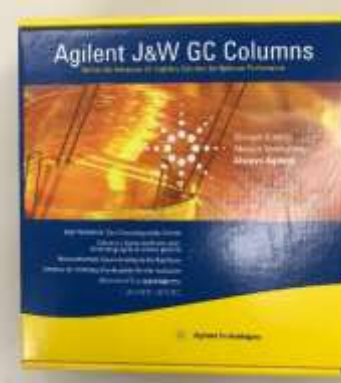


# Finding the Perfect Match: Practical Advice on GC Column Selection

Alexander Ucci  
Online Application Engineer  
September 12, 2019



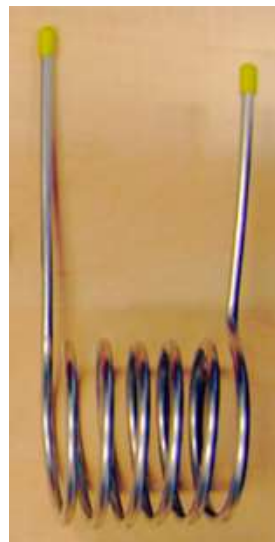
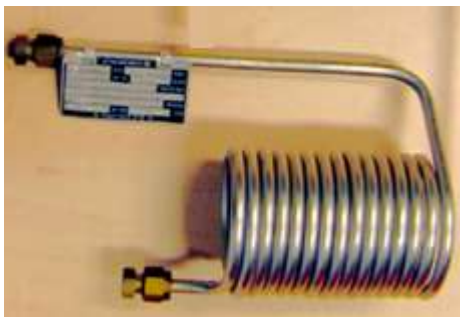
# Things to Consider when Choosing a Column

- Is it volatile enough to chromatograph by GC?
- Is it a gas or a liquid?
- How are we getting the sample injected?
- What is the sample matrix?
  - Can we do sample clean up?
- Is it an established method?
- What do we know about the analytes?
- **What else *may* be present in the sample?**



# Some Column History

**1969:**



**1977:** First glass capillary columns



**1979:** First WCOT fused silica capillary columns

# Capillary Column Types

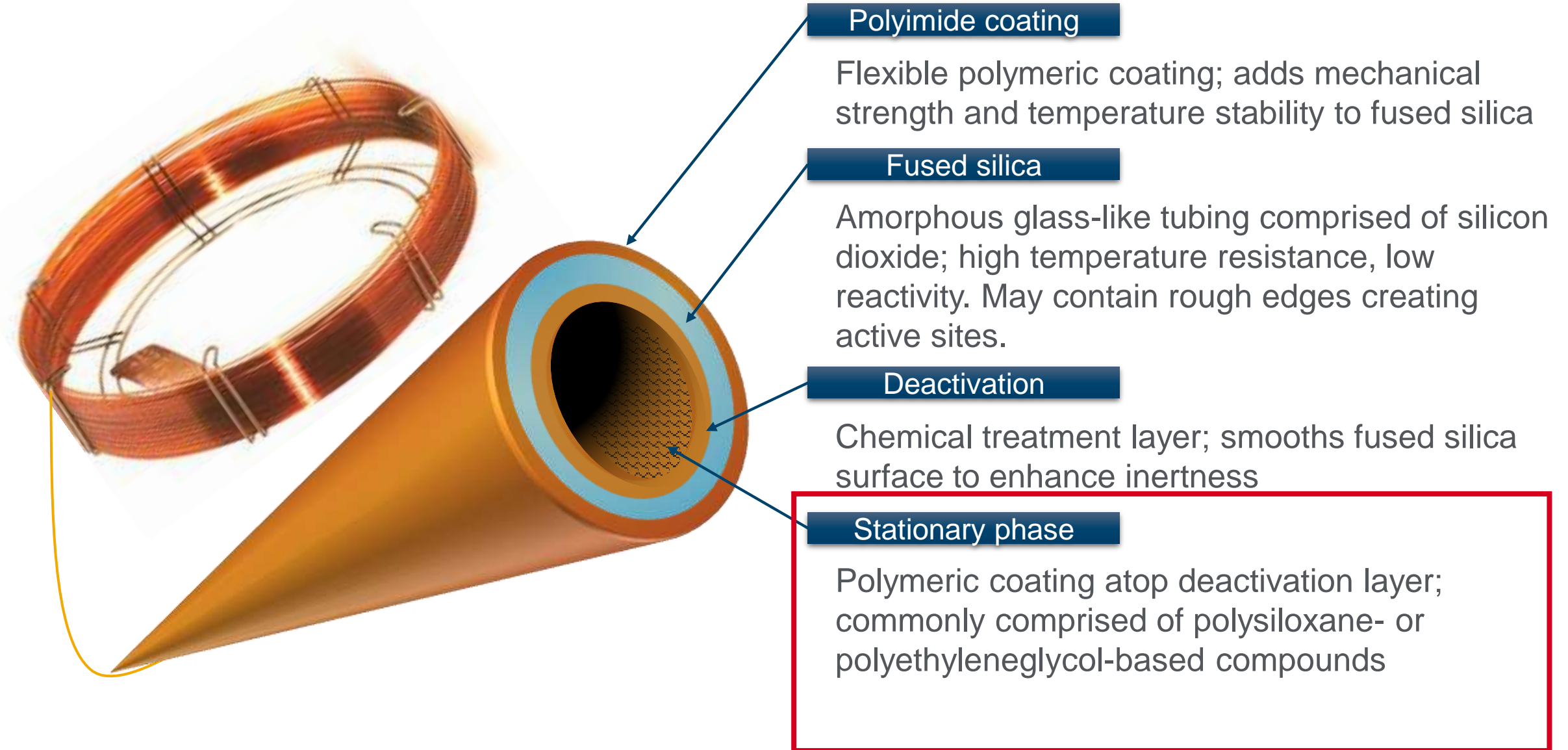
## Porous Layer Open Tube (PLOT)



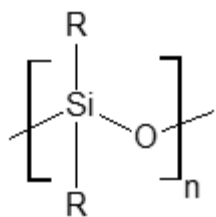
## Wall Coated Open Tube (WCOT)



# Column Construction



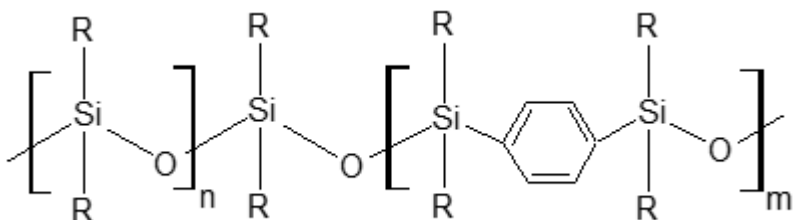
# Stationary Phase Polymers



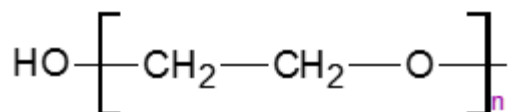
Siloxane

R= methyl, phenyl, cyanopropyl, trifluoropropyl

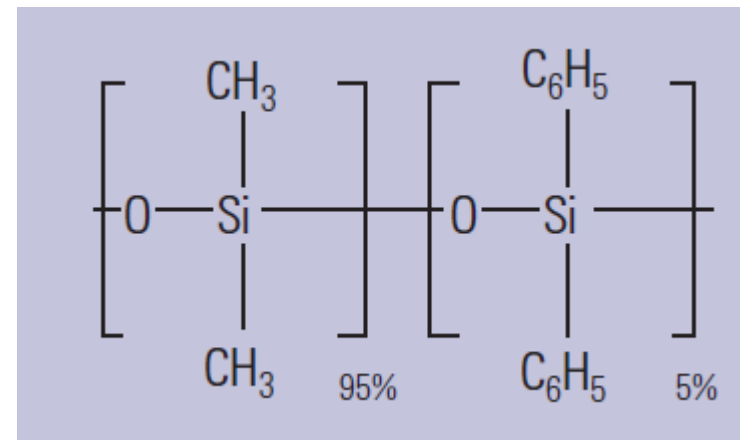
% = number of sites on silicon atoms occupied



Siarylene backbone



Polyethylene glycol



Structure of Agilent J&W HP-5ms

5% phenyl / 95% methyl

# Polyethylene Glycol Phases



- Less stable than polysiloxanes
- Unique separation characteristics
- Popular WAX phases from Agilent
  - Most inert: Agilent J&W DB-WAX UI
  - High temperatures: Agilent J&W DB-HeavyWAX
  - Application specific: Agilent J&W DB-FATWAX UI

# Another New Column: J&W DB-HeavyWAX

## The WAX column you've been waiting for!

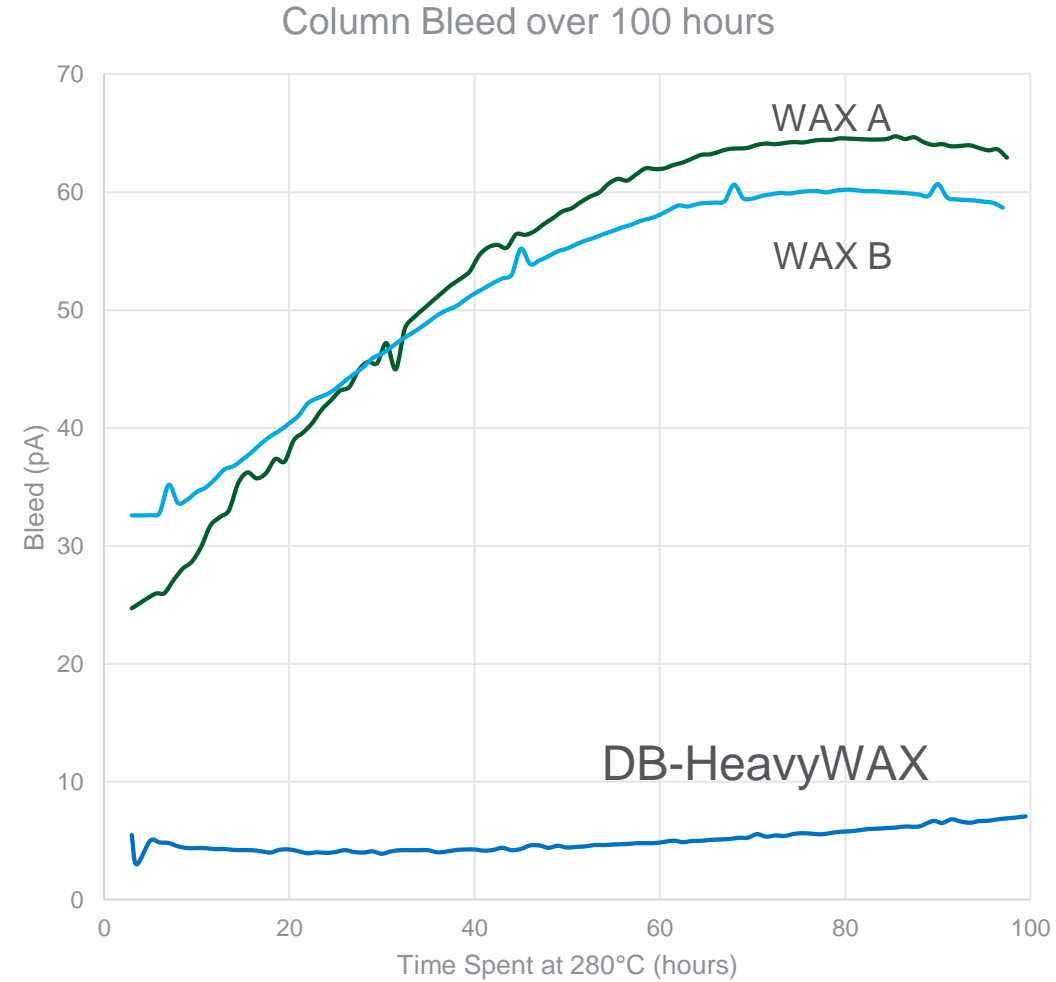
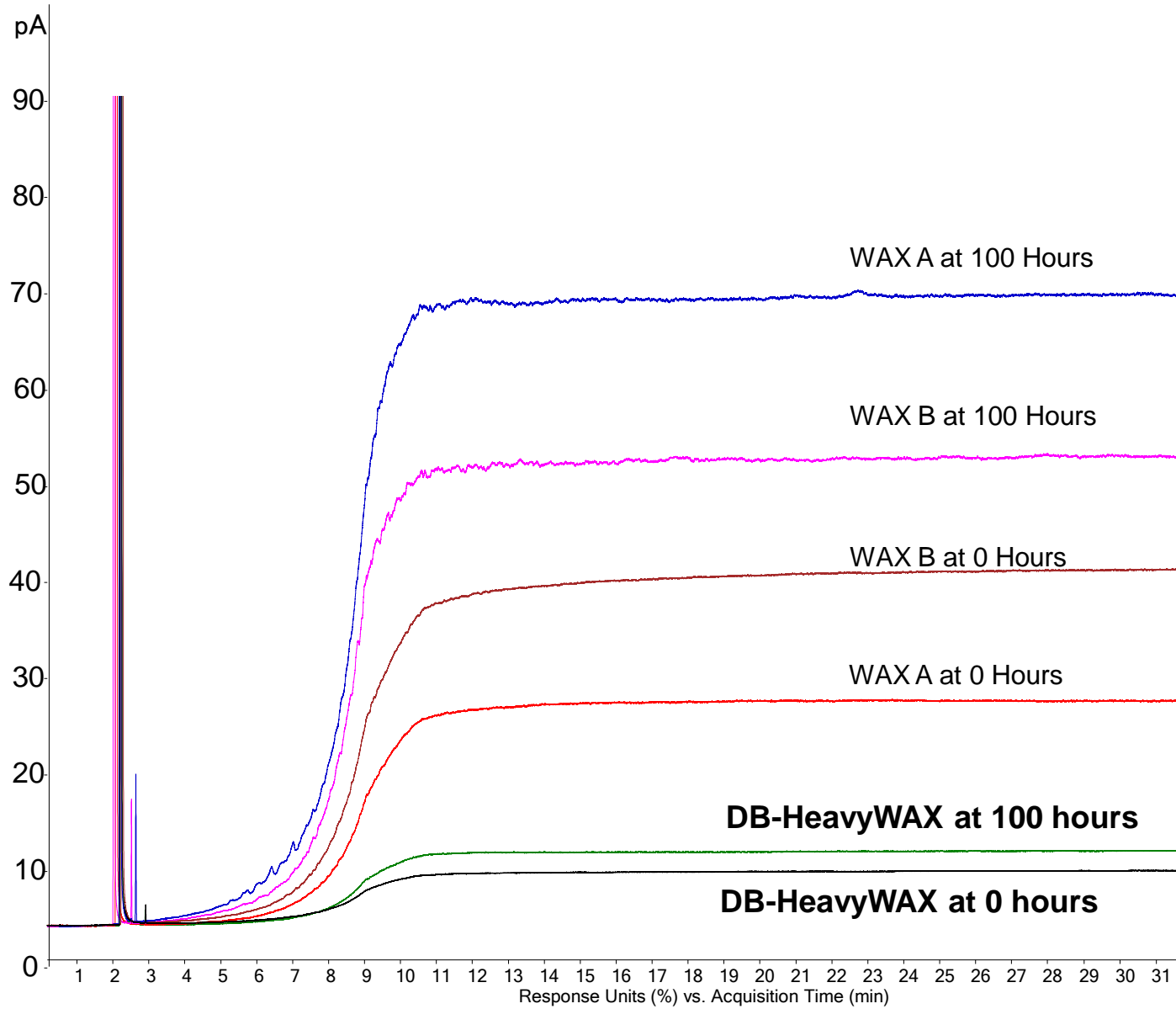
- Increase temperature range
  - 280 °C isothermal
  - 290 °C programmed
- Increased thermal stability
  - Stable retention times
  - Consistent peak order
- Lower bleed
  - Greater sensitivity for “heavier” compounds
  - Increased analyte range
  - Can decrease analysis time
  - Safely bake out column



[www.agilent.com/chem/db-heavywax](http://www.agilent.com/chem/db-heavywax)



# Bleed Summary at 280°C Over 100 Hours



# JW Column Portfolio- DB, HP, CP, VF

Low Polarity			Mid Polarity			High Polarity		
CP-Sil 2	DB & HP-1MS UI	DB & HP-5MS UI	DB-XLB	DB-225MS	DB-ALC1	HP-88	DB-WAX	DB-WAX UI
DB-MTBE	DB & HP1-MS	DB & HP5-MS	VF-XMS	DB-225	DB-Dioxin	CP-Sil 88	DB-WAX ETR	DB-HeavyWAX
CP-Select CB MTBE	VF-1MS	VF-5MS	DB-35MS UI	CP-Sil 43 CB	DB-200	DB-23	HP-INNOWax	DB-FATWAX UI
	DB & HP-1	DB & HP-5	DB & VF-35MS	VF-1701 MS	VF-200MS	VF-23 MS	VF-WAX MS	
	CP-Sil 5 CB	CP-Sil 8 CB	DB & HP-35	DB-1701	DB-210		CP-WAX 57 CB	
	Ultra 1	Ultra 2	DB & VF-17MS	CP-Sil 19 CB	DX-4		DB & HP-FFAP	
	DB-1HT	VF-DA	DB-17	HP-Blood Alcohol			DB-WAX FF	
	DB-2887	DB-5.625	HP-50+	DB-ALC2			CP-FFAP CB	
	DB-Petro/PONA	DB & VF-5HT	DB-17HT	DX-1			CP-WAX 58 FFAP CB	
	CP-Sil PONA CB	CP-Sil PAH CB	DB-608				CP-WAX 52 CB	
	DB-HT SimDis	Select Biodiesel	DB-TPH				CP-WAX 51	
	CP-SimDis	SE-54	DB-502.2				CP-Carbowax 400	
	CP-Volamine		HP-VOC				Carbowax 20M	
	Select Mineral Oil		DB-VRX				HP-20M	
	HP-101		DB-624				CAM	
	SE-30		VF-624MS				CP-TCEP	
			CP-Select 624 CB					
			DB-1301					
			VF-1301MS					
			CP-Sil 13 CB					

Agilent J&W has over 50 different stationary phase offerings

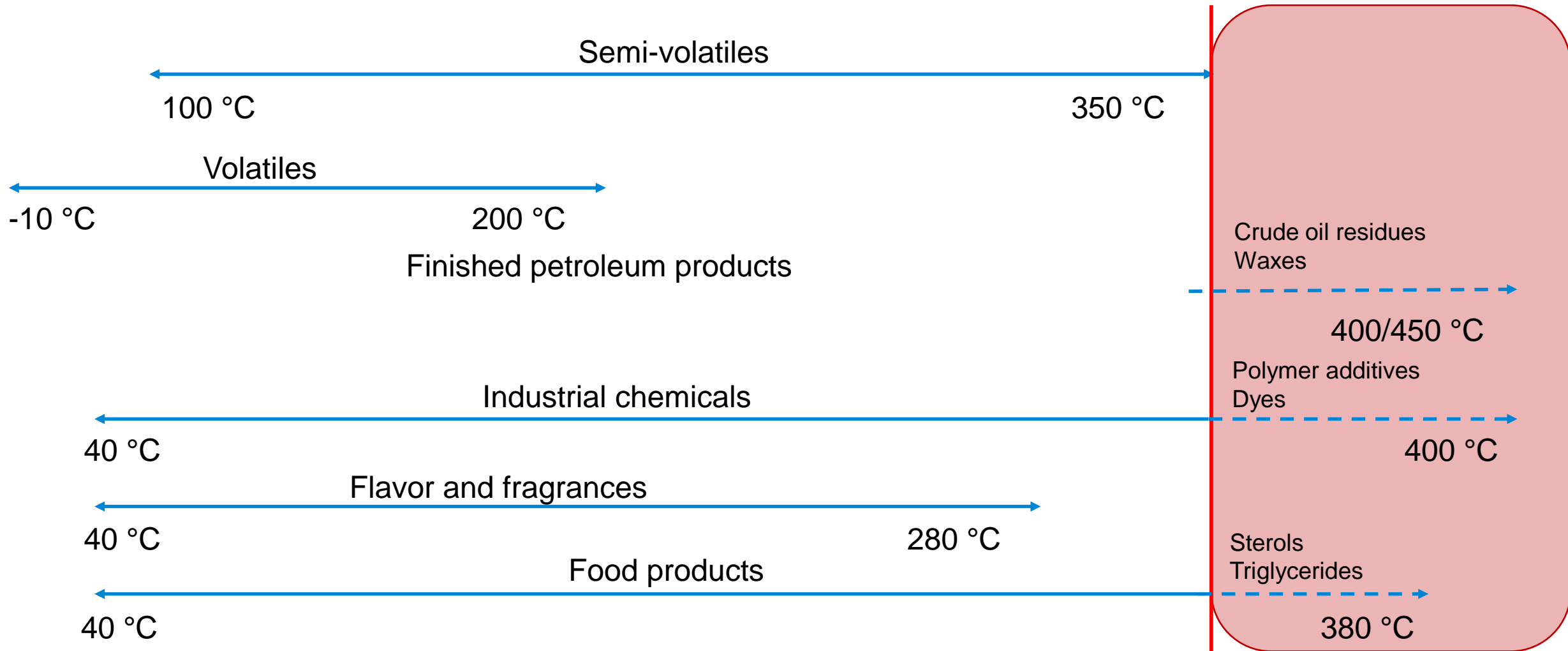
# Specialty Phases

Columns developed for particular applications

- Examples:
  - DB-8270D UI
  - DB-624 UI <467>
  - DB-MTBE
  - DB-TPH
  - DB-BAC1 UI & DB-BAC2 UI
  - DB-HT SimDis
  - Select Low Sulfur
  - CP-Volamine
  - Select PAH
  - DB-EUPAH
  - DB-CLP1 & DB-CLP2
  - DB-Select 624 UI <467>
  - CP-LowOx
  - Select Permanent Gases
  - And more!



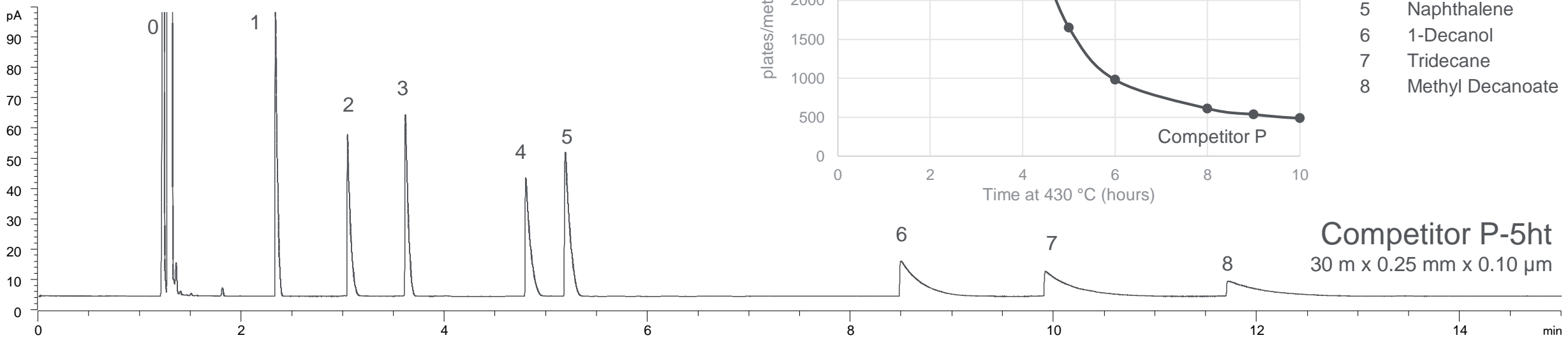
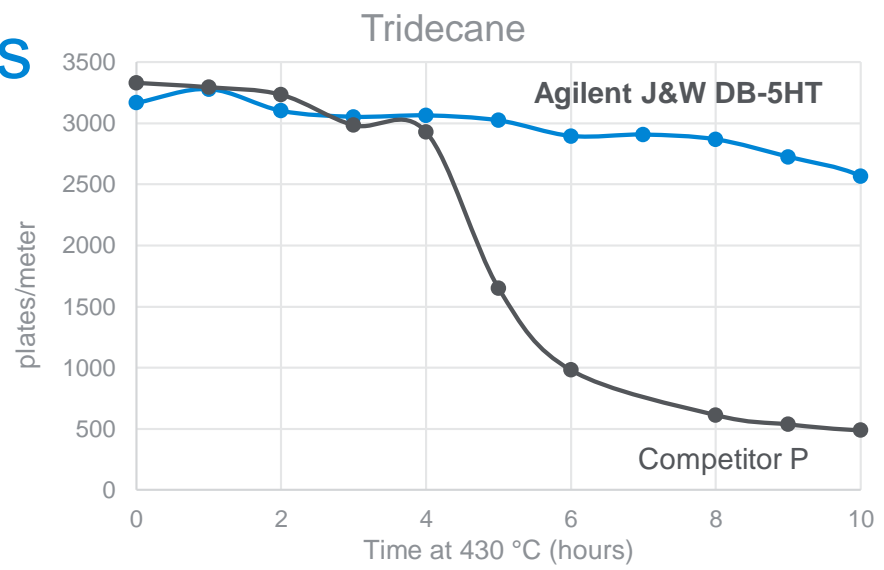
# High Temperature Applications



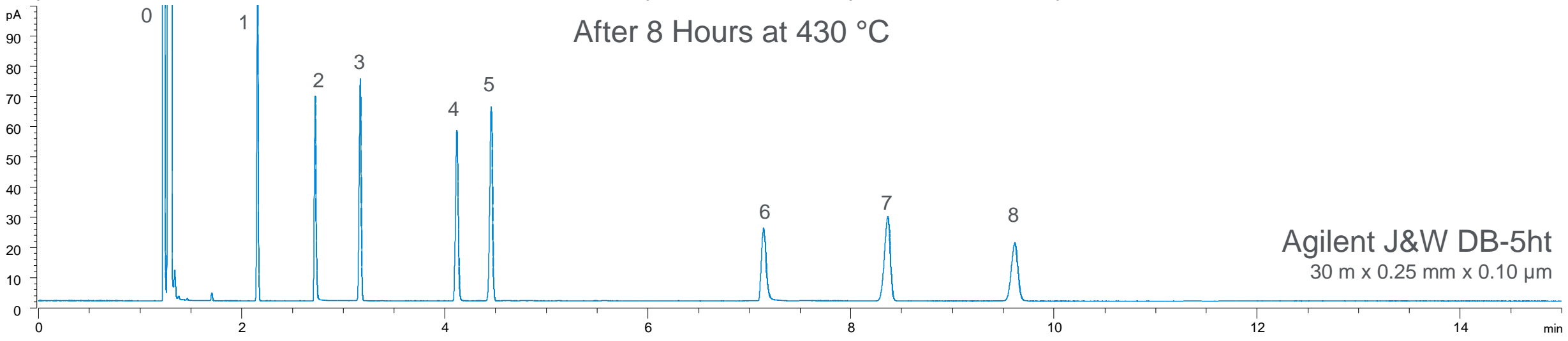
**If you aren't going above 350 °C, you don't need a high temperature column**

# Agilent's High Temperature Columns

Peak	Name
0	Methane
1	Decane
2	1-Octanol
3	2,6-Dimethylphenol
4	2,6-Dimethylaniline
5	Naphthalene
6	1-Decanol
7	Tridecane
8	Methyl Decanoate

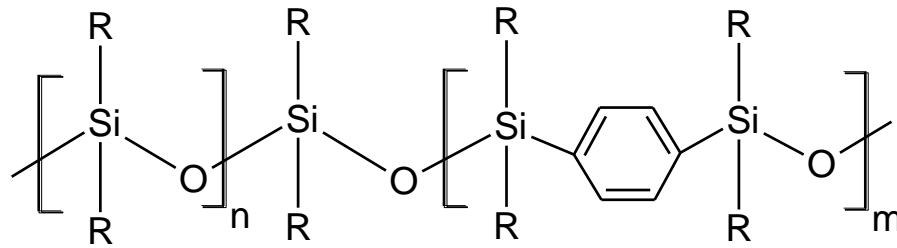


After 8 Hours at 430 °C



# Low Bleed Phases

- Phases tailored to “mimic” currently existing polymers  
Examples: DB-5ms, DB-35ms, DB-17ms, VF-1701ms

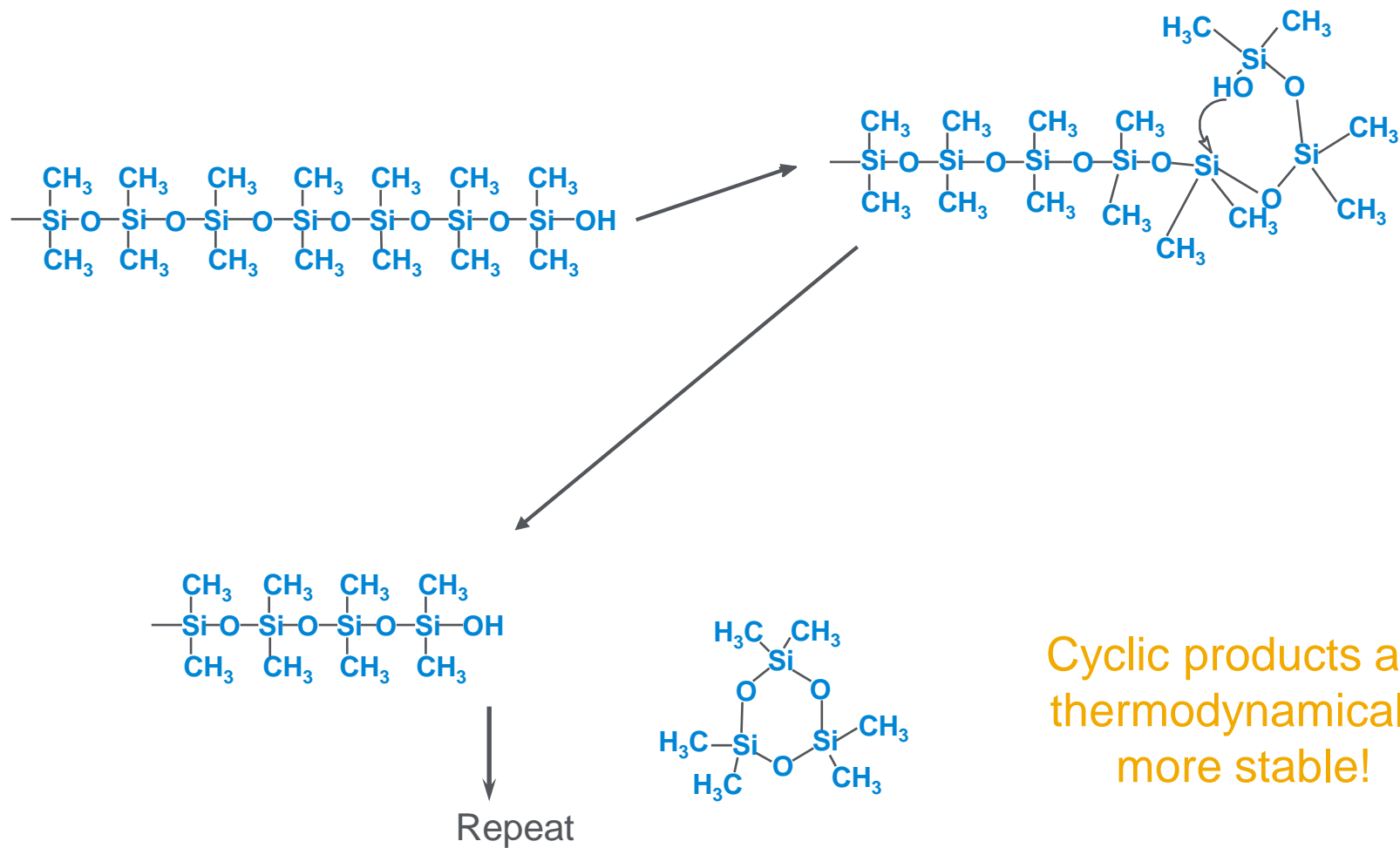


Siarylene backbone

- New phases unrelated to any previously existing polymers  
Examples: DB-XLB
- Optimized manufacturing processes  
Examples: DB-1ms, HP-1ms, HP-5ms, VF-5ms

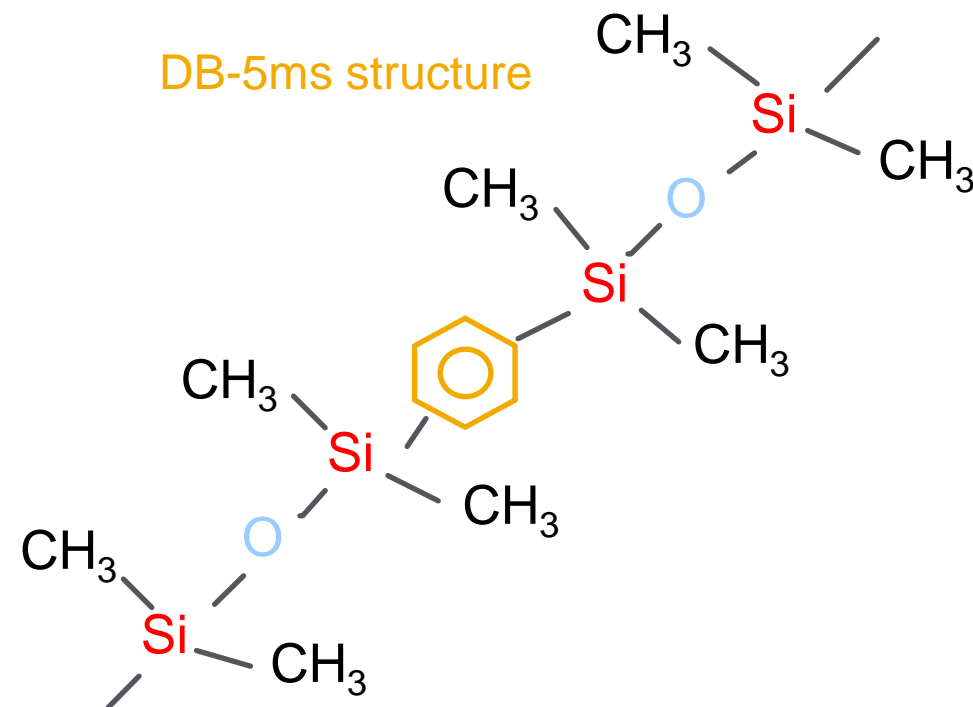
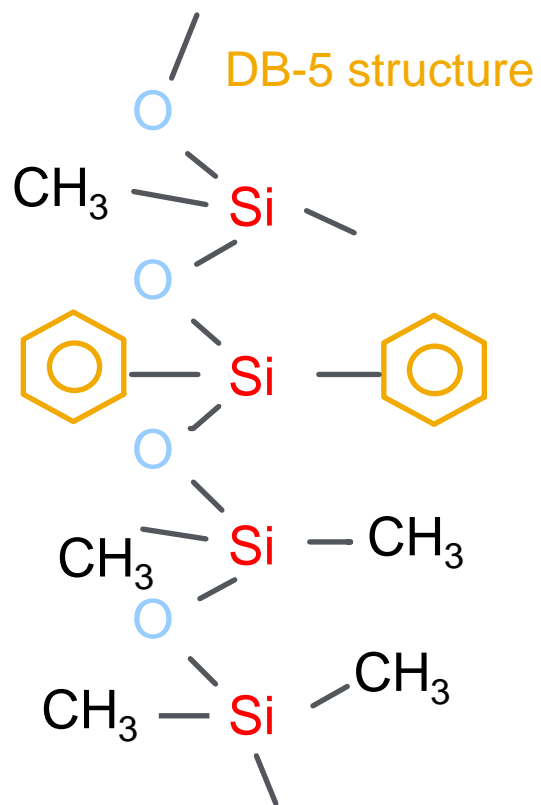
# What is Column Bleed?

“Back biting” mechanism of product formation



Cyclic products are thermodynamically more stable!

# Agilent J&W DB-5ms Structure



DB-5ms:

- Increased stability
- Different selectivity
- Optimized to match DB-5 as much as possible



# DB-5ms vs. DB-5 Selectivity

Solid line: Agilent J&W **DB-5ms**

**30 m x .25 mm id x .25 μm**

Dashed line: Agilent J&W **DB-5**

**30 m x .25 mm id x .25 μm**

Oven: 60 °C isothermal

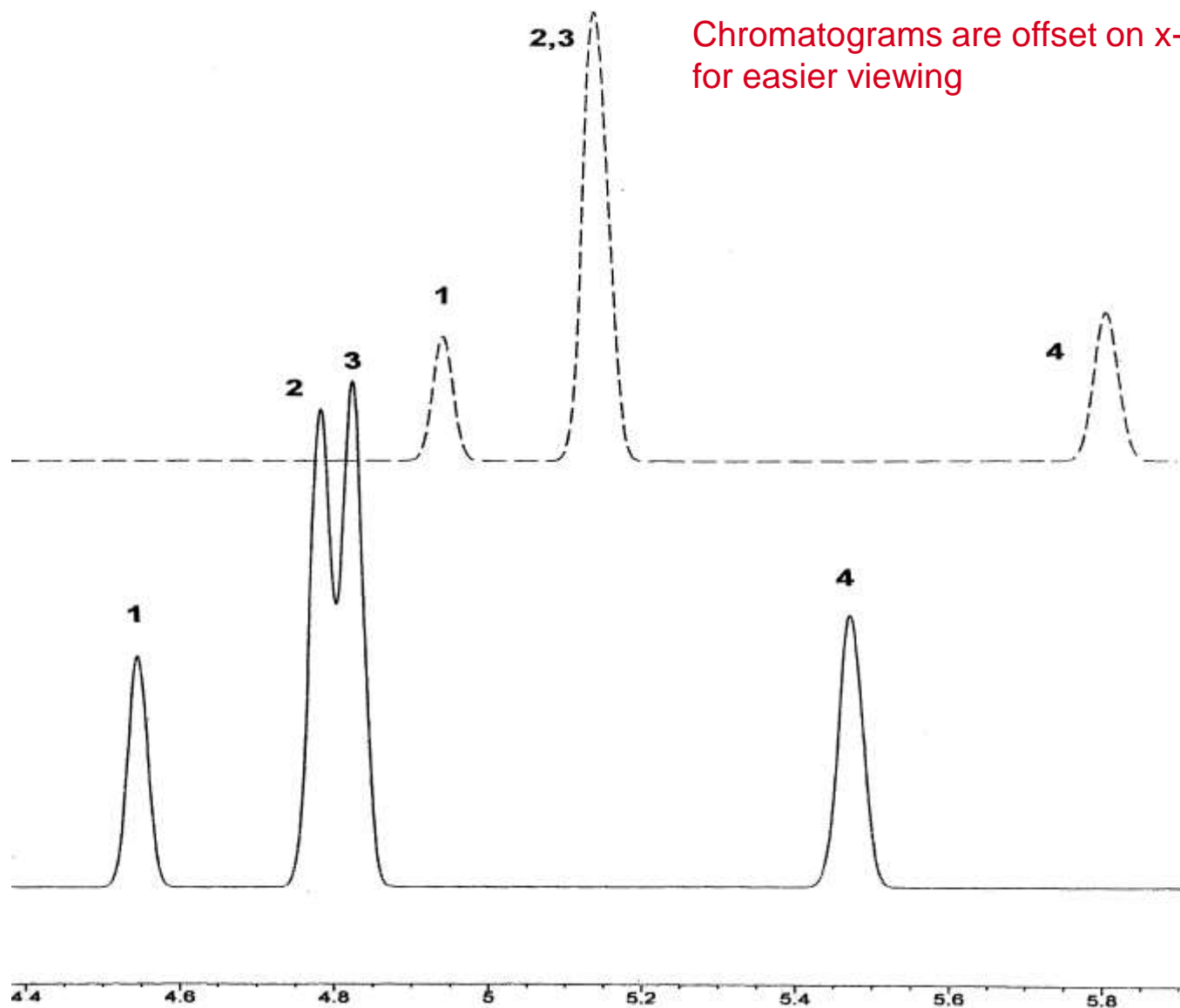
Carrier gas: H<sub>2</sub> at 40 cm/sec

1: Ethylbenzene

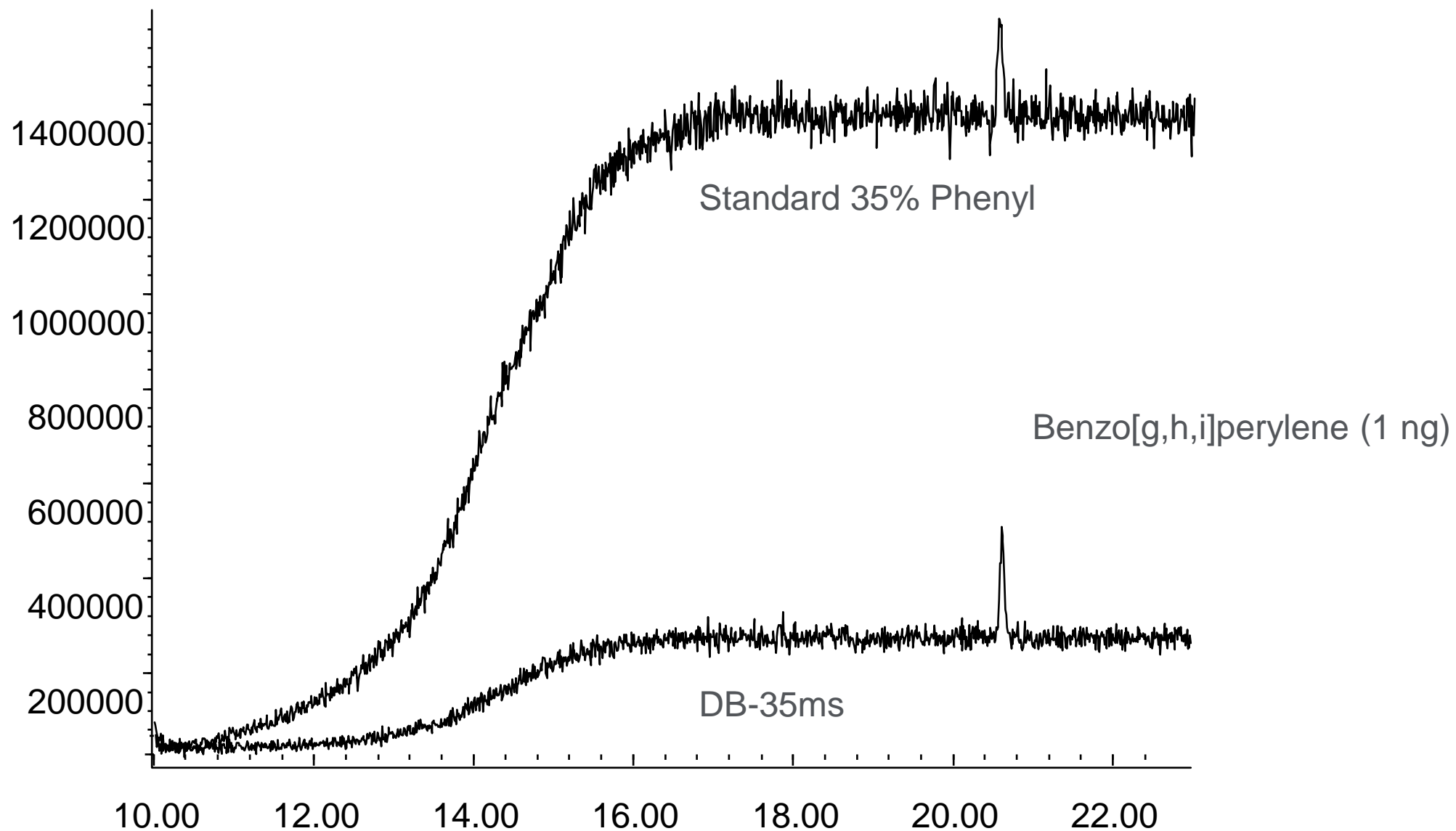
2: m-Xylene

3: p-Xylene

4: o-Xylene



# Comparison of Agilent J&W DB-35MS vs Standard DB-35



# Agilent Ultra Inert GC Columns

## Column inertness: what does it mean?

- Easier to describe “lack of inertness”
  - Peak tailing (reversible interaction)
  - Loss of compound all together (irreversible interaction)
- A high level of flow path inertness will produce peaks from active compounds that are not degraded and will look “normal”/symmetrical
- The negative effects the column has towards challenging compounds
  - Acids
  - Bases
  - Hydrogen bonding
  - i.e. 2,4-DNP, Endrin, etc.

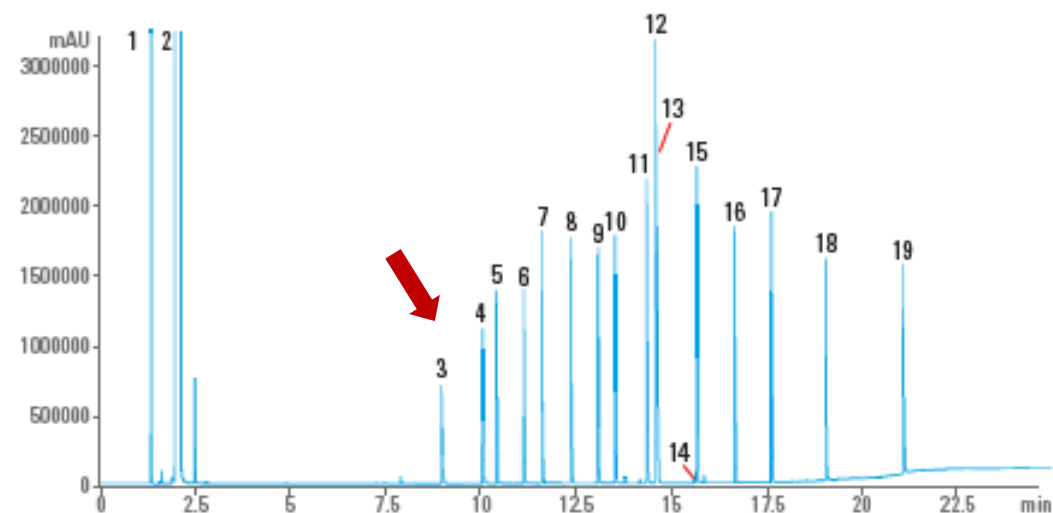
# DB-WAX Ultra Inert and Free Fatty Acid

Peak identification:

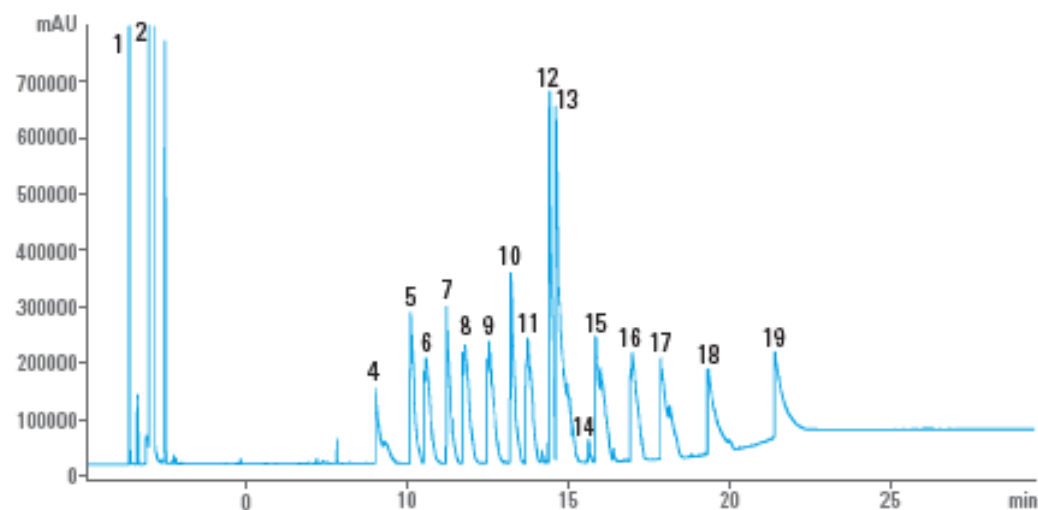
1. Methane
2. Acetone (solvent)
3. Acetic acid
4. Propionic acid
5. Isobutyric acid
6. Butyric acid
7. Isovaleric acid
8. Valeric acid
9. 4-Methylvaleric acid
10. Hexanoic acid
11. 4-Methylhexanoic acid
12. 2-Ethylhexanoic acid
13. Heptanoic acid
14. Pyruvic acid
15. Octanoic acid
16. Nonanoic acid
17. Decanoic acid
18. Undecylenic acid
19. Myristic acid (Tetradecanoic)

## Competitive comparison: free fatty acids

DB-WAX Ultra Inert GC column  
30 m x 0.25 mm id, 0.25 µm (p/n 122-7032UI)



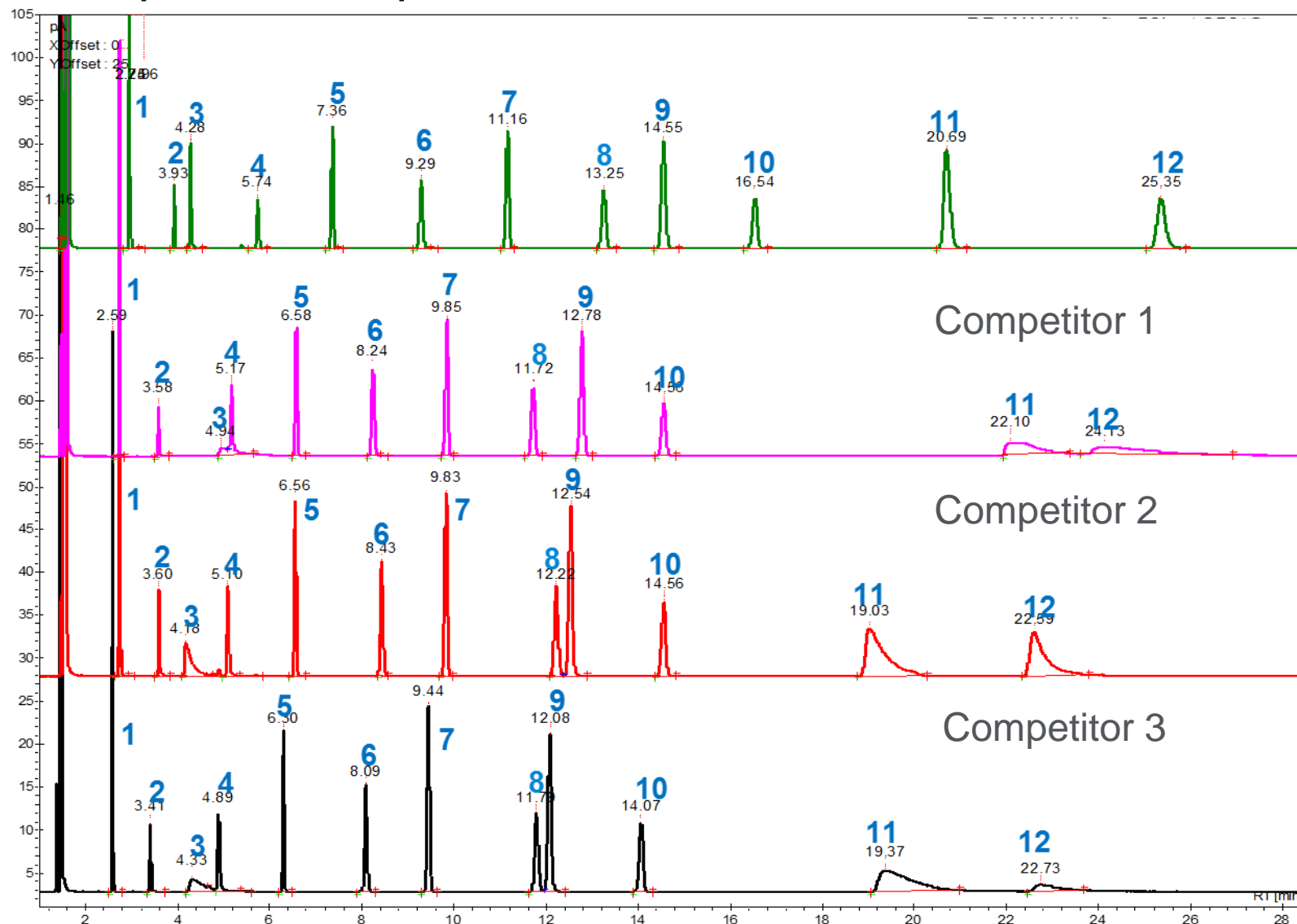
Standard WAX GC column  
30 m x 0.25 mm id, 0.25 µm



Application note: 5991-6709EN

# Let's Make a Better Wax Column: DB-WAX Ultra Inert

Competitor comparison DB-WAX UI test mix after 50 hours at 250 °C



## Compound I.D.

\*. Methane

1. 5-Nonanone

2. Decanal

3. Propionic Acid

4. Ethylene Glycol

5. Heptadecane

6. Aniline

7. Methyl Dodecanoate

8. 2-Chlorophenol

9. 1-Undecanol

10. Nonadecane

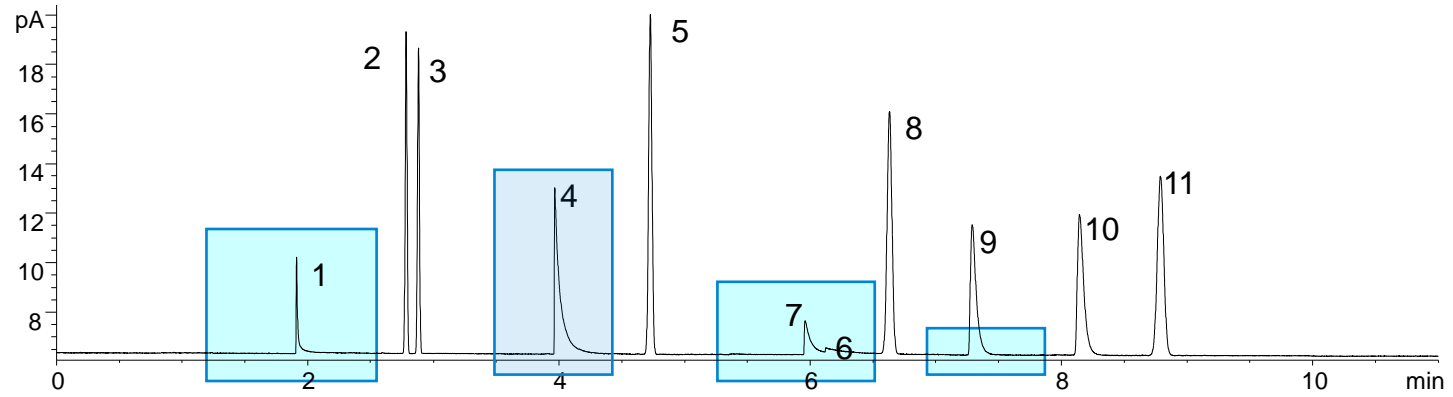
11. 2-Ethylhexanoic Acid

12. Ethyl Maltol

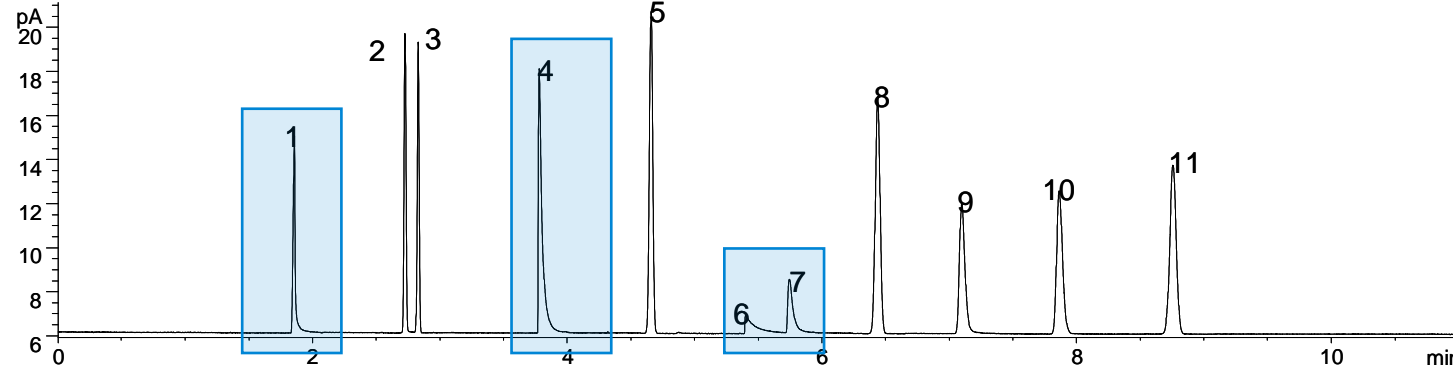
Application Note: 5991-6683EN

# Ultra Inert Test Mix – DB-5MS Ultra Inert v. Competitors

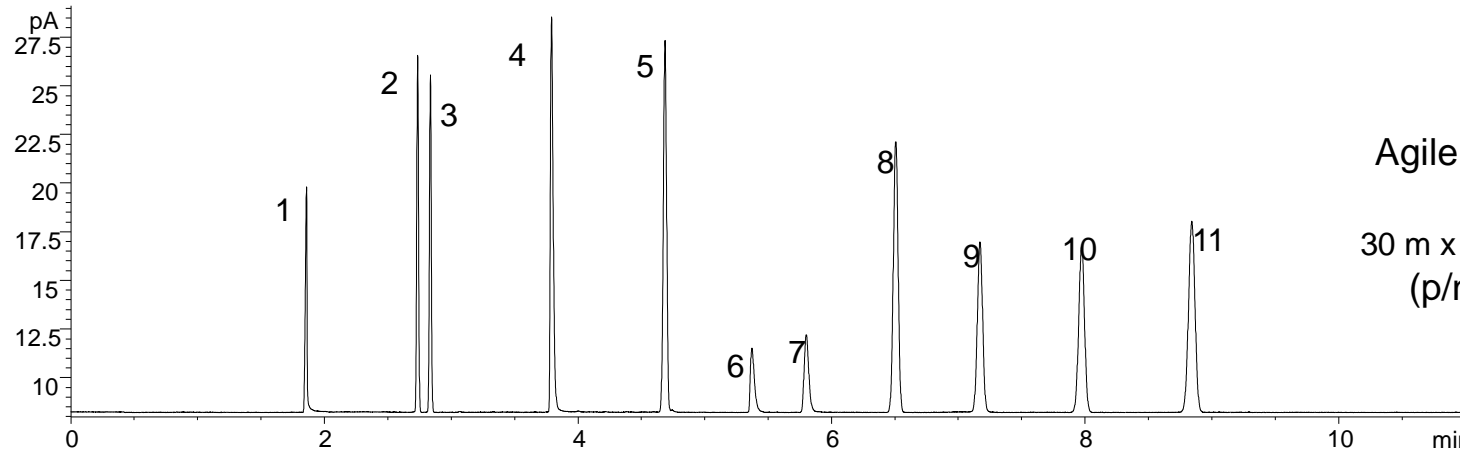
1. 1-Propionic acid
2. 1-Octene
3. n-Octane
4. 4-Picoline
5. n-Nonane
6. Trimethyl phosphate
7. 1,2-Pentanediol
8. n-Propylbenzene
9. 1-Heptanol
10. 3-Octanone
11. n-Decane



Competitor column



Competitor column



Agilent J&W DB-5ms  
Ultra Inert  
30 m x 0.25 mm x 0.25  $\mu$ m  
(p/n 122-5532UI)

# Why is Stationary Phase Type Important?

$$R_s = \frac{\sqrt{N}}{4} \left( \frac{k}{k+1} \right) \left( \frac{\alpha - 1}{\alpha} \right)$$

Influence on  $\alpha$

$$\alpha = \frac{k_2}{k_1}$$

$k_2$  = partition ratio of 2<sup>nd</sup> peak  
 $k_1$  = partition ratio of 1<sup>st</sup> peak

# Selectivity

- Relative spacing of the chromatographic peaks
- The result of all non-polar, polarizable and polar interactions that cause a stationary phase to be more or less retentive to one analyte than another





# Optimizing Selectivity ( $\alpha$ )

- Match analyte polarity to stationary phase polarity
  - “Like dissolves like”
- Take advantage of unique interactions between analyte and stationary phase functional groups

# Analyte Polarity

Nonpolar molecules - generally composed of only carbon and hydrogen and exhibit no dipole moment (Straight-chained hydrocarbons (n-alkanes))

Polar molecules - primarily composed of carbon and hydrogen but also contain atoms of nitrogen, oxygen, phosphorus, sulfur, or a halogen (Alcohols, amines, thiols, ketones, nitriles, organo-halides, etc. Includes dipole-dipole interactions and H-bonding)

Polarizable molecules - primarily composed of carbon and hydrogen, but also contain unsaturated bonds (Alkenes, alkynes and aromatic compounds)

# Selectivity Interactions

- Dispersion
- Dipole
- Hydrogen bonding

# Dispersion Interaction ( $\Delta H_{\text{vap}}$ )

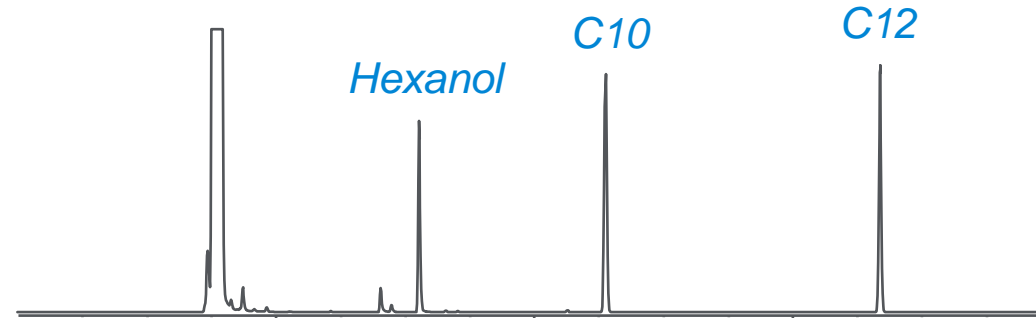
- Separation by differences in analyte heat of vaporizations (  $\Delta H_{\text{vap}}$  )
- Heat necessary to convert a liquid into a gas (at the same temperature)



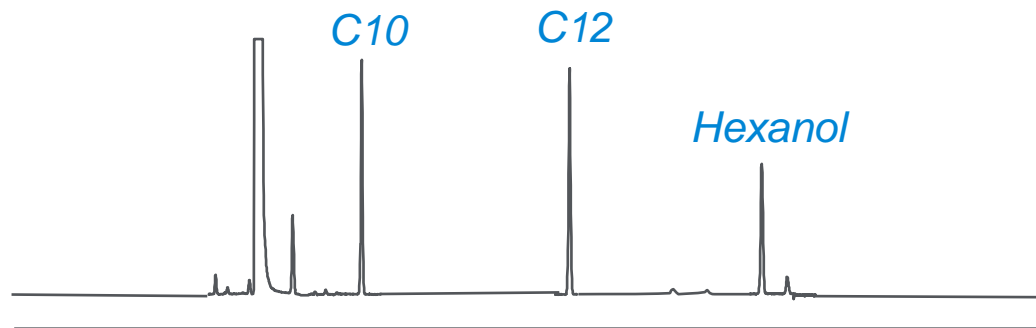
# Dispersion Interaction

## Solubility And Retention

Hexanol 158°C  
Decane 174°C  
Dodecane 216°C



100% Methyl  
(non-polar)



100% PEG  
(polar)

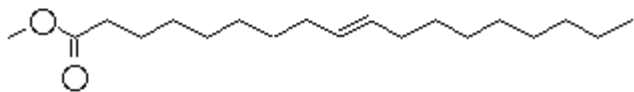
30 m x 0.32 mm id, 0.25  $\mu$ m  
He at 35 cm/sec  
50–170 °C at 15 °C/min

# Dispersion Interaction ( $\Delta H_{\text{vap}}$ )

Vapor pressure: good approximation

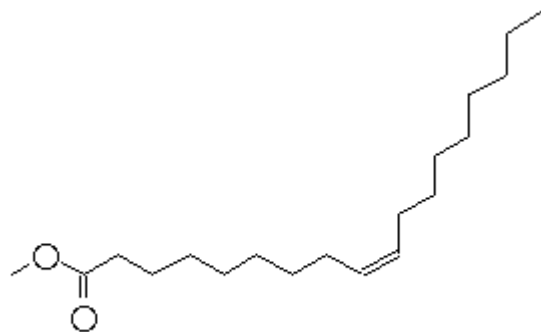
Boiling point: poor approximation

# Dipole Interaction



C18:1 (Methyl *trans*-9-octadecenoate)

B.Pt. 186°C



C18:1 (Methyl *cis*-9-octadecenoate)

B.Pt. 186°C

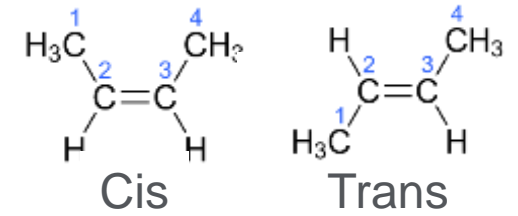
Smaller differences require a stronger dipole phase

# DB-FastFAME

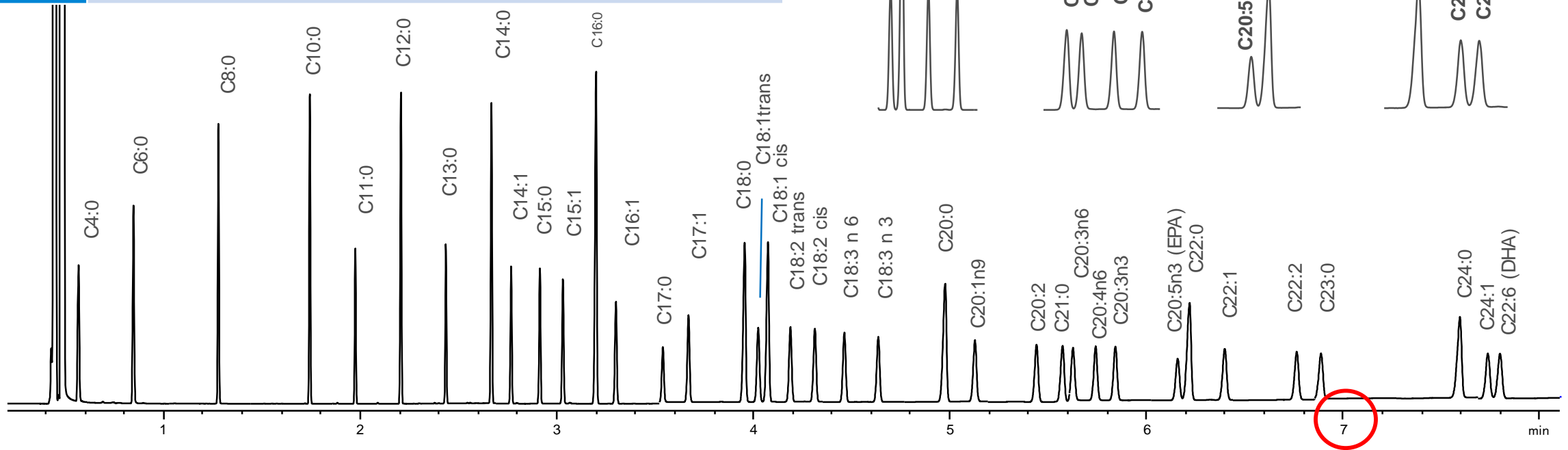
20 m x 0.18 mm x 0.20 μm

<b>Column</b>	Agilent J&W DB-FastFAME, 20 m x 0.18 mm, 0.20 μm
<b>Gas</b>	Hydrogen, 28 psi, constant pressure mode
<b>Inlet</b>	Split/splitless, 250 °C, split ratio 50:1
<b>Oven</b>	80 °C (0.5 min), 65 °C/min to 175 °C, 10 °C/min to 185 °C (0.5 min), 7 °C/min to 230 °C
<b>FID</b>	280 °C, Hydrogen: 40 mL/min; Air: 400 mL/min; make-up gas: 25 mL/min.
<b>Injection</b>	1 μL

Strong interaction between cis isomers and the dipoles of the cyano propyl ligands. That allows the trans to elute after the cis isomers.



$R_s \geq 1.95$  for cis/trans isomers

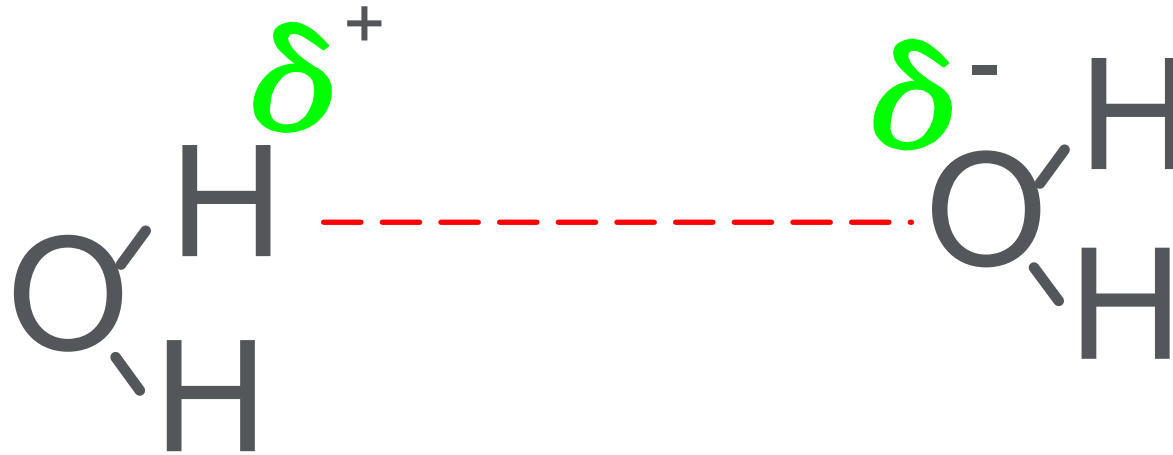


Application note: 5991-8706EN



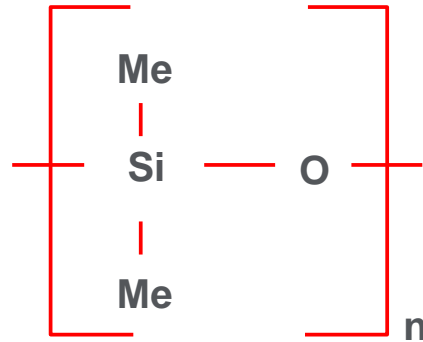
# Hydrogen Bonding Interaction

Dipole-Dipole interaction with hydrogen bound to oxygen or nitrogen interacting with an oxygen or nitrogen-atom



# Nonpolar Phases

Characterized by 100% polydimethylsiloxanes such as HP-1, DB-1, DB-1ms, HP-1ms, VF-1ms, CP-Sil 5 CB

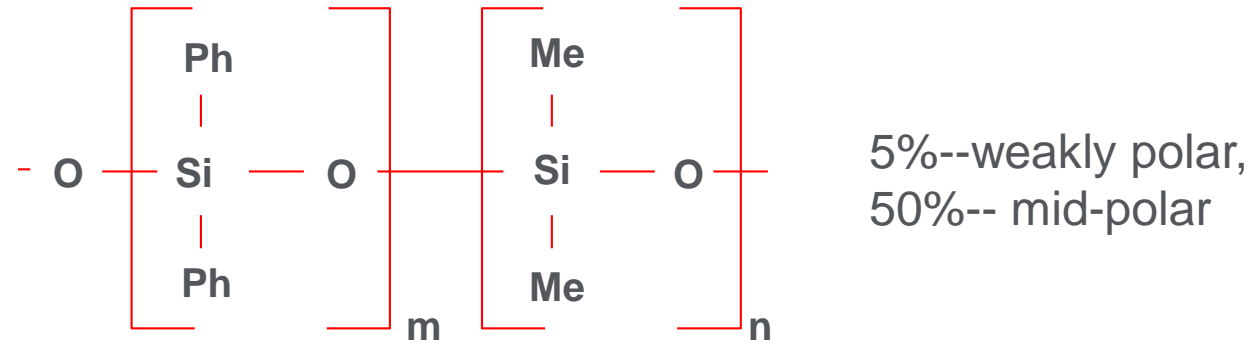


Separation mechanisms:

- Dispersion only

# Polarizable Phases

Typified by phenyl substituted siloxanes, substituted at 5–50% (HP-5, HP-5ms, DB-35, DB-35ms, DB-17, DB-17ms)

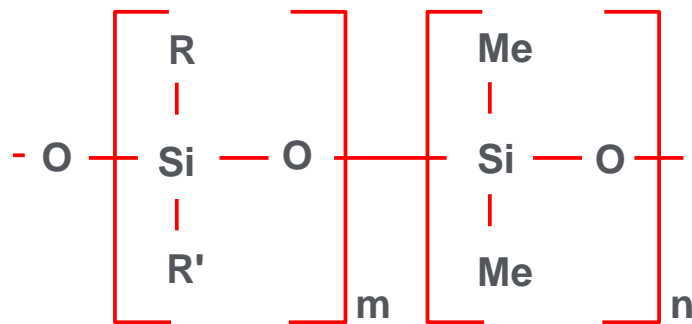


Separation mechanisms:

- Dispersion
- Inducible dipole at phenyl groups

# Strong Dipole Phases

Typified by cyanopropyl or trifluoropropyl substituted siloxanes, substituted 6–50% (DB-1701, DB-1301, DB-200, DB-23, DB-225)



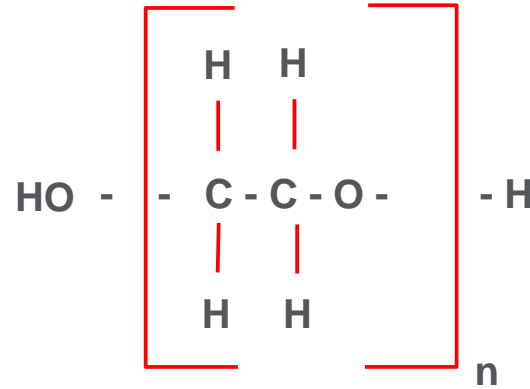
R = cyanopropyl or trifluoropropyl  
R' = phenyl or methyl

## Separation mechanisms:

- Dispersion
- Inducible dipole at phenyl groups
- Strong permanent dipole
- Hydrogen bonding

# Hydrogen Bonding Phases

Typified by polyethylene glycol polymers (HP-Innowax, DB-WAX UI, DB-HeavyWAX, DB-FFAP, VF-WAXms, CP-WAX52CB...)



Separation mechanisms:

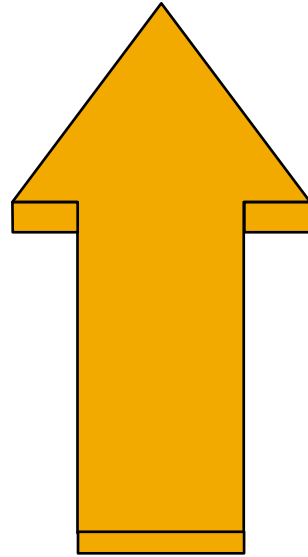
- Dispersion
- Strong permanent dipole
- Hydrogen bonding

# Selectivity

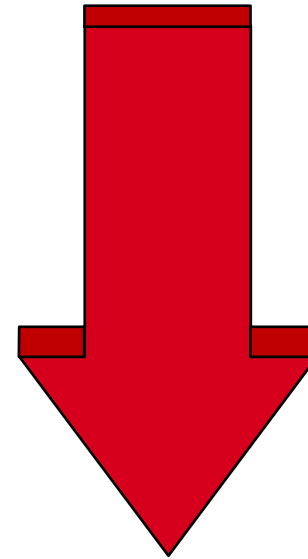
## Interaction strengths

Phase	Dispersion	Dipole	H Bonding
Methyl	Strong	None	None
Phenyl	Strong	None	Weak
Cyanopropyl	Strong	Very Strong	Moderate
Trifluoropropyl	Strong	Moderate	Weak
PEG	Strong	Strong	Moderate

# Polarity



Polarity



Stability  
Temperature Range

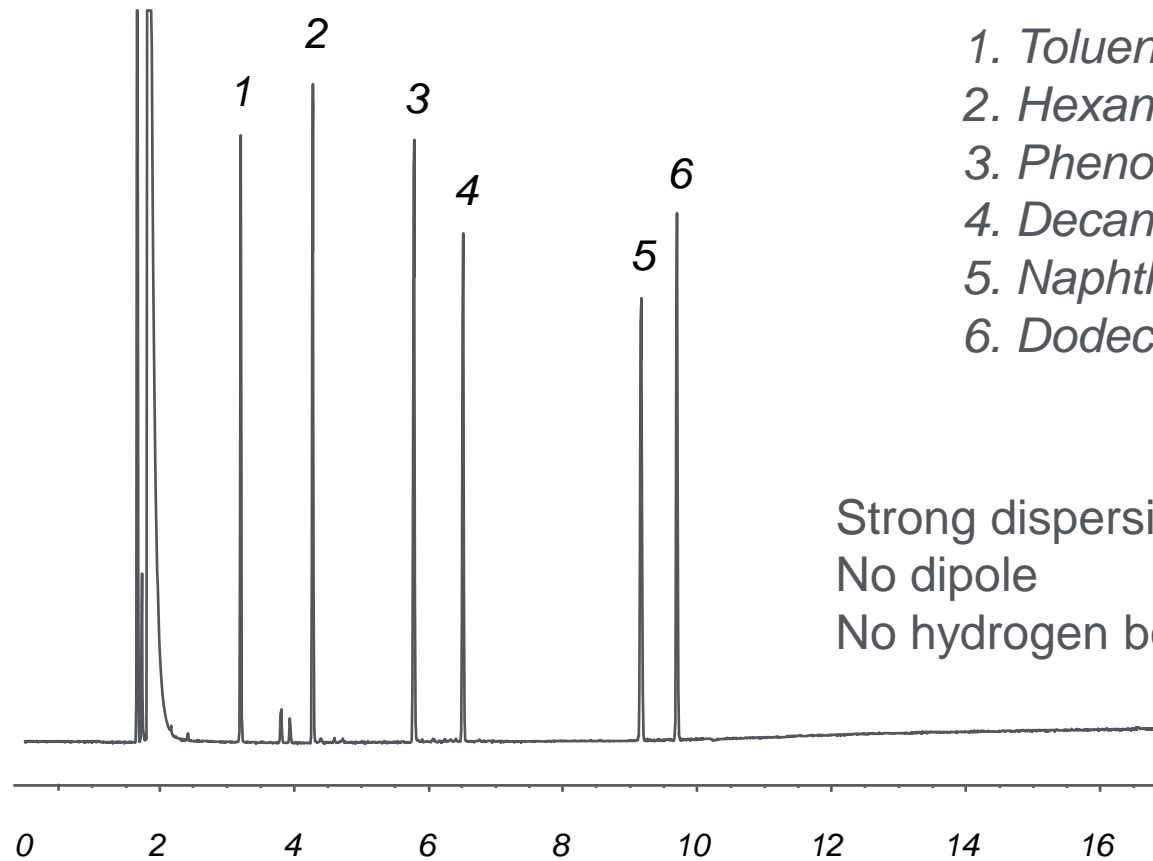


# Compounds & Properties

Compounds	Polar	Aromatic	Hydrogen Bonding	Dipole
Toluene	no	yes	no	induced
Hexanol	yes	no	yes	yes
Phenol	yes	yes	yes	yes
Decane	no	no	no	no
Naphthalene	no	yes	no	induced
Dodecane	no	no	no	no



# 100% Methyl Polysiloxane

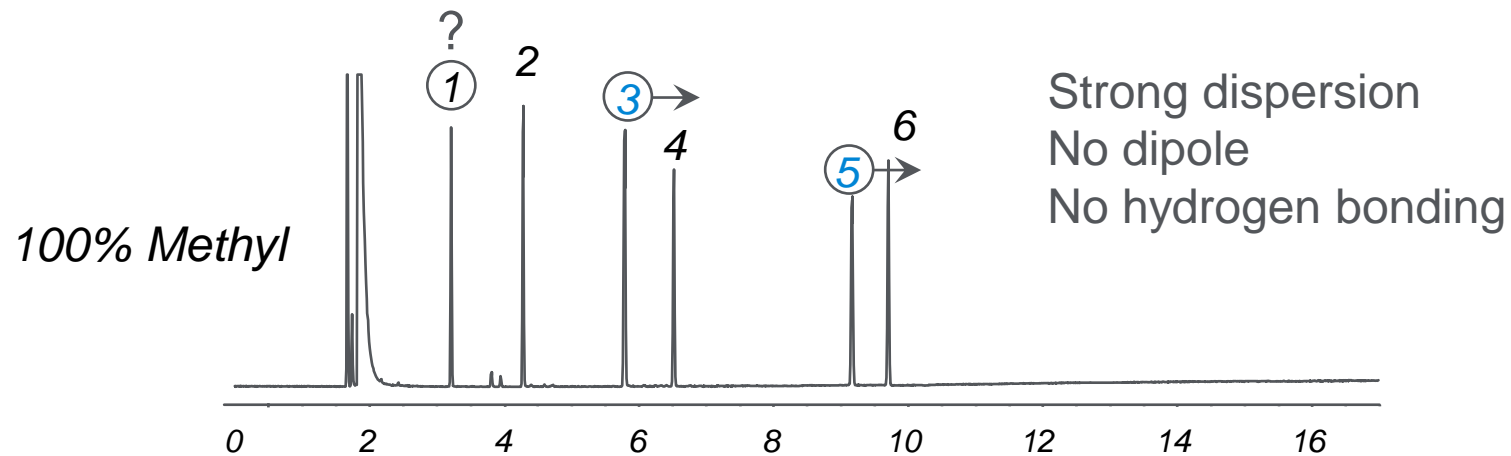
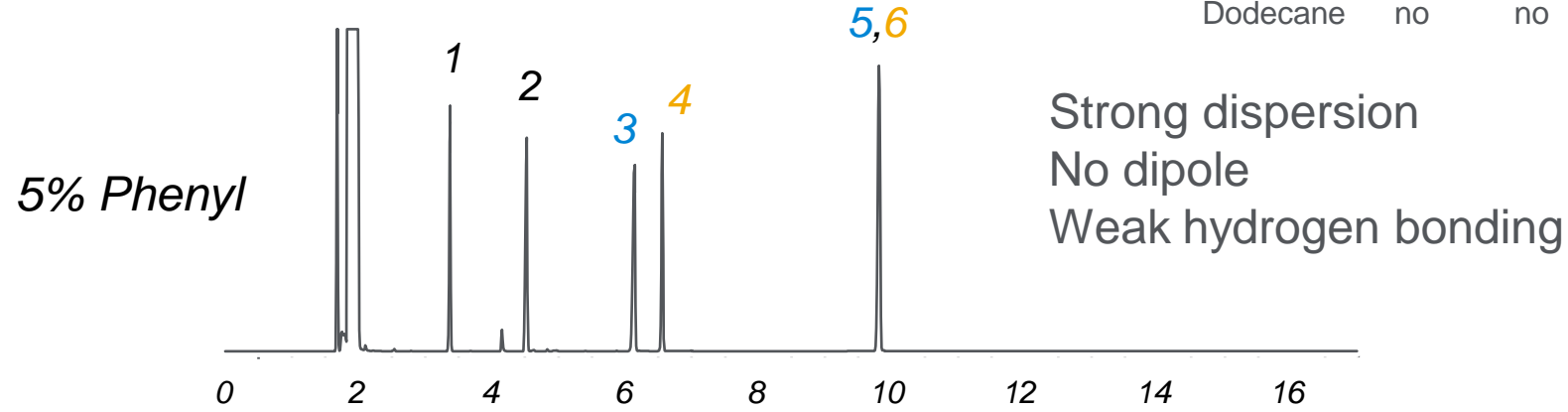


1. Toluene	110 °C
2. Hexanol	156 °C
3. Phenol	182 °C
4. Decane (C10)	174 °C
5. Naphthalene	218 °C
6. Dodecane (C12)	216 °C

Strong dispersion  
No dipole  
No hydrogen bonding

# 5% Phenyl

Compounds	Polar	Aromatic	Hydrogen Bonding	Dipole
Toluene	no	yes	no	induced
Hexanol	yes	no	yes	yes
Phenol	yes	yes	yes	yes
Decane	no	no	no	no
Naphthalene	no	yes	no	induced
Dodecane	no	no	no	no

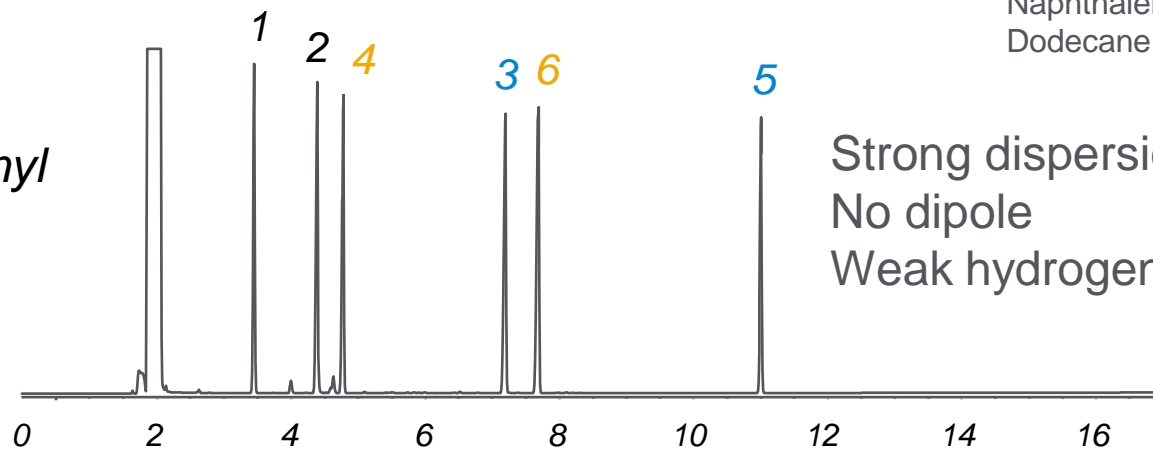


1. Toluene
2. Hexanol
3. Phenol
4. Decane (C10)
5. Naphthalene
6. Dodecane (C12)

# 50% Phenyl

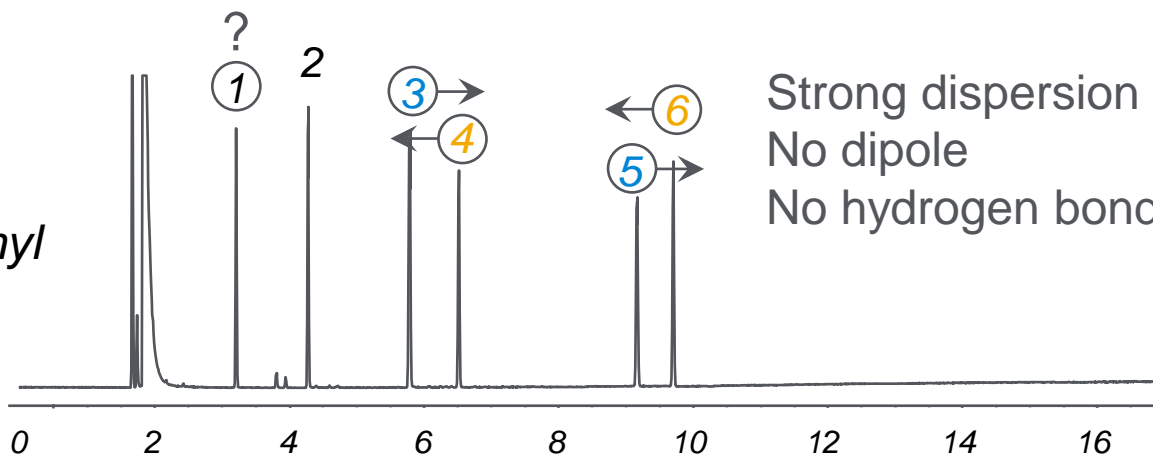
Compounds	Polar	Aromatic	Hydrogen Bonding	Dipole
Toluene	no	yes	no	induced
Hexanol	yes	no	yes	yes
Phenol	yes	yes	yes	yes
Decane	no	no	no	no
Naphthalene	no	yes	no	induced
Dodecane	no	no	no	no

50% Phenyl



Strong dispersion  
No dipole  
Weak hydrogen bonding

100% Methyl



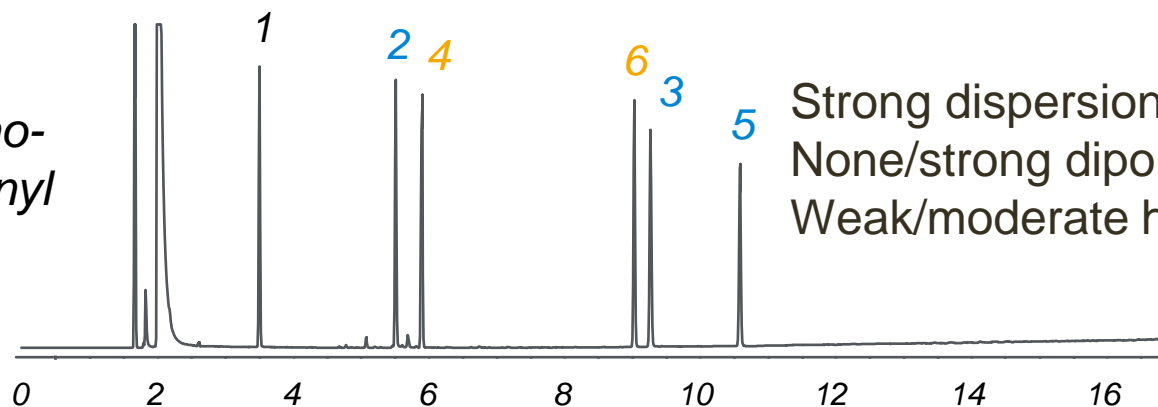
Strong dispersion  
No dipole  
No hydrogen bonding

- 1. Toluene 110 °C
- 2. Hexanol 156 °C
- 3. Phenol 182 °C
- 4. Decane (C10) 174 °C
- 5. Naphthalene 218 °C
- 6. Dodecane (C12) 216 °C

# 14% Cyanopropylphenyl

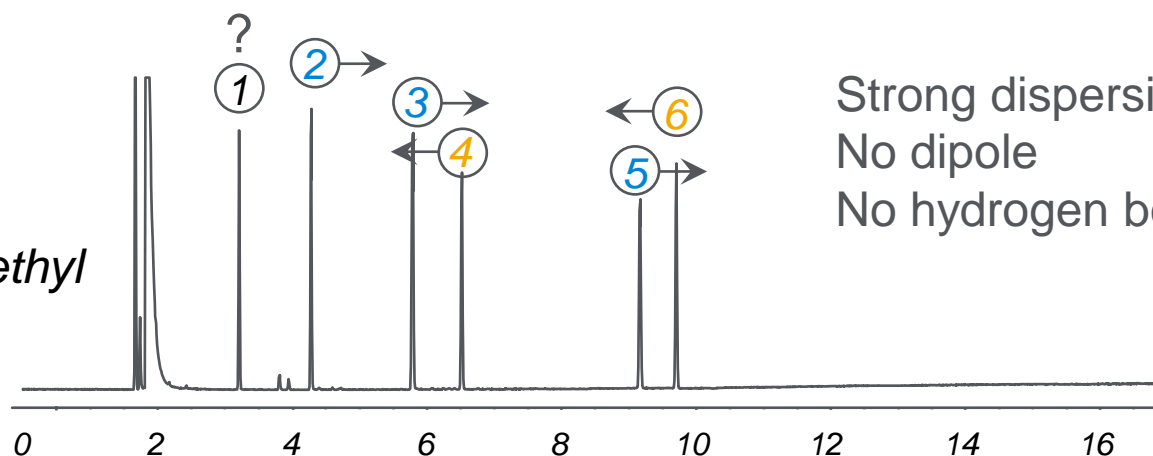
Compounds	Polar	Aromatic	Hydrogen Bonding	Dipole
Toluene	no	yes	no	induced
Hexanol	yes	no	yes	yes
Phenol	yes	yes	yes	yes
Decane	no	no	no	no
Naphthalene	no	yes	no	induced
Dodecane	no	no	no	no

14% Cyano-  
propylphenyl



Strong dispersion  
None/strong dipole (phenyl/cyanopropyl)  
Weak/moderate hydrogen bonding (phenyl/cyanopropyl)

100% Methyl

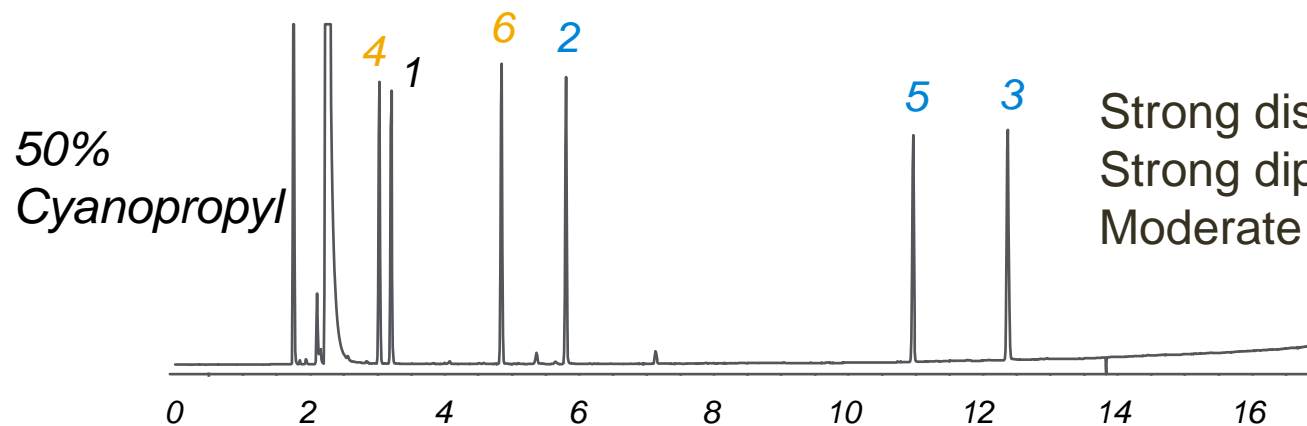


Strong dispersion  
No dipole  
No hydrogen bonding

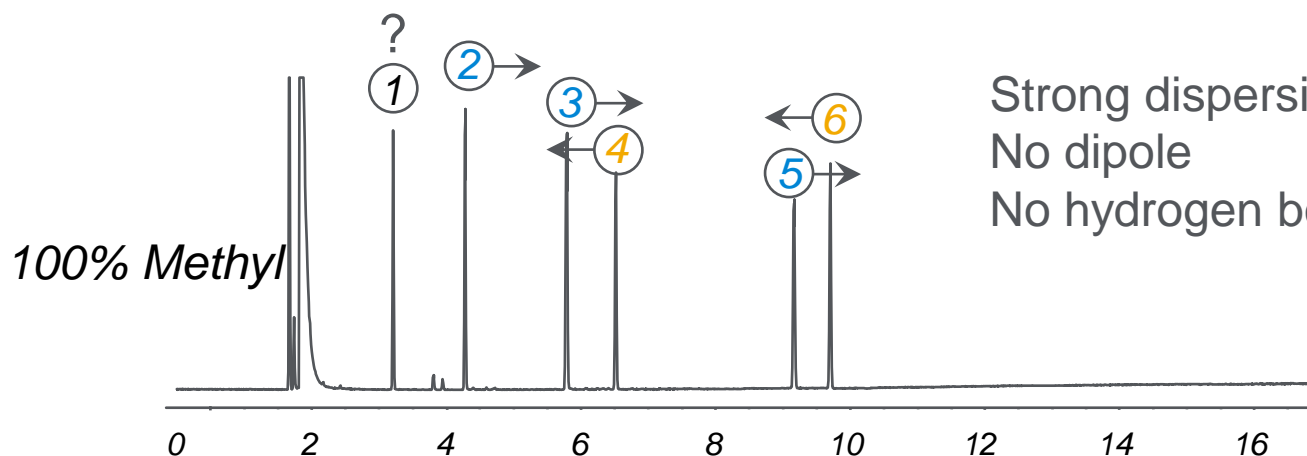
1. Toluene
2. Hexanol
3. Phenol
4. Decane (C10)
5. Naphthalene
6. Dodecane (C12)

# 50% Cyanopropyl

Compounds	Polar	Aromatic	Hydrogen Bonding	Dipole
Toluene	no	yes	no	induced
Hexanol	yes	no	yes	yes
Phenol	yes	yes	yes	yes
Decane	no	no	no	no
Naphthalene	no	yes	no	induced
Dodecane	no	no	no	no



Strong dispersion  
Strong dipole  
Moderate hydrogen bonding

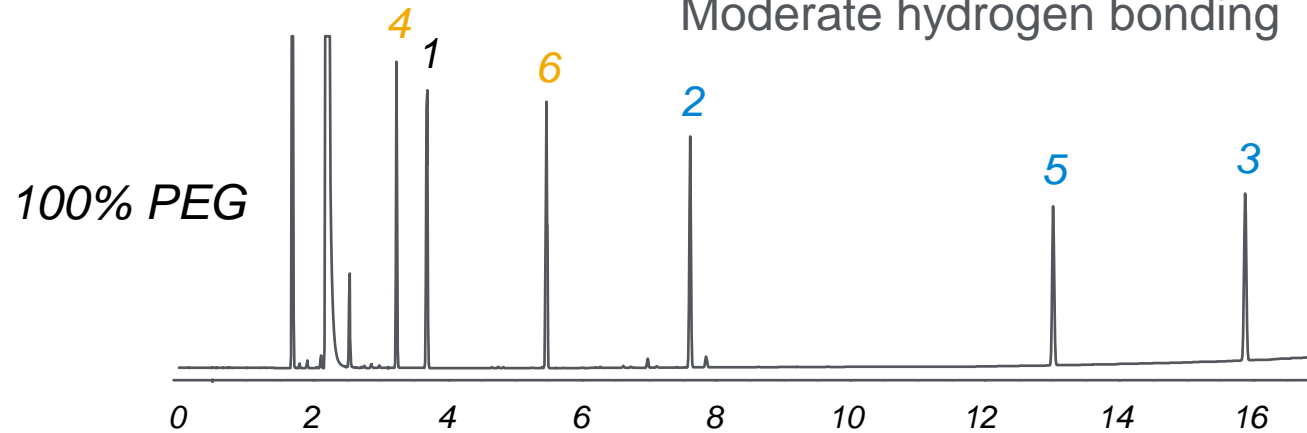


Strong dispersion  
No dipole  
No hydrogen bonding

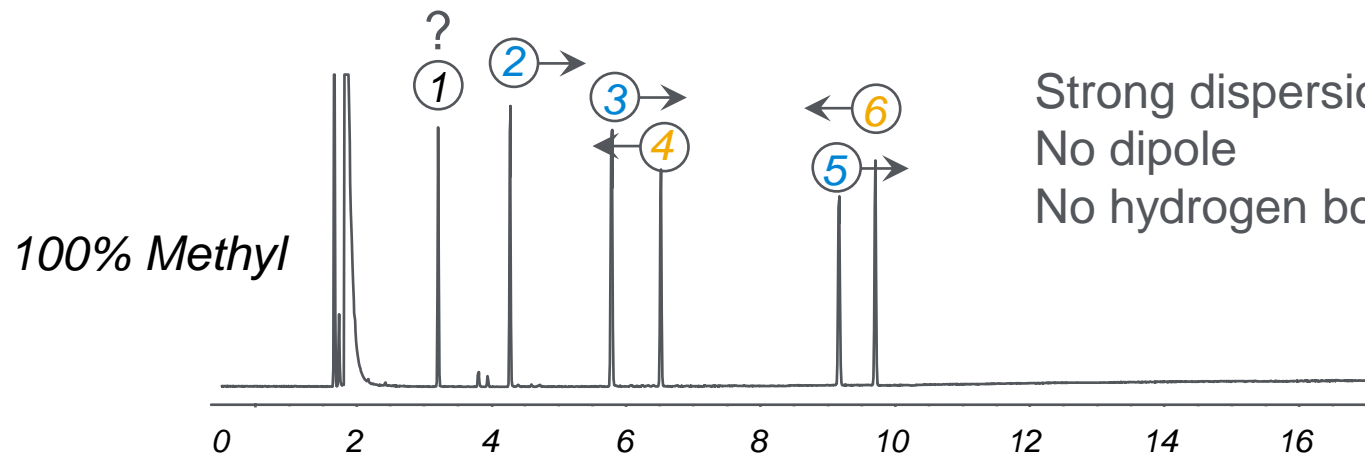
1. Toluene
2. Hexanol
3. Phenol
4. Decane (C10)
5. Naphthalene
6. Dodecane (C12)

# 100% Polyethylene Glycol

Strong dispersion  
 Strong dipole  
 Moderate hydrogen bonding



Compounds	Polar	Aromatic	Hydrogen Bonding	Dipole
Toluene	no	yes	no	induced
Hexanol	yes	no	yes	yes
Phenol	yes	yes	yes	yes
Decane	no	no	no	no
Naphthalene	no	yes	no	induced
Dodecane	no	no	no	no



Strong dispersion  
 No dipole  
 No hydrogen bonding

1. Toluene
2. Hexanol
3. Phenol
4. Decane (C10)
5. Naphthalene
6. Dodecane (C12)

# Stationary Phase Selection

## Part 1

- Existing information
- Selectivity
- Polarity
- Critical separations
- Temperature limits



# Agilent Bond Elut Sample Cleanup Products

Solid Phase Extraction  
cartridges and plates



Synthetic Chem Elut S

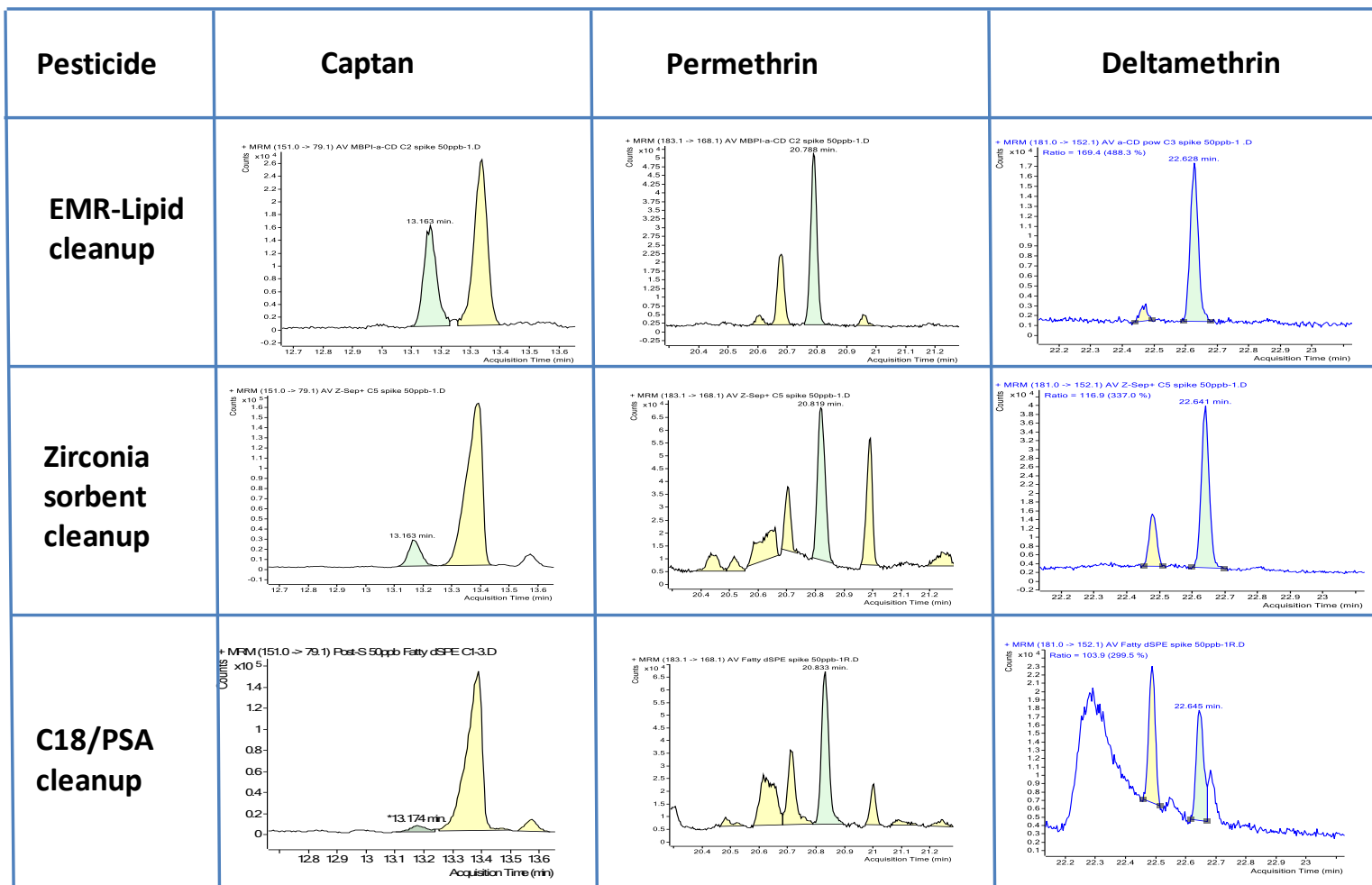
Filtration cartridges  
and plates



Captiva EMR Lipid



# Captiva EMR–Lipid Cleanup Improves Analytes S/N Ratio and Integration Accuracy on GC/MS(/MS) of Pesticides in Olive Oil



# Stationary Phase Selection

## Part 2

- Capacity
- Analysis time
- Bleed
- Versatility
- Selective detectors

# Column Dimensions

- Inner diameter
- Length
- Film thickness



# Column Diameter

## Capillary Columns

id (mm)	Common Name
0.53	Megabore
0.45	High speed megabore
0.32	Wide
0.20–0.25	Narrow
0.18	Minibore

# Column Diameter

## Theoretical Efficiency

	id (mm)	N/m
	0.10	11905
	0.18	6666
	0.20	5941
	0.25	4762
	0.32	3717
$k = 5$	0.53	2242

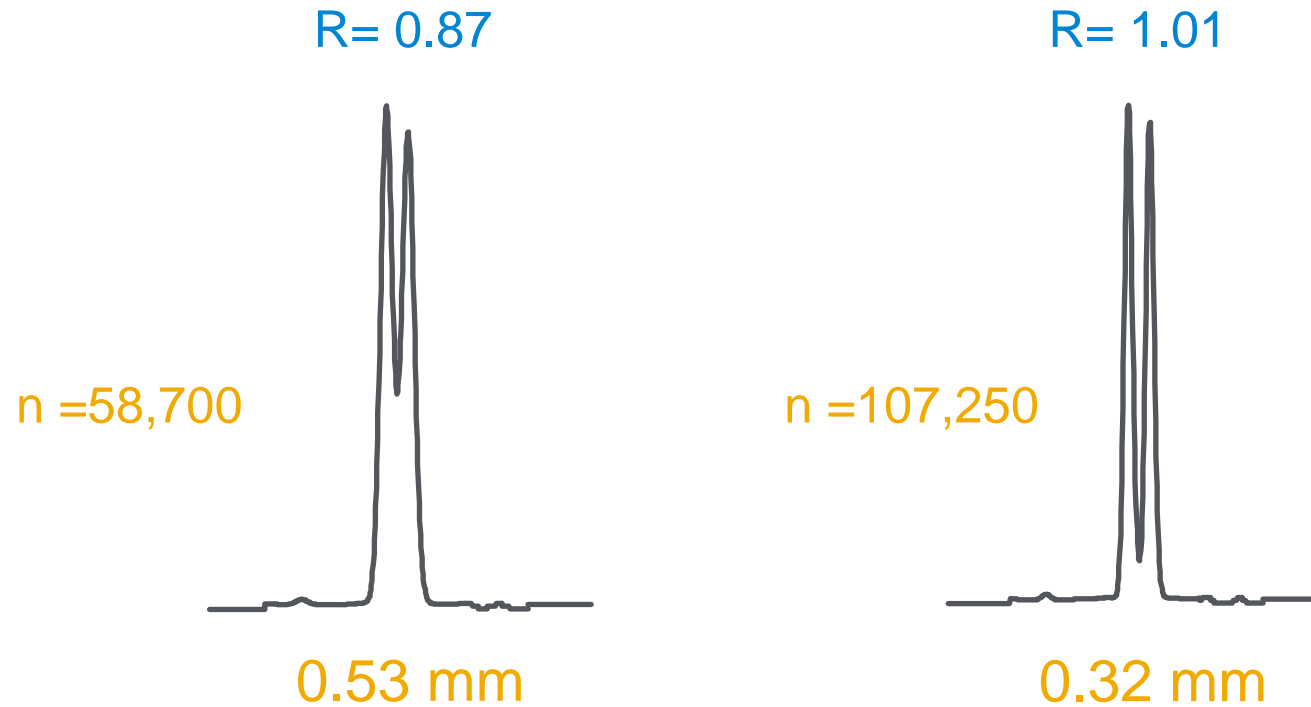
# Efficiency and Resolution Relationship

$$\sqrt{N} \propto R_s$$

efficiency **x 4** = resolution **x 2**

# Column Diameter

Resolution (180 °C isothermal)



Square root of resolution is inversely proportional to column diameter

# Column Diameter

Inlet head pressures for 30 meter column (helium)

id (mm)	Pressure (psig)
0.10	225–250
0.20	25–35
0.25	15–25
0.32	10–20
0.53	2–4

Hydrogen would produce about half the amount of pressure



# Column Diameter

Capacity (0.25  $\mu\text{m}$  film thickness)

id (mm)	Capacity (ng)
0.20	50–100
0.25	75–150
0.32	125–250
0.53	200–400

Like polarity phase/solute

# Column Diameter

## Carrier gas flow rate

Smaller diameters for low flow situations  
(e.g., GC/MS)

Larger diameters for high flow situations  
(e.g., purge & trap, headspace, gas sample valve)

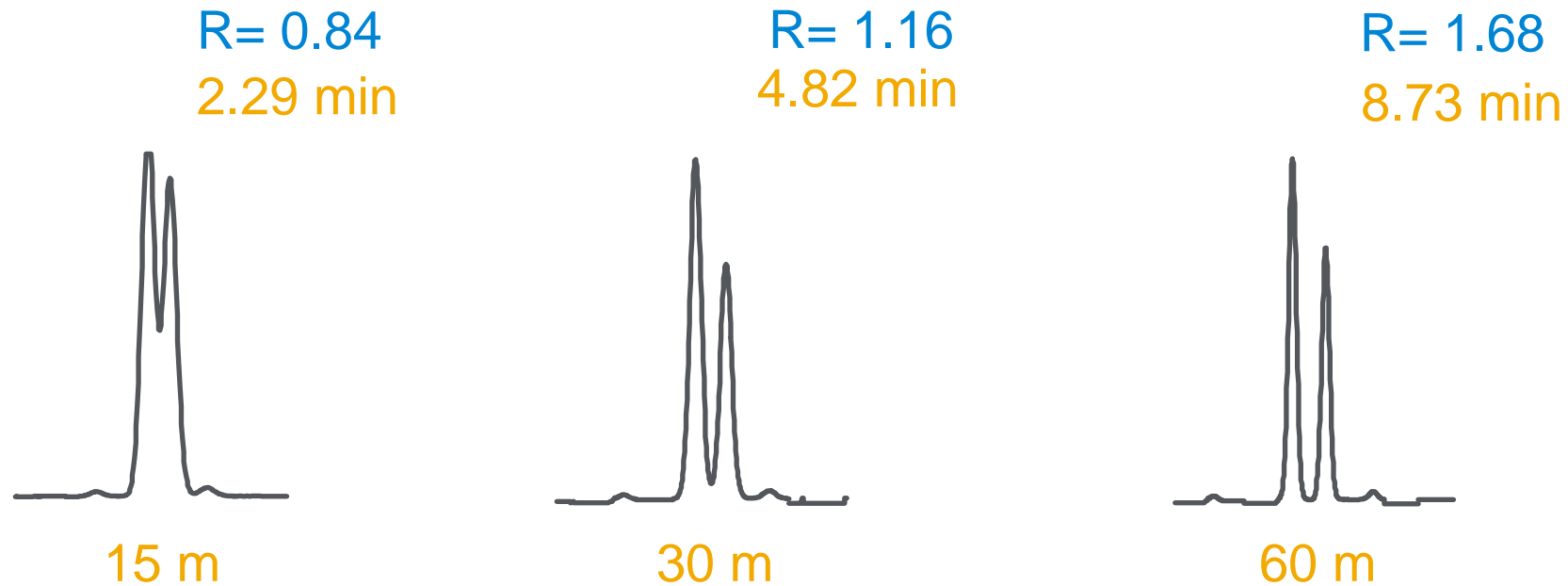
# Column Length

Most common: 15–60 meters

Available: 5–200 meters

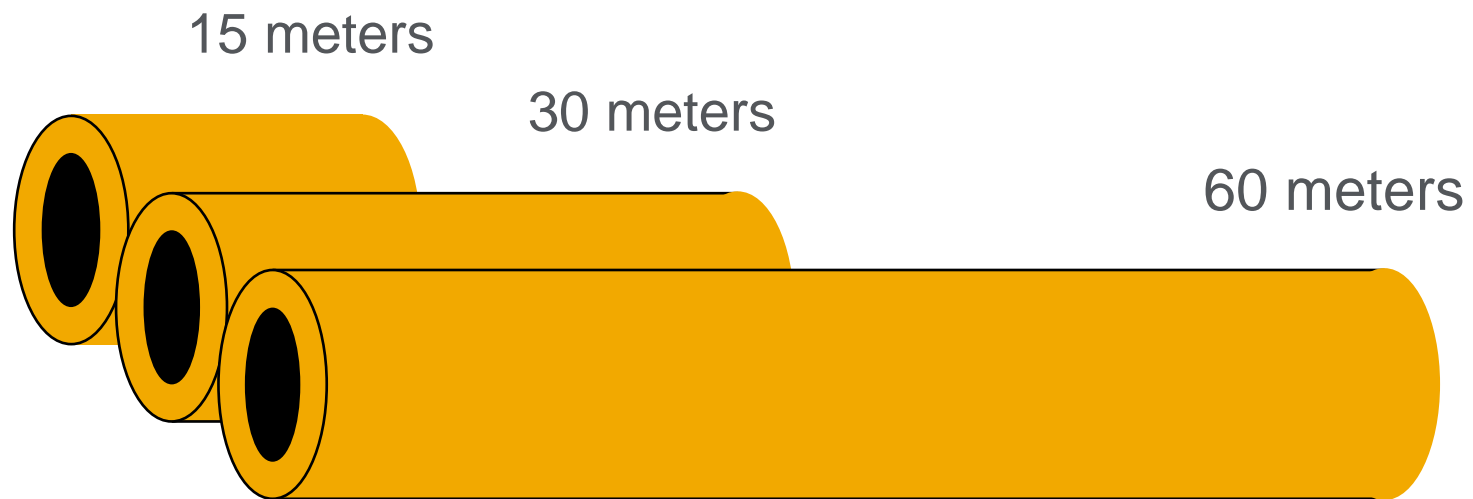
# Column Length

Resolution and retention 210°C isothermal



Resolution is proportional to the square root of column length  
Isothermal: retention is proportional to length  
Temperature program: 1/3–1/2 of isothermal values

# Column Length and Cost



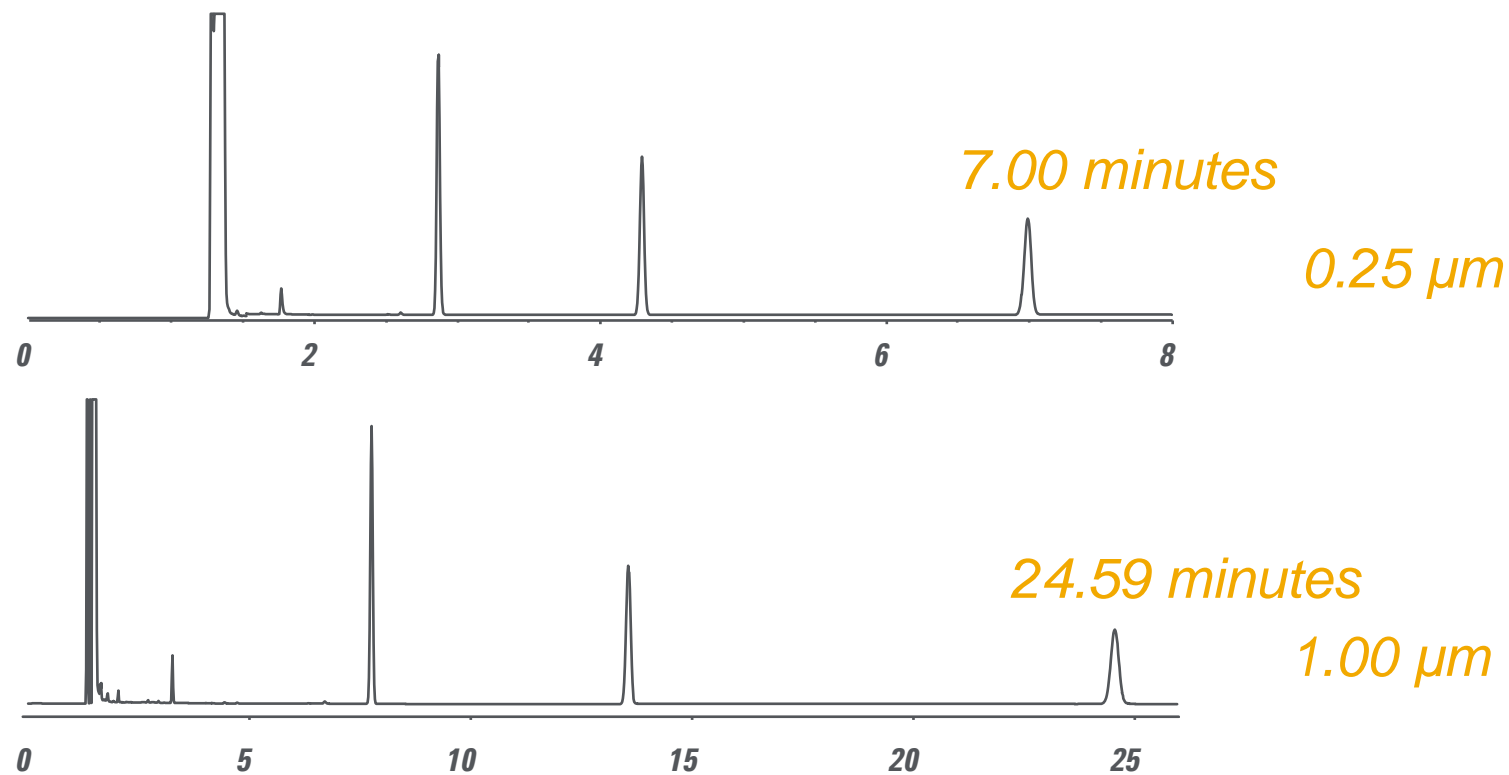
# Film Thickness

Most common: 0.1–3.0  $\mu\text{m}$

Available: 0.1–10.0  $\mu\text{m}$



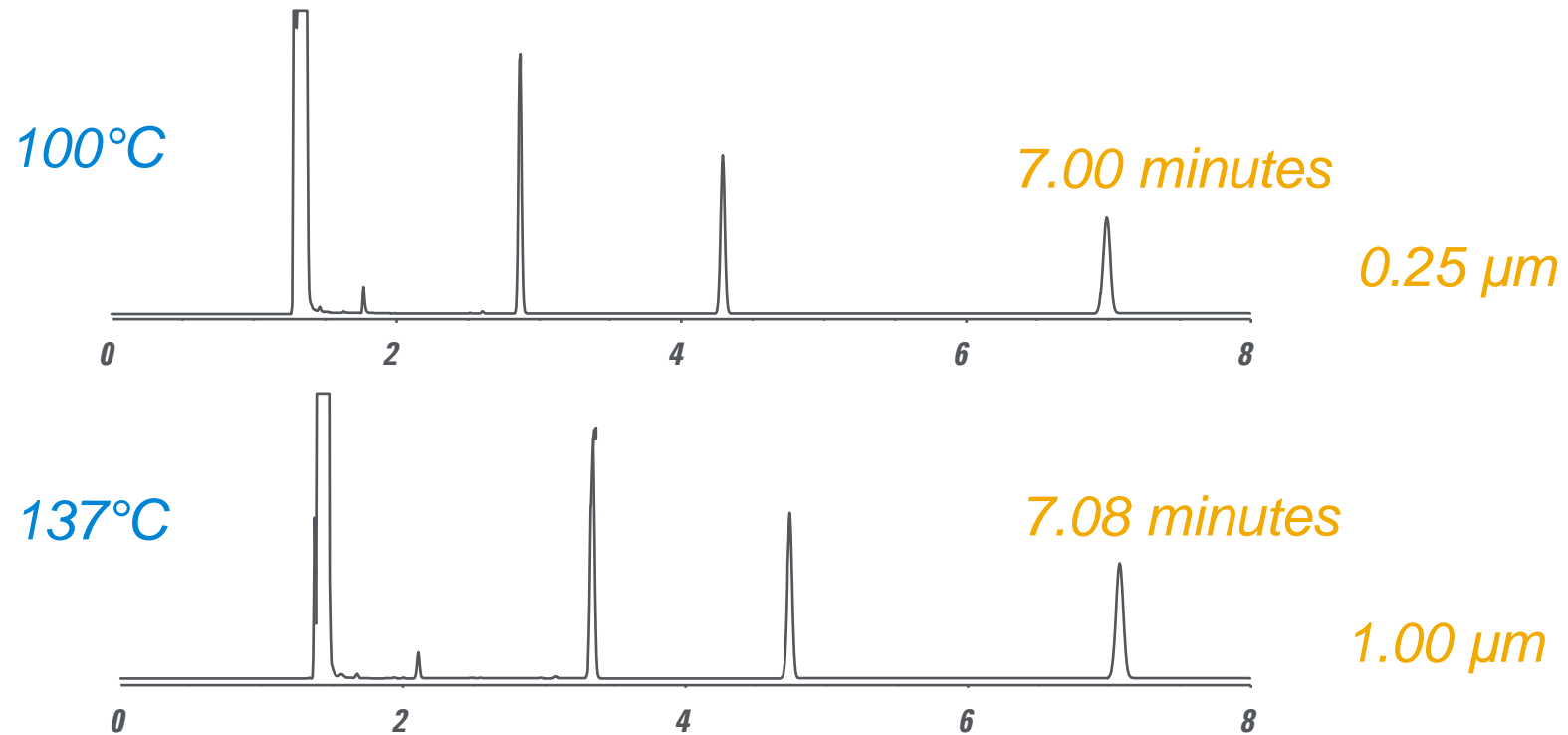
# Film Thickness and Retention (100°C Isothermal)



Isothermal: Retention is proportional to film thickness  
Temperature program: 1/3–1/2 of isothermal values

# Film Thickness

## Equal Retention: Isothermal



Agilent J&W DB-1, 30 m x 0.32 mm id  
He at 37 cm/sec  
C10, C11, C12



# Film Thickness and Resolution

When solute  $k < 5$

$d_f$



R

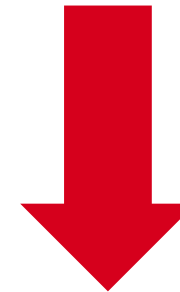


When solute  $k > 5$

$d_f$

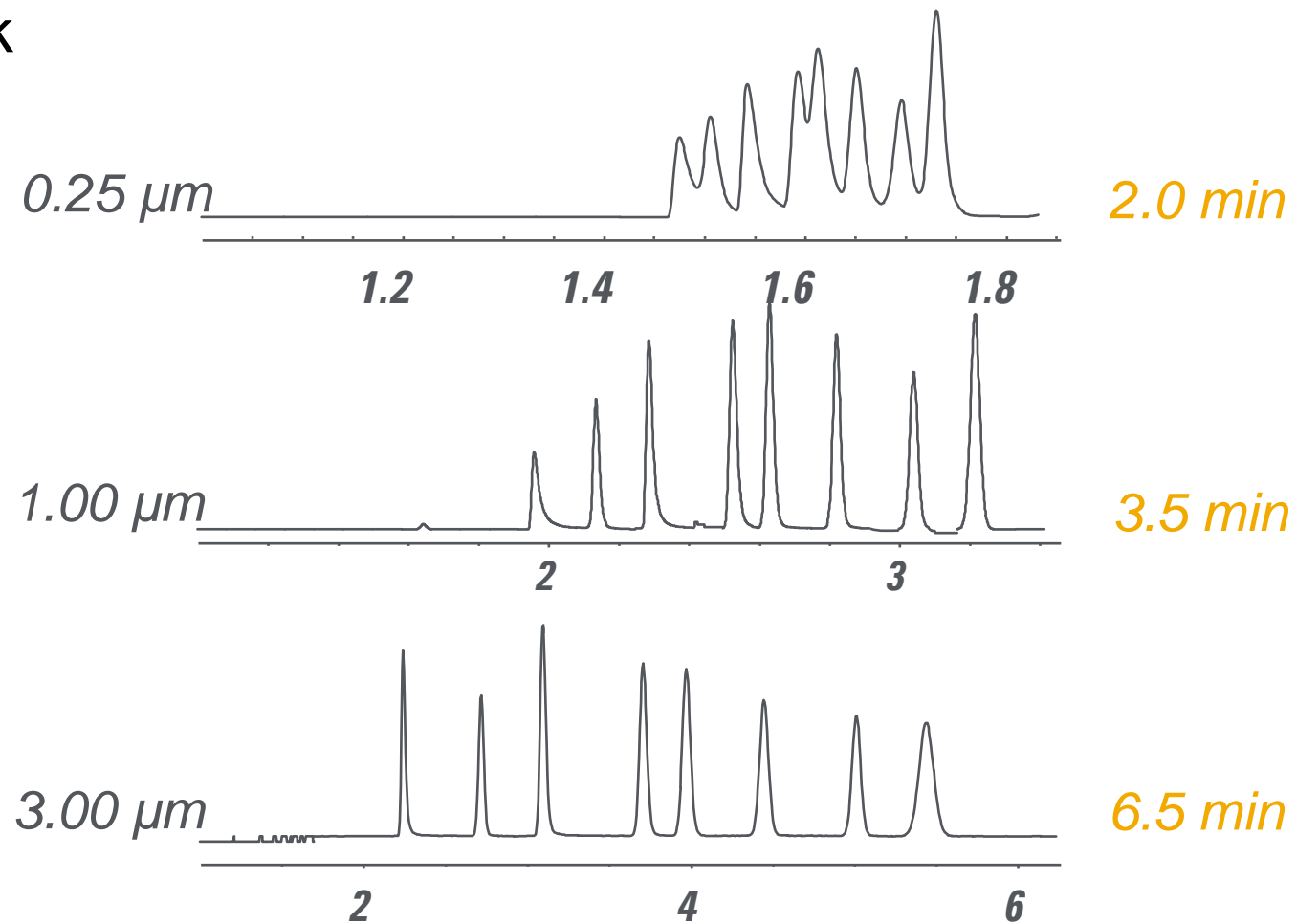


R



# Film Thickness

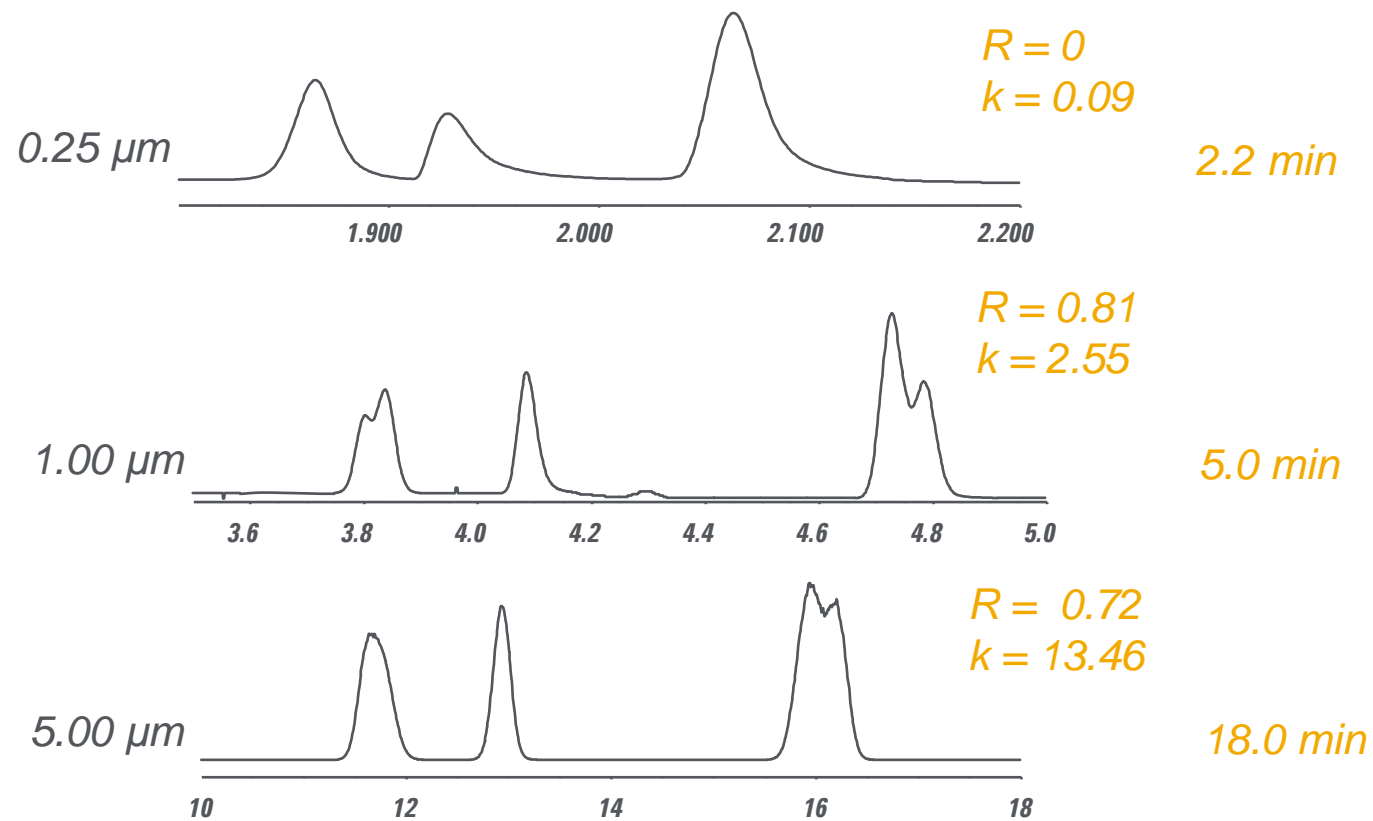
## Resolution at low k



DB-1, 30 m x 0.32 mm id  
40 °C isothermal, He at 35 cm/sec  
Solvent mixture

# Film Thickness

## Resolution at high k



DB-1, 30 m x 0.32 mm id  
40 °C isothermal, He at 35 cm/sec  
Solvent mixture

# Film Thickness (Capacity)

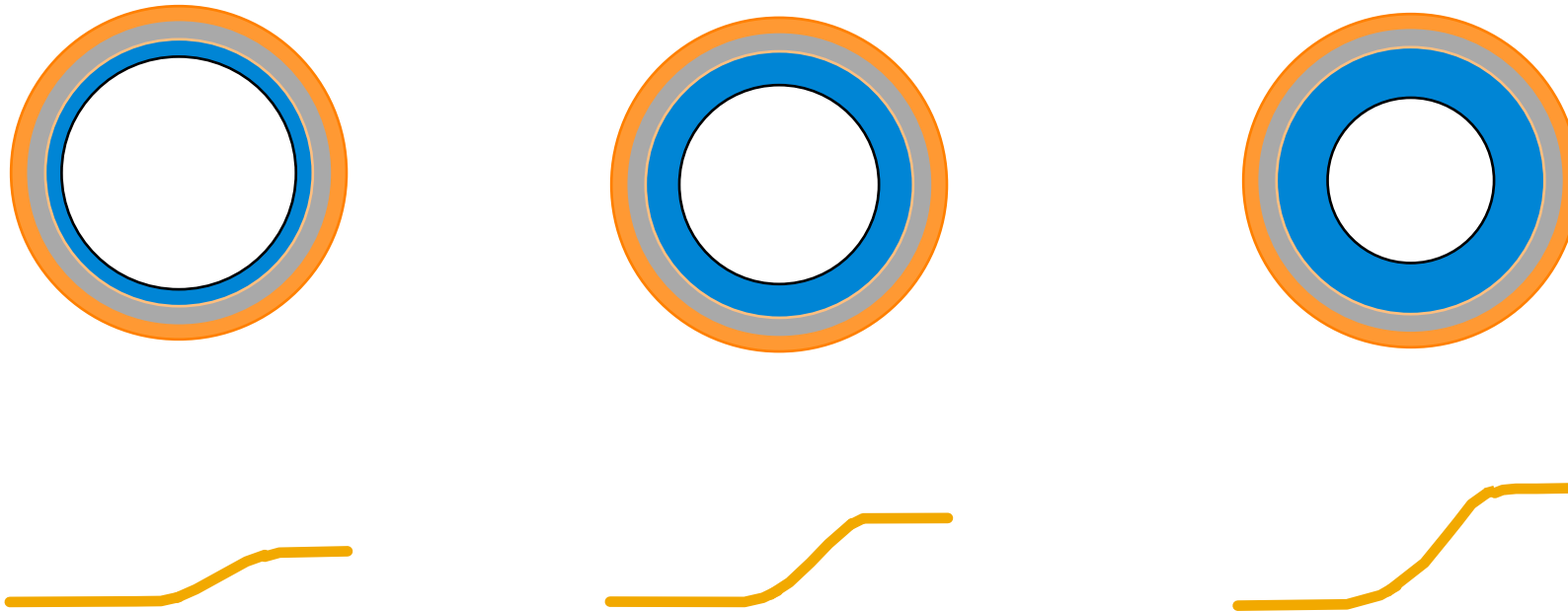
Thickness ( $\mu\text{m}$ )	Capacity (ng)
0.10	50-100
0.25	125-250
1.0	500-1000
3.0	1500-3000
5.0	2500-5000

0.32 mm column id

Like polarity phase/solute

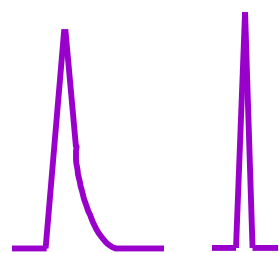
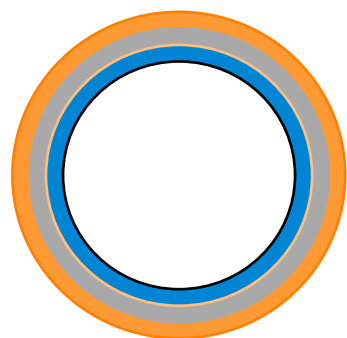
# Film Thickness (Bleed)

More stationary phase = More degradation products



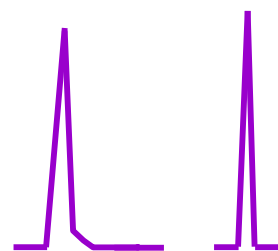
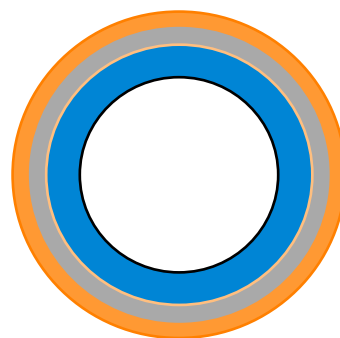
# Film Thickness (Inertness)

0.25  $\mu\text{m}$



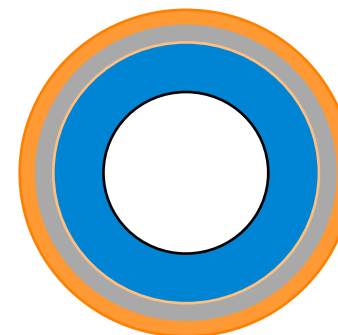
active inactive

1.0  $\mu\text{m}$



active inactive

3.0  $\mu\text{m}$



active inactive

# Column Dimensions

## Diameter summary

To Increase	Make Diameter
Resolution	Smaller
Retention	Smaller
Pressure	Smaller
Flow rate	Larger
Capacity	Larger

# Column Dimensions

## Length summary

To Increase	Make Length
Resolution	Longer
Retention	Longer
Pressure	Longer
Cost	Longer



# Column Dimensions

## Film thickness summary

To Increase

Make Film

---

Retention

Thicker

Resolution ( $k < 5$ )

Thicker

Resolution ( $k > 5$ )

Thinner

Capacity

Thicker

Inertness

Thicker

Bleed

Thicker

# Example of Changing Dimensions to Achieve Faster Chromatography Before

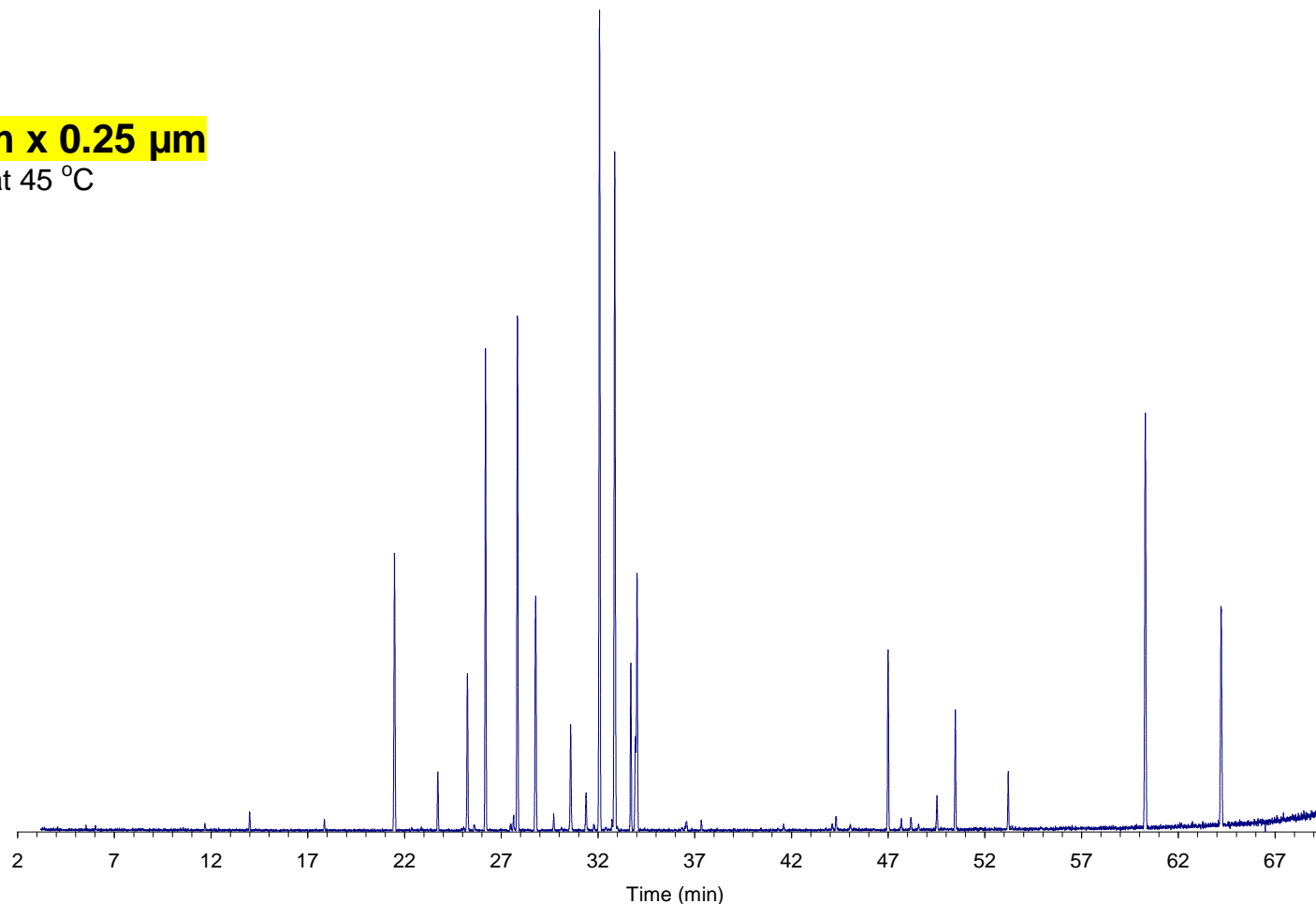
**Column: DB-WAX 30 m x 0.25 mm x 0.25  $\mu$ m**

Carrier: Helium at 25.4 cm/sec measured at 45 °C

Oven: 45 °C for 2 min  
45 to 250 °C at 3 °C/min  
250 °C for 34 min

Injector: Split 1:30, 250 °C  
1  $\mu$ L of 1:35 oil in Acetone

Detector: MSD full scan at m/z 40–500  
250 °C transfer line



# Example of Changing Dimensions to Achieve Faster Chromatography After

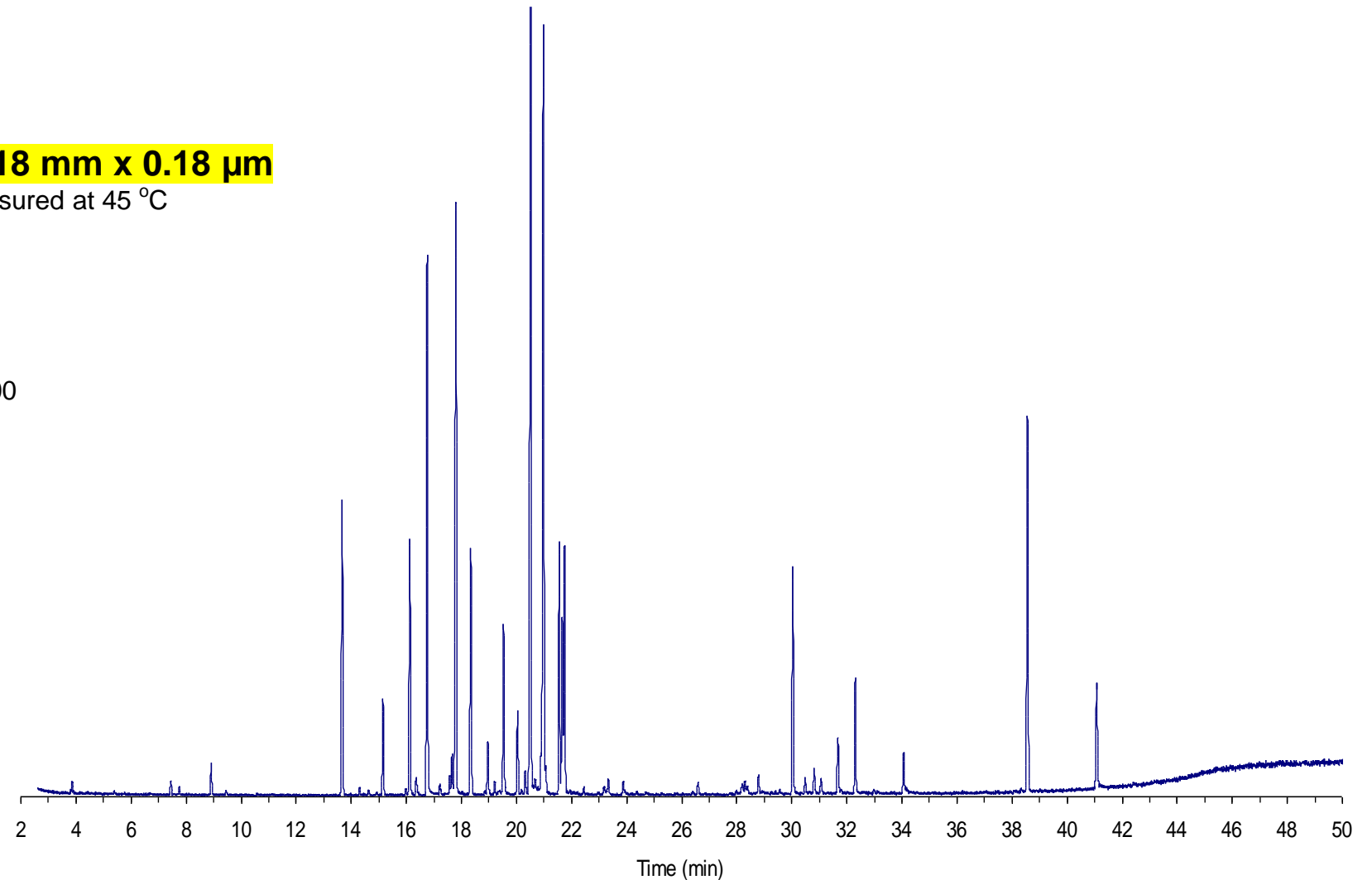
**Column: DB-WAX 20 m x 0.18 mm x 0.18  $\mu$ m**

Carrier: Helium at 26.3 cm/sec measured at 45 °C

Oven: 45 °C for 1.28 min  
45 to 250 °C at 4.67 °C/min  
250 °C for 21.81 min

Injector: Split 1:30, 250 °C  
1  $\mu$ L of 1:35 Oil in Acetone

Detector: MSD full scan at m/z 40–500  
250 °C transfer line



# Conclusions

- Understand the sample
- Is it volatile and thermally stable enough to chromatograph by GC?
- Try to match polarity – **oil and water don't mix!**
- Look for unique characteristics of compounds and match them to a phase
- If you have the correct selectivity, change the dimensions to improve resolution – **consider a smaller id**
- If you need better peak shape for difficult compounds, try the '**UI**' version
- Look for available information for a particular application

**Call Tech Support!**

# Contact Agilent Chemistries and Supplies Technical Support



1-800-227-9770 Option 3, Option 3:

**Option 1 for GC and GC/MS columns and supplies**

Option 2 for LC and LC/MS columns and supplies

Option 3 for sample preparation, filtration, and QuEChERS

Option 4 for spectroscopy supplies

Option 5 for chemical standards

**Available in the USA and Canada 8–5, all time zones**



[gc-column-support@agilent.com](mailto:gc-column-support@agilent.com)

[lc-column-support@agilent.com](mailto:lc-column-support@agilent.com)

[spp-support@agilent.com](mailto:spp-support@agilent.com)

[spectro-supplies-support@agilent.com](mailto:spectro-supplies-support@agilent.com)

[chem-standards-support@agilent.com](mailto:chem-standards-support@agilent.com)