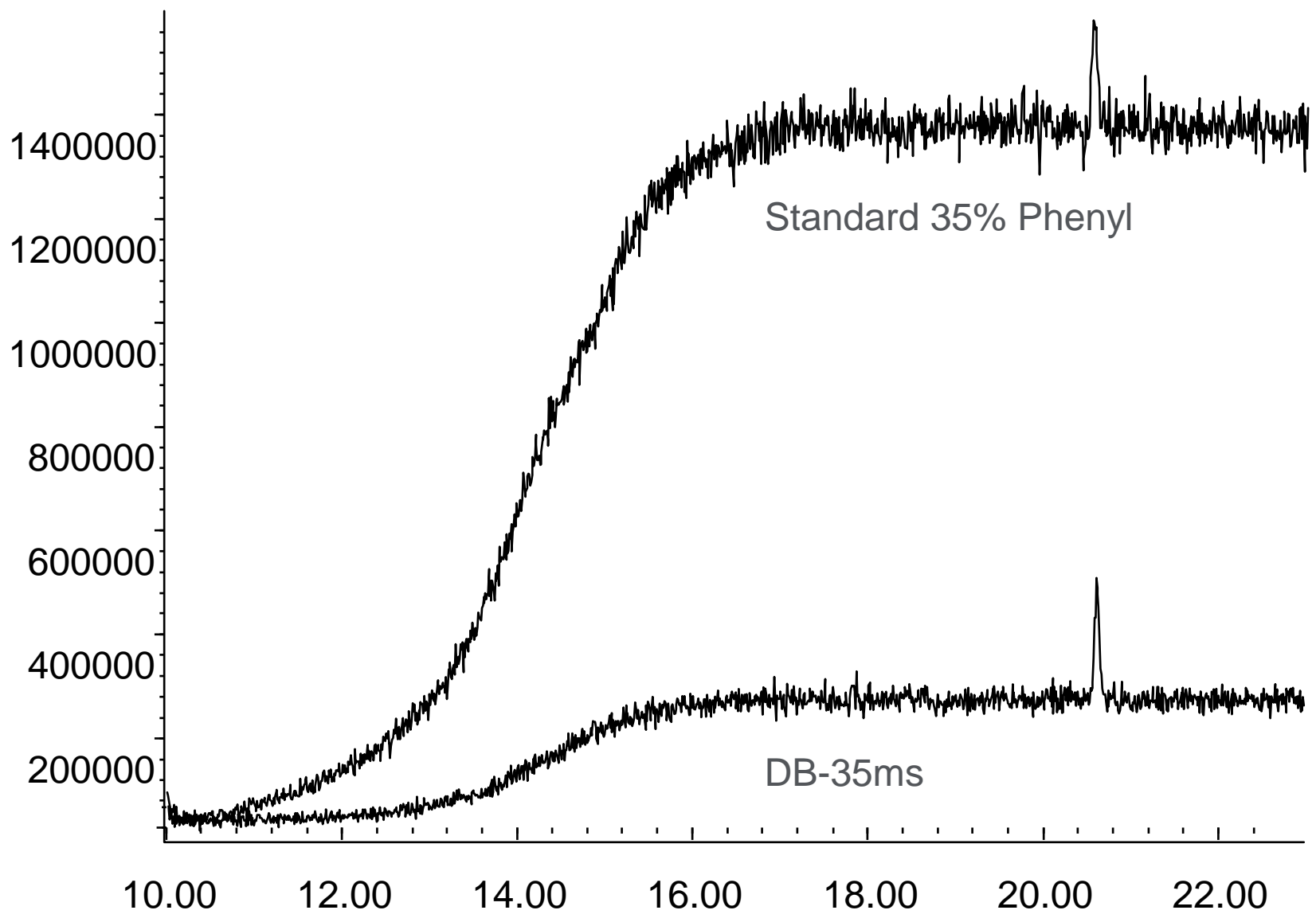
Cyclic products are thermodynamically more stable!

DB-5ms Structure



DB-35MS VS STANDARD 35% PHENYL

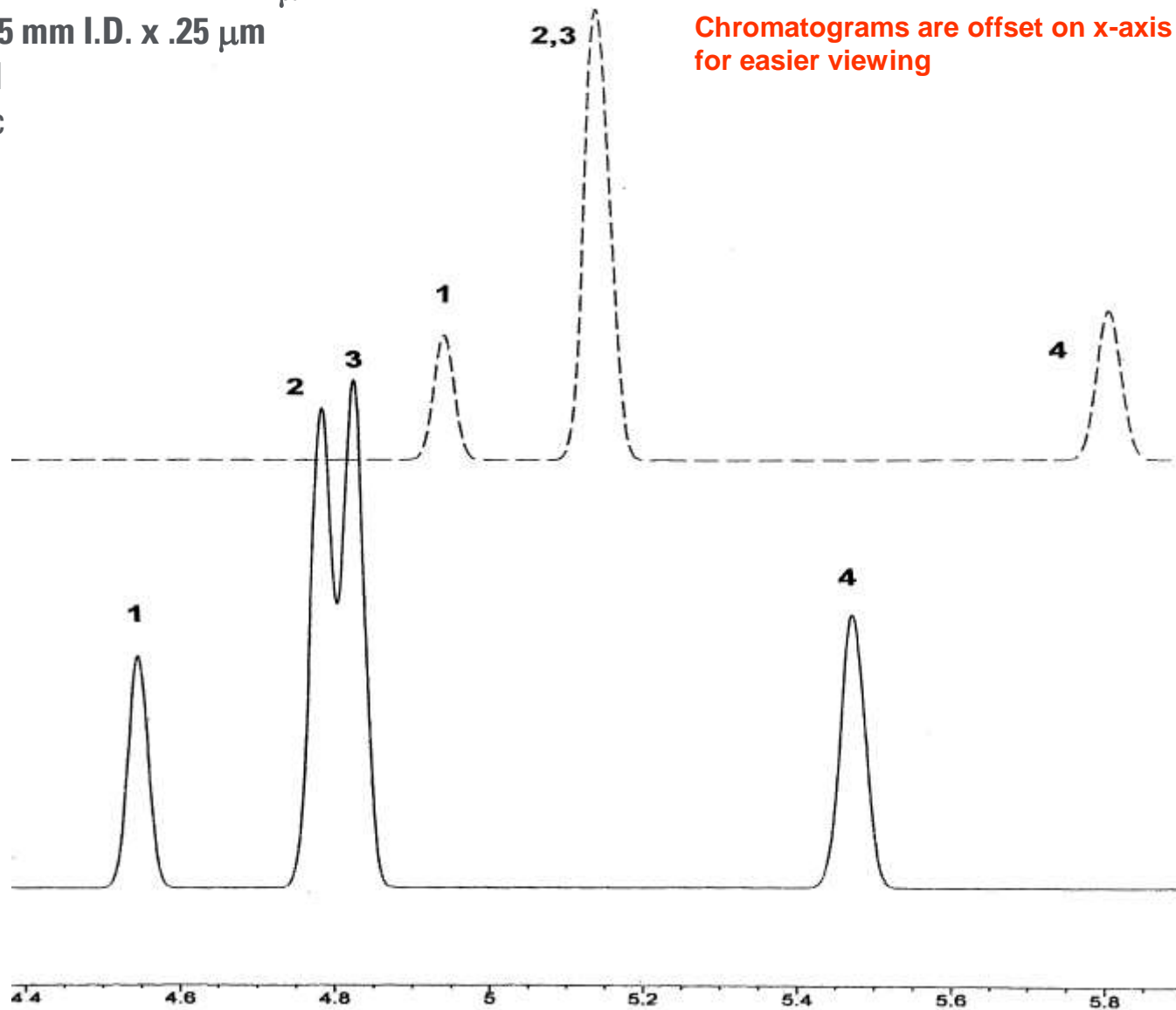
Benzo[g,h,i]perylene, 1ng



Solid line: DB-5ms 30 m x .25 mm I.D. x .25 μ m
Dashed line: DB-5 30 m x .25 mm I.D. x .25 μ m
Oven: 60° C isothermal
Carrier gas: H₂ at 40 cm/sec

- 1: Ethylbenzene
- 2: m-Xylene
- 3: p-Xylene
- 4: o-Xylene

Chromatograms are offset on x-axis for easier viewing



Why is stationary phase type important?

$$R_s = \frac{\sqrt{N}}{4} \left(\frac{k}{k+1} \right) \left(\frac{\alpha-1}{\alpha} \right)$$

Influence on α

$$\alpha = \frac{k_2}{k_1}$$

k_2 = partition ratio of 2nd peak
 k_1 = partition ratio of 1st peak

Selectivity

- Relative spacing of the chromatographic peaks
- The result of all non-polar, polarizable and polar interactions that cause a stationary phase to be more or less retentive to one analyte than another

Optimizing Selectivity (α)

Match analyte polarity to stationary phase polarity

- 'like dissolves like'

Take advantage of unique interactions between analyte and stationary phase functional groups

Analyte Polarity

Nonpolar Molecules - generally composed of only carbon and hydrogen and exhibit no dipole moment (Straight-chained hydrocarbons (n-alkanes))

Polar Molecules - primarily composed of carbon and hydrogen but also contain atoms of nitrogen, oxygen, phosphorus, sulfur, or a halogen (Alcohols, amines, thiols, ketones, nitriles, organo-halides, etc. Includes dipole-dipole interactions and H-bonding)

Polarizable Molecules - primarily composed of carbon and hydrogen, but also contain unsaturated bonds (Alkenes, alkynes and aromatic compounds)

Selectivity Interactions

- Dispersion
- Dipole
- Hydrogen bonding

Dispersion Interaction

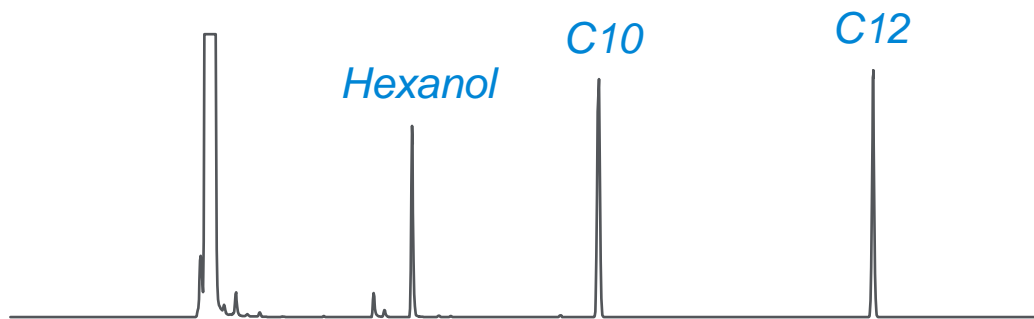
$$\Delta H_{\text{vap}}$$

- Separation by differences in analyte heat of vaporizations (ΔH_{vap})
- Heat necessary to convert a liquid into a gas (at the same temperature)

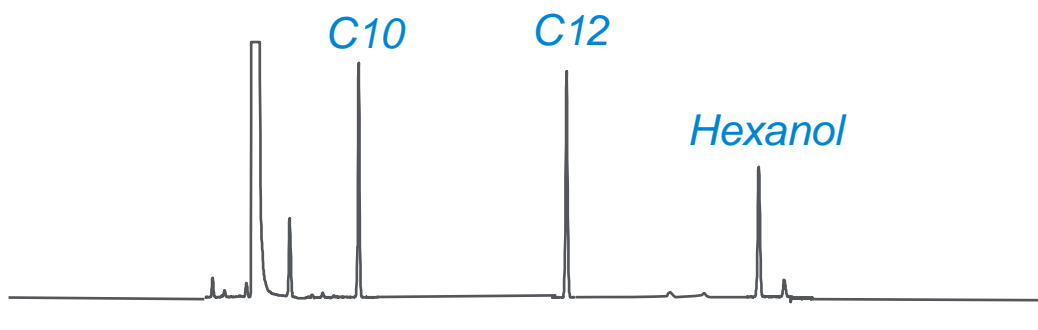
Dispersion Interaction

Solubility And Retention

Hexanol	158°C
Decane	174°C
Dodecane	216°C



100% Methyl
(non-polar)



100% PEG
(polar)

30 m x 0.32 mm ID, 0.25 μ m
He at 35 cm/sec
50-170°C at 15°/min

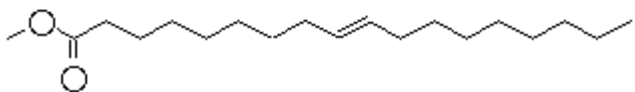
Dispersion Interaction

$$\Delta H_{\text{vap}}$$

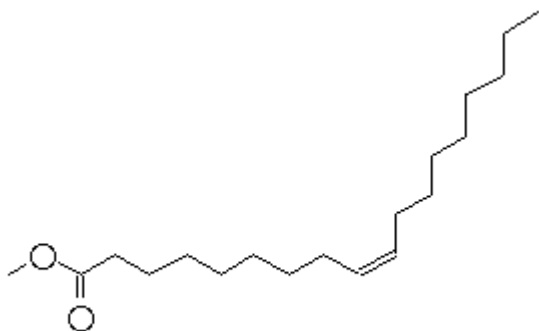
Vapor pressure: good approximation

Boiling point: poor approximation

Dipole Interaction



C18:1 (Methyl *trans*-9-octadecenoate)
B.Pt. 186°C



C18:1 (Methyl *cis*-9-octadecenoate)
B.Pt. 186°C

Smaller differences require a stronger dipole phase

Fames – 37 Component Standard

Column: DB-23
60 m X 0.25 mm X 0.15 µm

Agilent P/N 122-2361

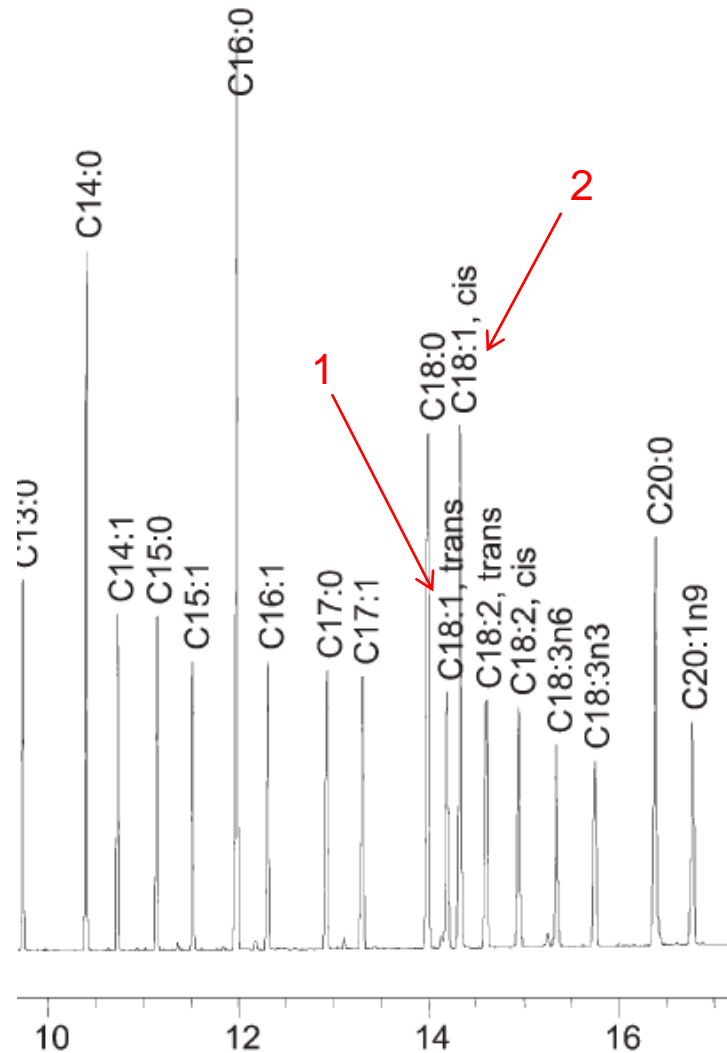
Carrier: He , 33 cm/sec @ 50°C

Oven: 50°C for 1 min
25°C/min to 175 (no hold)
4°C/min to 230°C hold 5 min

Injector: 250°C, Split 50:1, 1µL

Detector: FID, 250°C

- 1 C18:1 (Methyl *trans*-9-octadecenoate)
- 2 C18:1 (Methyl *cis*-9-octadecenoate)



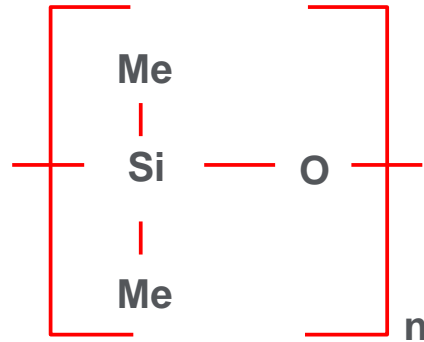
Hydrogen Bonding Interaction

Dipole-Dipole interaction with H bound to O or N interacting with an O or N



NONPOLAR PHASES

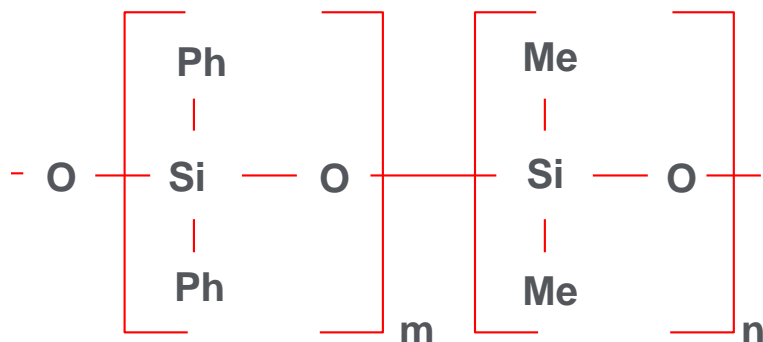
Typified by 100% polydimethylsiloxanes such as HP-1, DB-1, DB-1ms, HP-1ms, VF-1ms, CP-Sil 5 CB



Separation Mechanisms:
- Dispersion only

POLARIZABLE PHASES

Typified by phenyl substituted siloxanes, substituted at 5-50% (HP-5, HP-5ms, DB-35, DB-35ms, DB-17, DB-17ms)



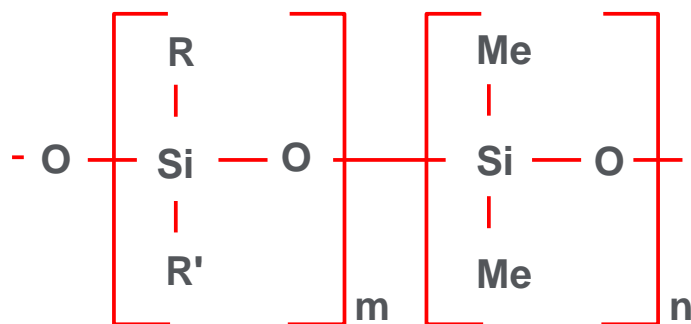
5%--weakly polar,
rest--mid polar

Separation Mechanisms:

- Dispersion
- Inducible dipole at phenyl groups

STRONG DIPOLE PHASES

Typified by cyanopropyl or trifluoropropyl substituted siloxanes, substituted 6-50% (DB-1701, DB-1301, DB-200, DB-23, DB-225)



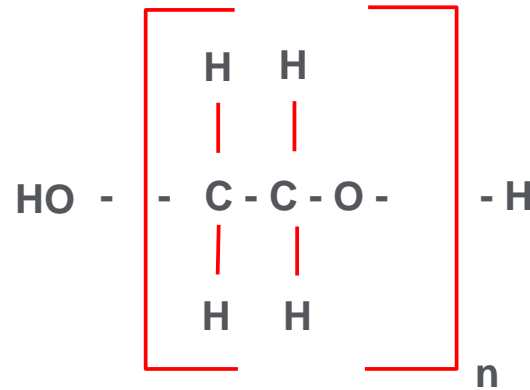
R = cyanopropyl or trifluoropropyl
R' = phenyl or methyl

Separation Mechanisms:

- Dispersion
- Inducible dipole at phenyl groups
- Strong permanent dipole
- Hydrogen bonding

HYDROGEN BONDING PHASES

Typified by polyethylene glycol polymers (Carbowax, HP-INNOWax, DB-WAX, DB-FFAP, VF-WAXms, CP-WAX52CB....)



Separation Mechanisms:

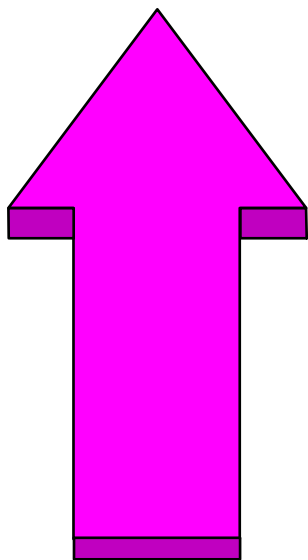
- Dispersion
- Strong permanent dipole
- Hydrogen bonding

Selectivity

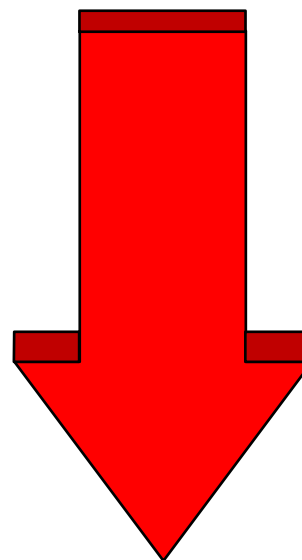
Interaction Strengths

Phase	Dispersion	Dipole	H Bonding
Methyl	Strong	None	None
Phenyl	Strong	None	Weak
Cyanopropyl	Strong	Very Strong	Moderate
Trifluoropropyl	Strong	Moderate	Weak
PEG	Strong	Strong	Moderate

Polarity



Polarity

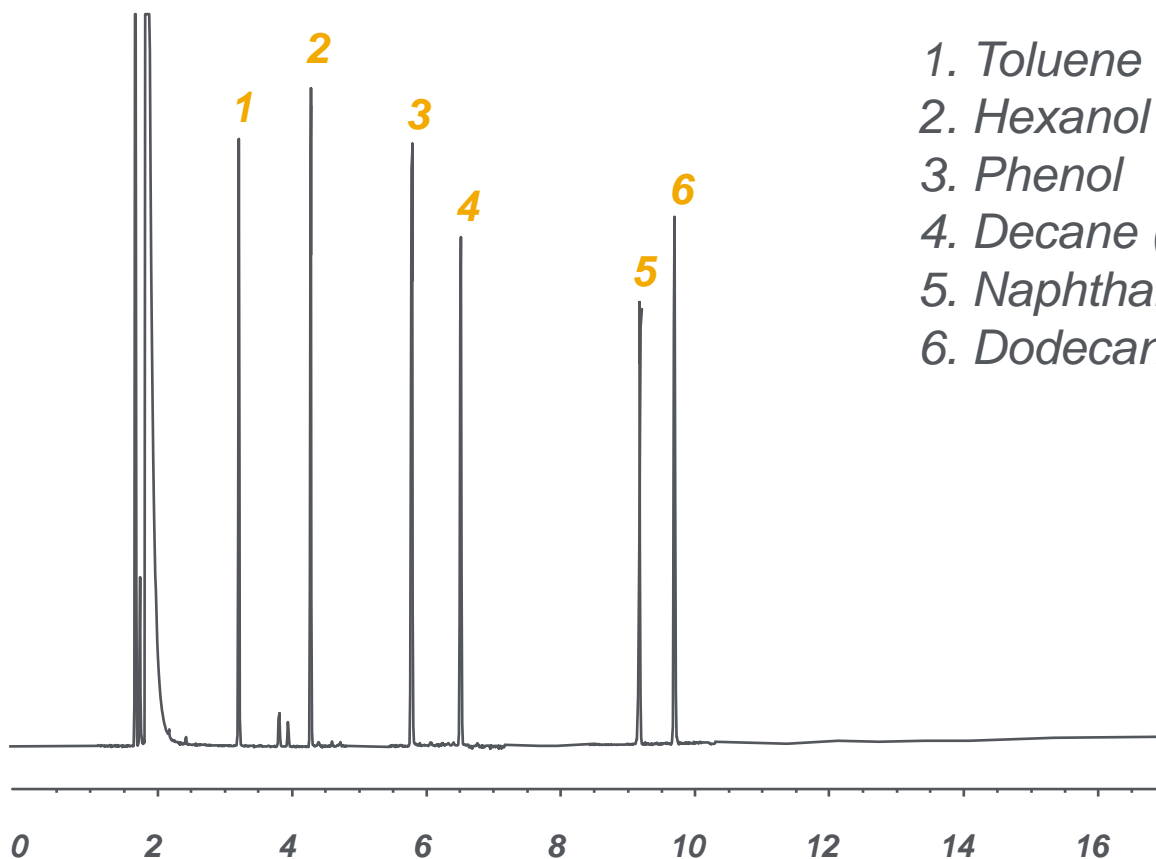


Stability
Temperature Range

Compounds & Properties

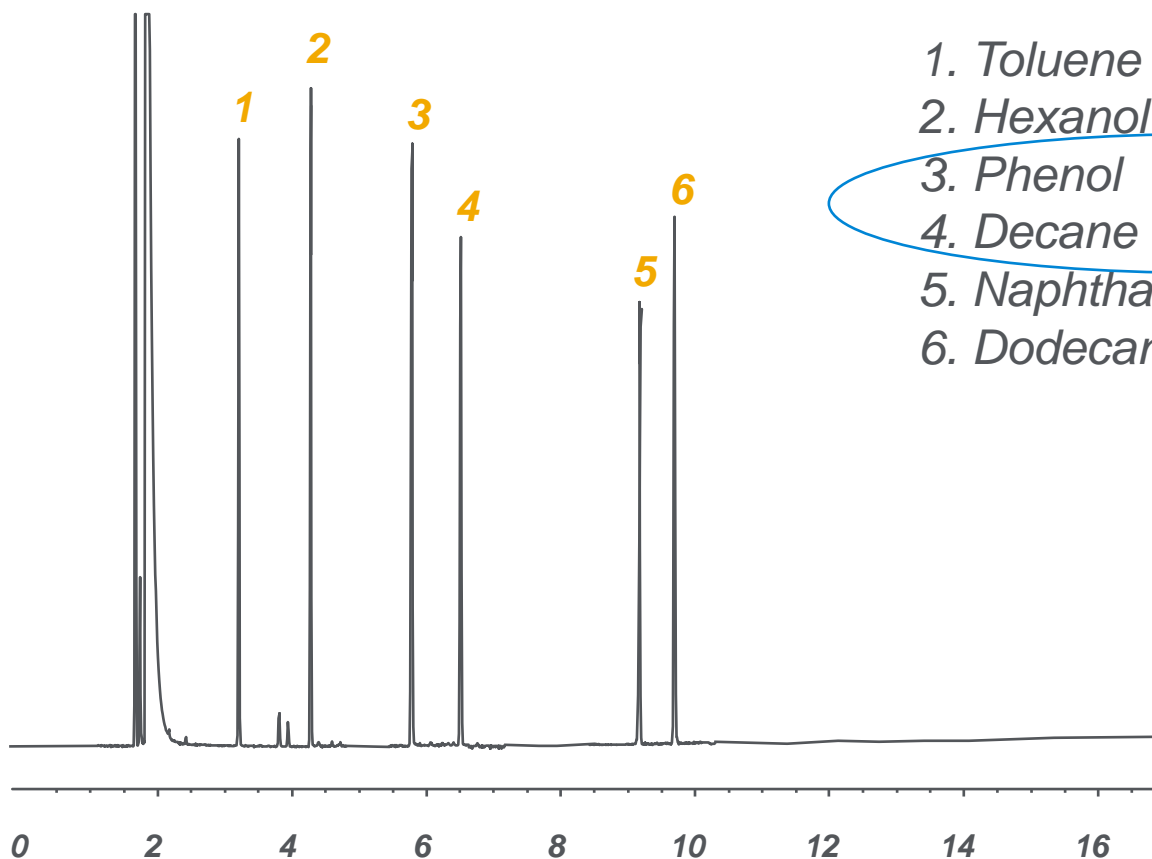
Compounds	Polar	Aromatic	Hydrogen Bonding	Dipole
Toluene	no	yes	no	induced
Hexanol	yes	no	yes	yes
Phenol	yes	yes	yes	yes
Decane	no	no	no	no
Naphthalene	no	yes	no	induced
Dodecane	no	no	no	no

100% Methyl Polysiloxane



Strong Dispersion
No Dipole
No H Bonding

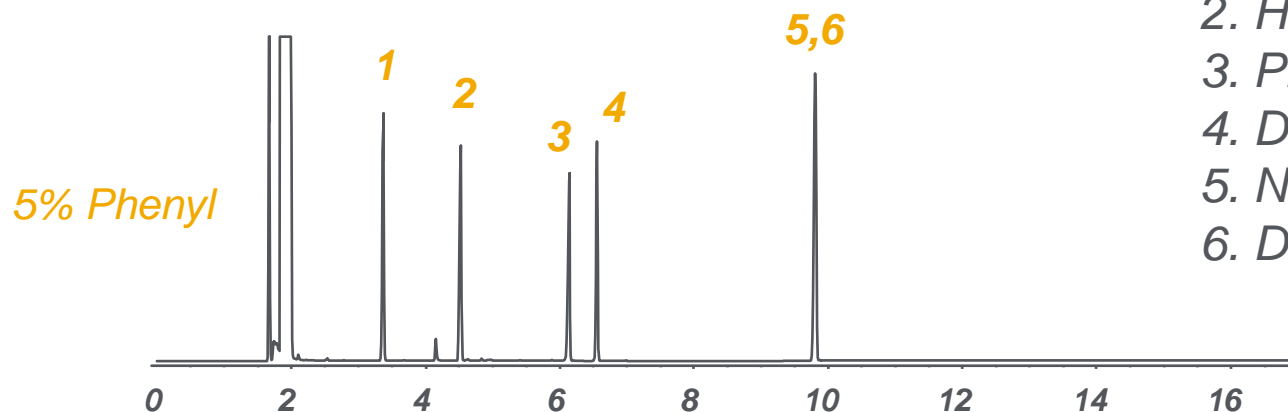
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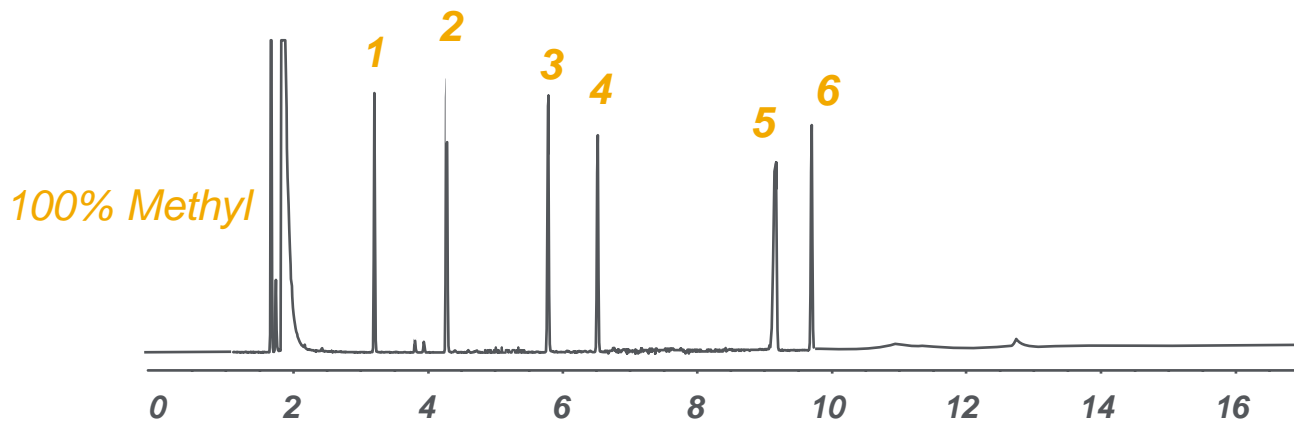
- | | |
|-------------------|-------|
| 1. Toluene | 110°C |
| 2. Hexanol | 158°C |
| 3. Phenol | 181°C |
| 4. Decane (C10) | 174°C |
| 5. Naphthalene | 218°C |
| 6. Dodecane (C12) | 216°C |

Strong Dispersion
No Dipole
No H Bonding

5% Phenyl



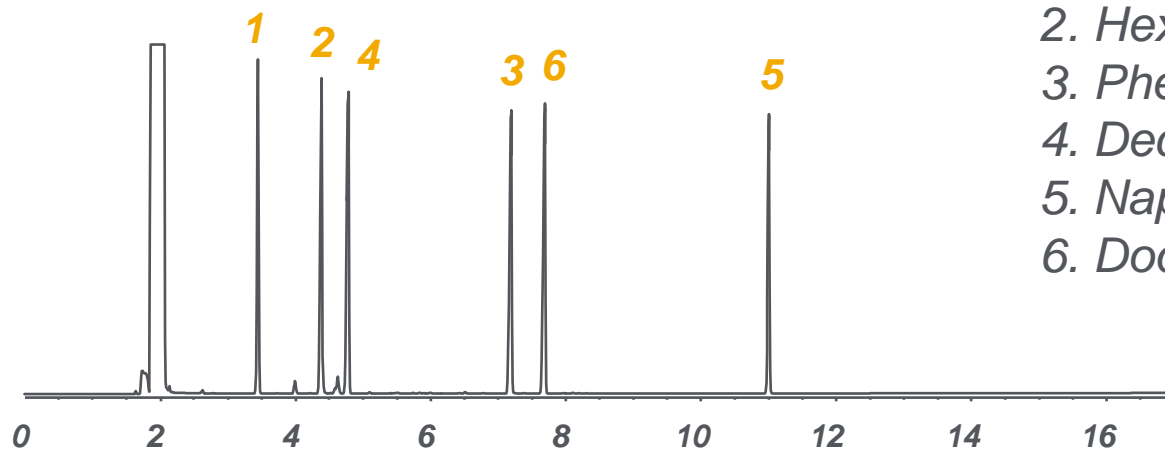
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Strong Dispersion
No Dipole
Weak H Bonding

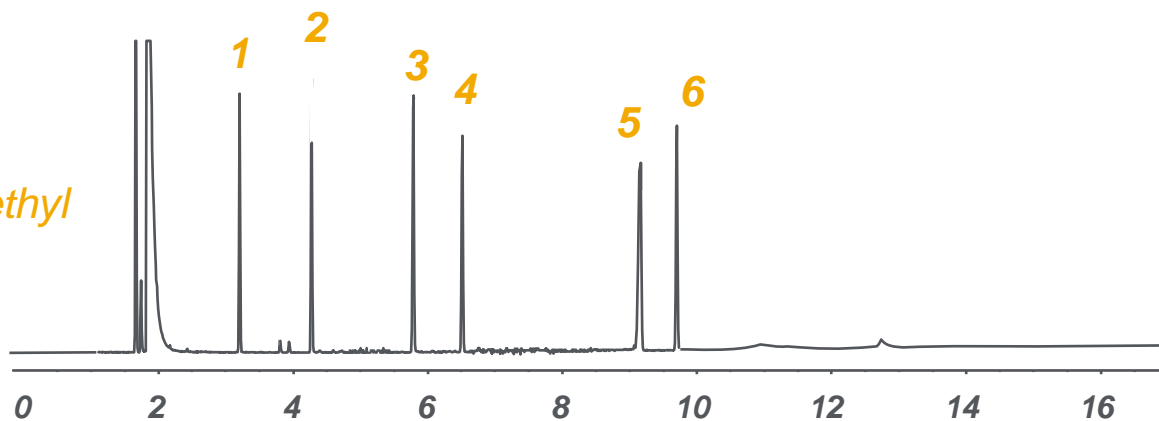
50% Phenyl

50%
Phenyl



1. Toluene 110°C
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6. Dodecane (C12) 216°C

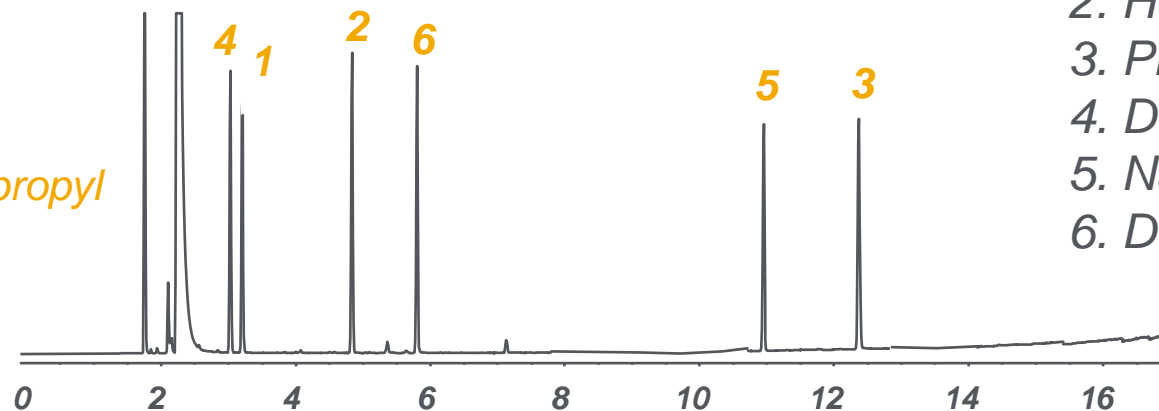
100% Methyl



Strong Dispersion
No Dipole
Weak H Bonding

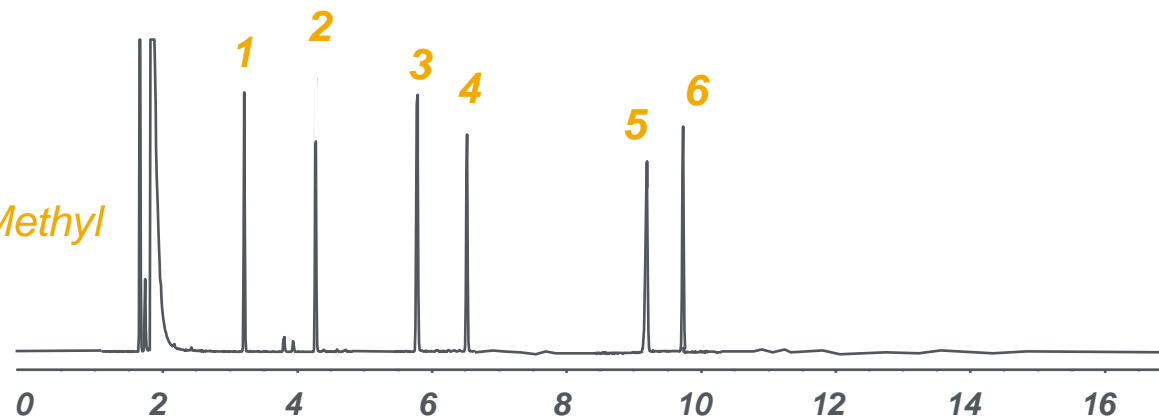
50% Cyanopropyl

50%
Cyanopropyl



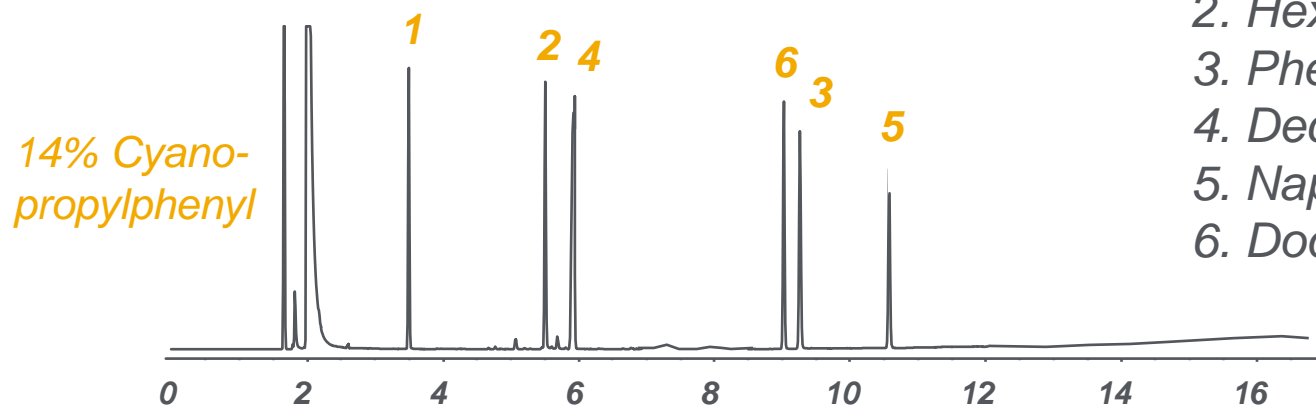
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100% Methyl

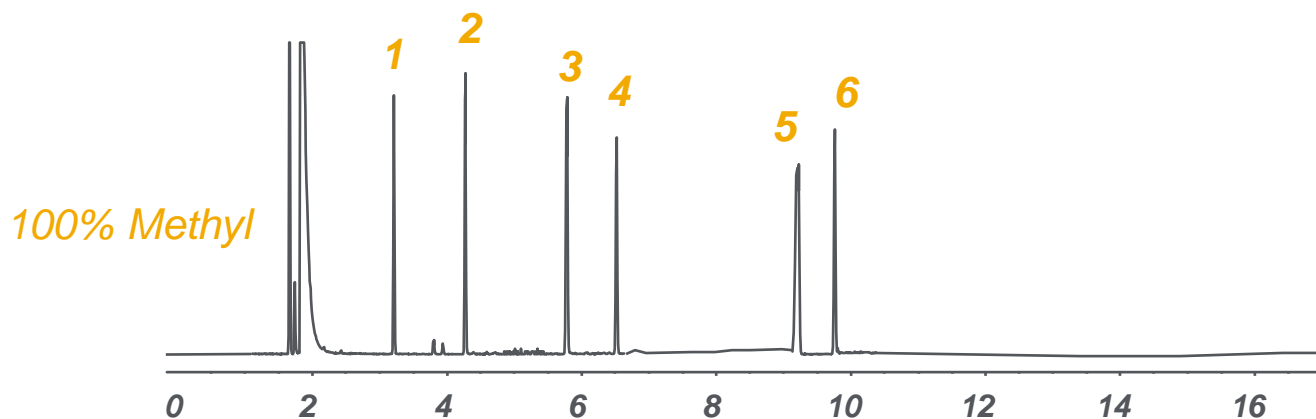


Strong Dispersion
Strong Dipole
Moderate H Bonding

14% Cyanopropylphenyl



1. Toluene	110°C
2. Hexanol	158°C
3. Phenol	181°C
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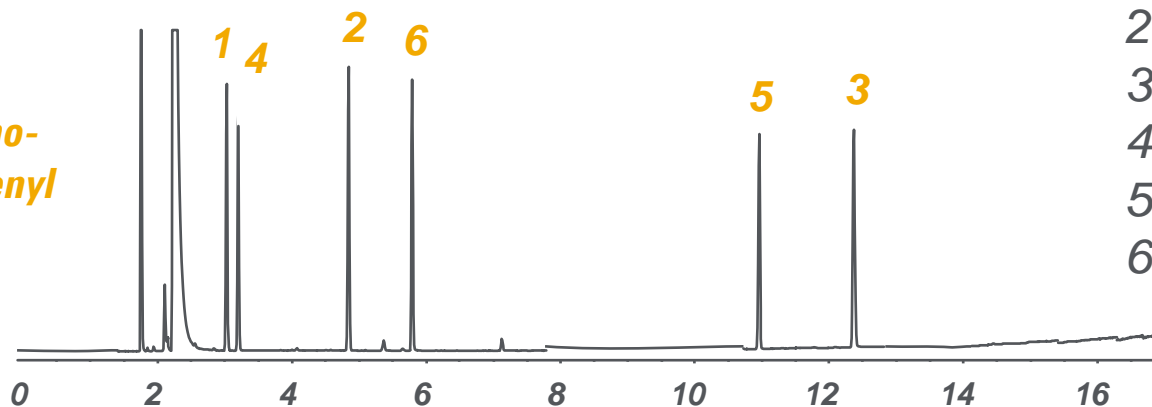


Strong Dispersion
None/Strong Dipole (Ph/CNPr)
Weak/Moderate H Bonding (Ph/CNPr)

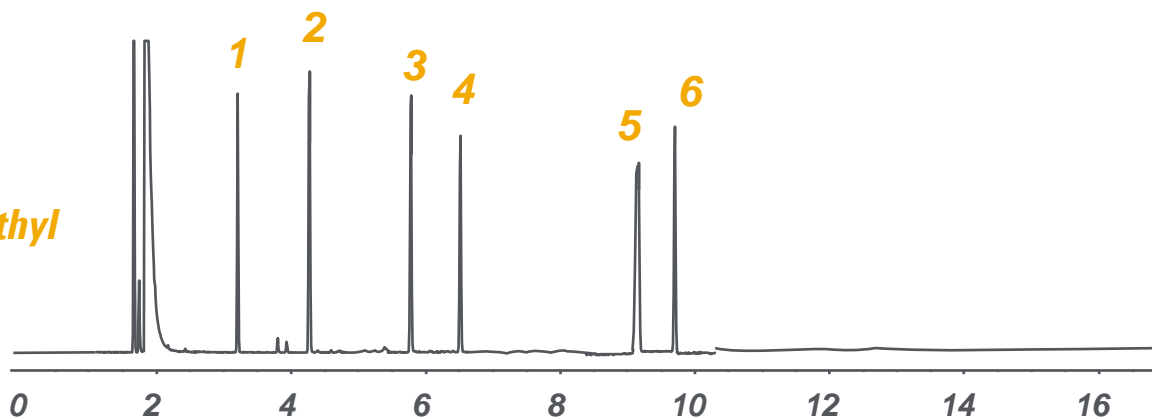
50% Cyanopropylphenyl

1. Toluene
2. Hexanol
3. Phenol
4. Decane (C10)
5. Naphthalene
6. Dodecane (C12)

50% Cyano-
propylphenyl

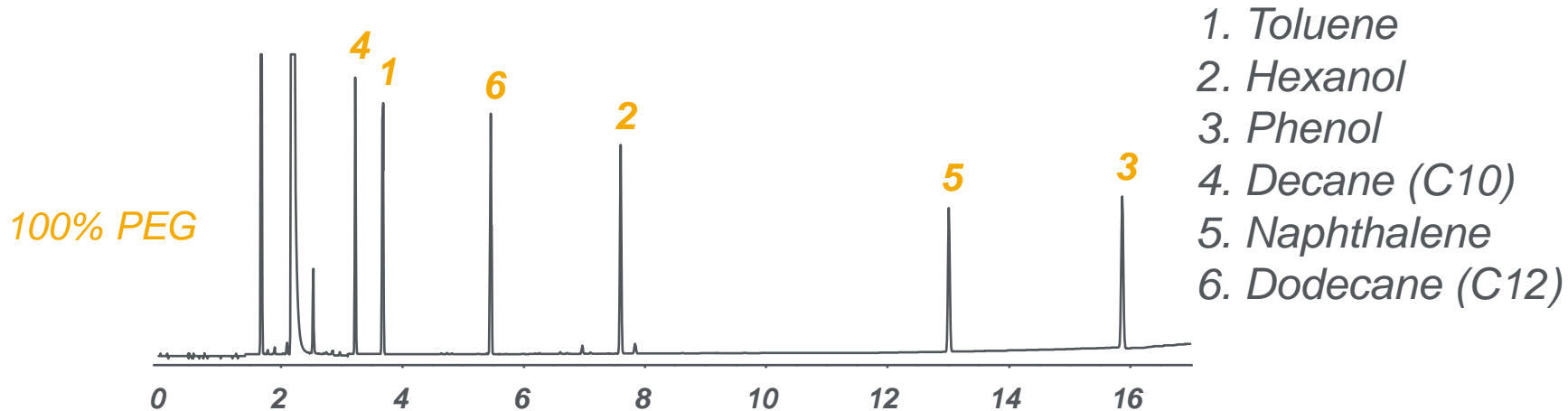


100% Methyl

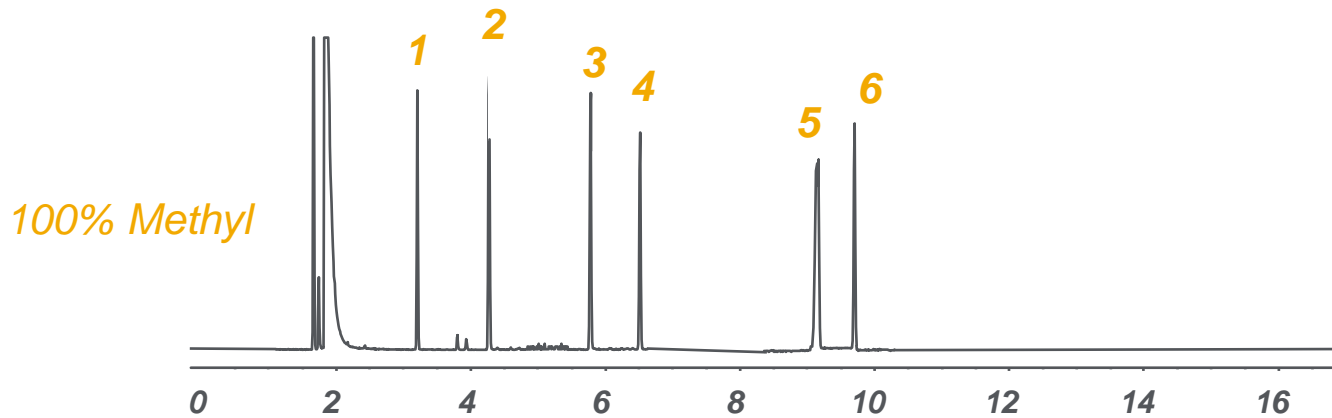


Strong Dispersion
None/Strong Dipole (Ph/CNPr)
Weak/Moderate H Bonding (Ph/CNPr)

100% Polyethylene Glycol



1. Toluene
2. Hexanol
3. Phenol
4. Decane (C10)
5. Naphthalene
6. Dodecane (C12)



Strong Dispersion
Strong Dipole
Moderate H Bonding

Stationary Phase Selection

Part 1

- Existing information
- Selectivity
- Polarity
- Critical separations
- Temperature limits

Stationary Phase Selection

Part 2

- Capacity
- Analysis time
- Bleed
- Versatility
- Selective detectors

Column Dimensions

- Inner diameter
- Length
- Film Thickness

Column Diameter

Capillary Columns

I.D. (mm)	Common Name
0.53	Megabore
0.45	High speed Megabore
0.32	Wide
0.20-0.25	Narrow
0.18	Minibore

Column Diameter

Theoretical Efficiency

I.D. (mm)	N/m
0.10	11905
0.18	6666
0.20	5941
0.25	4762
0.32	3717
0.53	2242

$k = 5$

Efficiency and Resolution Relationship

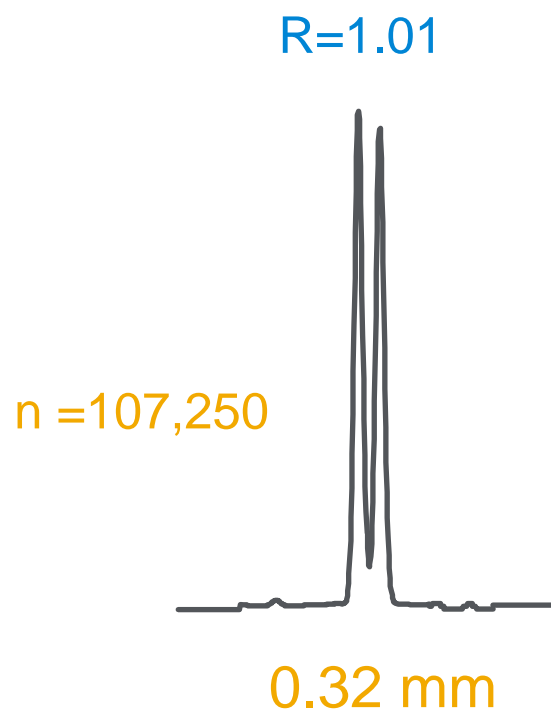
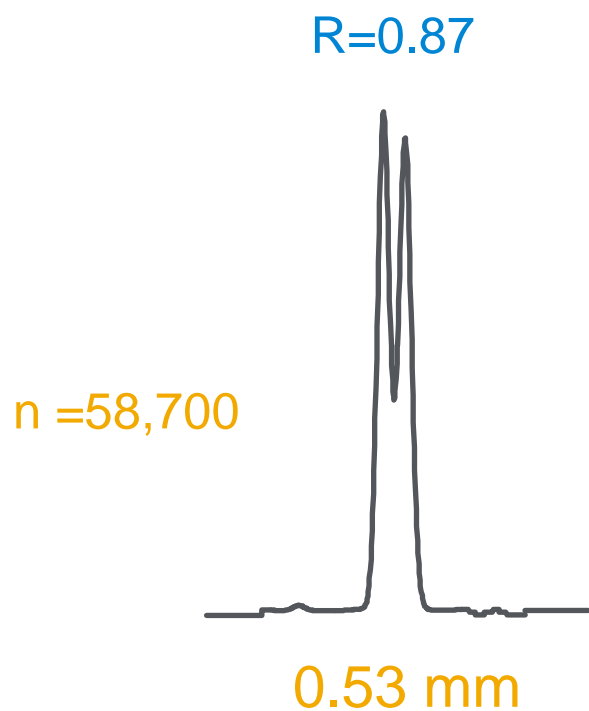
$$\sqrt{N} \propto R_s$$

Efficiency X 4 = Resolution X 2

Column Diameter

Resolution

180°C isothermal



Square root of resolution is inversely proportional to column diameter

Column Diameter

Inlet Head Pressures

Helium

I.D (mm)	Pressure (psig)
0.10	225-250
0.20	25-35
0.25	15-25
0.32	10-20
0.53	2-4

30 meters

Hydrogen pressures x 1/2

Column Diameter

Capacity

Like Polarity Phase/Solute

I.D. (mm)	Capacity (ng)
0.20	50-100
0.25	75-150
0.32	125-250
0.53	200-400

0.25 μm film thickness

Column Diameter

Carrier Gas Flow Rate

Smaller diameters for low flow situations
(e.g., GC/MS)

Larger diameters for high flow situations
(e.g., purge & trap, headspace, gas sample valve)

Column Length

Most common: 15-60 meters

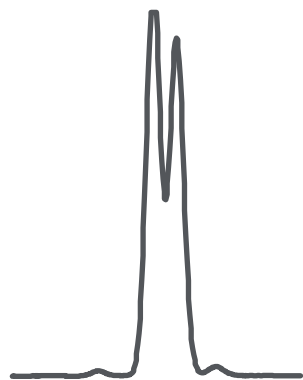
Available: 5-200 meters

Column Length

Resolution and Retention
210°C isothermal

R=0.84

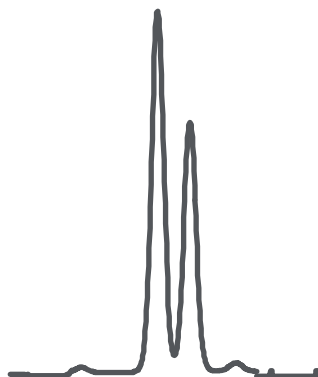
2.29 min



15 m

R=1.16

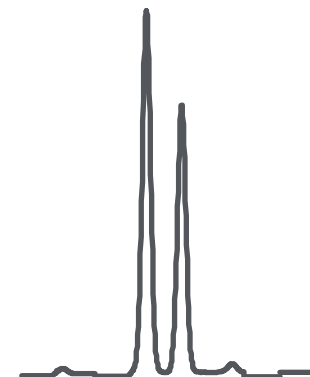
4.82 min



30 m

R=1.68

8.73 min



60 m

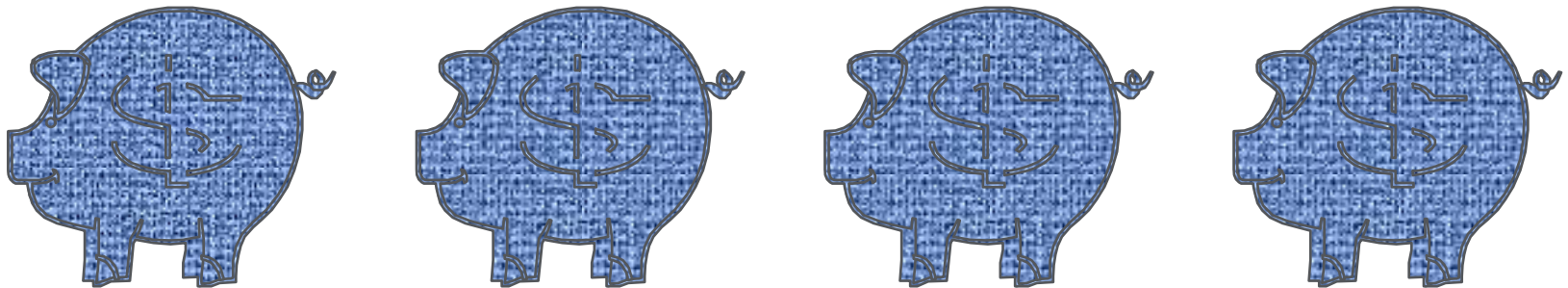
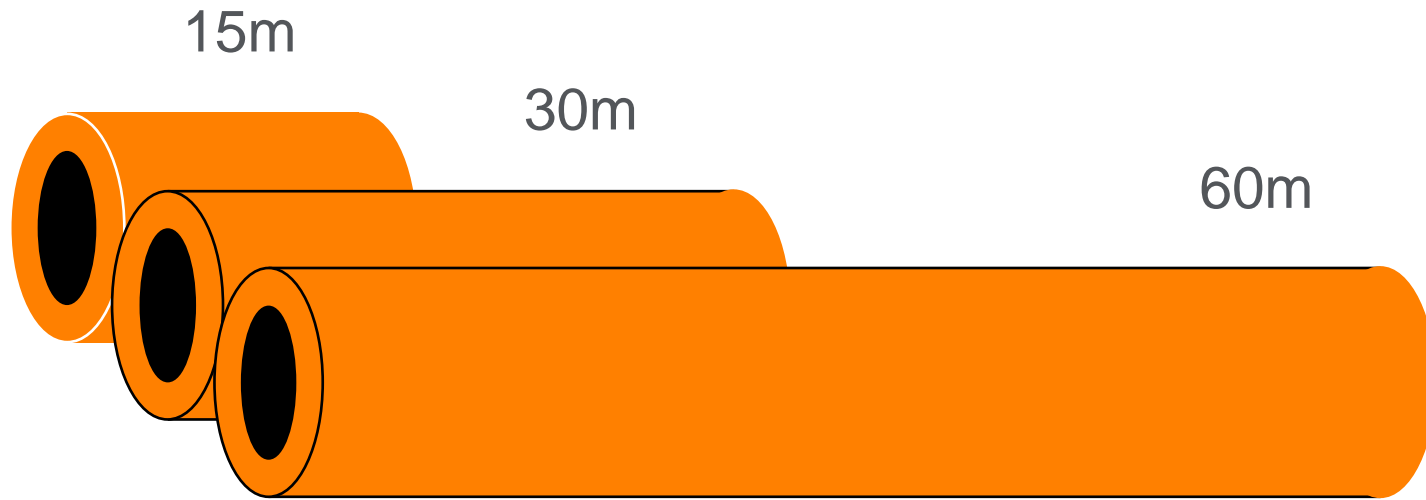
Resolution is proportional to the square root of column length

Isothermal: Retention is proportional to length

Temperature program: 1/3-1/2 of isothermal values

Column Length

Cost



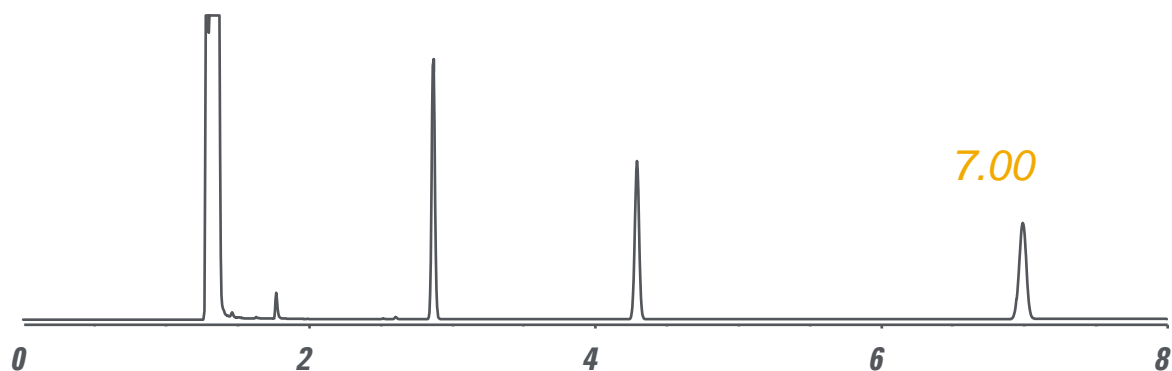
Film Thickness

Most common: 0.1-3.0 μm

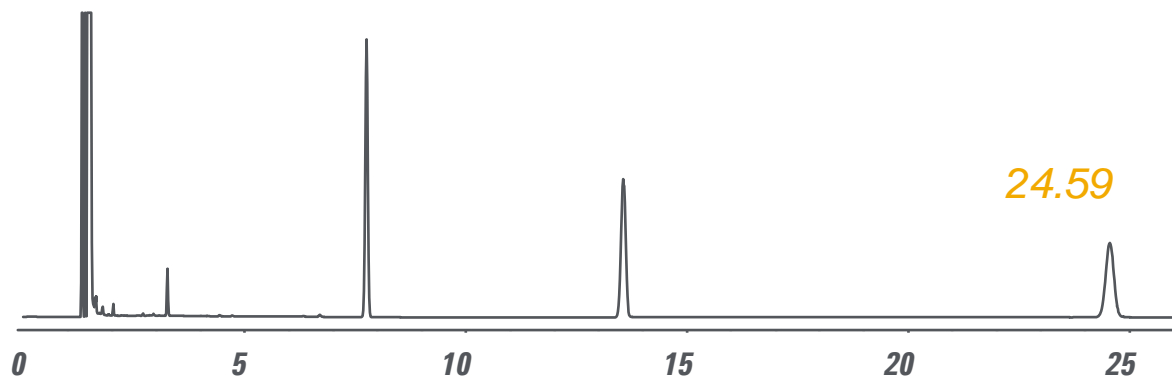
Available: 0.1-10.0 μm

Film Thickness

Retention
100°C Isothermal



0.25 μm

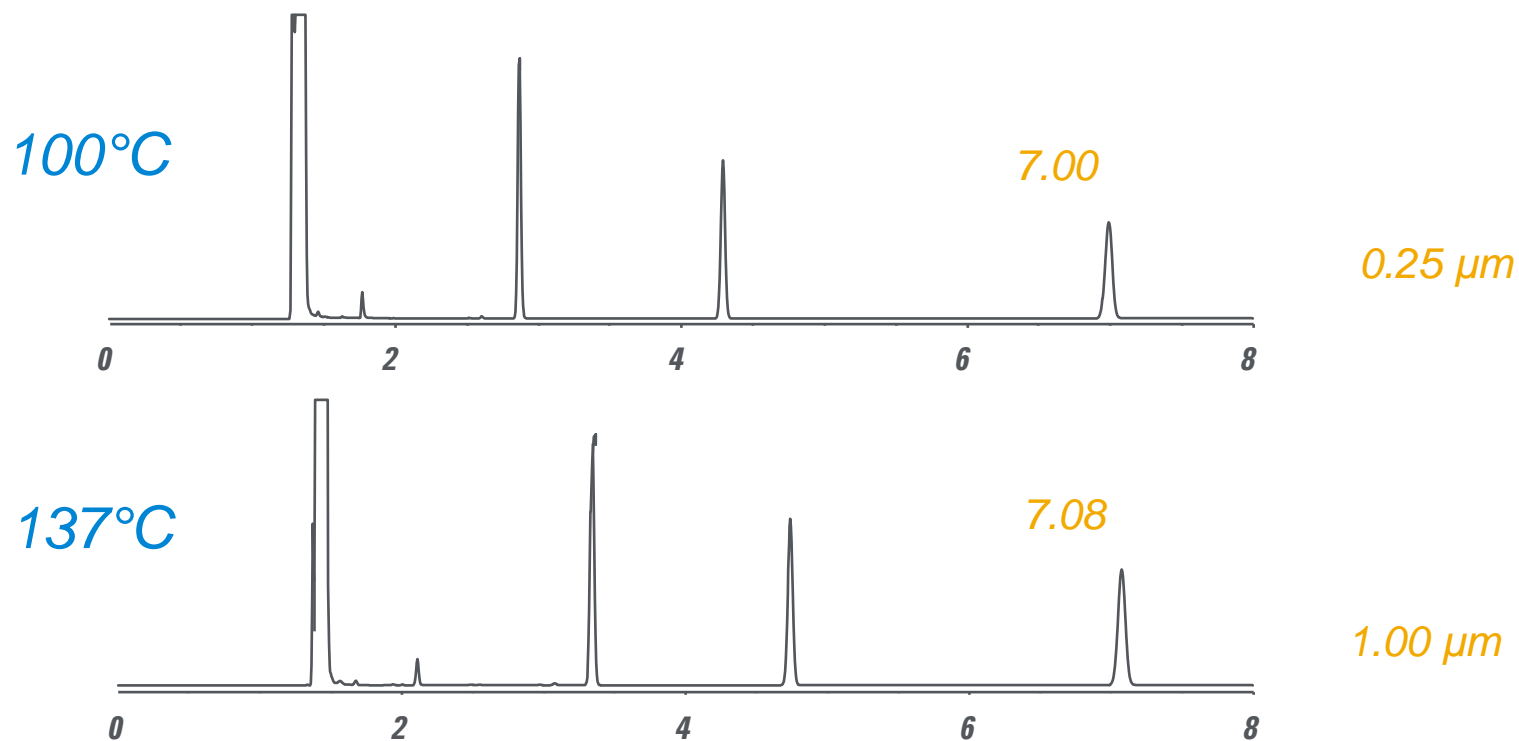


1.00 μm

Isothermal: Retention is proportional to film thickness
Temperature program: 1/3-1/2 of isothermal values

Film Thickness

Equal Retention: Isothermal

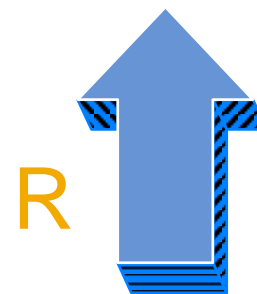
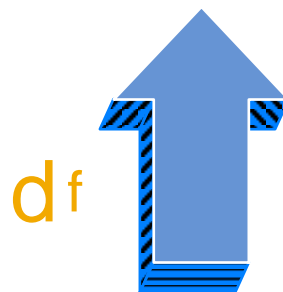


DB-1, 30 m x 0.32 mm ID
He at 37 cm/sec
C10, C11, C12

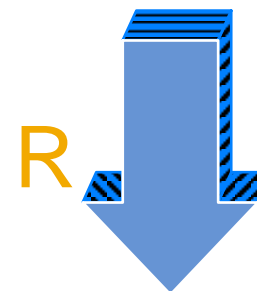
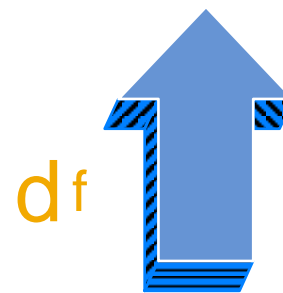
Film Thickness

Resolution

When solute $k < 5$

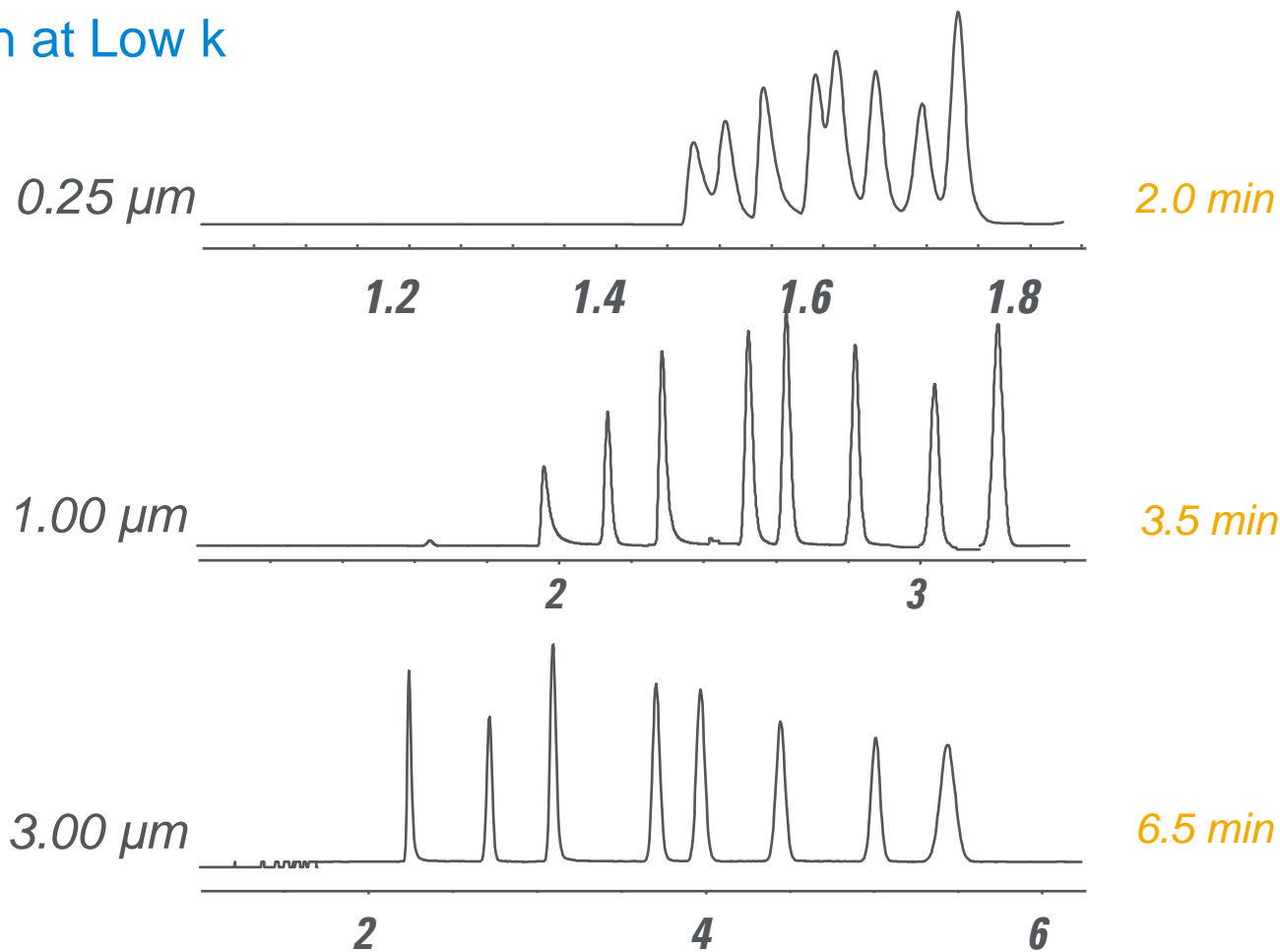


When solute $k > 5$



Film Thickness

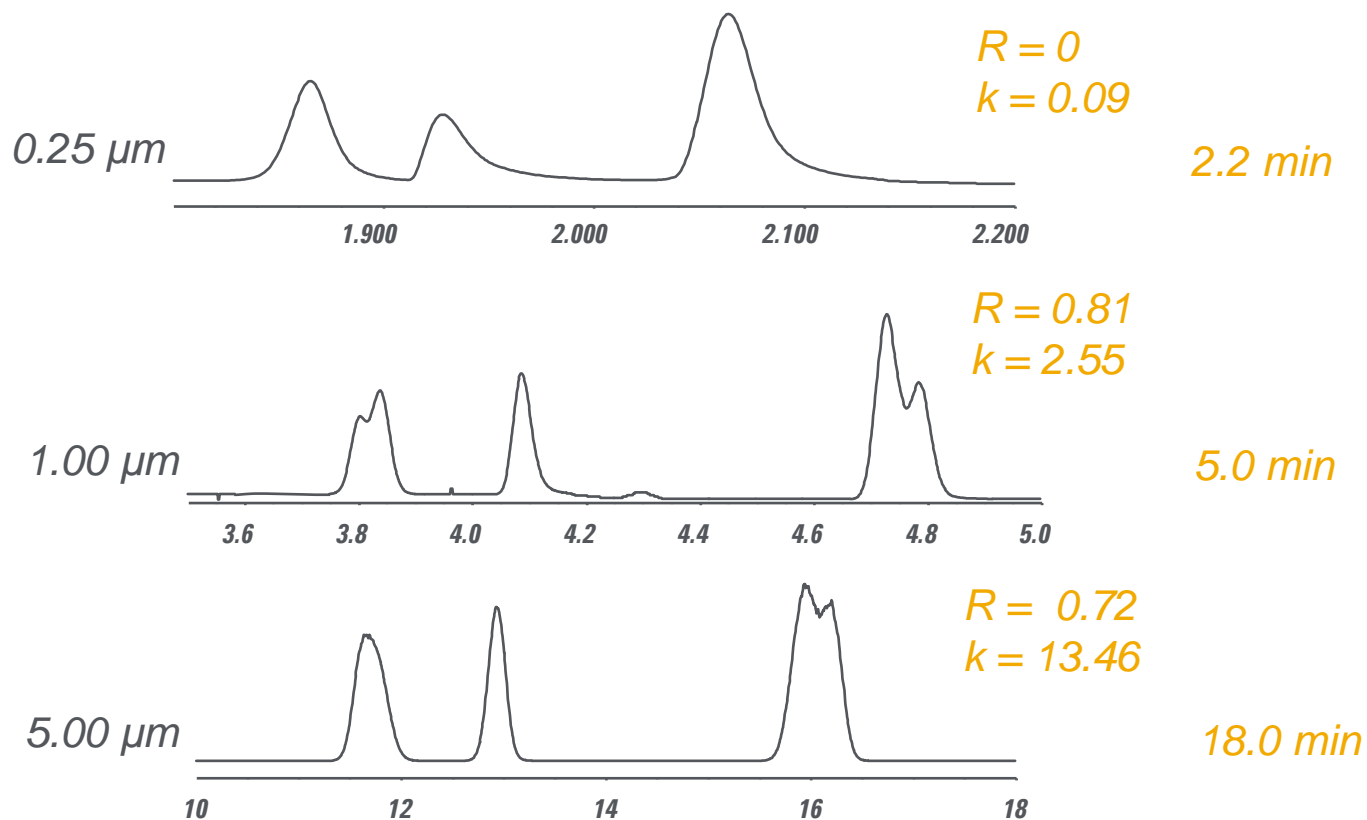
Resolution at Low k



DB-1, 30 m x 0.32 mm ID
40°C isothermal, He at 35 cm/sec
Solvent mixture

Film Thickness

Resolution at High k



DB-1, 30 m x 0.32 mm ID
40°C isothermal, He at 35 cm/sec
Solvent mixture

Film Thickness

Capacity

Like Polarity Phase/Solute

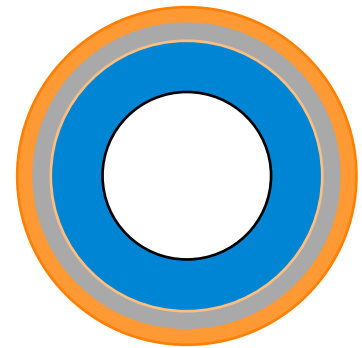
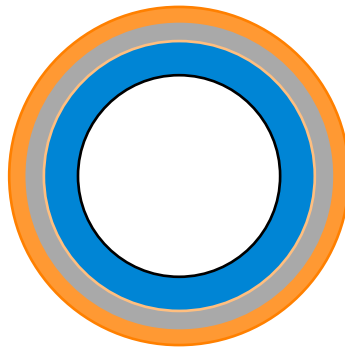
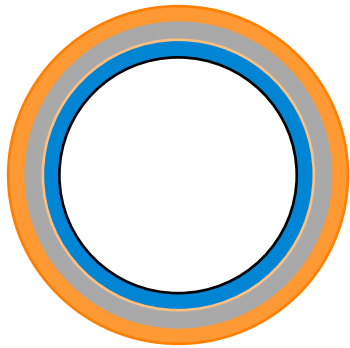
Thickness (um)	Capacity (ng)
0.10	50-100
0.25	125-250
1.0	500-1000
3.0	1500-3000
5.0	2500-5000

0.32 mm I.D.

Film Thickness

Bleed

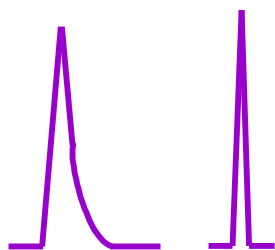
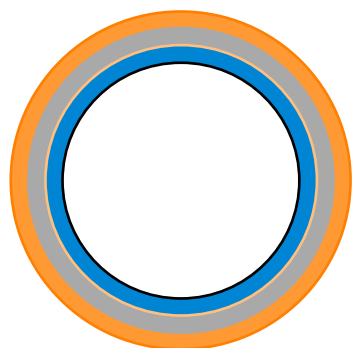
More stationary phase = More degradation products



Film Thickness

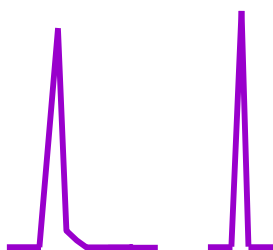
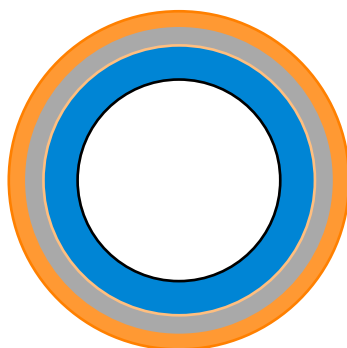
Inertness Summary

0.25



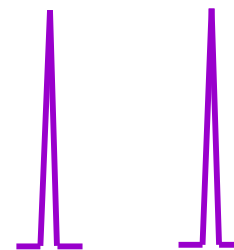
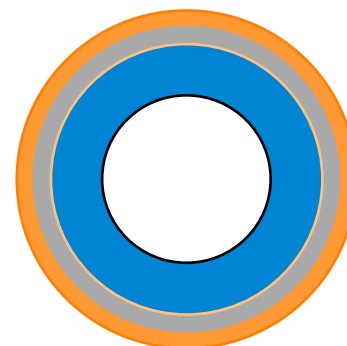
active inactive

1.0



active inactive

3.0



active inactive

Column Dimensions

Diameter Summary

To Increase

Make Diameter

Resolution

Smaller

Retention

Smaller

Pressure

Smaller

Flow rate

Larger

Capacity

Larger

Column Dimensions

Length Summary

To Increase

Make Length

Resolution

Longer

Retention

Longer

Pressure

Longer

Cost

Longer

Column Dimensions

Film Thickness Summary

To Increase

Make Film

Retention

Thicker

Resolution ($k < 5$)

Thicker

Resolution ($k > 5$)

Thinner

Capacity

Thicker

Inertness

Thicker

Bleed

Thicker

Column: DB-WAX 30 m X 0.25 mm X 0.25 μ m

Carrier: Helium at 25.4 cm/sec measured at 45°C

Oven: 45°C for 2 min

45 to 250°C at 3°C/min

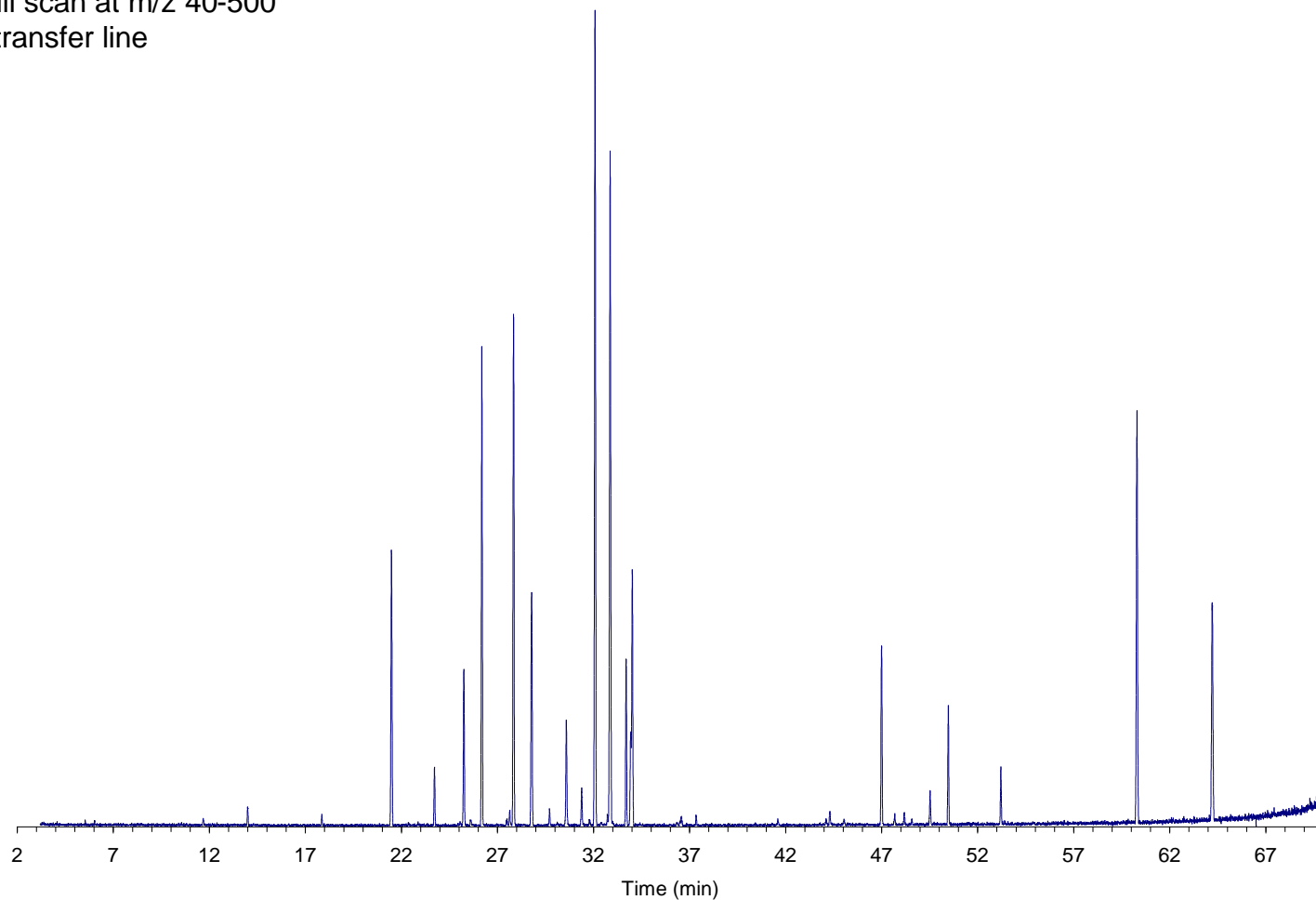
250°C for 34 min

Injector: Split 1:30, 250°C

1 μ L of 1:35 Oil in Acetone

Detector: MSD full scan at m/z 40-500

250°C transfer line



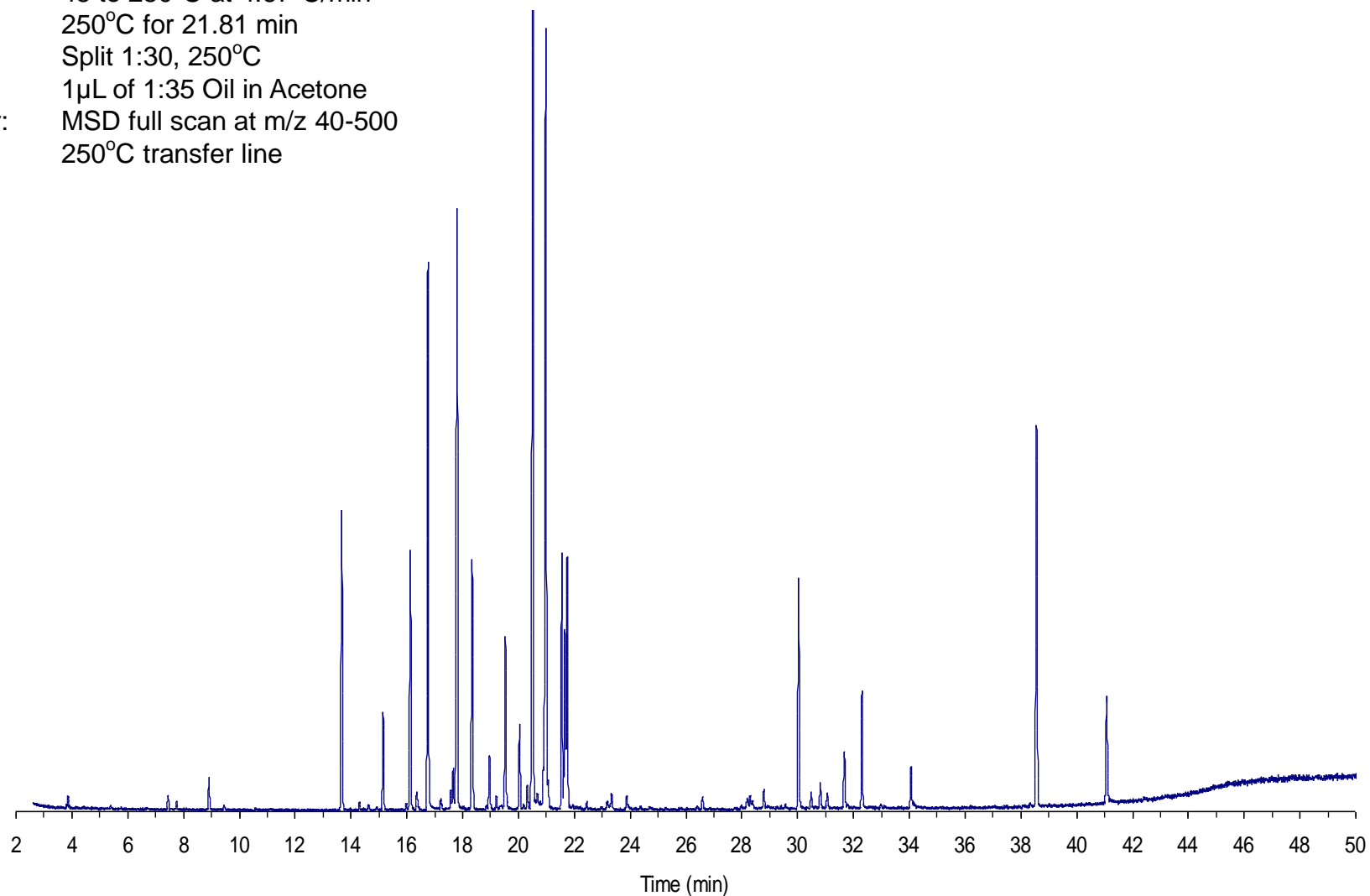
Column: DB-WAX 20m X 0.18mm X 0.18um

Carrier: Helium at 26.3 cm/sec measured at 45°C

Oven: 45°C for 1.28 min
45 to 250°C at 4.67°C/min
250°C for 21.81 min

Injector: Split 1:30, 250°C
1µL of 1:35 Oil in Acetone

Detector: MSD full scan at m/z 40-500
250°C transfer line



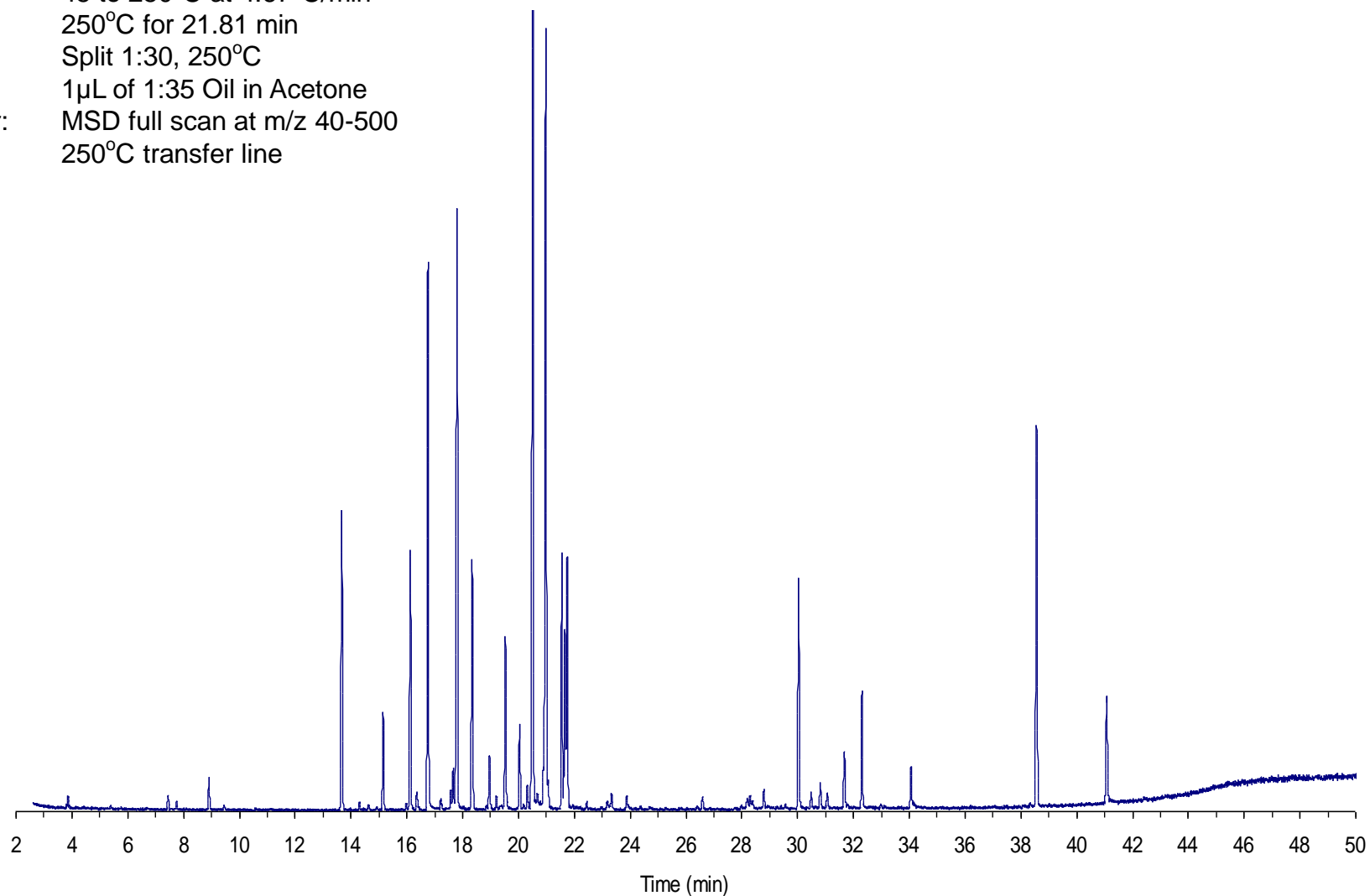
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250°C for 21.81 min

Injector: Split 1:30, 250°C
1µL of 1:35 Oil in Acetone

Detector: MSD full scan at m/z 40-500
250°C transfer line



Conclusions:

Understand the Sample

Is it volatile and thermally stable enough to chromatograph by GC?

Try to match polarity – **oil and water don't mix!**

Look for unique characteristics of compounds and match them to a phase

If you have the correct selectivity, change the dimensions to improve resolution – **consider a smaller ID**

If you need better peak shape for difficult compounds, try the '**UI**' version

Look for available information for a particular application

Call Tech Support!

References:

GC Column Selection Guide: [5990-9867EN](#)

Integrated Particle Trap PLOT columns: [5991-1174EN](#)

ScanView: Application Database

<https://community.agilent.com/docs/DOC-2118-software-supported-method-development-the-scanview-program>

Contact Agilent Chemistries and Supplies Technical Support



1-800-227-9770 Option 3, Option 3:

Option 1 for GC/GCMS Columns and Supplies

Option 2 for LC/LCMS Columns and Supplies

Option 3 for Sample Preparation, Filtration and QuEChERS

Option 4 for Spectroscopy Supplies

Available in the USA 8-5 all time zones



gc-column-support@Agilent.com

lc-column-support@agilent.com

spp-support@agilent.com

spectro-supplies-support@agilent.com