



Application News

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Spectroscopy – Inductively Coupled Plasma Mass Spectrometry

Elemental bioimaging of Al₂O₃/ZrO₂/TiO₂ nanoparticles in lung tissue Jennifer-Christin Müller¹, Michael Sperling¹, Uwe Karst¹

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Introduction

Nanotechnology is one of the key developments of the 21st century. Nanoparticles are used in a wide industrial. military and varietv of medical applications. Titanium dioxide (TiO₂) nanoparticles are one of the most frequently used nanomaterials, for example in sunscreens as UV blocker or as antibacterial textile coating. Aluminum oxide (Al₂O₂) nanoparticles are used as abrasionresistant coating in the automobile industry. In addition to that, grinding dusts can contain AI_2O_3 as well as zirconium dioxide (ZrO₂) nanoparticles.

Because of the small diameter, the nanoparticles can enter the body through the lungs and are able to cause inflammation or cell defects in the lung tissue. To be able to assess the long-term toxicity of nanoparticles, the distribution in affected organs plays an important role. Elemental bioimaging by means of laser ablation-inductively coupled plasma-mass spectrometry (LA-ICP-MS) is a useful tool to analyze various elements in tissue sections. Here, the distribution and concentration of mixed oxide nanoparticles in rat lung tissue was analyzed.

Material and methods

Animal study

The animal study and organ preparation were performed by IBE R&D gGmbH. The used nanoparticles were composed of aluminum(III) oxide (77% w/w), zirconium(IV) dioxide (12% w/w) and titanium(IV) dioxide (11% w/w) with a d50 value of approx. 35 nm. The nanoparticles were

intratracheally instilled into the left lung in 0.9% sodium chloride solution with an addition of 0.025 mg/ml lecithin. The instilled nanoparticle concentration was 4.8 mg per lung. The rats were euthanized three days after the instillation. The lungs were filled with an embedding media to avoid collapsing of the organ and snap-frozen afterwards. Tissue sections with a thickness of 10 μ m were prepared using a cryotome and subsequently mounted on a microscope slide.

Standard preparation

To quantify the nanoparticle concentration in the lung, an external calibration with matrix-matched standards was performed. Gelatin (10% w/w) was spiked the nanoparticle dispersion to obtain standards with different aluminum, titanium and zirconium concentrations. Thin sections with the same thickness as the tissue sample were prepared and analyzed with the same measurement parameters.

Analytical conditions

For elemental bioimaging of tissue samples, LA-ICP-MS is the analytical method of choice. Here, the ICPMS-2030 quadrupole mass spectrometer was coupled to a laser ablation system LSX-213 G2⁺ (Teledyne CETAC). The system was equipped with a Nd-YAG laser operating at a wavelength of 213 nm. The laser ablation was connected to the ICP-MS via Tygon[®] tubing attached to the expansion pipe. The ablation cell was flushed with helium.



Figure 1: Schematic drawing of the LA-ICP-MS setup with internal standard introduction.

Laser ablation	LSX-213 G2+	
Spot size	50 µm	
Scan speed	100 µm	
Shot frequency	20 Hz	
Cell gas flow	0.8 l/min	
ICP-MS	ICPMS-2030	
Plasma power	1.2 kW	
Sampling depth	5.0 mm	
Plasma gas	8.0 l/min	
Auxiliary gas	1.10 l/min	
Carrier gas	0.45 l/min	
Cell gas	6.0 ml/min	
Cell voltage	-21 V	

Table 1: LA and ICP-MS conditions.

To monitor the plasma stability, an internal standard containing 1 ng/g rhodium in 2% HNO₃ was introduced via the nebulizer. The used setup is schematically shown in Figure 1. The ICP-MS was equipped with copper sampler and skimmer. To avoid polyatomic interferences, it was operated in the collision mode with helium as collision gas. The isotopes ²⁷AI, ⁴⁷Ti, ⁴⁹Ti, and ⁹¹Zr were measured with an integration time of 0.1 s each, the isotopes ³¹P and ¹⁰³Rh were measured for 0.05 s each.

Results

The rhodium signal, which was used as internal standard, has a low standard deviation of approx. 4.4%, indicating a good plasma stability during the ablation process. Limits of detection and quantification for analysis with a laser spot size of 50 μ m (Table 2) were sufficient for the expected nanoparticle concentration in the lung tissue.

In Figure 2, the bright field microscopic image as well as the LA-ICP-MS images for the ³¹P, ²⁷Al, ⁴⁷Ti and ⁹¹Zr distribution are shown. The endogenous element phosphorous can be used to illustrate the tissue structure with the larger air-filled bronchial tubes and smaller bronchioles and alveoli, because of its occurrence for example in phospholipids.

Isotope	LOD [µg/g]	LOQ [µg/g]
²⁷ AI	65.1	217.0
⁴⁷ Ti	1.4	4.8
⁴⁹ Ti	1.6	5.2
⁹¹ Zr	1.4	4.6

Table 2: Limits of detection and quantification.

The aluminum, titanium and zirconium distributions are very similar, leading to the conclusion that the observed signals of those elements belong in fact to the instilled nanoparticles. With the matrix matched external calibration. aluminum concentrations up to 2500 µg/g were found. In contrast to that, titanium concentrations up to 500 µg/g, as well as zirconium concentrations up to 750 µg/g were found. These differences can be explained with the composition of the mixed oxide nanoparticles. The nanoparticle distribution appears to be relatively homogenous.

Some hotspots with highest concentrations can be found in lung regions with denser tissue. Those areas correlates with higher intensities in the ³¹P distribution. In contrast to that, the nanoparticle concentration in the tissue near the bronchial tubes and bronchioles is much lower, indicating a possible cleaning effect of the lung.



Figure 2: Bright field microscopic image (top left) as well as LA-ICP-MS images of the ³¹P, ²⁷Al, ⁴⁷Ti, and ⁹¹Zr distribution in lung tissue three days after instillation of nanoparticles. The white dotted line indicates the borders of the tissue.

Conclusion

The nanoparticle distribution in rat lung tissue was successfully visualized by means of LA-ICP-MS. The measured phosphorous distribution can be used to visualize the lung tissue. The colocalization of the three elements aluminum, titanium and zirconium, leads to the conclusion, that the observed signals can be unambiguous assigned to the instilled nanoparticles. In addition to that, the particle concentration could be determined using matrix-matched gelatin standards.

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